

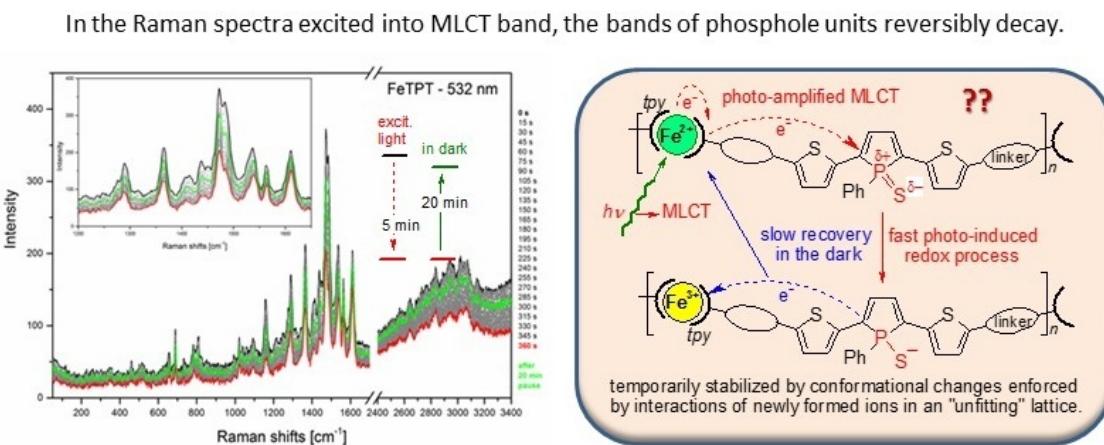
Novel Metallo-Supramolecular Polymers with 1-Thioxophosphole Main-Chain Units and Remarkable Photoinduced Changes in Their Resonance Raman Spectra

Ivana Sloufova ¹, Tereza Urvalkova ^{1,2}, Muriel Hissler ² and Jiri Vohlidal ^{1,*}

¹ Charles University, Faculty of Science, Department of Physical and Macromolecular Chemistry, Hlavova 2030, 128 40 Prague 2, Czech Republic

² University Rennes, CNRS, ISCR - UMR 6226, 35000 Rennes, France

Correspondence: vohlidal@natur.cuni.cz; sloufovi@natur.cuni.cz; muriel.hissler@univ-rennes1.fr



Supplementary Information

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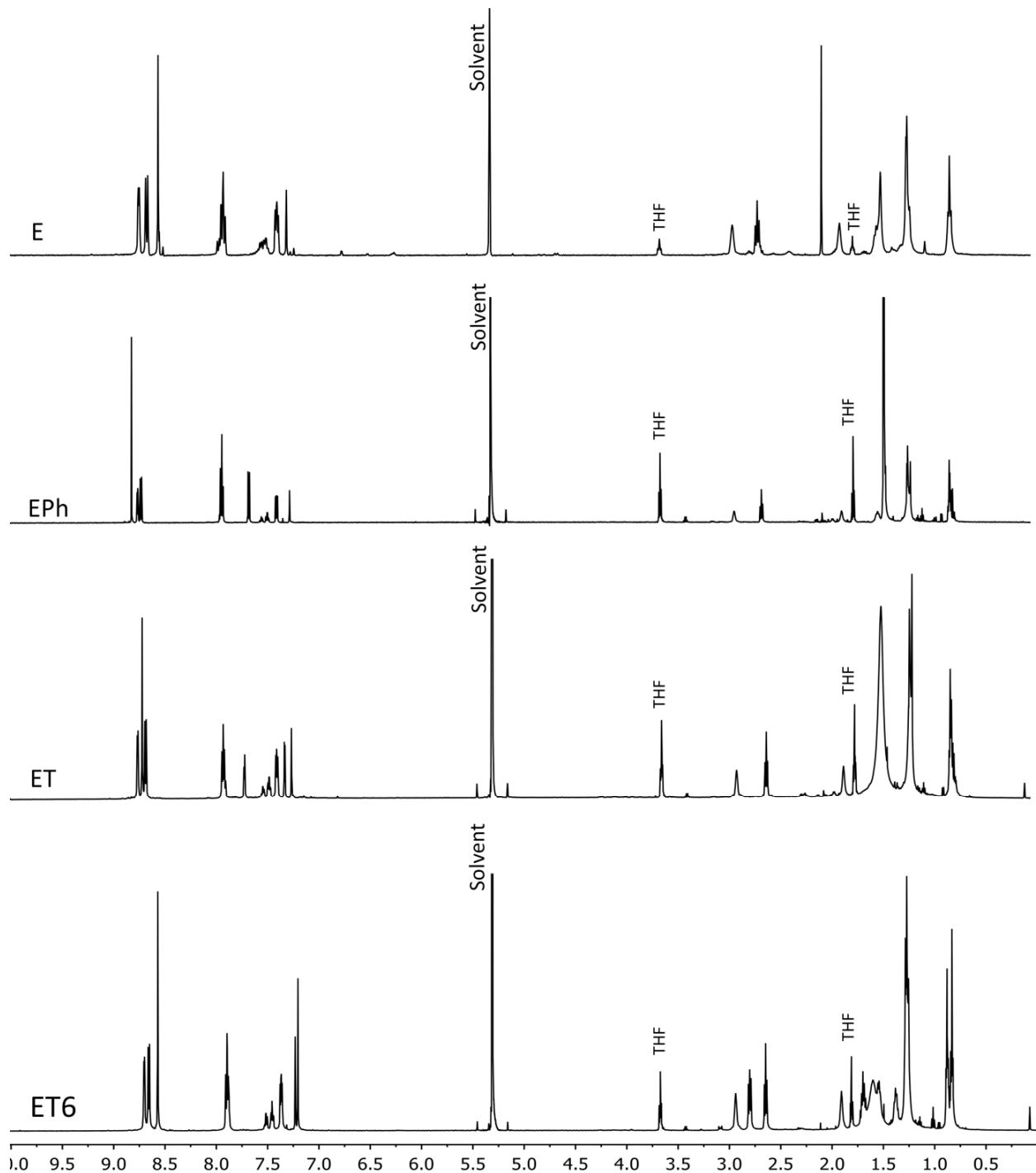


Figure S1. ¹H NMR spectra of prepared unimers (measured in CD_2Cl_2).

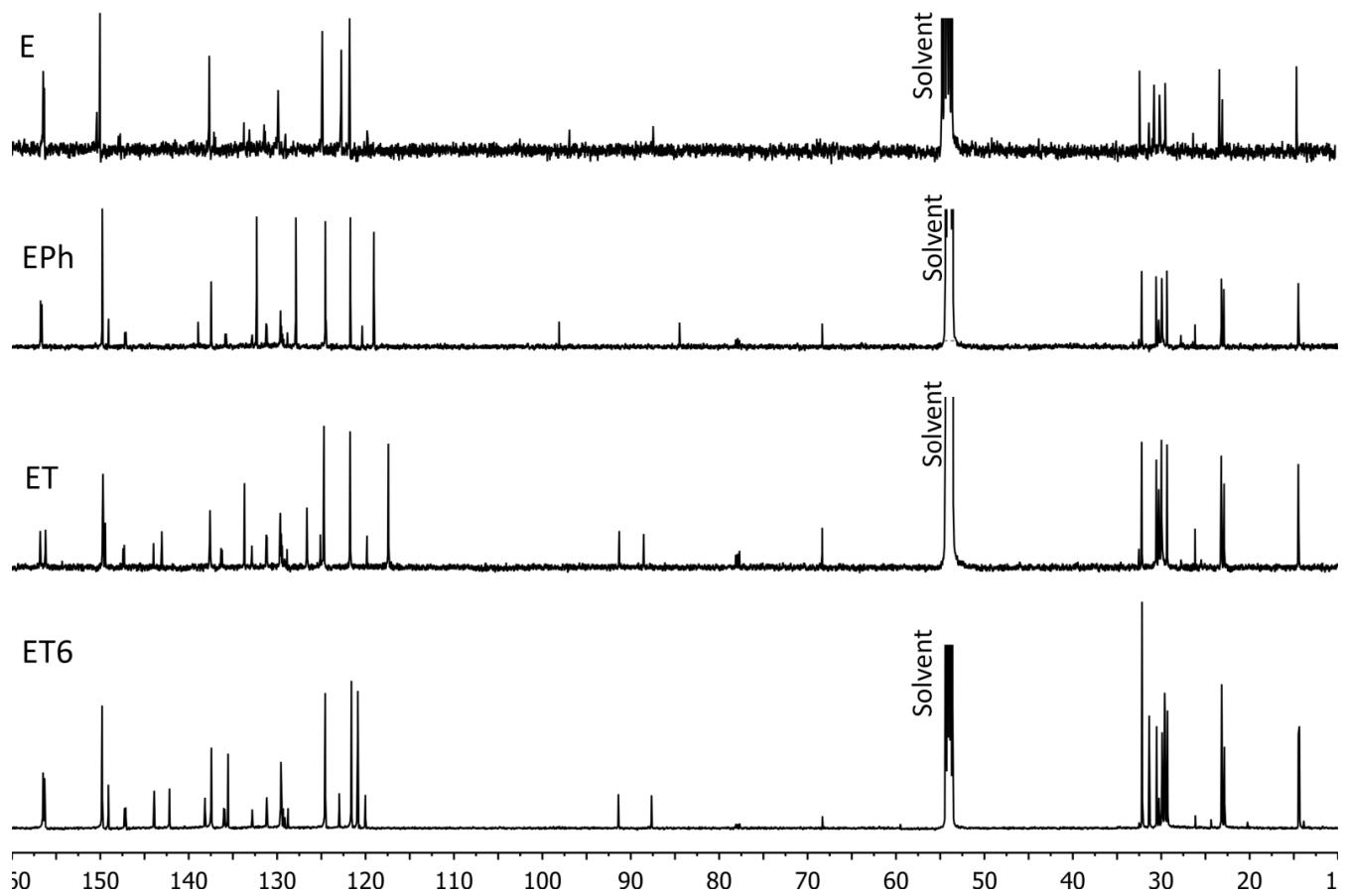


Figure S2. ^{13}C NMR spectra of prepared unimers (measured in CD_2Cl_2).

Monitoring of Zn-MSP and Fe-MSP assembly in solution

A measured volume of a solution ($2 \times 10^{-3} \text{ M}$) of Zn^{2+} or Fe^{2+} perchlorate in chloroform/acetonitrile mixed solvent was added into a unimer (**U**) solution ($2 \times 10^{-5} \text{ M}$) in the same solvent. The metal to unimer (M^{2+}/U) mole ratio: $r = \text{M}^{2+}/\text{U}$ was varied from 0 to 3. The UV/vis absorption and the photoluminescence emission spectra of solutions were measured at room temperature one day after preparation.

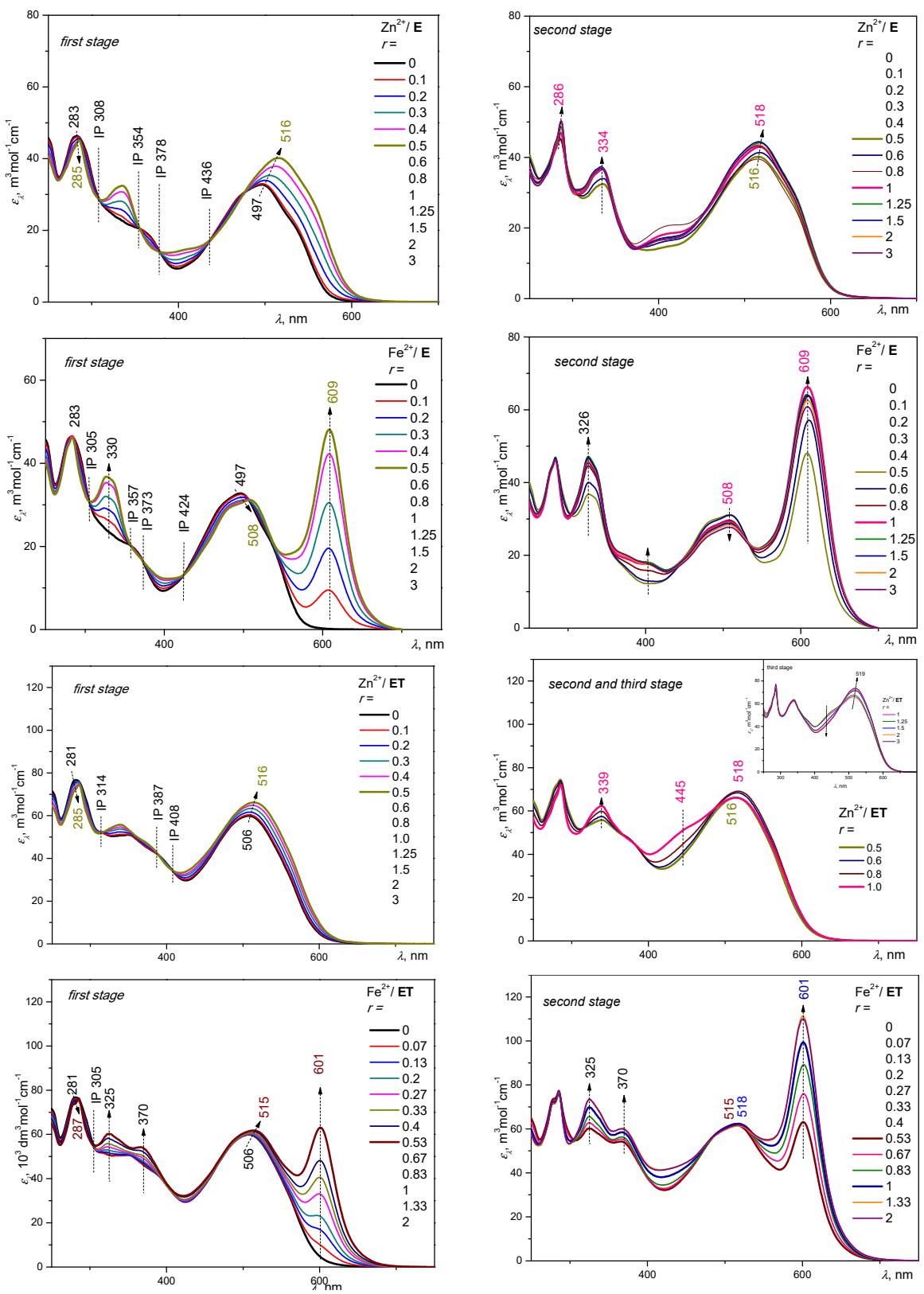


Figure S3-a. Changes in absorption spectra accompanying a stepwise assembly of MSPs from unimers **E** and **ET**.

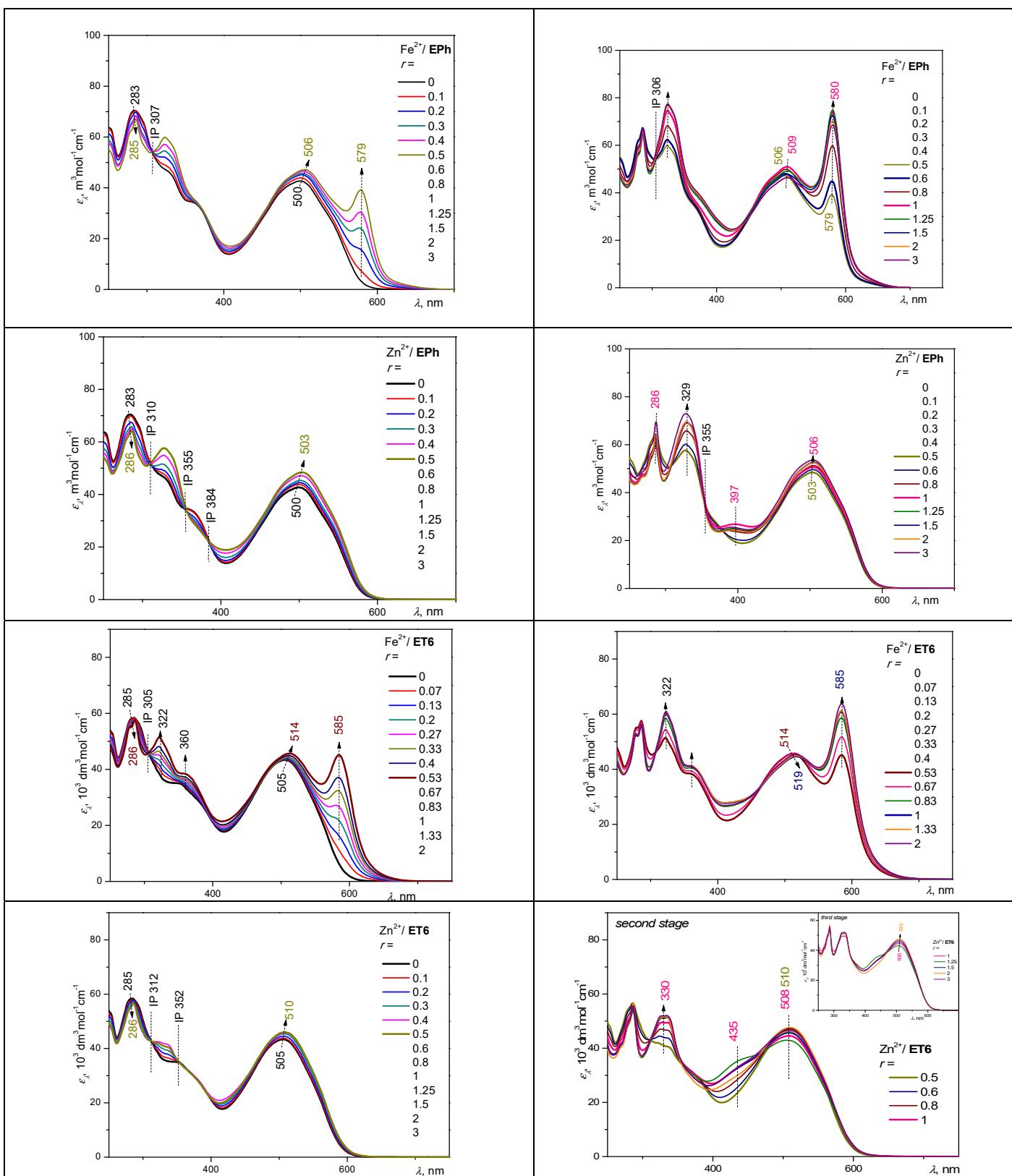


Figure S3-b. Changes in absorption spectra accompanying a stepwise assembly of MSPs from unimers **EPh** and **ET6**.

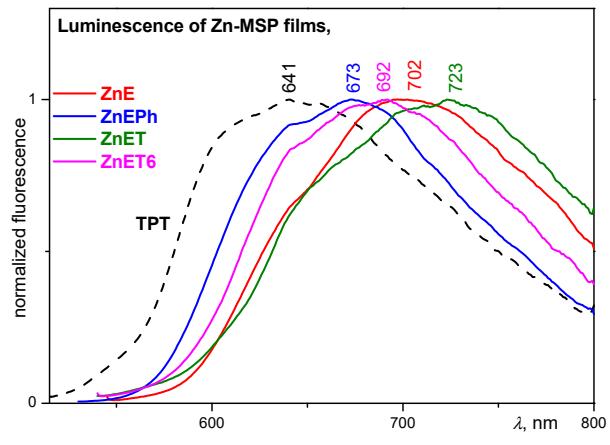
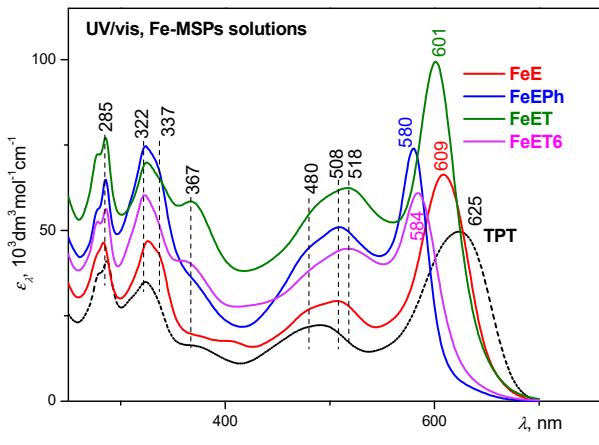
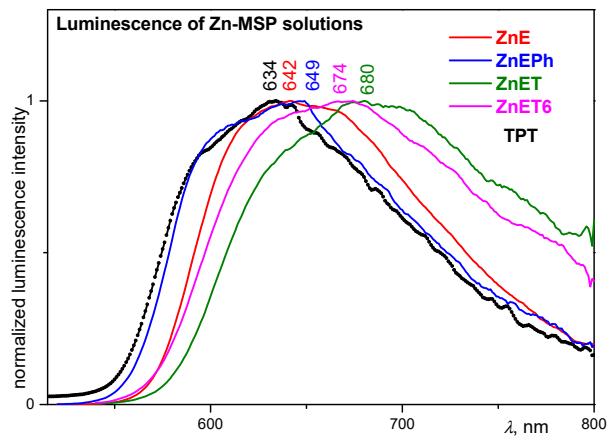
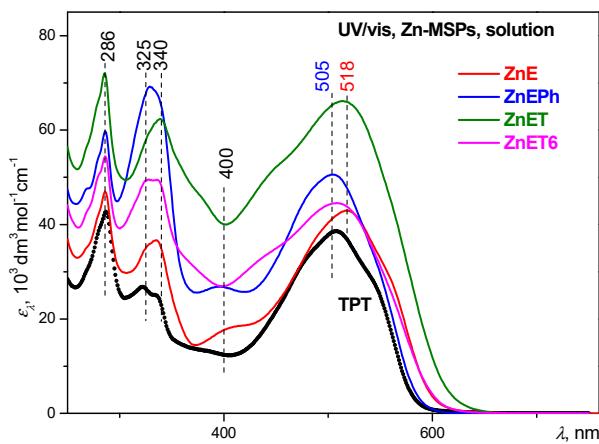
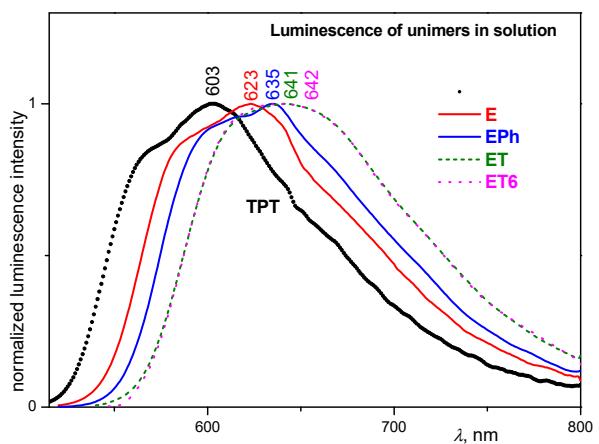
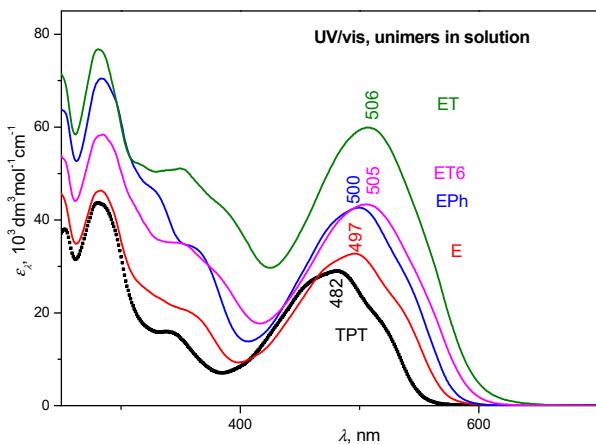


Figure S4. The UV/vis and photoluminescence solution spectra of unimers

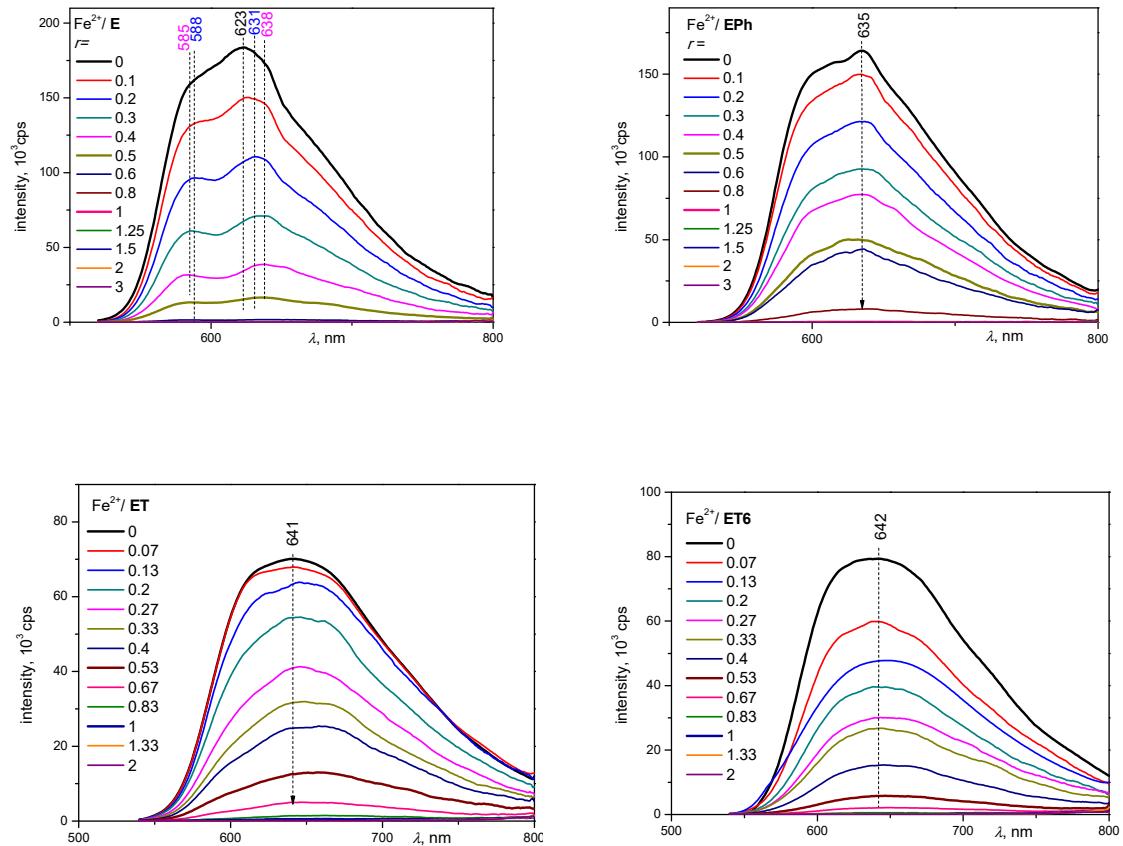


Figure S6. The unimer luminescence quenching in response to the unimer binding to Fe^{2+} ions. The luminescence totally disappears when the ratio r approaches the value of 1.

Thanks to the quenched luminescence, the resonance Raman spectra of Fe-MSPs can be easily measured, which is impossible in the case of ZnMSPs.

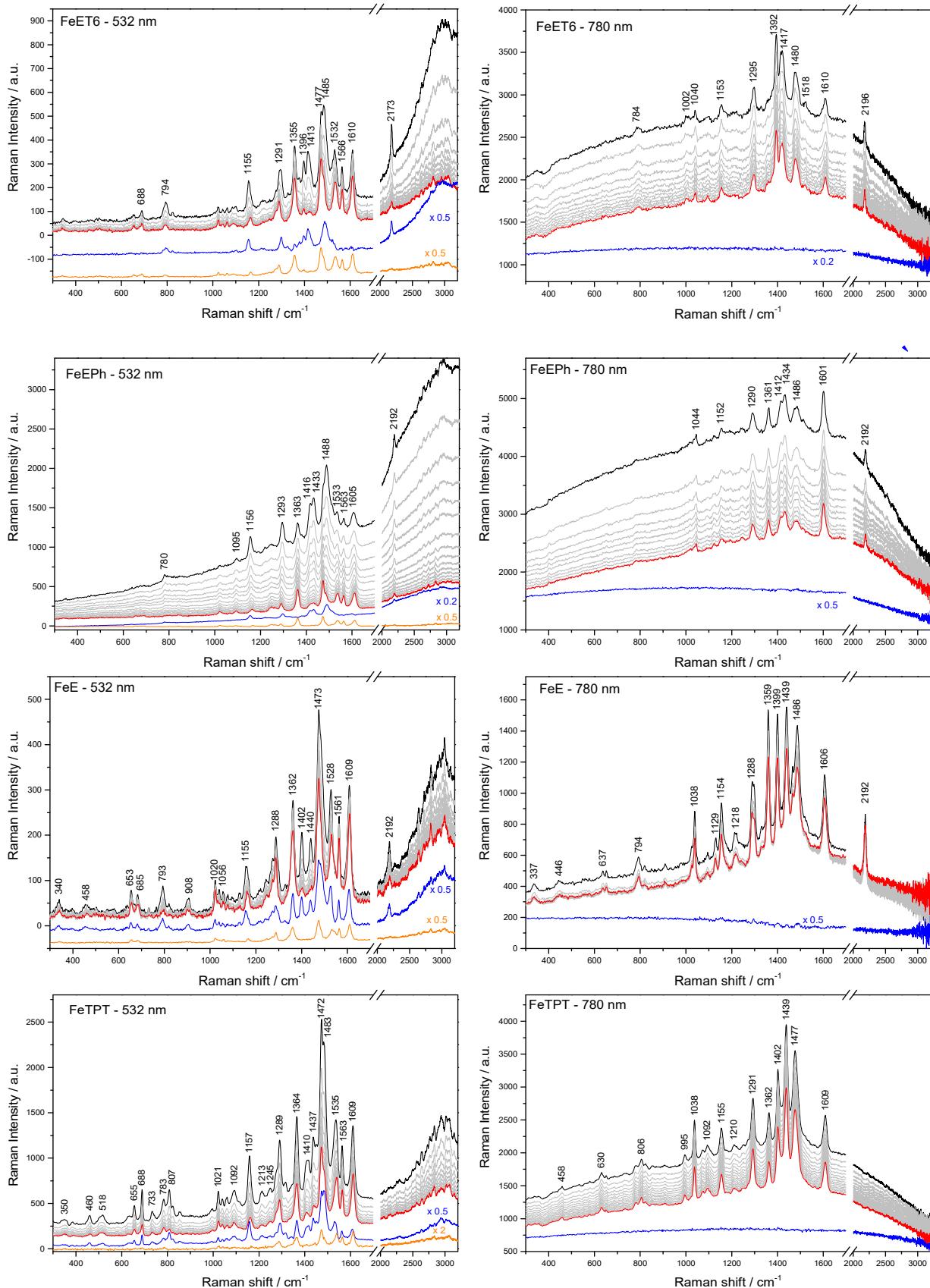


Figure S7. Time evolution spectral sets of the resonance (532 nm, 0.1 mW) and off-resonance (780 nm, 5 mW) Raman spectra; accumulation time 15 s; total time 6 - 9 min. (For FeET see the main part.

Processing of Raman spectral sets by Factor analysis (FA)

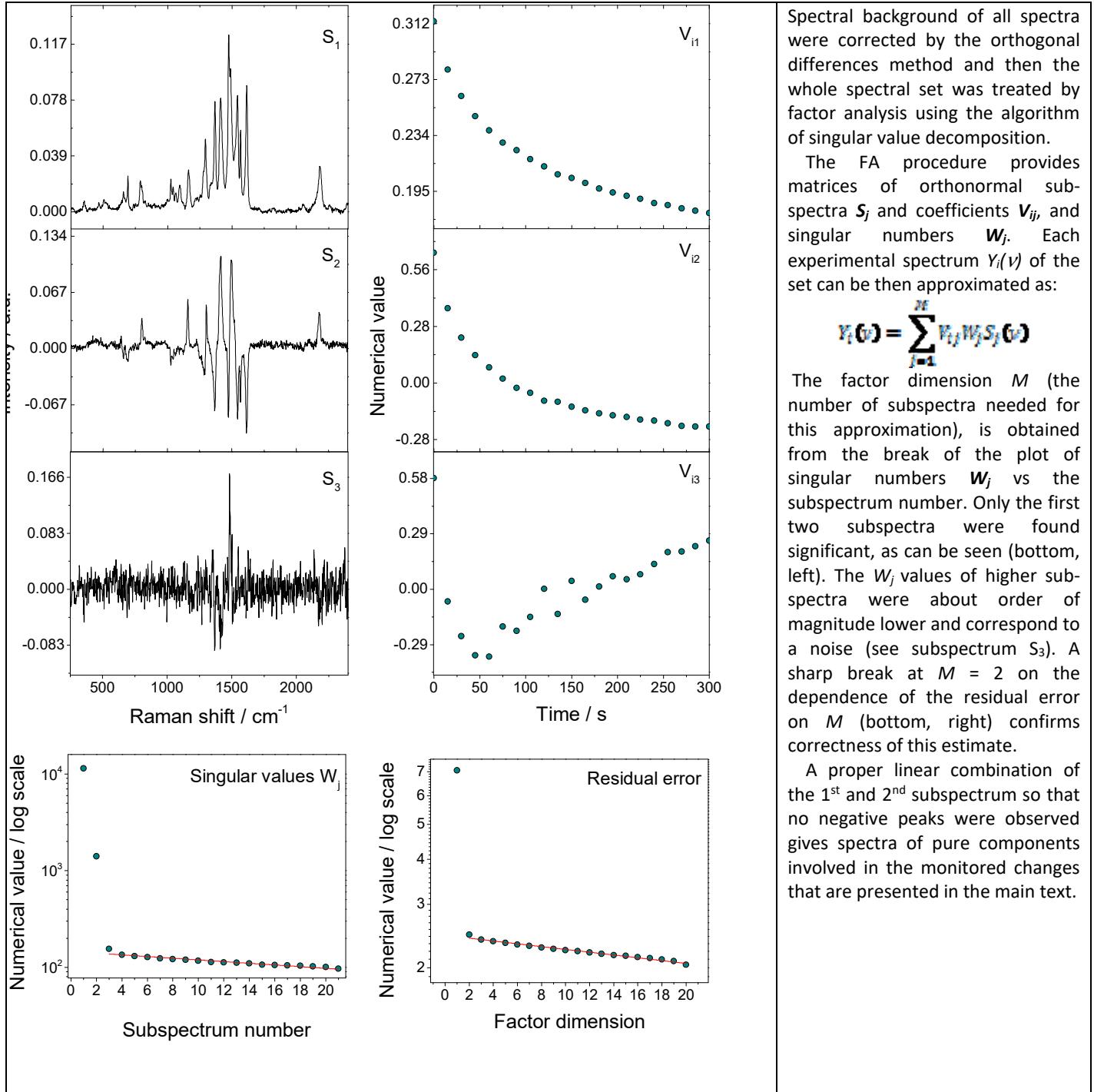


Figure S8. Results of factor analysis of time evolution spectral set of FeET, excitation 532 nm.

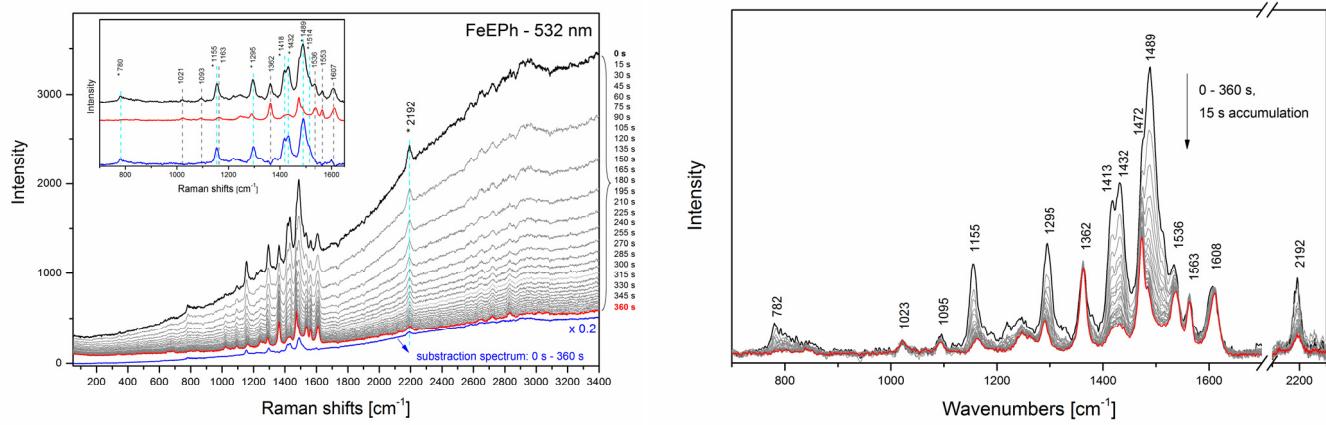


Figure S9. Left: Raman spectral set for **FeEPH** (laser power **0.1 mW**); INSET: **black**: starting spectrum; **red**: spectrum after 360 s; **blue**: difference spectrum **black – red**. Right: The spectral set after subtraction of fluorescence background to better see changes in spectral pattern.

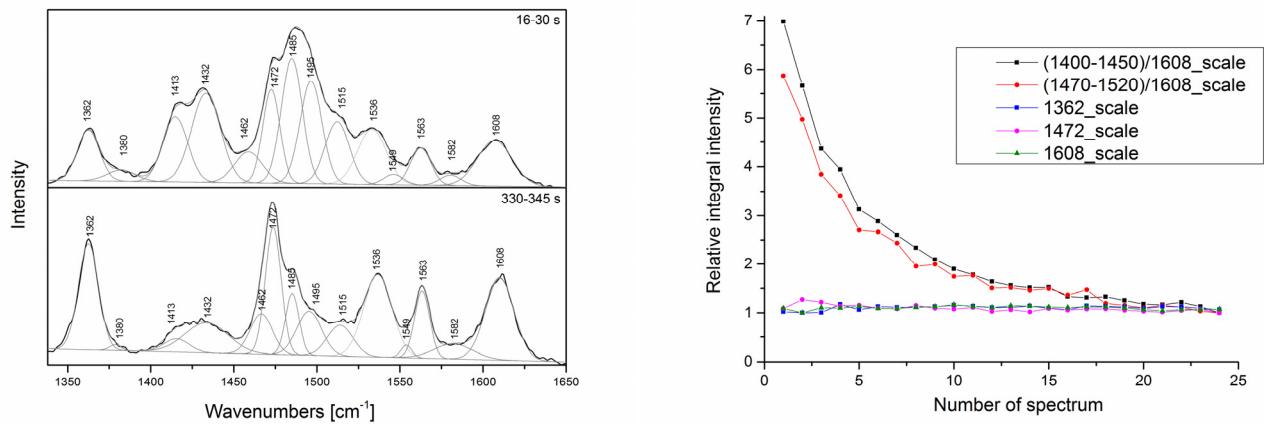


Figure S10. Deconvolution of the second and next-to-the-last spectrum of **FeEPH** and kinetics of obtained bands.

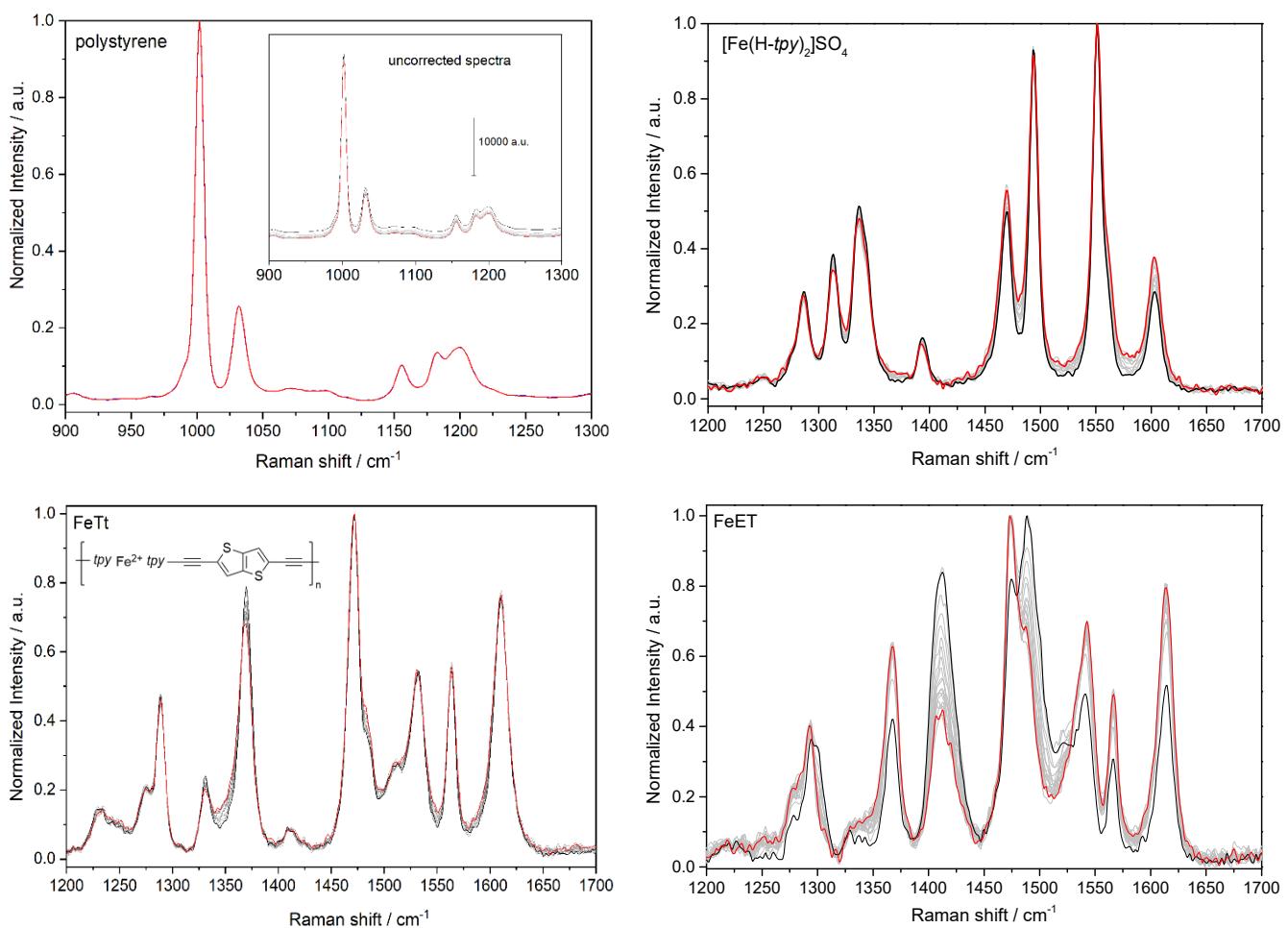


Figure S11 – Overlaid baseline corrected spectra of the spectral sets of indicated compounds, normalized to the highest spectral band; black – first measured spectrum (0 – 15 s), red – last measured spectrum (300 – 315 s).

The above figures demonstrate the reliability of the baseline corrections of spectra by the method of orthogonal differences. The uncorrected RS of polystyrene, thermally as well as photochemically stable material, provide identical RS after this correction. The resonance spectra of simple complex salt $[Fe(H-tpy)_2]SO_4$ and the 1-thioxophosphole-free polymer **FeTt** have shown a slight thermal broadening of Raman bands, whereas the RRS of **FeET** corrected in the same manner clearly showed convincing changes over time, which cannot be caused by baseline corrections.