

## 1 Appendix – Supplementary material

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Sample	Abbreviation	Molecular weight (kDa)
High molecular weight hyaluronan	HMW HYA	1255 ± 90
Low molecular weight hyaluronan	LMW HYA	309 ± 4
Diethylaminoethyl-dextran hydrochloride	DEAED	729 ± 32

3 **Table S1.** Molecular weights of all polysaccharides determined by SEC-MALLS technique

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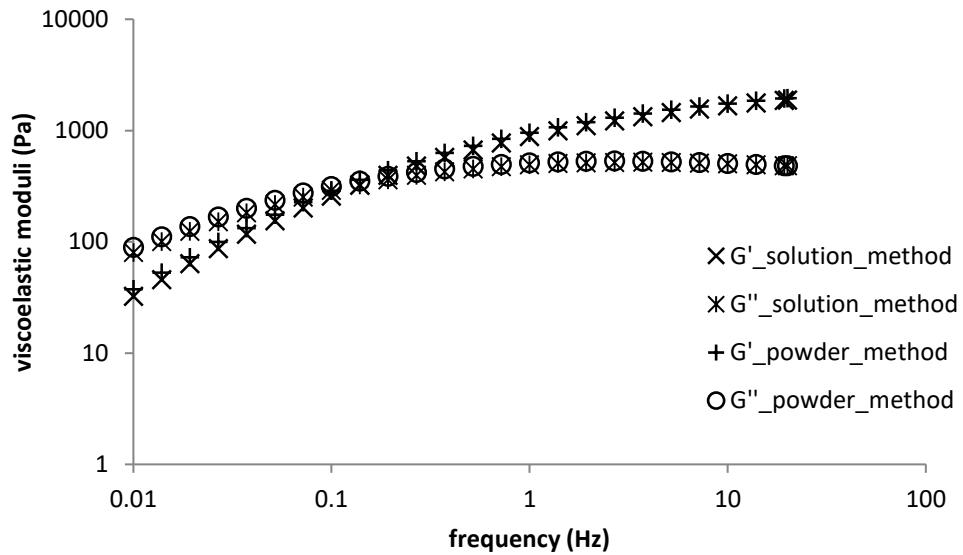
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pH range	Used chemicals	Buffer name
3,5 – 5,5	CH <sub>3</sub> COONa·3 H <sub>2</sub> O; CH <sub>3</sub> COOH	Acetate buffer
6 – 7,5	KH <sub>2</sub> PO <sub>4</sub> ; NaOH	Phosphate buffer
8 – 9	HCl; Na <sub>2</sub> B <sub>4</sub> O <sub>7</sub> ·10 H <sub>2</sub> O	Borate buffer
9,5 – 11	NaOH; Na <sub>2</sub> B <sub>4</sub> O <sub>7</sub> ·10 H <sub>2</sub> O	Borate buffer

7      **Table S2.** Substances used for buffers preparation for study of pH effect

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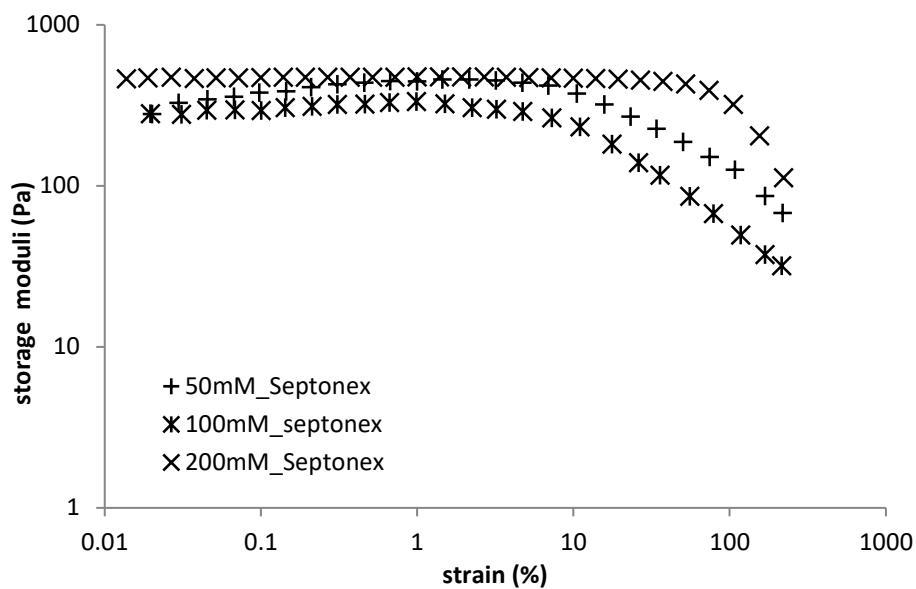


**Figure S1.** Comparison of viscoelastic properties of H1 sample prepared by different ways (solution vs. powder method)

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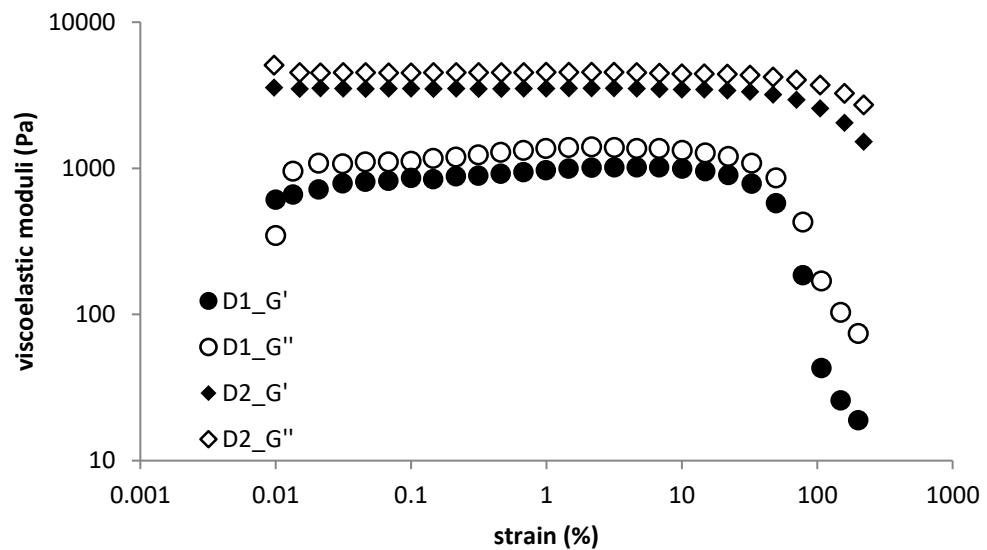
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**Figure S2.** Storage moduli ( $G'$ ) for HYA hydrogels prepared – concentration dependence

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**Figure S3.** Strain sweep for DEAED hydrogels – concentration dependence

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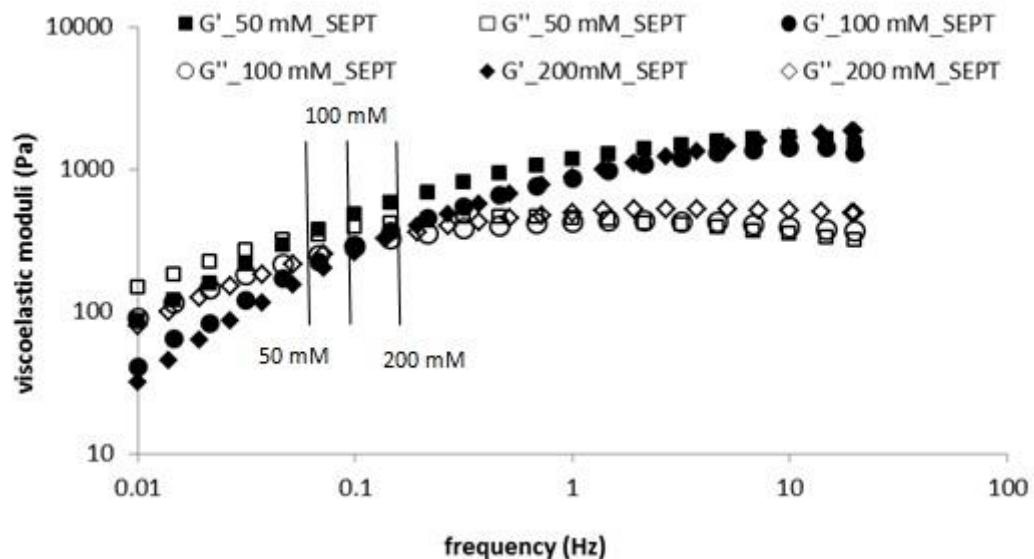
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Sample	End of linear viscoelastic region (%)
H1	19.30
H2	7.75
H3	5.91
H4	1.94
H5	4.77
H6	10.18
D1	21.98
D2	70.32

17      **Table S3.** The end of linear viscoelastic region range for hydrogels prepared from HMW and LMW  
18      hyaluronan

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**Figure S4.** Frequency sweep for H1, H2, H3 samples

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Sample name	G <sub>1</sub> (Pa)	λ <sub>1</sub> (s)	G <sub>2</sub> (Pa)	λ <sub>2</sub> (s)	G <sub>3</sub> (Pa)	λ <sub>3</sub> (s)	G <sub>4</sub> (Pa)	λ <sub>4</sub> (s)	G <sub>5</sub> (Pa)	λ <sub>5</sub> (s)
H1	803.1	0.009	607.2	0.069	509.3	0.352	286.6	1.824	72.2	12.240
H2	588.9	0.010	389.1	0.061	485.0	0.265	354.9	1.586	100.5	11.800
H3	508.7	0.010	435.2	0.072	508.5	0.340	464.3	1.735	194.8	11.830
H4	1576.0	0.010	542.2	0.082	142.8	0.209	78.3	0.780	20.3	1.976
H5	648.0	0.010	170.9	0.056	97.7	0.153	22.5	1.729	12.1	12.350
H6	511.3	0.009	228.0	0.074	72.7	0.456	18.5	1.958	9.6	12.010
D1	12170	0.013	953.1	0.140	471.5	0.478	16.39	1.856	6.262	8.379
D2	31380	0.009	4900	0.093	1954	0.488	365.6	1.872	30.79	11.470

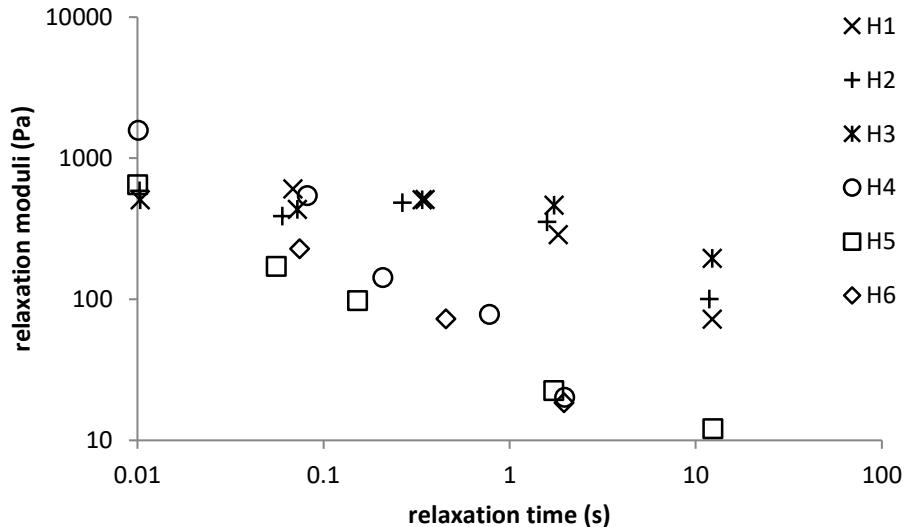
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**Table S4.** Relaxation spectra parameters for all tested hydrogels

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Sample name	Zero-shear viscosity (Pa.s)	Shear rate at the end of Newton plateau (1/s)
H1	1499 ± 105	0.03
H2	54 ± 6	0.09
H3	14.3 ± 0.7	0.06
H4	10.8 ± 0.4	0.29
H5	3.29 ± 0.08	2.94
H6	1.12 ± 0.05	0.20
D1	396 ± 8	2.00
D2	2715 ± 212	1.36

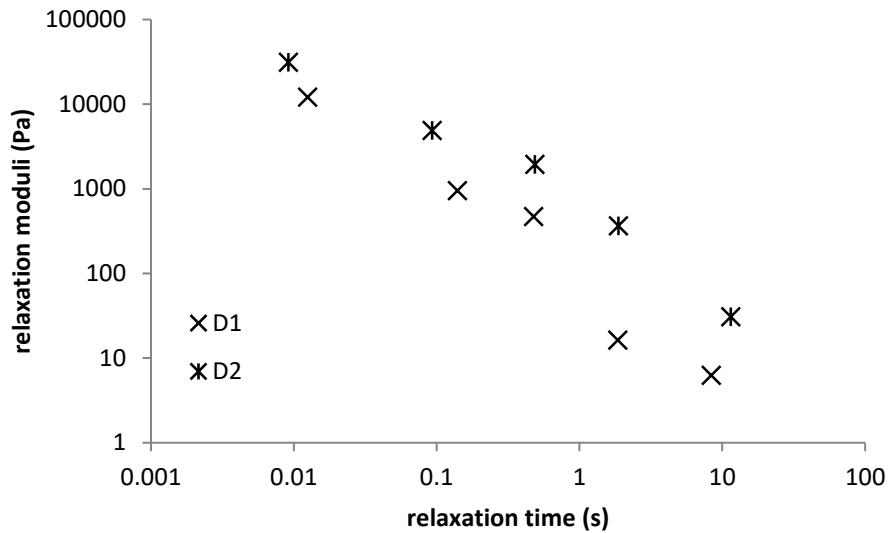
**Table S5.** Zero-rate viscosity and Newtonian plateau for all studied hydrogels



**Figure S5.** Relaxation spectra for H1-H6 samples

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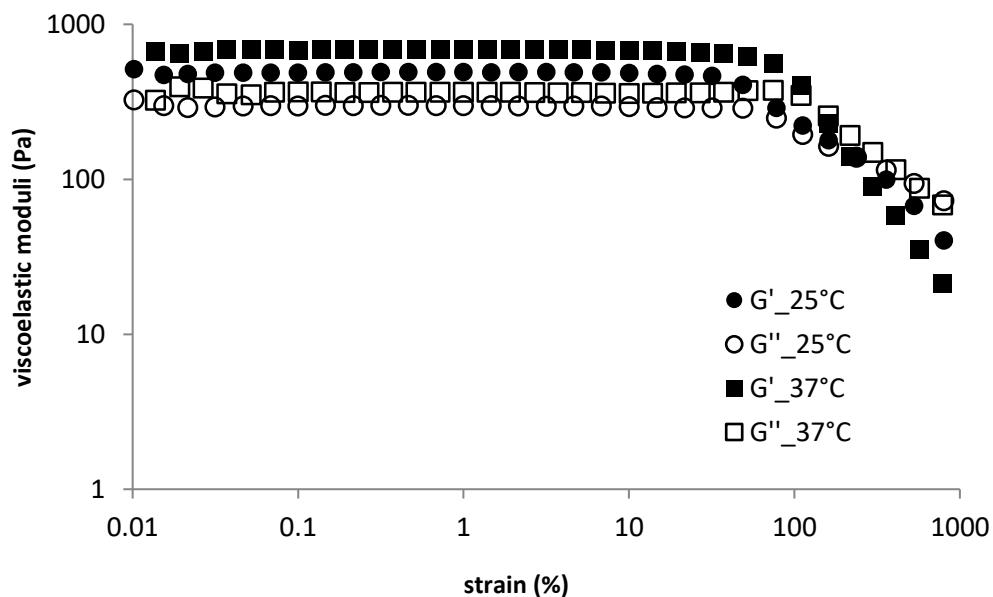
**Figure S6.** Discrete relaxation spectra for DEAED hydrogels – concentration dependence

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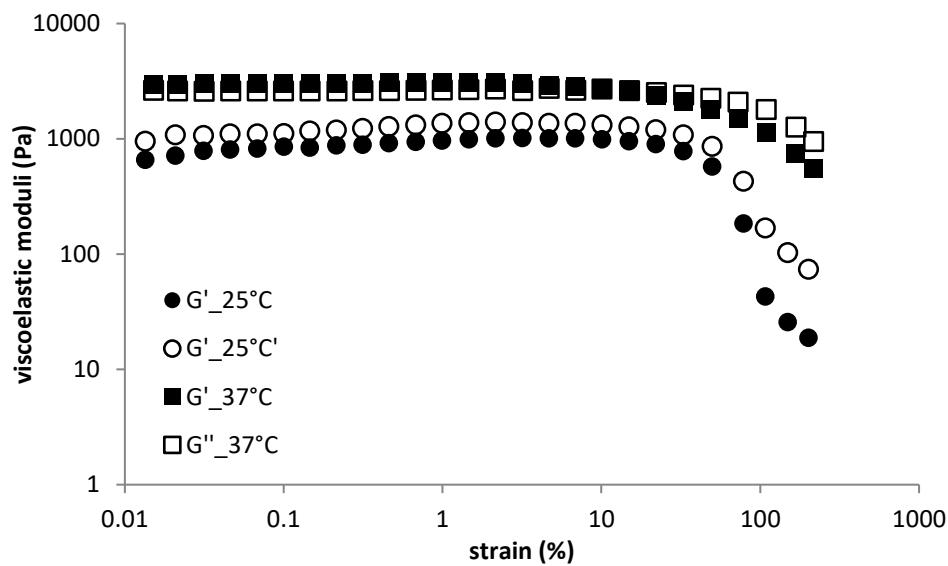
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**Figure S7.** Strain sweep for H1 sample - temperature dependence

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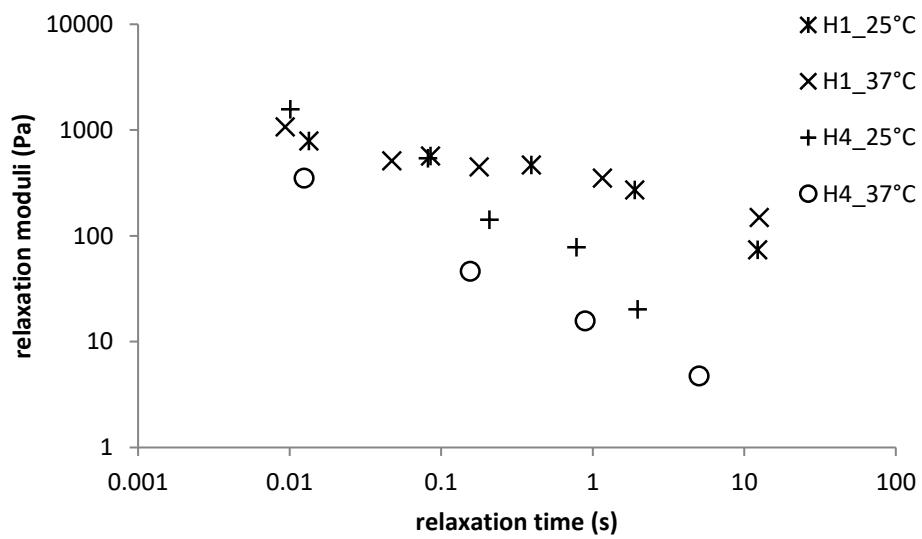


**Figure S8.** Strain sweep for D1 sample – temperature dependence

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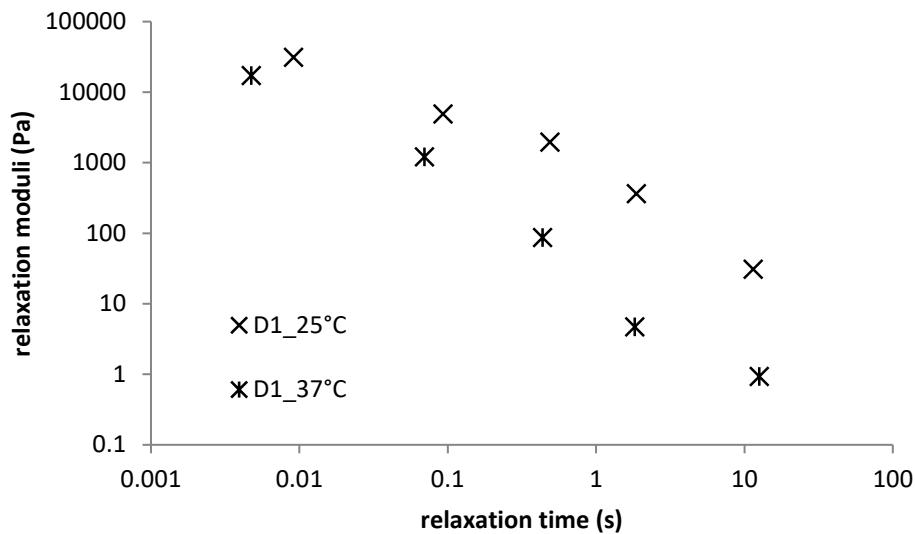


**Figure S9.** Discrete relaxation spectra for H1 and H4 samples – temperature dependence

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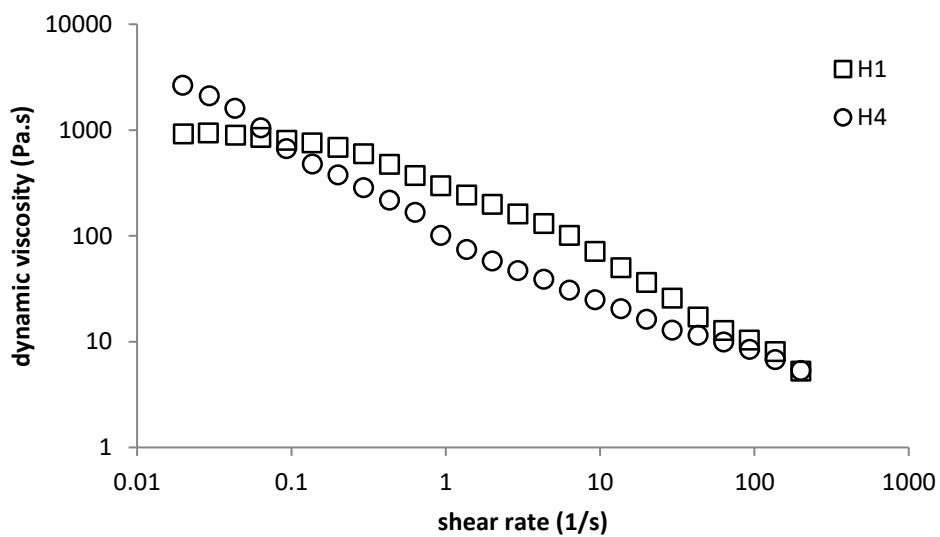
**Figure S10.** Discrete relaxation spectra for D1 sample – temperature dependence

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**Figure S11.** Flow properties of H1 and H4 samples (37 °C)

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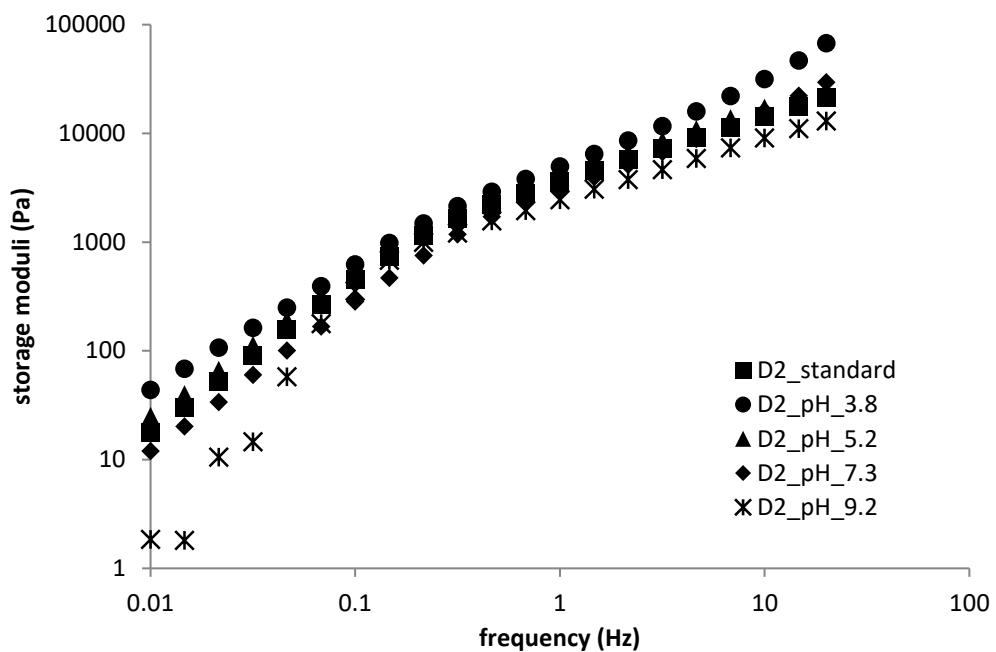
Sample	Linear viscoelastic region (%)				Viscoelastic moduli G'/G'' (Pa)			
	CaCl <sub>2</sub>	MgCl <sub>2</sub> ·6 H <sub>2</sub> O	FeCl <sub>3</sub>	Original samples	CaCl <sub>2</sub>	MgCl <sub>2</sub> ·6 H <sub>2</sub> O	FeCl <sub>3</sub>	Original samples
H1	75.63	75.15	103.59	19.30	277/198	544/377	1163/308	481/268
H2	50.83	225.96	32.15	4.75	1414/719	66/64	177/88	289/79
H3	29.28	15.66	14.94	6.91	415/201	635/216	300//99	42/13
H4	100.71	221.79	No gel	1.94	65/126	30/74	No gel phase detected	210/105
H5	149.95	216.91	No gel	4.77	109/199	7.6/30	No gel phase detected	168/36
H6	21.63	No gel phase detected	4.69	10.18	169/241	No gel phase detected	321/162	64/19
D1	1.51	14.99	10.15	21.98	7544/794 9	3321/466 4	1075/162 5	899/1206
D2	1.47	69.86	47.20	70.32	666/737	2216/255 3	699/835	2939/401 5

58      **Table S6.** Linear viscoelastic region ranges for hydrogels. Describe the effect of multivalent ions to  
 59      range of linear viscoelastic region.

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**Figure S12.** Storage moduli ( $G'$ ) for D2 sample – pH dependence

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Sample	Mesh size (nm)			
	original sample	CaCl <sub>2</sub>	MgCl <sub>2</sub> ·6H <sub>2</sub> O	FeCl <sub>3</sub>
H1	23.23	17.80	16.57	16.01
H2	24.55	15.82	15.93	20.55
H3	23.01	15.23	17.64	21.54
H4	39.00	22.39	22.08	25.98
H5	25.62	16.93	21.91	19.43
H6	27.77	15.40	19.43	23.17
D1	8.94	7.68	6.55	6.22
D2	7.12	-	9.29	13.82

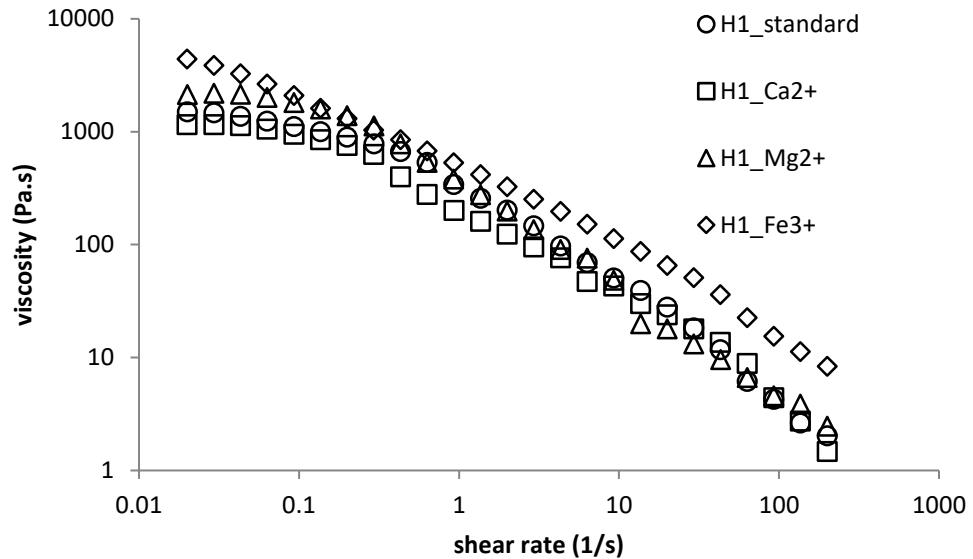
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**Table S7.** Calculated values for the mesh size of phase-separated hydrogels. Describe the effect of multivalent ions to mesh size in comparison with mesh size of original samples

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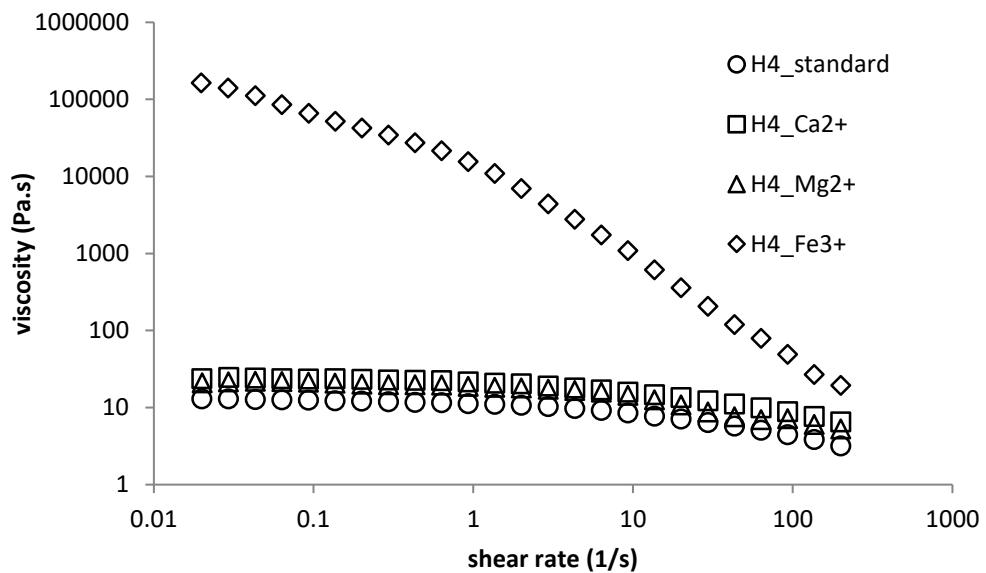
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**Figure S13.** Flow properties of H1 sample – the effect of multivalent ions

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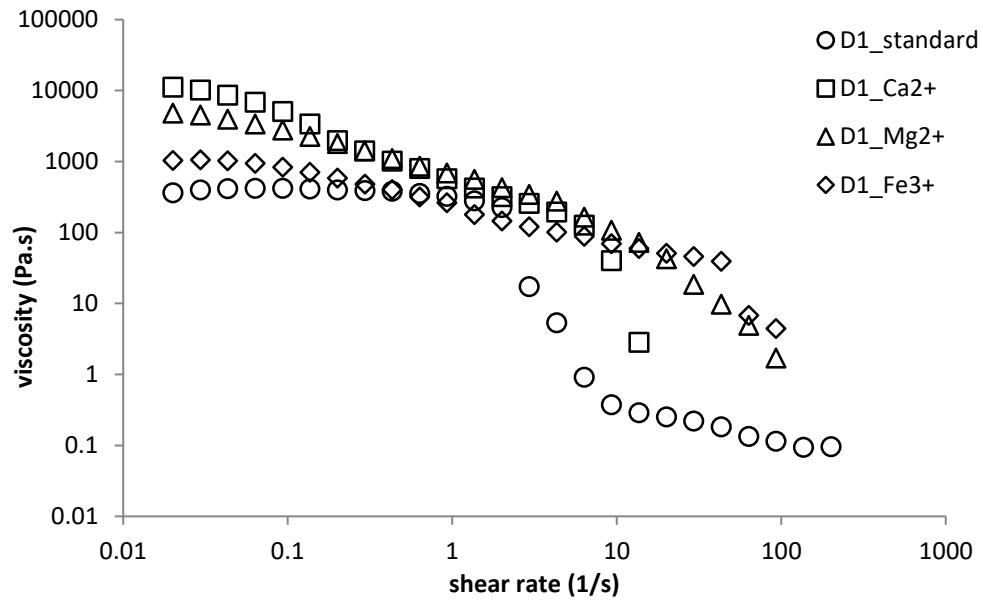
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**Figure S14.** Flow properties of H4 sample – the effect of multivalent ions

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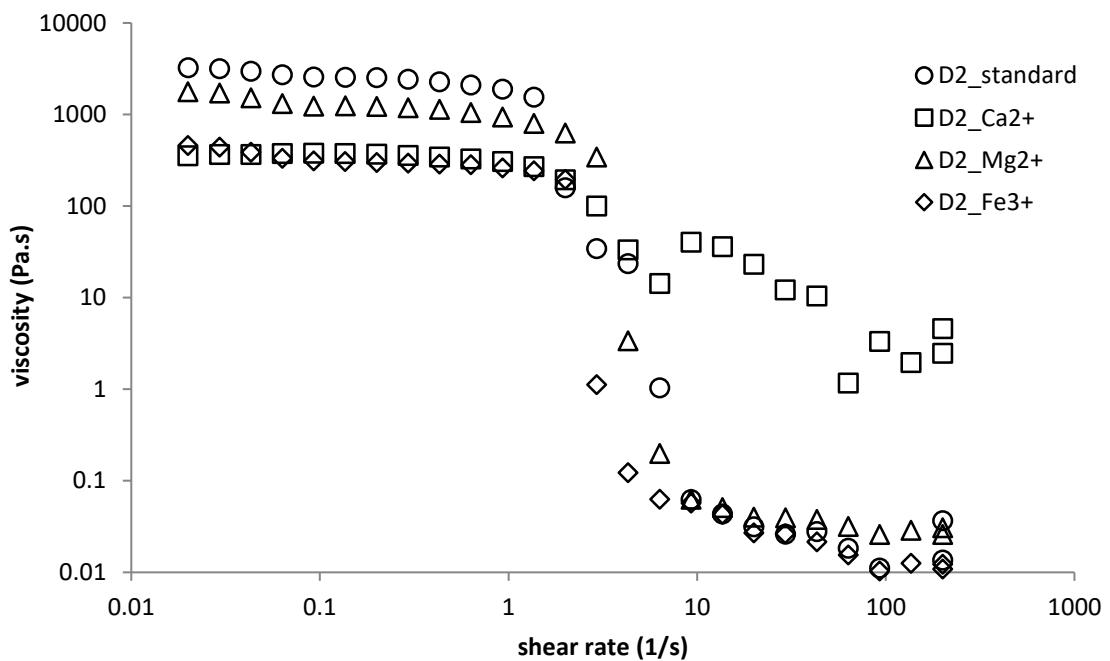
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**Figure S15.** Flow properties of D1 sample – the effect of multivalent ions

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**Figure S16.** Flow properties of D2 sample – the effect of multivalent ions

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