Supplementary Materials: Low Band Gap Donor-Acceptor Type Polymers Containing 2,3-Bis(4-(decyloxy)phenyl)pyrido[4,3-b]pyrazine as Acceptor and Different Thiophene Derivatives as Donors

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Figure S1. (a) ¹H NMR spectrum of 2,5-diromopyrido-3,4-diamine (1) in DMSO. Solvent peak at δ = 2.49 ppm is marked by "x", water peak at δ = 3.33 ppm is marked by "y"; (**b**) ¹³C NMR spectrum of **1** in DMSO. Solvent peak at δ = 40.76 ppm is marked by "X".



Figure S2. (a) ¹H NMR spectrum of 1,2-bis(4-(decyloxy)phenyl)ethane-1,2-dione (2) in CDCl₃. Solvent peak at δ = 7.26 ppm is marked by "x"; (b) ¹³C NMR spectrum of 2 in CDCl₃. Solvent peak at δ = 72.50 ppm is marked by "X".



Figure S3. (a) ¹H NMR spectrum of 5,8-dibromo-2,3-bis(4-(decyloxy)phenyl)pyrido[4,3-b]pyrazine (3) in CDCl₃. Solvent peak at δ = 7.26 ppm is marked by "x"; (b) ¹³C NMR spectrum of 3 in CDCl₃.Solvent peak at δ = 72.50 ppm is marked by "X".





Figure S4. (a) ¹H NMR spectrum of M1 in CDCl₃. Solvent peak at δ = 7.26 ppm is marked by "x"; (b) ¹³C NMR spectrum of M1 in CDCl₃. Solvent peak at δ = 72.50 ppm is marked by "X".



Figure S5. (a) ¹H NMR spectrum of M2 in CDCl₃. Solvent peak at δ = 7.26 ppm is marked by "x"; (b) ¹³C NMR spectrum of M2 in CDCl₃. Solvent peak at δ = 72.50 ppm is marked by "X".



Figure S6. (a) ¹H NMR spectrum of M3 in CDCl₃. Solvent peak at δ = 7.26 ppm is marked by "x"; (b) ¹³C NMR spectrum of M3 in CDCl₃. Solvent peak at δ = 72.50 ppm is marked by "X".



Figure S7. (a) ¹H NMR spectrum of M4 in CDCl₃. Solvent peak at δ = 7.26 ppm is marked by "x"; (b) ¹³C NMR spectrum of M4 in CDCl₃. Solvent peak at δ = 72.50 ppm is marked by "X".







Figure S8. Cyclic voltammetry (CV) curves of the monomers: (a) M2; (b) M3; and (c) M4.





Figure S9. CV curves of the polymers for p-type doping process at various scan rates: (**a**) P2; (**b**) P3; and (**c**) P4. Insert: graphs of scan rate vs. peak current density.





Figure S10. The first and the 1000th CV curve of the polymers: (a) P2; (b) P3; and (c) P4.



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Figure S11. Film thicknesses of the polymers deposited potentiostatically onto ITO electrode: (a) P1; (b) P2; (c) P3; and (d) P4.