

Supplementary data for:

# Sometimes the same, sometimes different: understanding self-assembly algorithms in coordination networks

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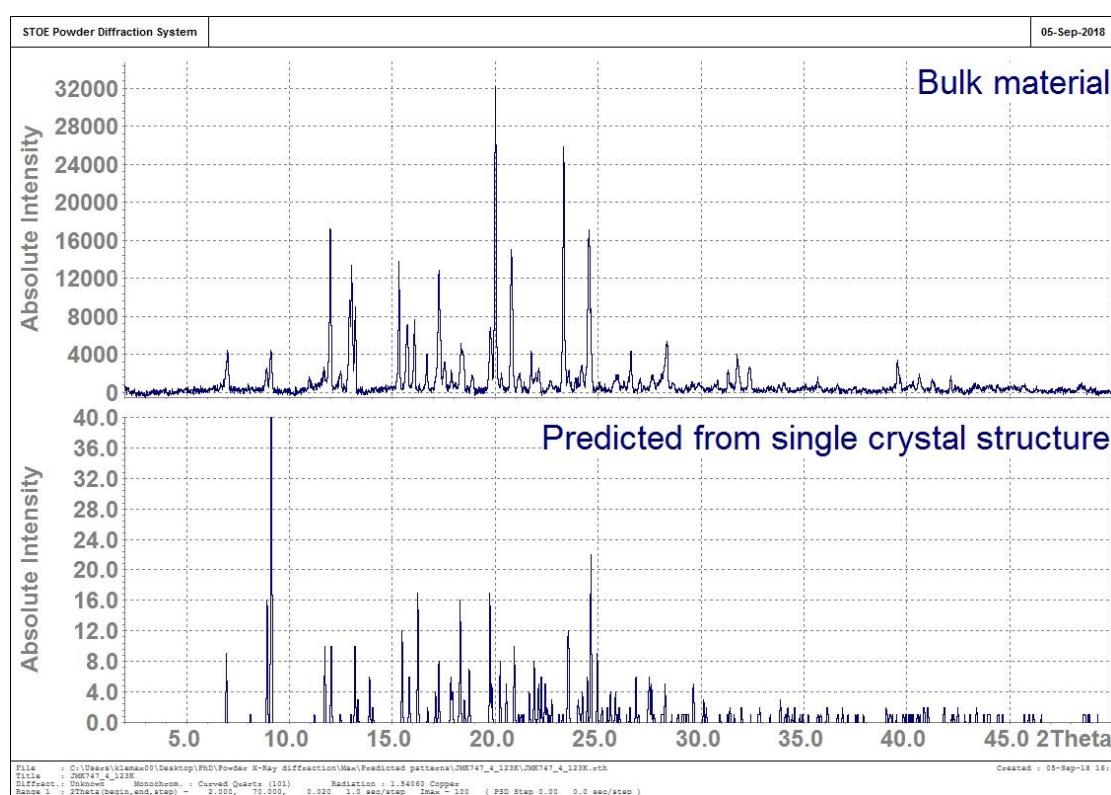


Fig. S1. Powder diffraction data for the bulk sample after removal of crystals from mother liquor compared to the powder pattern predicted from the single crystal structure of  $\{[\text{Co}(\text{NCS})_2(4)] \cdot 1.6\text{H}_2\text{O} \cdot 1.2\text{C}_6\text{H}_4\text{Cl}_2\}_n$ .

Sample Name **Max Klein / JMK302**  
Comment 10ug/mL in MeCN+0.1%TFA, measured in  
MeCN+0.1%TFA

Instrument maXis 4G  
Method 23 Direct\_pos\_higher.m

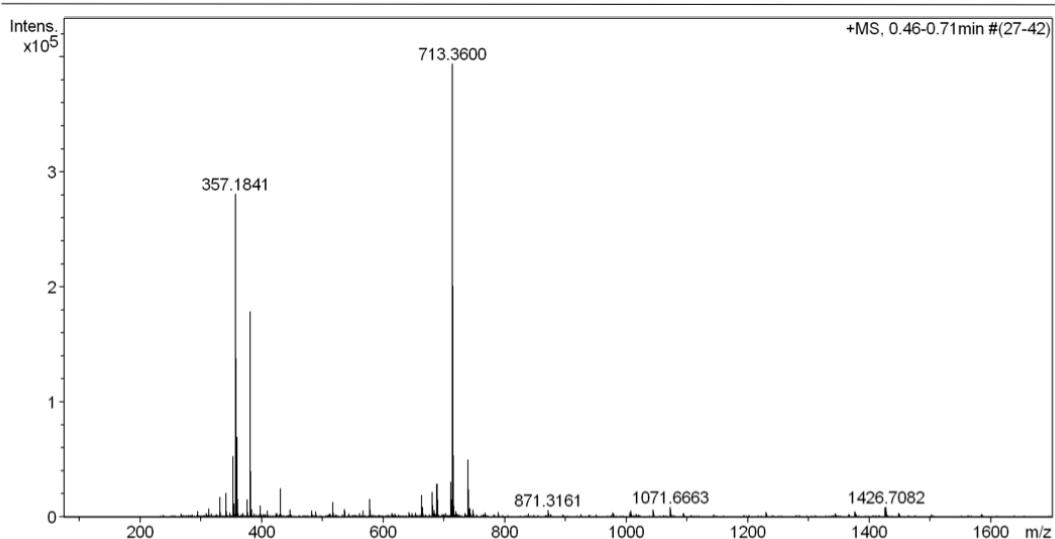


Fig. S2. High resolution electrospray (HR-ESI) mass spectrum of 3.

Sample Name **Max Klein / JMK822**  
Comment ~10 ug/mL in MeCN+0.1%TFA, analyzed in  
MeCN+0.1% TFA

Instrument maXis 4G  
Method 22 Direct\_pos\_mid.m

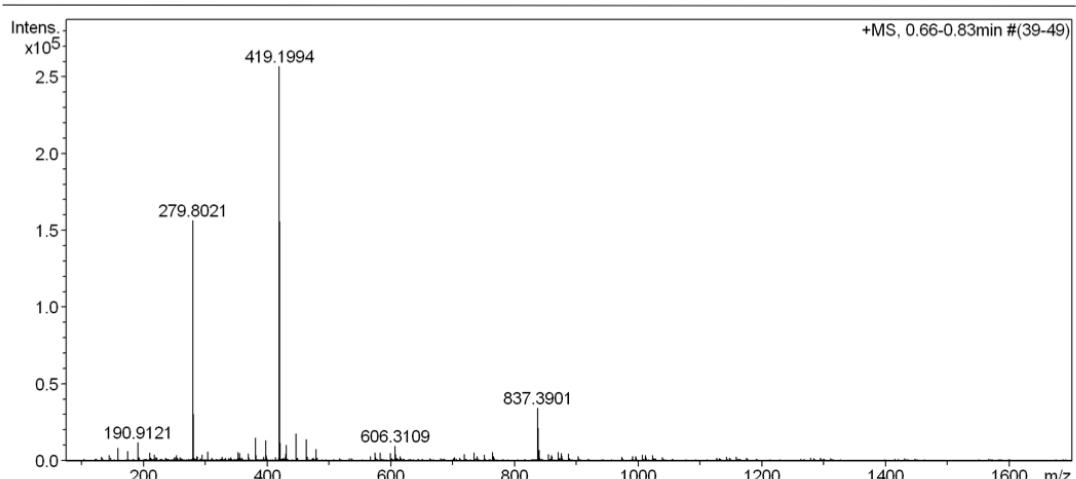


Fig. S3. High resolution electrospray (HR-ESI) mass spectrum of 4.

Sample Name	<b>Max Klein / JMK533</b>	Instrument	maXis 4G
Comment	10ug/mL in MeCN+0.1%TFA, measured in MeCN+0.1%TFA	Method	23 Direct_pos_higher.m

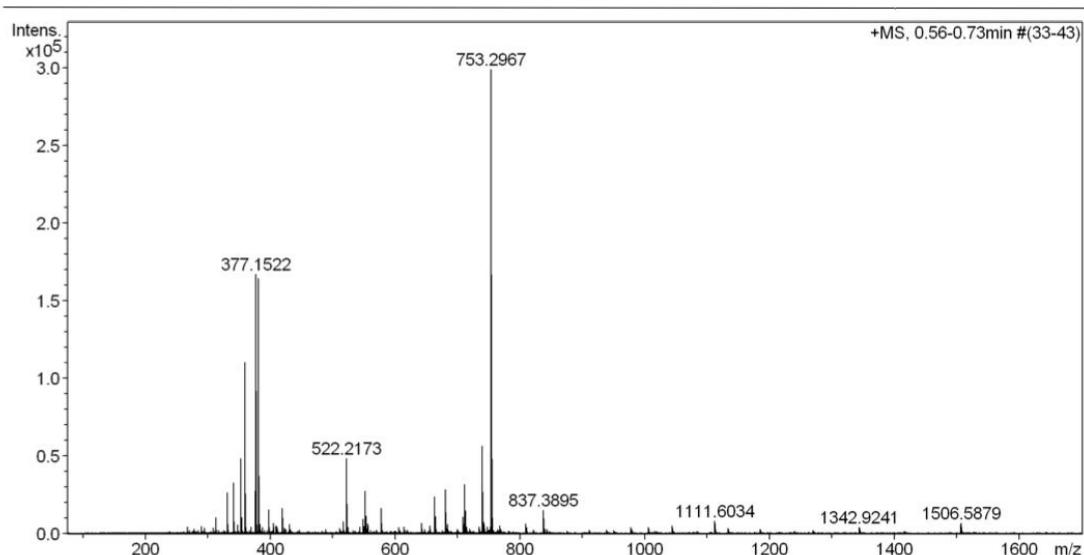


Fig. S4. High resolution electrospray (HR-ESI) mass spectrum of **5**.

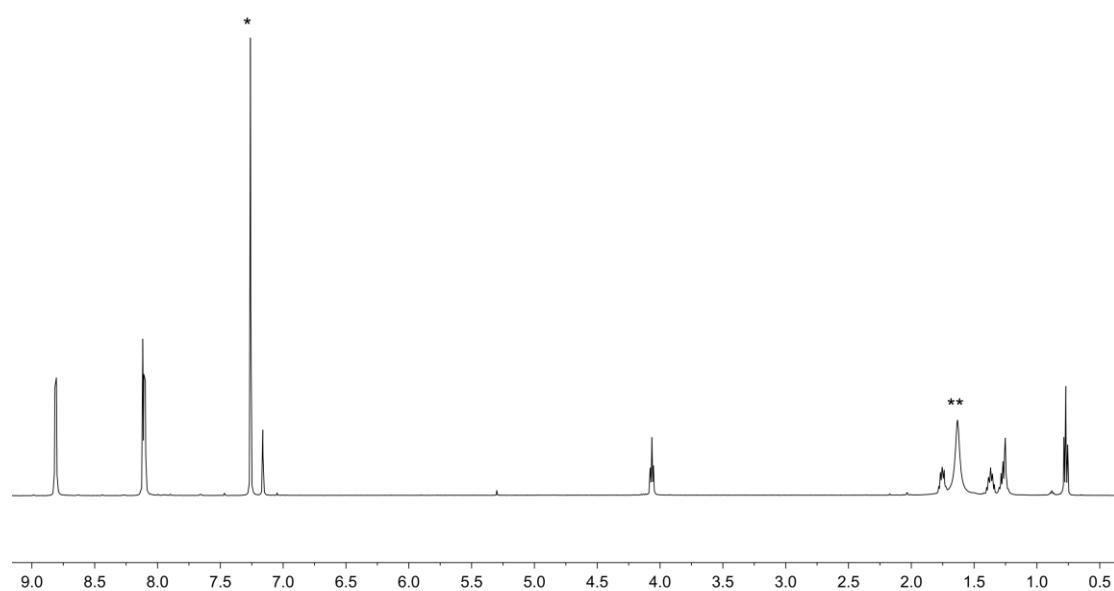


Fig. S5.  $^1\text{H}$  NMR spectrum of **3** (500 MHz,  $\text{CDCl}_3$ , 298 K). \* = residual  $\text{CHCl}_3$ ; \*\* =  $\text{H}_2\text{O}$ .

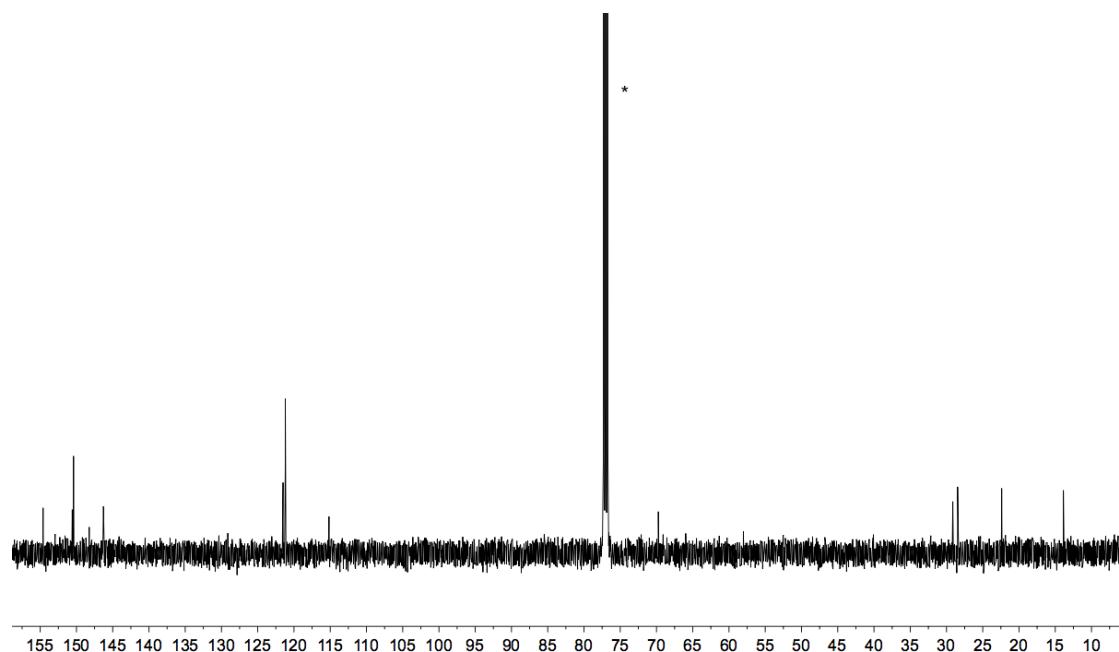


Fig. S6. <sup>13</sup>C NMR spectrum of **3** (126 MHz, CDCl<sub>3</sub>, 298 K). \* = CDCl<sub>3</sub>.

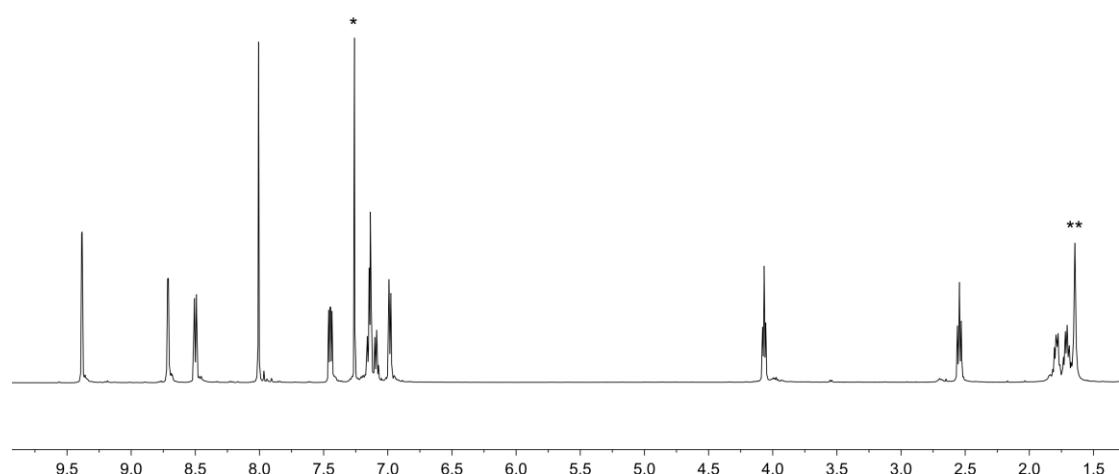


Fig. S7. <sup>1</sup>H NMR spectrum of **4** (500 MHz, CDCl<sub>3</sub>, 298 K). \* = residual CHCl<sub>3</sub>; \*\* = H<sub>2</sub>O.

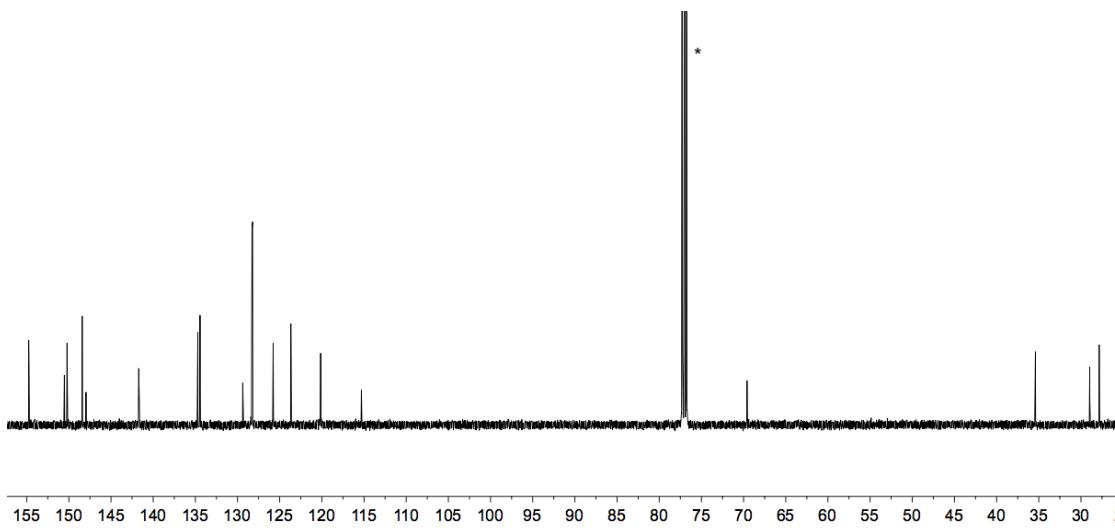


Fig. S8.  $^{13}\text{C}$  NMR spectrum of **4** (126 MHz,  $\text{CDCl}_3$ , 298 K). \* =  $\text{CDCl}_3$ .

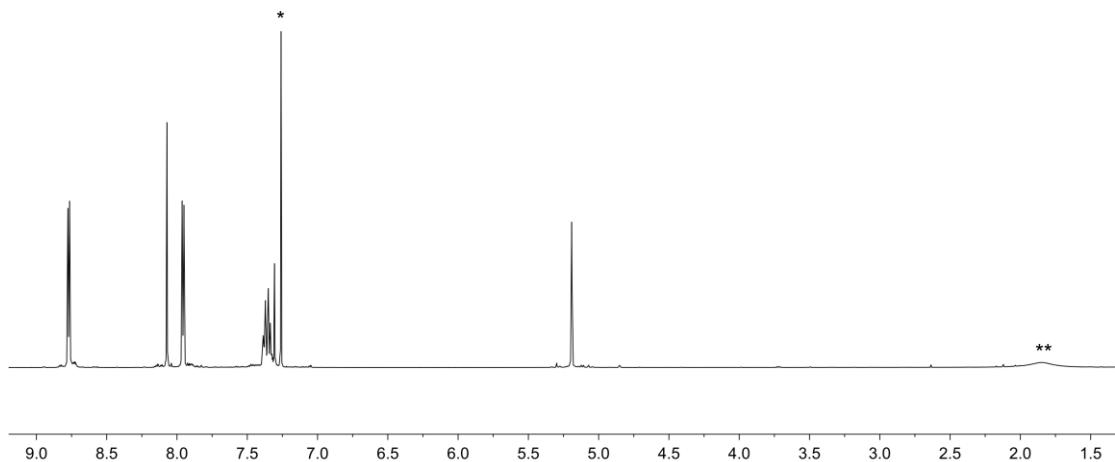


Fig. S9.  $^1\text{H}$  NMR spectrum of **5** (500 MHz,  $\text{CDCl}_3$ , 298 K). \* = residual  $\text{CHCl}_3$ ; \*\* =  $\text{H}_2\text{O}$ .

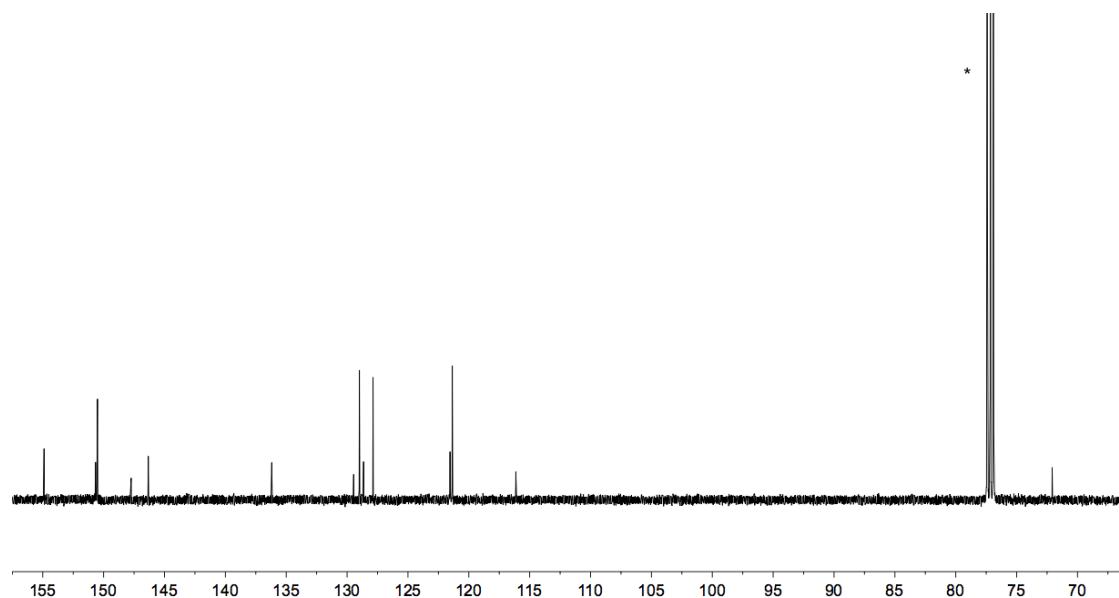


Fig. S10.  $^{13}\text{C}$  NMR spectrum of **5** (126 MHz,  $\text{CDCl}_3$ , 298 K). \* =  $\text{CDCl}_3$ .

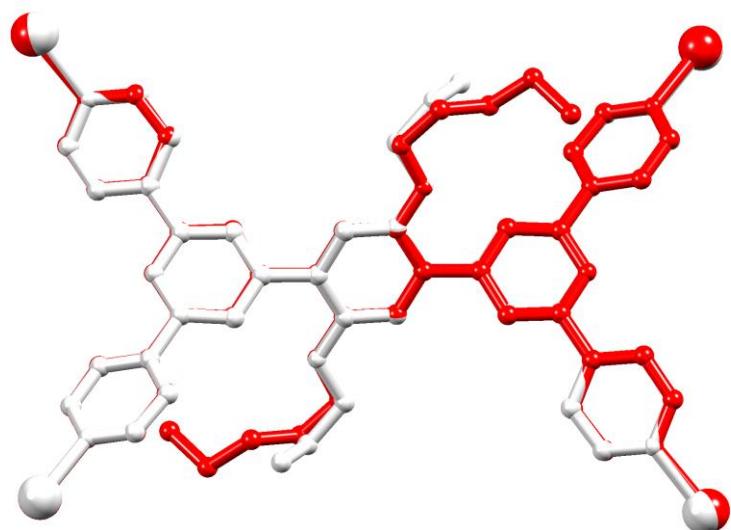


Fig. S11. Overlay of  $\{\text{Co}_4(\mathbf{3})\}$  (red) and  $\{\text{Co}_4(\mathbf{1d})\}$  (silver) units in  $\{[\text{Co}(\text{NCS})_2(\mathbf{3})] \cdot 0.8\text{C}_6\text{H}_4\text{Cl}_2\}_n$  and  $\{[\text{Co}(\text{NCS})_2(\mathbf{1d})] \cdot 2\text{C}_6\text{H}_4\text{Cl}_2\}_n$  [ref. 28]. In each overlay, the atoms of the central phenylene ring superimposed.

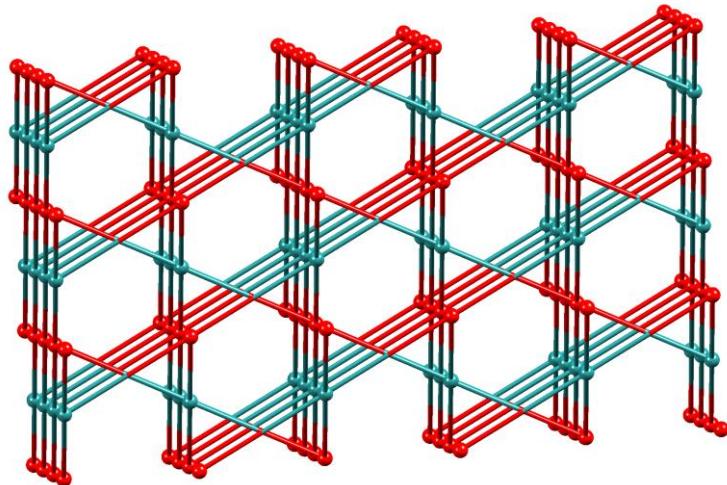


Fig. S12. Topological representation of the 3D net in  $\{[\text{Co}(\text{NCS})_2(\mathbf{3})] \cdot 0.8\text{C}_6\text{H}_4\text{Cl}_2\}_n$  generated using Mercury [34,35] with 4-connecting cobalt and ligand nodes.

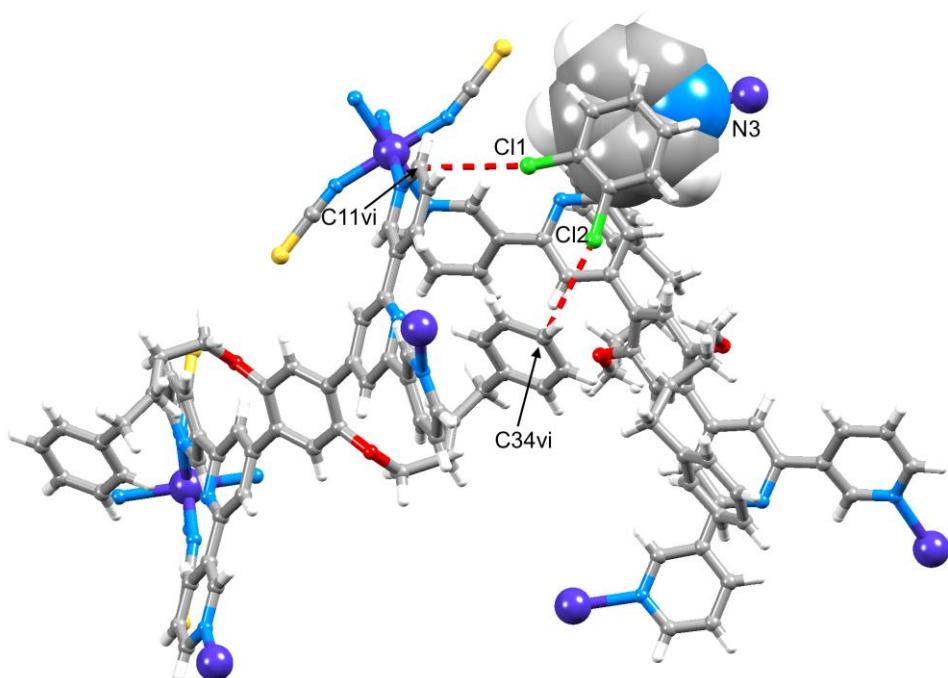


Fig. S13. Close contacts involving the 1,2-dichlorobenzene molecule in  $\{[\text{Co}(\text{NCS})_2(\mathbf{4})] \cdot 1.6\text{H}_2\text{O} \cdot 1.2\text{C}_6\text{H}_4\text{Cl}_2\}_n$ . Symmetry code vi =  $x, \frac{3}{2}-y, -\frac{1}{2}+z$  (see Fig. 3). Distances: Cl1...C11vi = 3.43, Cl2...C34vi = 3.42 Å. Corresponding Cl...H separations are 3.28 and 2.91 Å. See text for discussion of the π-stacking interaction.