

Synthesis of amphiphilic statistical copolymers bearing methoxyethyl and phosphorylcholine groups and their self-association behavior in water

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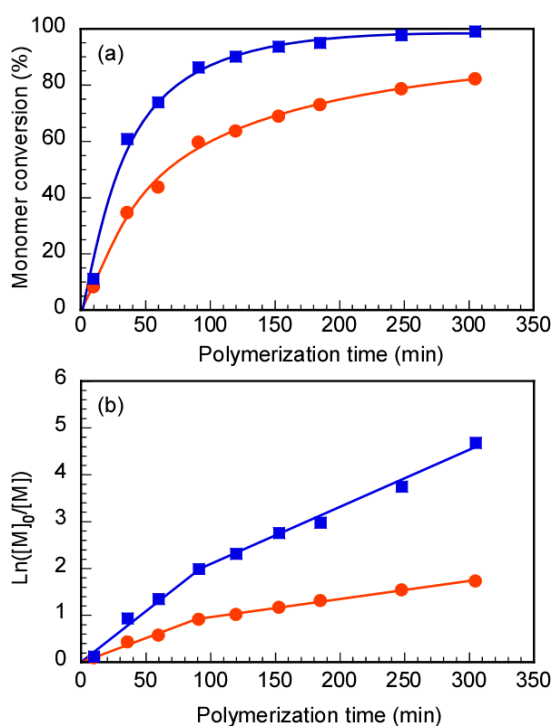


Figure S1. (a) Time-conversion and (b) the pseudo first-order kinetic plots for conventional free-radical polymerization of equimolar concentrations of MEA (●) and MPC (■) in methanol at 40°C. $[M]_0$ and $[M]$ were monomer concentrations at polymerization times of 0 and t min, respectively.

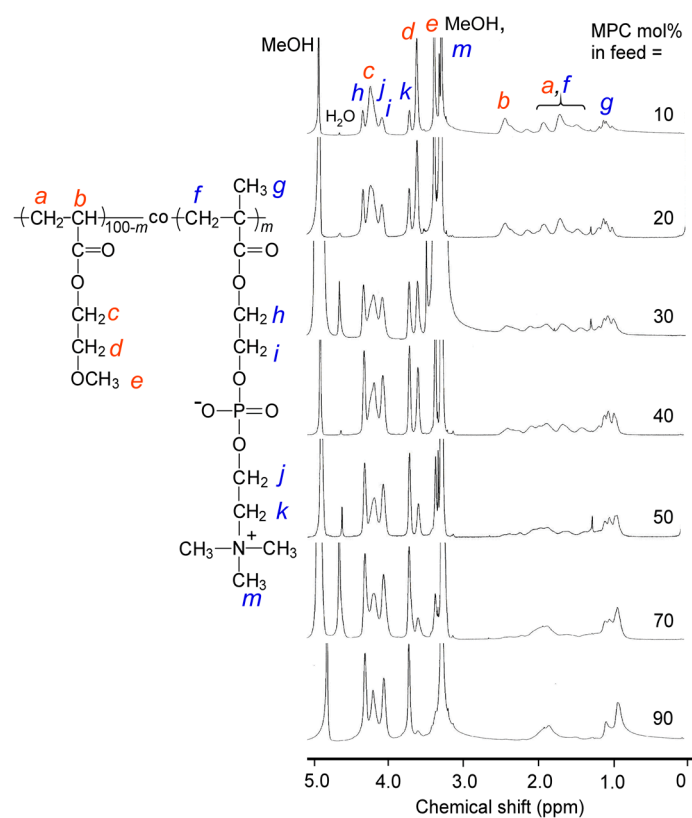


Figure S2. ^1H NMR spectra of $P(\text{MEA}/\text{MPC}_m)$ with various feed mol% of the hydrophilic MPC units in methanol- d_4 at room temperature.

Table S1. The Fineman–Ross parameters of the copolymers, as determined via ^1H NMR in methanol- d_4 at room temperature

Monomer in feed (mol%)		Monomer ratio in feed	Integral intensities of monomers in the copolymers		Content ratios of monomers in the copolymer	Parameters of Fineman–Ross equation	
MEA	MPC	$F = \frac{[M_{\text{MEA}}]_0}{[M_{\text{MPC}}]_0}$	MEA (I_1) ^a	MPC (I_2) ^b	$f = \frac{m_{\text{MEA}}}{m_{\text{MPC}}} = I_1/I_2$	F^2/f	$F(f-1)/f$
90	10	9.00	398.7	87.1	4.58	17.69	7.03
80	20	4.00	268.8	124.8	2.15	7.43	2.14
70	30	2.33	174.2	168.8	1.03	5.28	0.07
60	40	1.50	170.5	200.5	0.85	2.65	−0.26
50	50	1.00	99.3	219.2	0.45	2.21	−1.21
30	70	0.43	43.1	215.3	0.20	0.92	−1.71
10	90	0.11	9.89	214.8	0.05	0.27	−2.30

^a The integral intensities of the pendant methylene protons in the MEA units at 3.62 ppm. ^b

The integral intensities of the pendant methylene protons in the MPC units at 3.72 ppm.

Table S2. Composition of the copolymers, as estimated via ^1H NMR in methanol- d_4 at room temperature

Monomer in feed (mol)		m in feed ^a (mol%)	Integral intensities in the copolymers		m in the copolymer ^d (mol%)
MEA	MPC		MEA (I_1) ^b	MPC (I_2) ^c	
1.901	0.099	4.96	9.37	0.55	$5.54 \approx 6$
1.836	0.201	9.88	4.06	0.53	$11.55 \approx 12$
1.201	0.800	39.97	1.75	1.49	$45.99 \approx 46$

^a m in feed = $[M_{\text{MPC}}]_0 / ([M_{\text{MEA}}]_0 + [M_{\text{MPC}}]_0) \times 100$. ^b The integral intensities of the pendant methylene protons in the MEA units at 3.62 ppm. ^c The integral intensities of the pendant methylene protons in the MPC units at 3.72 ppm. ^d m in the copolymer = $I_2 / (I_1 + I_2) \times 100$.

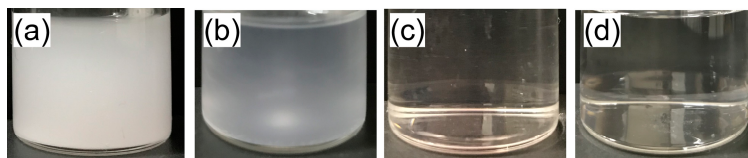


Figure S3. Photographs of (a) PMEA, (b) P(MEA/MPC₆), (c) P(MEA/MPC₁₂), and (d) P(MEA/MPC₄₆) solutions after dialysis using pure water.

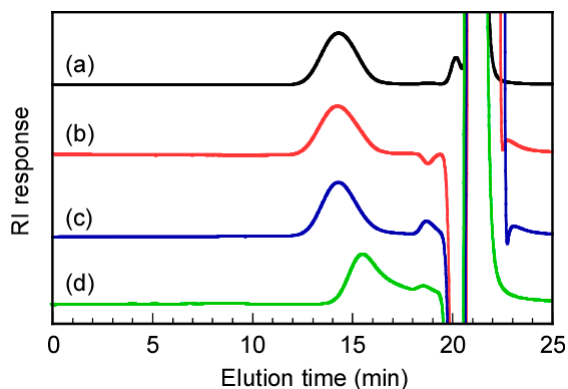


Figure S4. SEC elution curves for (a) PMEA, (b) P(MEA/MPC₆), (c) P(MEA/MPC₁₂), and (d) P(MEA/MPC₄₆) using methanol containing 0.1 M lithium perchlorate as the eluent at 40°C.

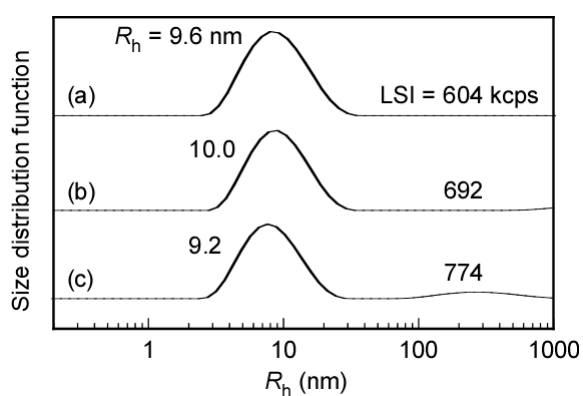


Figure S5. Hydrodynamic radius (R_h) distributions for (a) P(MEA/MPC₆), (b) P(MEA/MPC₁₂), and (c) P(MEA/MPC₄₆) in methanol at $C_p = 10$ g/L at 25°C.

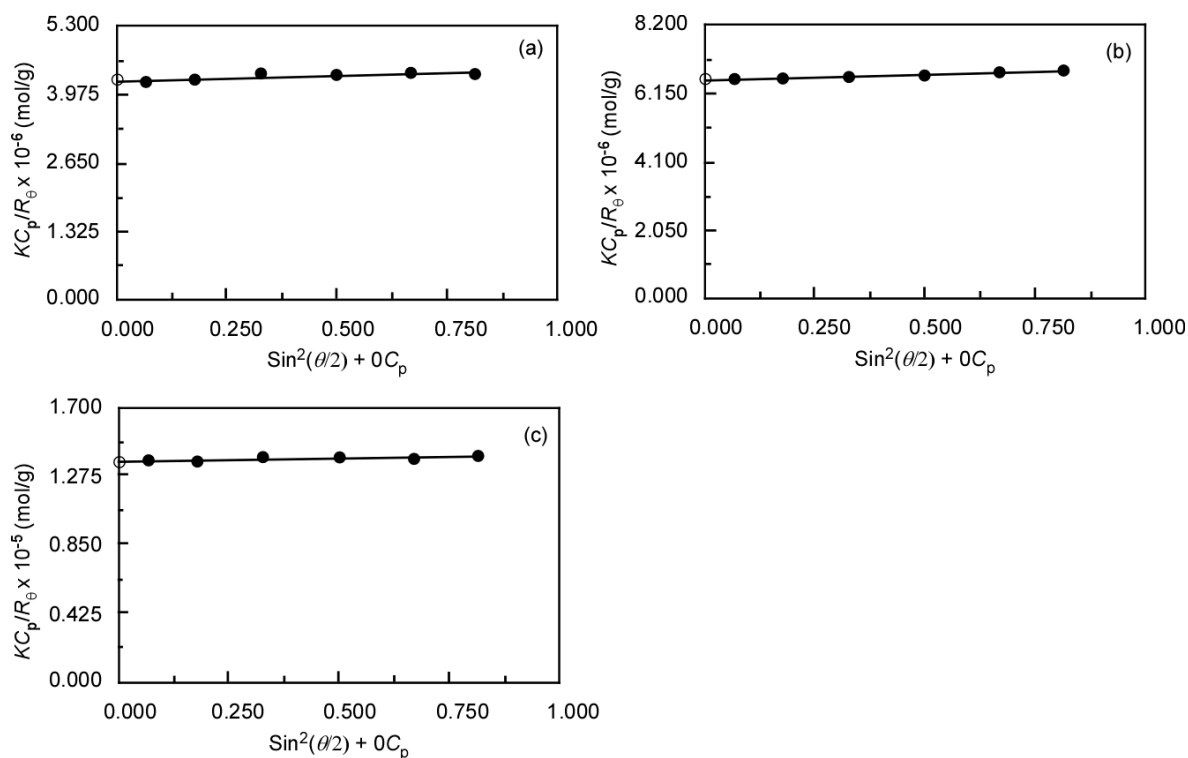


Figure S6. Zimm plots of (a) P(MEA/MPC₆), (b) P(MEA/MPC₁₂), and (c) P(MEA/MPC₄₆) in methanol at 25°C.

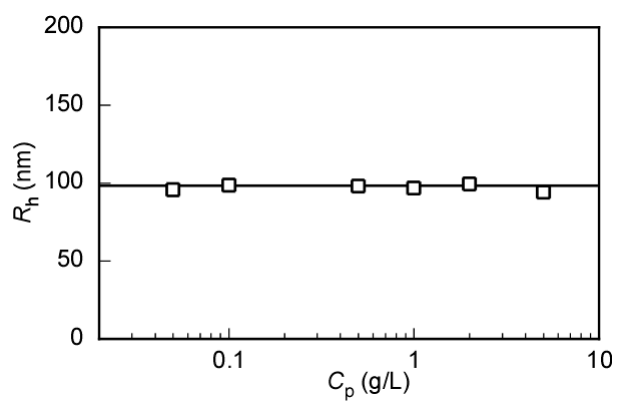


Figure S7. Hydrodynamic radius (R_h) of P(MEA/MPC₆) as a function of the polymer concentration (C_p) in water.

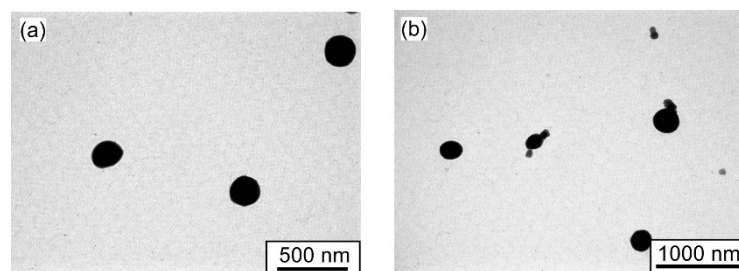


Figure S8. Transmission electron microscopy (TEM) images for P(MEA/MPC₆) at $C_p = 1.0$ g/L in water with different magnifications.

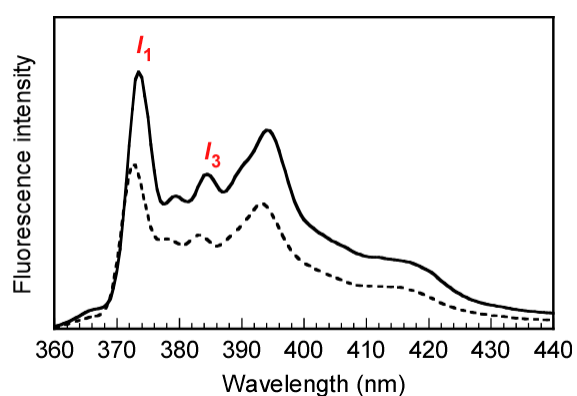


Figure S9. Fluorescence spectra of pyrene excited at 334 nm in water in the presence of P(MEA/MPC₆) at $C_p = 0.08$ (solid line) and 0.0012 g/L (dashed line).