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Review

Photoinduced Phase Transitions in α -, θ -, and κ -type ET Salts: Ultrafast Melting of the Electronic Ordering

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Abstract: Photoinduced phase transitions in organic compounds with strong electron correlation ET [bis(ethylenedithio)-tetrathiafulvalene)-based salts α -(ET)₂I₃, θ -(ET)₂RbZn(SCN)₄, κ -(*d*-ET)₂Cu[N(CN)₂Br] were discussed based, on time resolved optical pump-probe spectroscopy using ~150 fs mid-infrared pulse, 12 fs near infrared pulse, and sub-picosecond terahertz pulse. (i) In charge-ordered insulators α -(ET)₂I₃ and θ -(ET)₂RbZn(SCN)₄, we captured ultrafast snapshots of charge dynamics *i.e.*, immediate (ca. 15 fs) generation of a microscopic metallic state (or equivalently the microscopic melting of the charge order) which is driven by the coherent oscillation (period; 18 fs) of correlated electrons. Subsequently, condensation of the microscopic metallic state to the macroscopic scale occurs in α -(ET)₂I₃. However, in θ -(ET)₂RbZn(SCN)₄, such condensation is prevented by the large potential barrier reflecting the structural difference between the insulator and metal; (ii) In a Dimer–Mott insulator κ -(*d*-ET)₂Cu[N(CN)₂Br], photogeneration of the metallic state rises during ca. 1 ps that is much slower than the melting of charge order, because the photoinduced insulator to metal transition is driven by the intradimer molecular displacement in the dimer Mott insulator. The ultrafast dynamics of photoinduced insulator-metal transitions depend strongly on the molecular arrangement, reflecting various competing phases in the ET sheets.

Keywords: photoinduced phase transition; ultrafast spectroscopy; insulator-metal transition; charge order; Mott insulator

1. Introduction

The insulator to metal (I–M) transition is the most well known electronic phase transition in correlated electron systems, as described in the text book [1]. The I–M transition is a melting of the frozen charge, which is localized by Coulomb repulsion interaction U (on-site) and V (inter-site), *i.e.*, the charge-ordered (CO) insulator and Mott insulator. Figure 1a,b shows the schematic illustrations of the CO insulator (Figure 1a) and the Mott insulator (Figure 1b), respectively. Charges are localized in a pattern of 1, 0, 1, 0, 1, ... in the CO insulator and 1, 1, 1, 1, ... in the Mott insulator. The dramatic change of the electronic phase from insulator to metal attracts much attention in itself, but this phenomenon is also related to other exotic phase transitions involving superconductivity, ferroelectricity, and ferromagnetism. Many examples of I–M transition metal oxides [2] and/or low dimensional organic conductors [3,4] (Figure 2a). In addition, photoexcitation can induce I–M transition [5–16] (Figure 2b).

Figure 1. Schematic illustrations of (a) CO insulator and (b) Mott insulator. U, V, and t respectively represent on-site and inter-site Coulomb energies and an intermolecular transfer integral.

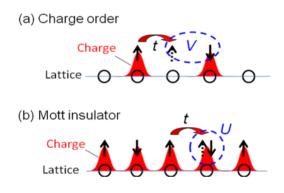
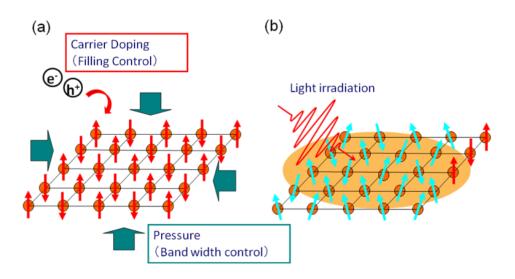


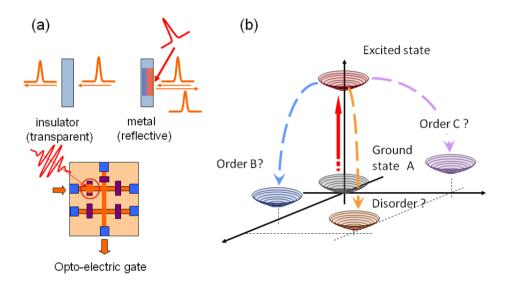
Figure 2. (a) Schematic illustrations of filling control (FC) and bandwidth control (BC) I–M transition and (b) the photoinduced I–M transition.



The photoinduced I–M transitions (PIMTs) are dynamic melting of the electron ordering in a Mott insulator and the CO insulator. The Mott and the CO insulators sometimes compete with the dielectrically and/or magnetically ordered phases or the superconducting state [1–4].

Therefore, PIMT is expected to induce electronic or magnetic order. Ultrashort laser technology enables us to realize high-speed switching in terabit/s of optical, electronic and/or magnetic properties (Figure 3a). Furthermore, discovery of new material phases (orders B and C in Figure 3b) as non-equilibrium states on the free energy surface with multi stabilities is expected.

Figure 3. Schematic illustrations of (**a**) optical switch and (**b**) photoinduced phase transition from the ground state to non-equilibrium states on the free energy surface with multiple stabilities.



Layered organic compounds $(ET)_2X$ [ET; bis(ethylenedithio)tetrathiafulvalene, hereafter ET, X; anion (acceptor)] are typical of organic compounds with strong electron correlation [3,4]. The series of this compound consists of ET (donor, D) and acceptor (A) sheets, as portrayed schematically in Figure 4. A characteristic feature of this organic salt is that the difference in the molecular arrangement in the donor sheet shows various polytypes such as θ -, α -, and κ -types, as depicted in Figure 5. In D₂A-type charge transfer (CT) salts, the highest occupied molecular orbital band of D sheets (average charge of a D molecule is +0.5) is 3/4-filling. In such salts, a CO insulator, in which the electrons (or holes) are sometimes localized on the molecular site in a pattern as shown by a red circle, is stabilized at low temperature (Figure 5a,b).

The I–M transition is an electronic transition. However, such a dramatic change of the electronic state necessarily affects the lattice, although the manner of electron–lattice interaction depends strongly on the kind of compound. The I–M transition in the typical ET salts α -(ET)₂I₃ (Figure 5a, CO I–M transition temperature; $T_{co} = 135$ K) [17–22] shows very small structural change [P1(CO), P1 (metal)]. However, the I–M transition in another CO compound, θ -(ET)₂RbZn(SCN)₄ (Figure 5b, $T_{co} = 195$ K) [23–25], shows the large symmetry breaking of the molecular arrangement upon the I–M transition (C2(CO), I222(metal)), which is induced by changes of the intermolecular dihedral angle. An interesting aspect of the CO is ferroelectricity. Our recent results reveal that a CO state in α -(ET)₂I₃

shows ferroelectricity, which is attributable to the breaking of the inversion symmetry as a result of charge disproportionation, not by a structural deformation [26]. Such ferroelectricity induced by the electronic origin (electronic ferroelectricity [27–29]) is advantageous for high speed switching or modulation of dielectric properties.

Figure 4. Crystal structure of (ET)₂X consisting of donor (ET) layers and acceptor layers.

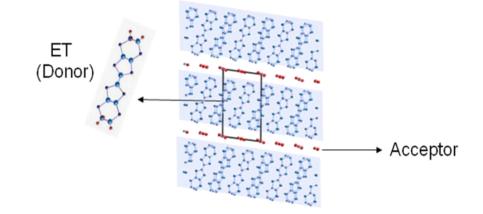
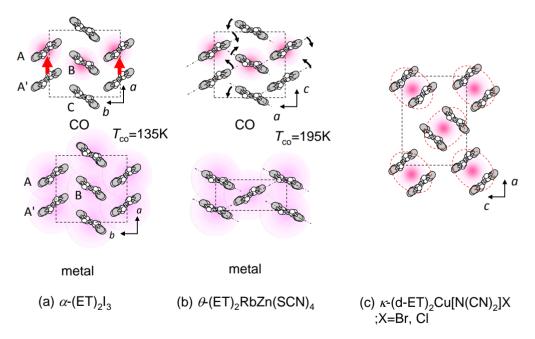


Figure 5. Schematic illustration of the molecular arrangement for (**a**) α -(ET)₂I₃; (**b**) θ -(ET)₂RbZn(SCN)₄; (**c**) κ -(d-ET)₂Cu[N(CN)₂]X (X = Br, Cl). Red circles and dashed rectangles respectively denote the localized charge and the unit cell. Red arrows indicate the electric dipole moment induced by the CO.



Other layered ET salts κ -(d-ET)₂Cu[N(CN)₂]Br and κ -(d-ET)₂Cu[N(CN)₂]Cl presented in Figure 5c are Dimer–Mott (DM) insulators [30–35]. In κ -type ET salts, the ET layer has a unit comprising a pair of ET molecules (ET dimer). If an effective on-site Coulomb energy U_{dimer} on each dimer site is larger than a critical value, which is usually of the order of the bandwidth and proportional to interdimer (or equivalently inter-site) transfer energy t, then electrons are localized on each dimer site in a Hubbard model, forming a Dimer–Mott insulator.

Various electronic properties have been observed using physical and chemical methods for tuning of bandwidth. The BC phase diagram of κ type (ET)₂X salt, comprising a Mott insulator and metal (superconductor) phases, separated by a characteristic curved boundary, is drawn by changing chemical pressure through deuteration of ET and substitutions of X, or by changing of external pressure as discussed in a later section.

In this study, we investigated the fundamental mechanism of photoinduced melting of the CO and the Mott insulators for different polytypes θ , α , and κ type salts. The differences in molecular re-arrangement in the thermal I–M transition for respective polytypes are evident. Therefore, such an approach is effective for clarifying the roles of the electron correlation and the electron-phonon interaction in the PIMT.

This review is organized as follows; in Section 2, experimental setups for various methods of time resolved spectroscopy in near-infrared, mid-infrared (IR) and terahertz (THz) regions for investigating the PIMT will be described. In Section 3, PIMT in CO insulators α -(ET)₂I₃ and θ -(ET)₂RbZn(SCN)₄ will be discussed, based on results of near-IR and mid-IR pump-probe spectroscopy with time resolution of 200 fs (3.1.–3.3.) [15,36], optical pump-THz probe spectroscopy (3.4.) [37], high-time resolution spectroscopy using infrared 3-optical cycle (12 fs) pulses (3.5.) [38]. The PIMT in the Dimer–Mott insulator κ -(d-ET)₂Cu[N(CN)₂]Br is discussed in Section 4 [39]. In Section 5, we summarize the results described above.

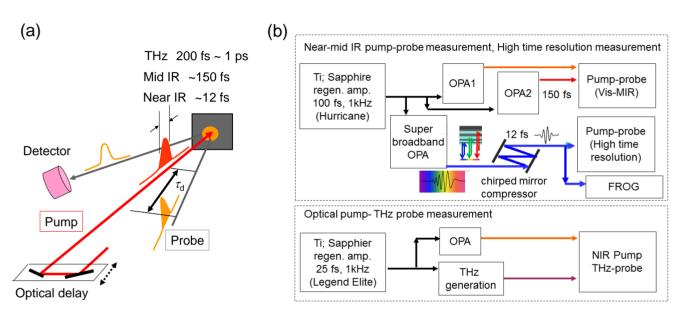
2. Experimental Section

Single crystals of α -(ET)₂I₃ (~1.5 × 1.5 × 0.2 mm), θ -(ET)₂RbZn(SCN)₄ (~1 × 1 × 0.2 mm) and deuterated κ -(d-ET)₂Cu[N(CN)₂]Br (~1 × 1 × 0.5 mm) and κ -(d-ET)₂Cu[N(CN)₂]Cl (~1 × 1 × 0.5 mm) were prepared using a previously reported procedure [17,23,31]. A schematic illustration of pump-probe spectroscopy (Figure 6a) and diagram of the experiments (Figure 6b) are presented. Photoinduced changes of the electronic states in the sample were measured by the probe lights, which are reflected or transmitted from the sample. For near and mid-IR pump-probe measurements, the fundamental output from a Ti: Sapphire regenerative amplifier (Hurricane; Spectra-Physics), operating at 1 kHz, 800 nm, with a pulse width of *ca*. 100 fs, was used for the excitation of optical parametric amplifiers (OPAs) or the generation of THz pulse.

For the near- and mid-IR pump-probe measurements, signal, idler pulses generated in type II β -BaB₂O₄ (BBO) and differential frequency generation (DFG) between them were used as pump and probe lights. Energies of pump and probe beams are, respectively, $E_{ex} = 0.89$ eV and $E_{pr} = 0.1-0.8$ eV. Time resolution of the near- mid IR pump-probe measurement was approximately 200 fs, determined by the pulse width of the signal, idler and DFG pulses (100–150 fs). For pump-probe spectroscopy with higher time resolution, super broadband 2nd stage OPA with a degenerate type I configuration in BBO crystal [40], which was pumped by the Ti: sapphire amplifier, was constructed. In such degenerate OPA, phase matching condition is satisfied in 1.2–2 µm, resulting in the super broadband OPA. Such super broadband pulses were compressed using the computer controlled active mirror compressor or chirped mirror compressor. The center wavelength and pulse width were, respectively, measured by the frequency resolved time gating (FROG) measurement as 12 fs, and 1400 nm. In this

measurement, time resolution of the pump-probe measurement was evaluated as 15 fs by the cross correlation at the 10- μ m-thick BBO at the sample position in the cryostat.

Figure 6. (a) Schematic illustration of pump-probe (reflection detected) spectroscopy. τ_d represents the time delay between the pump and the probe pulses. Time resolution of the measurement setups are, respectively, 200 fs [near-IR, mid-IR (0.1–0.8 eV), 12 fs [near-IR (0.65–0.9 eV)], and ~1 ps [THz (2–25 meV)]; (b) Experimental diagrams of visible-mid IR pump-probe measurement, high time resolution measurement and optical pump and THz probe measurement.



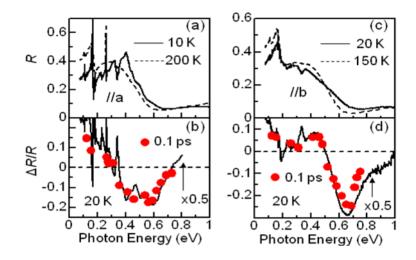
For the optical-pump THz-probe experiment, a THz probe pulse was generated by the second-order nonlinear effect in ZnTe crystal (thickness, 3 mm) and GaP crystal (thickness, 0.5 mm). We employed the 25 fs pulse from a Ti: sapphire regenerative amplifier (Legend Elite USX; Coherent Inc.) as an excitation source for generating a THz pulse. The terahertz pulse was focused on the 1.2-mm diameter sample surface. The residual beam was used for generating a 1400 nm pump beam in the OPA. The pumping diameter on the sample was ca. 2 mm. The transmitted THz probe was detected using an electro-optical (EO) sampling method. The probing energy region and the time resolution were, respectively, 0.5–7 THz and ca. 1 ps.

3. Photoinduced Melting of CO and Metallic Domain Formation in α -(ET)₂I₃ and θ -(ET)₂RbZn(SCN)₄ [15,36–39]

3.1. Mid-IR Response Reflecting the Photoinduced I–M Transition [15,36]

Figure 7a,c respectively shows the polarized reflectivity spectra of α -(ET)₂I₃ and of θ -(ET)₂RbZn(SCN)₄ for the electric field of light perpendicular to the molecular stack [*a*-axis in α -(ET)₂I and *c*-axis in θ -(ET)₂RbZn(SCN)₄]. A reflection band at less than 0.8 eV was assigned to the CT transition between the ET molecules [41–43]. Reflectivity (*R*) spectra for temperatures $T > T_{co}$ and $T < T_{co}$ are shown respectively as dashed and solid curves. Open circles in Figure 7b,d shows the transient reflectivity change $\Delta R/R$ spectra at 20 K.

Figure 7. Reflectivity spectra of (**a**) α -(ET)₂I₃; (**c**) θ -(ET)₂RbZn(SCN)₄ [15]. $\Delta R/R$ spectra measured at $t_d = 0.1$ ps (20 K) are shown as red circles for (**b**) α -(ET)₂I₃; (**d**) θ -(ET)₂RbZn(SCN)₄. Excitation energy E_{ex} and intensity I_{ex} are, respectively, 0.89 eV and 0.1 mJ/cm². Differential spectra (R_M - R_I)/ R_I reflecting the thermal I–M transition are shown in (**b**) and (**d**) as solid curves.



The pump ($E_{ex} = 0.89 \text{ eV}$) and probe lights were polarized perpendicular to the molecular stack. The $\Delta R/R$ spectra measured at $t_d = 0.1$ ps are analogous to the differential reflectivity spectra $(R_M - R_I)/R_I$ (solid curves in Figure 7b,d), in which R_M and R_I respectively represent the reflectivity of the high temperature {150 K [α -(ET)₂I], 200 K[θ -(ET)₂RbZn(SCN)₄]} metallic and CO insulator phases {20 K [α -(ET)₂I], 10 K [θ -(ET)₂RbZn(SCN)₄]}, which indicates that the PIMT occurs immediately after photoexcitation in both compounds. The spectral coincidence between $(R_M - R_I)/R_I$ and the transient reflectivity in the mid-infrared region shows that the optical excitation results in the generation of the metallic state, which is similar to the high-temperature metal.

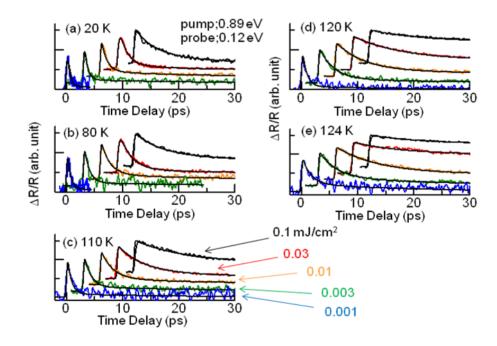
The magnitude of $\Delta R/R$ observed at 0.1–0.8 eV for $t_d = 0.1$ ps increases linearly with excitation intensity (I_{ex}) up to 0.1 mJ/cm². Considering the absorption coefficients {5000 cm⁻¹ [α – (ET)₂I₃], 15,000 cm⁻¹ [θ -(ET)₂RbZn(SCN)₄]} at 0.89 eV and unit cell volumes [1690 Å³ for α -(ET)₂I₃, 2050 Å³ for θ -(ET)₂RbZn(SCN)₄], the I_{ex} of 0.1 mJ/cm² corresponds to the excitation of one photon per approximately 600 [α -(ET)₂I₃], 1600 [θ -(ET)₂RbZn(SCN)₄] and donor molecules. The efficiency of the PIMT is evaluated as 250 [α -(ET)₂I₃], and 100 [θ -(ET)₂RbZn(SCN)₄] molecules/photon. According to these results, PIMT starts with the generation of the microscopic metallic clusters of 100–200 molecules (a scale of *ca*. 10 nm).

The ultrafast (<100 fs) PIMTs for both α -(ET)₂I₃ and θ -(ET)₂RbZn(SCN)₄ suggest that these responses are not driven by structural instability. Recent results of the pump-probe experiment using three-optical cycle (12 fs) IR pulses shows that the initial process occurs in the timescale of *ca*. 15 fs [38], as shown in Section 3.5. Such an ultrafast response indicates that the molecular rearrangement or structural change is unimportant and that electronic processes play a main role because the initial response is as fast as the timescale of the intermolecular CT (*ca*. 20–40 fs).

3.2. Recovery Dynamics of the Photoinduced Metallic State

In this section, relaxation processes of the photoinduced metallic state are discussed. Time evolutions of $\Delta R/R$ at 0.12 eV in α -(ET)₂I₃, reflecting the recovery dynamics of the photoinduced metallic state to the original CO state for various I_{ex} values and temperatures, are shown in Figure 8a–e. Figure 9a,b summarize the time evolutions [Figure 9a α -(ET)₂I₃ (20 K), Figure 9b α -(ET)₂I₃ (124 K)] together with that of θ -(ET)₂RbZn(SCN)₄ (20 K) (Figure 9c). The decay curves of α -(ET)₂I₃ depend strongly on I_{ex} and temperature. Figure 9b shows that the decay time of the dominant component lengthens with I_{ex} in the picosecond region. At 124 K, just below T_{co} , the decay time increases more markedly with I_{ex} from the sub-picosecond to the nanosecond time domain. The longer decay time for the larger I_{ex} near T_{co} suggests the inhomogeneous characteristics of the photoinduced metallic state, as portrayed in Figure 10, *i.e.*, the microscopic metallic domain is relaxed rapidly to the charge-ordered state for small I_{ex} . For large I_{ex} , the microscopic domains were condensed into the macroscopic domain to reduce the interfacial energy, accompanying the molecular rearrangement, as described later.

Figure 8. Time evolutions of $\Delta R/R$ at 0.12 eV in α -(ET)₂I₃, reflecting the recovery dynamics of the photoinduced metallic state to the original CO state for various I_{ex} values and temperatures are shown. Fitting curves [see the text (3.3)] are also shown as black curves.



Decay profiles for θ -(ET)₂RbZn(SCN)₄ (Figure 9c) were fitted using a two-component exponential curve $\Delta R(t)/R = A_{\text{fast}} \exp(-t/\tau_{\text{fast}}) + A_{\text{slow}} \exp(-t/\tau_{\text{slow}})$ whose decay times τ_{fast} , τ_{slow} [fractions $A_{\text{fast}}/(A_{\text{fast}} + A_{\text{slow}})$] are $\tau_{\text{fast}} = 0.2$ ps (0.83) and $\tau_{\text{slow}} = 2$ ps (0.17). The recovery dynamics are independent of I_{ex} (0.001–0.1 mJ cm⁻²) and temperature (20–150 K).

The thermodynamic natures of the photoinduced macroscopic metallic domain in α -(ET)₂I₃ will be described in the next section. Here, the difference between the relaxation dynamics of α -(ET)₂I₃ and that of θ -(ET)₂RbZn(SCN)₄ is considered. In θ -(ET)₂RbZn(SCN)₄, marked temperature and I_{ex} dependences were not observed, indicating that the condensation of the dense metallic domain did not

occur. These results are related with how different from each other the structures of the I-phase and M-phase are in the thermal I–M transition for both compounds: in the CO phase of θ -(ET)₂RbZn(SCN)₄, structural symmetry is reduced by the change in molecular rotation in the a-c plane and molecular tilts from the *b*-axis [23–25] (Figure 5b), modulating the intermolecular CT, although such structural modification is very small in the thermal I–M transition of α -(ET)₂I₃ [18,19] (Figure 5a). Considering the large molecular rearrangement is observed in θ -(ET)₂RbZn(SCN)₄ [23,24], potential barriers against molecular displacement, such as changes of dihedral angle, might block the evolution of macroscopic quasi-stable metallic domains. Consequently, the microscopic, unstable metallic domains return to the charge-ordered states within a few picoseconds (Figure 10b). The small structural difference and the small barrier between the CO and metallic phases in α -(ET)₂I₃ favor generation of the macroscopic metallic domains (Figure 10c).

Figure 9. Time evolutions of $\Delta R/R$ for [(**a**) α -(ET)₂I₃ (20 K), (**b**,**c**) α -(ET)₂I₃ (124 K)] are summarized together with that of θ -(ET)₂RbZn(SCN)₄ (20 K) (**c**).

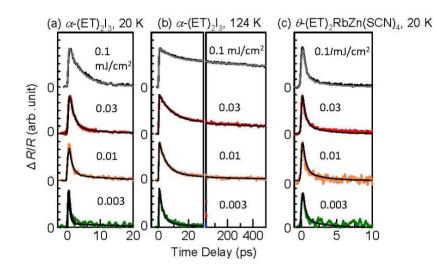
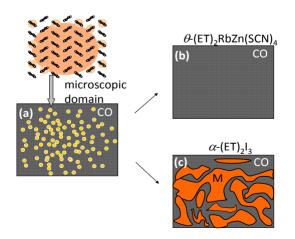


Figure 10. Schematic illustrations of (**a**) initially produced microscopic metallic domains; (**b**) fast recovery of the charge-ordered state in θ -(ET)₂RbZn(SCN)₄; (**c**) macroscopic metallic domain in α -(ET)₂I₃.



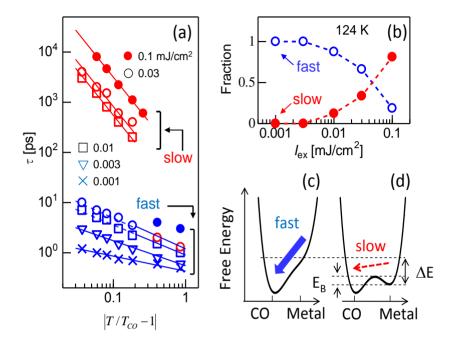
Accordingly, the primary processes of the PIMT in the CO ET salts consist of the following two steps: (i) generation of the microscopic domain by the electronic photo-doping response (α -(ET)₂I₃, θ -(ET)₂RbZn(SCN)₄) in <100 fs timescale and (ii) the condensation of the metallic domain accompanying the molecular rearrangement (α -(ET)₂I₃). Detailed dynamics of the microscopic metallic state in a shorter time region will be discussed in Section 3.5.

3.3. Thermodynamic Nature of the Photoinduced I–M Transition Critical Slowing Down in α -(*ET*)₂I₃. [15,36]

Another feature of the PIMT in α -(ET)₂I₃ is that the relaxation time (τ) of the metallic state shows critical slowing down (CSD) [44], indicating a reduction of the thermodynamic recovery force of the CO state near T_{co} , as shown in Figure 11c,d, *i.e.*, a large instability induces fast recovery for $T \ll T_{co}$ (Figure 11c). The slower recovery for $T \sim T_{co}$ is attributable to its slight instability (Figure 11d). The thermodynamic natures of the photoinduced phase transition have been extracted from the CSD [45,46]. As described in the previous section, the slower decay for larger I_{ex} is attributable to condensation of the short-lived microscopic domain to the macroscopic domain. The temperature dependence of the decay times of $\Delta R/R$, τ_{fast} and τ_{slow} , are shown as functions of reduced temperature $|T/T_{\text{co}} - 1|$ for various I_{ex} values in Figure 11a. Furthermore, τ_{fast} and τ_{slow} represent the time constants obtained from the fitting procedure, with three exponential functions whose time constants are $\tau_{\text{fast}} \sim 1$ ps, $\tau_{\text{middle}} \sim 15$ ps, and $\tau_{slow} \sim 1$ ns (Figures 8 and 9). Of the three components, the decay time of the small (ca. 5%) component τ_{middle} (~15 ps) is independent of temperature, indicating that τ_{middle} is unrelated to the I–M transition. The I_{ex} dependences of the relative fraction for the fast and slow components are presented in Figure 11b. The fast component was observed solely for small I_{ex} , whereas the slow component becomes dominant for large I_{ex} . Therefore, τ_{fast} and τ_{slow} are attributable to the relaxation of the microscopic and the macroscopic metallic domains. Figure 11a shows that, for $T < T_{co}$, τ_{fast} and $\tau_{\rm slow}$ increase with $|T/T_{\rm co} - 1|$.

According to the dynamic scaling theory (DST) of the second-order transition, the CSD observed in the relaxation time (τ) of the quasi-stable state can be represented by $\tau \propto |T/T_{co} - 1|^{-vz}$, where v and zrespectively signify critical exponents of the correlation length ξ and the dynamic critical exponent. For $I_{ex} > 0.01$ mJ cm⁻², τ_{slow} shows vz = 1.8, which is close to the calculated value (=2.1665), as evaluated by Monte Carlo simulation within the framework of the 2-D Ising model [47]. The slow component is therefore attributable to the macroscopic metallic domain. These results underscore that the thermodynamic characteristics of the microscopic domain differ from those of the macroscopic domain.

The concept of CSD is available for the second-order transition, although the thermal I–M transition in α -(ET)₂I₃ is the first-order phase transition. The reason for the observation of CSD-like behavior in the photoinduced metallic state is regarded as explained below. In a second-order phase transition, the energy barrier E_B (Figure 11c,d) between both phases is smaller than the energy scale of the critical behavior, *i.e.*, the energy difference $\Delta E \gg E_B$. In the thermal phase transition, ΔE is in the energy scale of the thermal fluctuation k_BT (*ca.* 10 meV). **Figure 11.** (a) Relaxation times of the photoinduced metallic state in α -(ET)₂I₃, τ_{fast} and τ_{slow} , are shown as a function of $|T/T_{\text{co}} - 1|$ for various I_{ex} ; (b) Fraction of the fast (open circles: $\tau_{\text{fast}} \sim 1$ ps) and slow (closed circles: $\tau_{\text{slow}} \sim 1$ ns) decay components are shown as a function of I_{ex} (124 K); (c,d) Schematic illustrations of free energy surface for (c) $T \ll T_{\text{co}}$ (fast relaxation) and (d) $T \sim T_{\text{co}}$ (slow relaxation).



The energy scale of the fluctuation induced by the photoexcitation is extremely large (*ca.* 1000 meV). Therefore, $\Delta E \gg E_B$ might be satisfied for the photoinduced phase transition, even if the relation is not satisfied in the thermal phase transition because the photoinduced metallic state is far from the ground state. In other words, the observed CSD-like behaviors are not the true "critical" phenomena.

3.4. THz Spectroscopy of Photoinduced Phase Transition Electronic State of Photoinduced Metallic State [37]

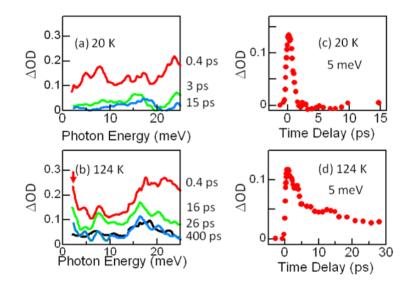
Spectroscopic measurement in THz or equivalently the far IR region is well known as a powerful technique for investigating the metallic state in transition metal oxides and low-dimensional organic conductors. To investigate the electronic state of the photoinduced metallic state with the lifetime of approximately one picosecond, transient THz spectra in a picosecond timescale should be obtained using the time resolved optical pump-THz probe spectroscopy.

In earlier sections [Section 3.1.–3.3], we described that the short-lived (*ca.* picosecond) microscopic metallic state was observed for weak excitation at $T \ll T_{co}$, whereas the macroscopic metallic state is stabilized in the timescale of >ns for strong excitation at $T \sim T_{co}$. Such a dramatic change of the lifetime, depending on I_{ex} and temperature, strongly suggests a difference between the electronic characteristics of the microscopic metal and that of the macroscopic metal. However, optical spectra in mid-IR region cannot probe such a difference.

Figure 12a,b shows transient terahertz (Δ OD, photoinduced absorption change) spectra observed respectively at 20 K (a) and 124 K (b) (t = 0.1 ps, $I_{ex} = 0.03$ mJ/cm²). Both spectra show an absorption increase in the THz region, indicating a closing of the optical gap. The time profiles of the Δ OD shown

in Figure 12c,d shows behaviors that are analogous to those of mid-IR response (Figures 8 and 9). However, the transient THz spectra in Figure 12a,b provide additional information, *i.e.*, broad absorption is detected at 20 K, although the Drude-like spectral weight exists in the low energy (<5 meV) region at 124 K (red arrow in Figure 12b). Considering that the time profile presented in Figure 12d reflects the recovery of the macroscopic metallic state, the Drude-like THz response detected at 124 K ($T \sim T_{co}$) corresponds to the macroscopic metallic state. The broad absorption increase at 20 K ($T \ll T_{co}$) is attributable to the microscopic metallic state.

Figure 12. (**a**,**b**) Transient absorption (Δ OD) spectra in α -(ET)₂I₃ at (**a**) 20 K; (**b**) 124 K in THz region ($E_{\text{ex}} = 0.89 \text{ eV}$, $I_{\text{ex}} = 0.03 \text{ mJ/cm}^2$); (**c**,**d**) Time evolutions detected at 5 meV [(**c**) 20 K; (**d**) 124 K].



It is worth noting that the Drude-like spectral weight detected at 124 K is not observed for the thermal I–M transition, indicating that the low energy electronic characteristics of the photoinduced metallic state differ from those of the high-temperature (HT) steady state metallic state. The electronic properties of the photoinduced macroscopic metallic state remain unclear. However, one possible explanation is that in the photoinduced macroscopic metallic state, the charge disproportionation (CD) is melted on the B–C molecular stacks in Figure 5 arising from the asymmetric structure, is markedly weakened in addition to the melting of the correlated CD on the A–A' stack. In the HT metallic state, however, the CD on the B–C stack remains [18].

3.5. Ultrafast Snapshot of Correlated Electron Dynamics 10 fs Spectroscopy of Photoinduced Phase Transition [38,41]

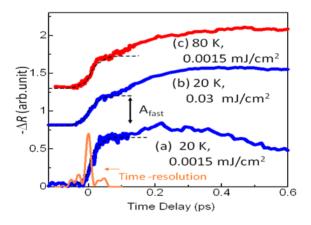
In earlier sections, the recovery dynamics of the photoinduced metallic state were discussed, based on the near-, mid-IR and THz spectroscopy with the time resolution of 200 fs to 1 ps. However, that is insufficient to capture the initial dynamics for the I–M transition. Here, ultrafast snapshots are captured using the 3-optical cycle 12 fs pulse in near IR region.

In 1/n filling system (*n*: 2, 3, 4...), the regularly localized charge patterns shown by Figure 1a,b are balanced electrostatically if the Coulomb repulsion interaction is sufficiently large. That is, a correlated insulator such as the CO insulator (Figure 1a) and the Mott insulator (Figure 2b). The FC I–M transition is the melting of such regularly localized charge patterns realized using chemical carrier doping. The mechanism of PIMT is sometimes understood using the concept of photo-carrier doping [10]. However, the primary dynamics immediately after the photoexcitation cannot be described by such static carrier redistribution. That requires a more dynamic picture because the PIMT is triggered by the generation of high-frequency electric polarization. Therefore, our fundamental question is "How does such high-frequency electronic oscillation engender the phase transition?"

Recent technology of femtosecond light pulses in the 10 fs region, which are roughly equal to the timescale of the intermolecular electronic motion, enables us to understand the early stage dynamics of PIMT in organic CT compound [38,48]. Here, we employed few-optical-cycle infrared pulses (pulse width, *ca.* 12 fs) in pump-probe measurements to investigate the early stage dynamics of electron motion in α -(ET)₂I₃.

Figure 13a–c shows the time evolution of $\Delta R/R$ measured at 0.7 eV at 20 K [(a) $I_{ex} = 0.0015$ mJ cm⁻²); (b) $I_{ex} = 0.03$ mJ/cm²] and 80 K [(c) $I_{ex} = 0.0015$ mJ/cm²] under excitation by 12 fs pulses, reflecting the generation dynamics of the photoinduced metallic state. The spectral range of the pump and the probe pulses generated by the degenerate OPA (indicated by the orange shaded area in Figure 13a) corresponds to the high-energy side of the intermolecular CT transition. The $\Delta R/R$ spectrum shows a marked decrease at *ca*. 0.7 eV, which indicates the occurrence of the PIMT. The $I_{ex} = 0.03$ mJ/cm² corresponds to a density of 1 photon/1500 molecules for 0.89 eV excitation. The timescale of the initial rise in the $\Delta R/R$ signal (<20 fs) is as fast as that of the correlated electron motion, as discussed later.

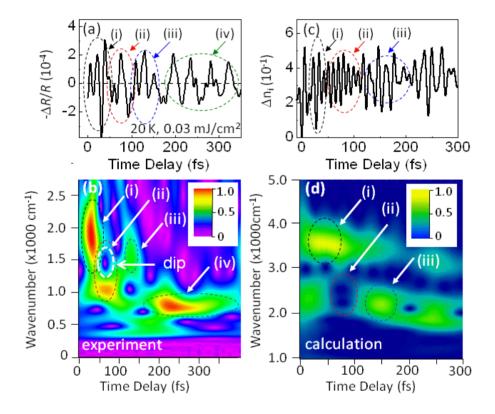
Figure 13. Time evolutions of $\Delta R/R$ in α -(ET)₂I₃ at 20 K for (**a**) $I_{\text{ex}} = 0.0015 \text{ mJ/cm}^2$; (**b**) 0.03 mJ/cm² (blue curves) and at (**c**) 80 K (red curve: $I_{\text{ex}} = 0.0015 \text{ mJ/cm}^2$). The orange curve represents the cross-correlation profile between the pump and the probe pulses as a response function G(t).



The high-frequency oscillating component of $\Delta R/R$ (20 K, $I_{ex} = 0.03 \text{ mJ/cm}^2$) was obtained using a Fourier high-pass (>300 cm⁻¹) filter (Figure 14a). In addition, a time-frequency spectrogram (Figure 14b) and time-resolved spectra (Figure 15b–e) were obtained by wavelet (WL) analysis of the

oscillating component. As indicated by the black dashed circle shown as (i) in Figure 14a,b, the spectrogram is dominated by the broad signal at *ca*. 1800 cm⁻¹ for $t_d <50$ fs. The signal peak exhibits a drastic frequency change with t_d between (ii) *ca*. 1000 cm⁻¹ and (iii) 1450 cm⁻¹ for $t_d = 50-200$ fs. Then, the (iv) signal approaches *ca*. 820 cm⁻¹ for $t_d > 200$ fs.

Figure 14. (a) High-frequency oscillating component of $\Delta R/R$ in α -(ET)₂I₃ at 20 K for $I = 0.03 \text{ mJ/m}^2$, which was obtained using a Fourier high-pass (>300 cm⁻¹) filter; (b) Time-frequency spectrogram calculated using wavelet analysis of the high-frequency oscillating component; (c) Calculated photoinduced charge-density modulation (Δn_i) at molecule A in the extended Peierls–Hubbard model with EMV coupling (see the text); (d) Spectrogram of calculated Δn [38].



It is worth noting that the broad spectrum at *ca*. 1800 cm⁻¹ exhibiting the spectral gap at *ca*. 1000 cm⁻¹ for $t_d = 30$ fs [Figure 15b and (i) in Figure 14a,b] is analogous to the steady state optical conductivity spectrum (Figure 15a). Therefore, the oscillation at 1800 cm⁻¹ is attributable to the electron oscillation reflecting the CO gap. This is the first observation of electron oscillation in the time domain. Figure 16 portrays a schematic illustration of such intermolecular electron oscillations. These electron oscillations are efficiently induced between the charge-rich sites (A and B sites) and between the charge-rich site and charge-poor site (B and A' sites) [38,41,42], as shown by the arrows in Figure 16 and Figure 15f.

Figure 15. (a) Steady state optical conductivity spectrum measured at 10 K in α -(ET)₂I₃; (b–e) Time-resolved spectra calculated from Figure 14a; (f–i) Schematic illustration of (f) intermolecular electron modulation and intramolecular vibrations (g) ν_3 , (h) ν_9 , and (i) ν_7 [38].

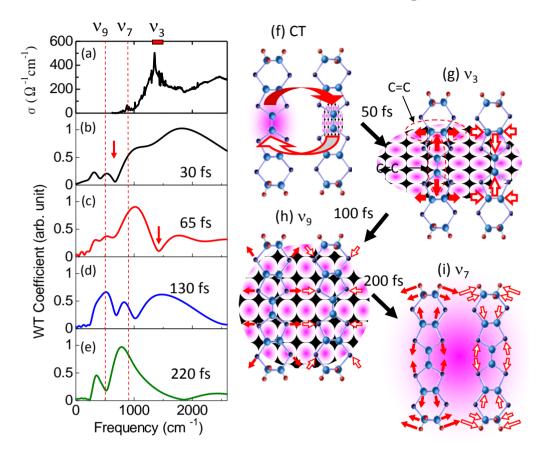
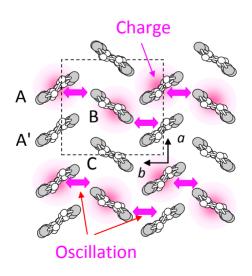


Figure 16. Schematic presentation of the electronic oscillation with a period of 18 fs in α -(ET)₂I₃. The oscillation occurs respectively between A (charge-rich) and B (charge-rich) sites and between B (charge-rich) and A' (charge-poor) sites.



Considering that C=C stretching modes v_3 (presented in Figure 15g) appear between 1352 and 1476 cm⁻¹ [49–51], the spectral hollow that appears at *ca*. 1450 cm⁻¹ [arrow in Figure 15c and (ii) in Figure 14a,b] is attributable to the destructive interference that prevails between the electron

oscillation and the intramolecular v_3 mode. The observation of such destructive interference, which is attributable to Fano antiresonance [52] at *ca*. 50 fs indicates that the electron oscillation begins to interact coherently with the intramolecular v_3 mode at the finite time delay. Considering the strong electron molecular vibration (EMV) coupling of the v_3 mode [49–51], it is reasonable to consider that the electron oscillation selectively induces the intermolecular anti-phase v_3 vibration, as shown in Figure 15f,g [49–51]. After 100 fs, the hollow at approximately 1450 cm⁻¹ disappears; a broad band appears at 1450 cm⁻¹, reflecting the electronic dephasing [Figure 15d and (iii) in Figure 14b]. Then, the disappearance of the broad 1450 cm⁻¹ mode indicates that the coherent v_3 vibration is dephased by interaction with other vibrations.

The other signals at *ca*. 500 cm⁻¹ (Figure 15d) and 820 cm⁻¹ (Figure 15e) after $t_d = 150$ fs are respectively attributable to the v₉ (508 cm⁻¹) (called breathing mode) and v₇ (896 cm⁻¹) modes, as illustrated in Figure 15h,i [49–51]. These modes also show large EMV coupling constants (v₃, 0.746; v₉, 0.476; v₇, 0.117; in dimensionless units). Figure 14b and Figure 15c,e suggest that sequential changes occurred in the interacting modes from the local C=C stretching mode (v₃) to the more delocalized breathing mode (v₉) and v₇ mode showing the pathway of e–e and e–ph interactions that leads to the I–M from the electron oscillation.

Yonemitsu *et al.* calculated the photoinduced electron dynamics by numerically solving the time-dependent Schrödinger equation for the exact many-body wave function for electrons coupled with classical intermolecular phonons and quantum intramolecular phonons, to consider the correlated electron motion and the coherent interplay between the electron and intramolecular v_3 vibration [38,41,42].

$$H = \sum_{\langle ij\sigma \rangle} \left[\left(t_{i,j} \pm \alpha_{i,j} u_{i,j} \right) c_{i\sigma}^{+} c_{j\sigma} + \text{H.c.} \right]$$

+ $U \sum_{i} n_{i\uparrow} n_{i\downarrow} + \sum_{\langle ij \rangle} V_{i,j} n_{i} n_{j} + \sum_{\langle ij \rangle} \left(K_{i,j} / 2 \right) u_{i,j}^{2} + \sum_{\langle ij \rangle} \left(K_{i,j} / 2 \omega_{i,j}^{2} \right) \dot{u}_{i,j}^{2}$
+ $g \sum_{i} \left(b_{i} + b_{i}^{+} \right) \left(n_{i} - 3/2 \right) + \omega_{b} \sum_{i} b_{i}^{+} b_{i}$ (1)

where $c_{i\sigma}^{+}$ creates an electron with spin σ at site *i*, $n_{i\sigma} = c_{i\sigma}^{+}c_{i\sigma}$, $n_i = \sum_{\sigma} n_{i\sigma}$, $u_{i,j}$ denotes the intermolecular phonon's displacement, b_i^{+} creates a quantum phonon of energy ω_b , and *g* is the EMV coupling strength. The other notation is standard, as introduced in an earlier report [41]. Using an eight-site periodic cluster and the same model parameters as those described in that report ($\omega_b = 0.36 \text{ eV}$, g = 0.0625 eV, and an oscillating electric field along the *b*-axis having a Gaussian profile of duration 5 fs and center frequency $\omega_b = 0.35 \text{ eV}$), we calculated the time evolution of n_i at molecule A (Figure 16) for $t_d < 300$ fs (Figure 14c) and conducted WL analysis (Figure 14d). The broad peak from 3000 to 4000 cm⁻¹ at around $t_d = 30$ fs corresponds to that of the calculated conductivity spectrum in that earlier report [41], indicating that the peak is attributable to the electron oscillation presented in Figure 16 [(i) in Figure 14c,d]. The difference between the experimental and calculated values of the conductivity peak (experiment, 1000–2000 cm⁻¹; calculation, 3000–4000 cm⁻¹) can be attributed to the small cluster size.

For $t_d = 50-100$ fs, the spectral hollow at 2000–3000 cm⁻¹ is attributable to the destructive interference between the correlated electron oscillation and the phonon [(ii) in Figure 14c,d]. Then, the phonon peak appears after $t_d = 100$ fs [(iii) in Figure 14c,d]. This calculated WL spectrogram that reflects the electron rv₃ vibration interference is qualitatively analogous to the experimentally obtained

result (Figure 14 b). For $t_d > 100$ fs, however, the observed WL spectrogram cannot be reproduced by the calculated one because the other vibrations are not considered in the calculation.

As indicated by the dashed curve in Figure 13a, the time required for the generation of the metallic state (τ_r) was evaluated as 15 fs at 20 K for $I_{ex} = 0.0015 \text{ mJ/cm}^2$ using Equation (2):

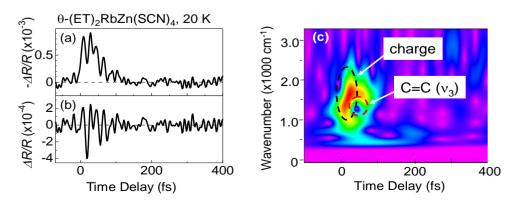
$$\Delta R(t)/R = \int_{-\infty}^{+\infty} A_{fast} \left[1 - \exp(-\frac{t/\tau}{\tau_r}) \right] G(t-t') dt'$$
⁽²⁾

where A_{fast} and G(t) respectively represent the coefficient and response function. Here, the cross-correlation profile between the pump and the probe pulses was used as G(t) (orange curve in Figure 13a). A τ_r value of 15 fs is comparable to the timescale of the electron oscillation (1800 cm⁻¹) in Figure 14a, indicating that the early stage dynamics of the PIMT are driven by the electronic response. This rapid generation process dominates the weak excitation ($I_{\text{ex}} = 0.015 \text{ mJ/cm}^2$) at 20 K.

However, the contribution of an additional slower (*ca.* 300 fs) growth component increases with I_{ex} and temperature, as indicated by the result for the strong excitation ($I_{ex} = 0.03 \text{ mJ/cm}^2$) at 20 K and at high temperature 80 K for $I_{ex} = 0.0015 \text{ mJ/cm}^2$, as presented in Figure 13c, reflecting the generation of the quasi-stable metallic state [15,37]. Considering that the coherent intramolecular vibrations are detected during this slower (*ca.* 300 fs) growth, it is inferred that the quasi-stable metallic state is generated by the interplay between the electron oscillation and the vibrations.

The time evolution of $\Delta R/R$ for θ -(ET)₂RbZn(SCN)₄ is presented in Figure 17. In the near-IR and mid-IR spectroscopy with the time resolution of 200 fs, reflectivity changes disappear at a timescale shorter than 1 ps, reflecting that the microscopic metallic state has not condensed and the CO state quickly recovers [15]. As indicated in the oscillating component (Figure 17b) of $\Delta R/R$ and the spectrogram (Figure 17c) in θ -(ET)₂RbZn(SCN)₄, the electronic oscillation at *ca*. 1800 cm⁻¹ and the Fano resonance were observed. However, the other oscillating structure detected for α -(ET)₂I₃ (500 cm⁻¹, 820 cm⁻¹) at a later time delay have not been detected at all. Considering that the macroscopic metallic state is initially driven by the coherent electronic oscillation. However, oscillation with lower frequencies (500 cm⁻¹, 820 cm⁻¹) is related to the condensation of the macroscopic metallic state in α -(ET)₂I₃.

Figure 17. (a) Time evolution of $\Delta R/R$ in θ -(ET)₂RbZn(SCN)₄; (b) Oscillating component of $\Delta R/R$ and (c) spectrogram.



The results presented above provide a clear microscopic view of the light-electron-lattice interactions leading up to the PIMT. It is worth noting that this view differs markedly from the traditional description of the e-ph interactions in the framework of thermodynamic analysis such as those obtained using a two-temperature model. The e-ph interaction begins to start before the electrons reach the quasi-equilibrium state. Moreover, finding the electronic coherence before the excited state is affected by the vibrations is important because the coherence of the correlated electron motion is indispensable for realizing optical manipulation of the electronic phase transition. Coherent control has been a central issue for the optical manipulation in few-body atomic systems. The interaction mode, which should be controlled, has not been specified in the PIMT because the initial dynamics are so fast and complicated. Observations of the electronic oscillation and the C=C vibration immediately after excitation will lead to the development of coherent control of the PIMT.

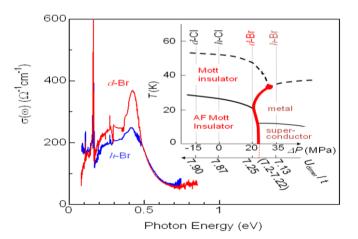
4. Photoinduced I–M Transition in Dimer–Mott Insulator κ-(d-ET)₂Cu[N(CN)₂]Br [35]

4.1. Bandwidth Control Mott Transition in κ-Type ET Salts

Another well-known layered ET salts with strong electron correlation is κ -(ET)₂X, as presented in Figure 5c [3,30]. This compound exhibits phase transitions from the DM insulator to metal and superconducting states by application of chemical or hydrostatic pressures [30–35]. In such BC control I–M transition, the kinetic energy of the electron is enhanced by increasing the intermolecular overlapping of electronic wave functions.

Around the characteristic curved phase boundary (1st order phase transition line) in the phase diagram (inset in Figure 18) for the partially deuterated sample κ -[(*d*-ET)_x(*h*-ET)_{1-x}]₂Cu[N(CN)₂]Br, [(*d*-ET) and (*h*-ET) show deuterated and non-deuterated ET molecules], phase separation occurs [30–35]. For our experiment, we used the fully deuterated κ -(*d*-ET)₂Cu[N(CN)₂]Br (*d*-Br) and κ -(*d*-ET)₂Cu[N(CN)₂]Cl (*d*-Cl) which are positioned as insulators, respectively, at lesser or greater distance from the first-order I–M phase boundary, although hydrogenated κ -(*h*-ET)₂Cu[N(CN)₂]Br (*h*-Br) is a metal.

Figure 18. Optical conductivity $\sigma(\omega)$ spectra of *d*-Br (red) and *h*-Br (blue) at 10 K. The inset shows the phase diagram of κ -(ET)₂X as functions of external pressure (ΔP) and $U_{\text{dimer}/t}$, which is drawn by changing the deuteration of ET and substitutions of X.



4.2. Mid-Infrared Pump-Probe Spectroscopy of κ -(d-ET)₂Cu[N(CN)₂]Br (d-Br) and κ -(d-ET)₂Cu[N(CN)₂]Cl (d-Cl)

Optical conductivity $\sigma(\omega)$ spectra of the DM insulator *d*-Br and metallic *h*-Br are obtained by carrying out a Kramers–Kronig transformation of the reflectivity spectra, as presented in Figure 18. The peak at 0.45 eV and a lower energy shoulder are respectively attributable to the intradimer (bonding to anti-bonding) transition and interdimer Hubbard gap transition [34,35]. Insets of Figure 19a,b show a differential spectrum between *h*-Br and *d*-Br at 10 K ($\Delta R_{h-d}/R_d$; red curve) and that for *d*-Br between 80 K and 10 K ($\Delta R_{temp}/R_{10}$; blue curve), which respectively reflect the I–M transition and the temperature increase. The characteristic increase in $\Delta R_{h-d}/R_d$ around <0.4 eV (red shade) is attributable to a Drude-like reflection, although the decrease in $\Delta R_{temp}/R_{10}$ around 0.2–0.6 eV (blue shade) reflects the thermal excitation of carriers against the small activation energy. In these compounds, the thermal I–M transition does not occur.

Figure 19. $\Delta R/R$ spectra at $t_d = 0.1$ ps and 2 ps for *d*-Br (**a**) and *d*-Cl (**b**) under excitation of 0.89 eV. Insets of (**a**) and (**b**) present differential spectra between *h*-Br and *d*-Br (a, $\Delta R_{h-d}/R_d$; red curve) at 10 K and between 80 K and 10 K (b, $\Delta R_{temp}/R_{10}$; blue curve) of *d*-r [39].

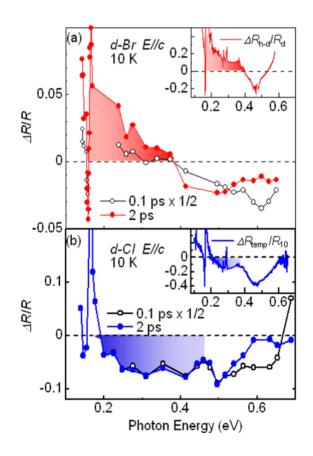


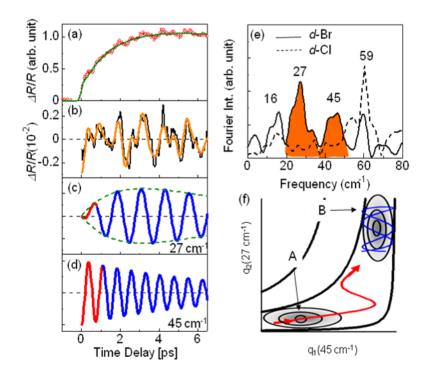
Figure 19a,b show transient reflectivity ($\Delta R/R$) spectra of *d*-Br and *d*-Cl, respectively at time delays t_d of 0.1 ps and 2 ps, under excitation at 0.89 eV. Actually, intradimer bonding to anti-bonding transition mainly contributes to $\sigma(\omega)$ at 0.89 eV. For *d*-Br, immediately after photo-irradiation, $\Delta R/R$

shows a negative value attributable to bleaching of the intradimer transition by the photo-carrier generation. Then, the amplitude of the Drude-like reflection at <0.4 eV increases. The $\Delta R/R$ spectrum at $t_d = 2$ ps is analogous to $\Delta R_{h-d}/R_d$, indicating the occurrence of PIMT. However, for *d*-Cl, $\Delta R/R$ decreases at <0.4 eV for any time delay, indicating that the PIMT does not occur at all. The efficiency of PIMT for *d*-Br was estimated as 50 dimers/photon from the fact that $\Delta R/R$ at 0.3 eV was approximately 10% of $\Delta R_{h-d}/R_d$; the photon density was 0.1 mJ/cm² (*i.e.*, 1 photon/500 molecules).

4.3. Dynamics of Charge and Lattice in the Dimer–Mott Insulator

Figure 20a shows time evolutions of the $\Delta R/R$ for *d*-Br observed at 0.26 eV, which indicate generation of the metallic state. The time profile, which reflects metallization, is reproduced by the Equation $\frac{\Delta R(t)}{R} = -A \exp(-\frac{t}{\tau_g}) + B \exp(-\frac{t}{\tau_1})$, as shown by the solid green curve, where A = -1.44, $\tau_g = 1.1$ ps, B = 2.44, and $\tau_1 = 70$ ps. It is worth noting that the generation time of the metallic state τ_g is considerably longer than the timescale (*ca.* 40 fs) of the inter-site charge transfer, suggesting that the I–M transition occurs because of the intermolecular motion and not the electronic process.

Figure 20. (a) Time evolutions of $\Delta R/R$ for *d*-Br detected at 0.26 eV under excitation at 0.89 eV (red circles with green fitted curve); (b) The time profile of the oscillating component of $\Delta R/R$ for *d*-Br (0.89 eV excitation); (c,d) The 27 cm⁻¹ (c) and 45 cm⁻¹ (d) cosine-like oscillations, which were obtained through analysis (see the text); (e) Fourier transformed spectra of the oscillation for *d*-Br and *d*-Cl; (f) Classical trajectories representing coherent phonon dynamics in the potential surface. A and B respectively denote the excited Mott insulator state and the photoinduced metallic state [39].



The oscillating structure in the time evolution of $\Delta R/R$ at 0.26 eV also indicates the contribution of the intermolecular coherent motion to the PIMT. The time profile and Fourier transformed spectrum of the oscillating component $\Delta R_{\rm osc}(t)/R$ for *d*-Br are shown respectively in Figure 20b–e. Analyses were conducted using the Equation $\Delta R_{\rm osc}(t) = \sum_{i=1}^{4} A_i (1 - \exp(-\frac{t}{\tau_{\rm ri}})) \exp(-\frac{t}{\tau_{\rm i}}) \cos(\omega_i t - \phi_i)$, and the

oscillating frequencies ω_i (i = 1-4) were estimated as 16, 27, 45, and 59 cm⁻¹. Here, τ_{ri} , τ_L and ϕ_i respectively present the rise and decay time constant and the initial phase for each oscillation. Actually, ϕ_i is smaller than 0.1 π , indicating a cosine-like oscillation, which is detected in the photoinduced phase transition, and is attributable to the displacive excitation of the coherent phonon in the photoexcited state or the photoinduced metallic state [53], and not in the ground state. Among these oscillations, the 27 cm⁻¹ and 45 cm⁻¹ modes are not detected in *d*-Cl, as shown by the dashed curve in Figure 20e. Accordingly, these two modes are related to the PIMT. It is worth noting that the 27 cm⁻¹ mode shows a finite rise time of $\tau_{r2} = 1$ ps (Figure 20c), which is identical to the timescale of metallization t_{o} , whereas the 45 cm⁻¹ mode appears instantaneously (Figure 20d). Therefore, as shown by the classical trajectories in the $q_1(45 \text{ cm}^{-1})-q_2(27 \text{ cm}^{-1})$ potential surface (Figure 20f), the 45 cm⁻¹ mode phonon is regarded as induced along the q_1 axes (red curve) immediately after generation in the photoexcited insulator state (A). Then, the 27 cm⁻¹ mode oscillates along the q₂ axis (blue curve) in the metallic state (B). Observation of the 27 cm⁻¹ mode in the Raman spectrum of metallic *h*-Br [54] supports the consideration presented above. Possible candidates for such coherent low-frequency phonons are intradimer molecular motions that effectively modulate the t_{dimer} , *i.e.*, the intradimer molecular displacements such as stretching and libration, which are triggered by bonding to the anti-bonding excitation, causing a decrease in t_{dimer} and $U_{dimer}/2$; this results in metallization.

Next, we roughly evaluate the photoinduced intradimer molecular displacement and corresponding change in U_{dimer} (ΔU_{dimer}) more quantitatively. Here, U_{dimer} and t of pristine samples are calculated as $U_{\text{dimer}} \cong 2t_{\text{dimer}}, t \cong |t_p + t_q|/2$, where t_p and t_q are anisotropic interdimer transfer integrals calculated using the extended Huckel method with structural data [55,56], although a more precise formula for U_{dimer} is presented [57,58]. In the past, U_{dimer}/t for the phase boundary was also estimated from precise studies of the phase diagram of κ -[(h-ET)_{1-x}(d-ET)_x]₂Cu[N(CN)₂]Br [31]. Considering that U_{dimer}/t values for *d*-Br and the I–M boundary are calculated as 7.25 and 7.2–7.22, respectively, as presented in Figure 18, observation of the PIMT reveals that the change of U_{dimer}/t upon PIMT is approximately 0.4–0.7%. Therefore, ΔU_{dimer} is estimated as approximately 0.4%–0.7% because the photoinduced change in U_{dimer}/t is mainly attributable to the ΔU_{dimer} . Such a small change of U_{dimer} is indicative of small molecular displacements. If we simply assume that the intradimer relative molecular distances are changed for reducing t_{dimer} to ca. $U_{\text{dimer}}/2$, then the magnitude of the displacement is as small as 0.05% of the molecular distance in the dimer. Furthermore, U_{dimer}/t for d-Cl is 7.87. Therefore, ΔU_{dimer} must be greater than 9% to induce the PIMT. A very large displacement of 0.5% is necessary to obtain such a large ΔU_{dimer} value, which might be the reason why the PIMT is not detected in d-Cl. These results and discussion demonstrate the optical modulation of U_{dimer} and the resultant PIMT in the DM insulator κ -(*d*-ET)₂Cu[N(CN)₂]Br. Under intradimer excitation, the PIMT is driven by the intradimer molecular displacements, which decrease U_{dimer} by approximately 0.4%–0.7%. Moreover, the mechanism of the PIMT strongly depends on the excitation photon energy [39], *i.e.*, the excitation

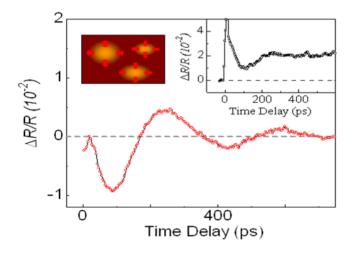
of the Hubbard band (~0.3 eV) shows FC type I–M transition. That was also analyzed theoretically [57,58].

Very recently, higher time-resolution measurement using 3-optical cycle 12 fs IR pulse showed that the intense intramolecular C=C stretching vibration (oscillation period = 25 fs) is driven by strong EMV coupling in the dimer system for κ -(*d*-ET)₂Cu[N(CN)₂]Br. After the rapid damping of this coherent C=C motion, various intramolecular and intermolecular modes are also driven, which might be induced by the nonlinear EMV coupling. Finally, low-frequency (<100 cm⁻¹) intradimer modes modulate the *t*_{dimer} and *U*_{dimer}. Details of primary dynamics faster than 100 fs will be described at a later opportunity.

4.4. Slower Dynamics of the Photoinduced Metallic State

At the end of this section, we discuss the slower dynamics occurring after the PIMT is completed. The photoinduced I–M domain wall (DW), which is in a highly non-equilibrium state, is expected to exhibit novel properties near the critical end point of the first-order phase boundary in Figure 18 [30–33]. The inset of Figure 21 shows time evolutions of $\Delta R/R$ at 0.31 eV, indicating generation of the metallic state. The most impressive feature of the time evolution is the oscillation of the structure with a period of 340 ps, which is estimated from the oscillating component shown by red circles.

Figure 21. Time evolutions of $\Delta R/R$ (inset) and the oscillating component detected at 0.31 eV for the slow time domain, together with a schematic illustration of the oscillating photoinduced metallic domain.



In some compounds exhibiting photoinduced structural transitions, such oscillations in time domain with the period of several tens~hundreds ps have been attributable to the coherent acoustic phonon as a propagation of a shock wave. The period of such a coherent acoustic phonon strongly depends on the detection energy, reflecting the dispersion of the acoustic phonon [59]. However, in this case, the frequency of the oscillation is independent of the probe energy. Moreover, the amplitude spectrum is analogous to the $\Delta R_{h-d}/R_d$ (not shown). Accordingly, such a long-period oscillation is attributable not to propagating or standing acoustic waves, but to the coherent modulation of the metallic volume fraction, suggesting oscillating DW motion. Such interpretation using a DW picture is supported by the real-imaging observation of the phase separation near the phase boundary in the BC–Mott transition [60]. The DW oscillation such as this is regarded as driven by the thermodynamic instability of the photoinduced state. Usually, it is disturbed by the I–M energy difference, potential barrier, and de-phasing process occurring via an electron–lattice interaction. However, for the PIMT in *d*-Br, the I–M energy difference and the barrier height between phases are tuned to be very small near the characteristic curved phase boundary and the critical endpoint at 35–40 K because the molecular displacement with the I–M transition is very small. In fact, the oscillating amplitude shows a maximum at approximately 25 K.

5. Summary

In this review, photoinduced I–M transitions in CO organic insulators α -(ET)₂I₃, θ -(ET)₂RbZn(SCN)₄, DM insulator κ -(*d*-ET)₂Cu[N(CN)₂Br] are discussed, based on results of pump-probe spectroscopy in near-IR, mid-IR, and THz regions in the timescale of 10 fs–100 ps.

The initial process of the PIMT is understood as an immediate (*ca.* 15 fs) generation of microscopic metallic domain that consists of *ca.* 100 molecules. Then, this initially generated microscopic metallic domain is condensed to a macroscopic metallic domain to reduce an I–M interfacial energy in a timescale of *ca.* 200 fs for α -(ET)₂I₃. On the other hand, condensation to the macroscopic metal does not occur in θ -(ET)₂RbZn(SCN)₄ because the structural symmetry of the metallic state differs markedly from that of the insulating state, *i.e.*, the potential barrier in the free energy surface for the first-order phase transition is sufficiently large. For both compounds, the initial electronic response is regarded as playing a main role in the generation of the microscopic domains, whereas the electron–phonon interaction or the molecular rearrangement effectively contributes to condensation to the macroscopic metallic state is driven by the coherent electron oscillation. Snapshots of such electronic oscillation and the Fano interference between the electronic and intramolecular C=C vibration were taken for the first time ever reported.

In the Dimer–Mott insulator κ -(*d*-ET)₂Cu[N(CN)₂Br], which is positioned near the I–M phase boundary in the phase diagram of bandwidth Mott transition, intradimer bonding anti-bonding excitation induces an photoinduced I–M transition. However, in this compound, the I–M transition occurs in the timescale of 1 ps that is much slower than that for the melting of CO described above. The photoinduced I–M transition in κ -(*d*-ET)₂Cu[N(CN)₂Br] is driven by the intradimer distortions induced by the intradimer excitation, resulting in the reduction of the intradimer kinetic energy t_{dim} ~ U_{dim} /2. Recent results with higher time-resolution measurement using 3-optical cycle 12 fs IR pulse shows that the intense intramolecular C=C stretching vibration causes intermolecular modes inducing I–M transition. In the near future, we will report on the faster dynamics of the light-correlated electron interaction measured by the sub 10 fs carrier-envelope phase (CEP) stabilized pulse in these organic conductors.

Another interesting feature has been detected in DM insulator κ -(ET)₂Cu₂(CN)₃, which is positioned near the phase boundary between the DM and the ferroelectric CO insulators. We observed the photoinduced growth of the ferroelectric CO cluster under the excitation of the DM insulator. There are few studies of such photoinduced reconstruction of the electronic ordering [61], although the photoinduced melting of electronic ordering such as CO and the DM insulator has been reported. The photoinduced phase transition of such a new type will be discussed at a later opportunity.

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