



Supplementary Materials: Ionic Conductivity and Assembled Structures of Imidazolium Salt-Based Block Copolymers with Thermoresponsive Segments

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Figure S1. ¹H NMR spectra of (a) *N*-methyl ethylimidzolium bromide (MEI-Br) and (b) MEI-NTf₂ in DMSO-*d*₆.



Figure S2. ¹³C NMR spectra of (a) MEI-Br and (b) MEI-NTf₂ in DMSO-d₆.



Figure S3. ¹H NMR spectra of (a) poly(NVI-Br) and (b) poly(NVI-NTf₂) in DMSO-d₆.



Figure S4. SEC traces of poly(NVI-Br)₆₈-*b*-poly(NIPAM)₃₂ (blue line) and poly(NVI-Br) macro-CTA (red line).



FigureS5.¹³CNMRspectraofpoly(NVI-NTf2)-b-poly(NIPAM) in DMSO-d6.

(a) poly(NVI-Br)-b-poly(NIPAM) and (b)



Figure S6. TG curves of poly(NVI-Br)-*b*-poly(NIPAM) and poly(NVI-NTf₂)-*b*-poly(NIPAM) under nitrogen atmosphere.



Figure S7. Temperature-dependent ionic conductivity of $poly(NVI-NTf_2)_{48}$ -*b*-poly(NIPAM)₅₂. Acetone solution of the block copolymer was casted onto a platinum electrode and dried at 40 °C for 2 h. After it was allowed at room temperature overnight, the sample was dried at 90 °C for 2 h.



Figure S8. DSC curves of (a) poly(NVI-NTf2) and (b) poly(NVI-NTf2)-b-poly(NIPAM).



Figure S9. DLS profiles of poly(NVI-Br)₂₁-*b*-poly(NIPAM)₇₉ in aqueous solution (polymer conc. = 2.0 mg/mL) at 25 °C and 40 °C.



Figure S10. SFM (a-b) height and (c-d) phase images of poly(NVI-Br)-*b*-poly(NIPAM)s; NVI-Br/NIPAM = (a, c) 68/32 and (b, d) 21/79, respectively. The samples were prepared by the drop casting of the methanol solutions of the block copolymers onto mica substrates.

Macro-CTA (Mn, Mw/Mn)	[M]0/ [macro-CTA]0	Conv. ^{d)} / Yield ^{e)} (%)	<i>M</i> ^{<i>n</i> f)} (theory)	Mn ^{d)} (NMR)	<i>M</i> ^{<i>n</i> g)} (SEC)	M _w /M _n ^{g)} (SEC)	n : m ^{d)}
15,400, 1.27	50 ^{b)}	53/78	18,900	19,500	20.600	1.39	68:32
7,600, 1.11	50 ^c)	94/75	13,700	13,000	9,100	1.26	48:52
14,000, 1.30	400 ^c)	89/70	60,000	46,900	18,000	1.38	21:79

Table S1. Synthesis of poly(NVI-Br)-*b*-poly(NIPAM) by RAFT polymerization of NIPAM using the dithiocarbamate-terminated poly(NVI-Br) macro-CTA with AIBN in methanol ^a).

^{a)} Polymerization was carried out at 80 °C for 6h, $[M]_0 = 0.1$ g/mL. ^{b)} [macro-CTA] $_0/[I]_0 = 2$. ^{c)} [macro-CTA] $_0/[I]_0 = 0.5$. ^{d)} Calculated by ¹H NMR in DMSO-*d*₆. ^{e)} Diethyl ether-insoluble part. ^{f)} The theoretical molecular weight ($M_{n, \text{ theory}}$) = (MW of M) × [M] $_0/[macro-CTA]_0 \times \text{conv.} +$ (MW of macro-CTA). ^{g)} Measured by SEC using poly (ethylene oxide) standards in H₂O/acetonitrile (50/50 vol% containing 0.05 M NaNO₃).

	H ₂ O	DMF	DMSO	MeOH	Acetone	THF
Poly(NVI-Br)	+	+	+	+	-	-
Poly(NVI-NTf2)	-	+	+	-	+	-
Poly(NIPAM)	+	+	+	+	+	+
Poly(NVI-Br)48-b-poly(NIPAM)52	+	-	+	+	-	-
Poly(NVI-NTf2)48-b-poly(NIPAM)52	-	+	+	+	+	-
	Dioxane	CHCl ₃	CH ₂ Cl ₂	AcOEt	Hexane	Ether
Poly(NVI-Br)	-	-	-	-	-	-
Poly(NVI-NTf2)	-	-	-	-	-	-
Poly(NIPAM)	+	+	+	+	-	-
	-					
Poly(NVI-Br)48-b-poly(NIPAM)52	-	-	-	-	-	-

Table S2. Solubility of homopolymers and block copolymers.

+ : Soluble at room temperature, - : Insoluble, ± : Partially soluble



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