



Suplementary materials for Two possible side reaction pathways during furanic etherification

Wenting Fang, Hualei Hu, Zhongsen Ma, Lei Wang and Yajie Zhang *

Catalyst characterization

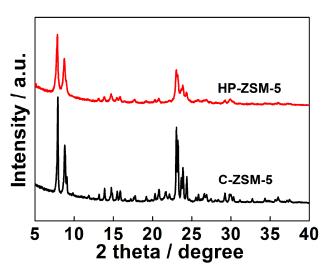


Figure S1. X-ray diffraction patterns of the catalysts. (HP-ZSM-5: hierarchical porous ZSM-5; C-ZSM-5: conventional ZSM-5.)

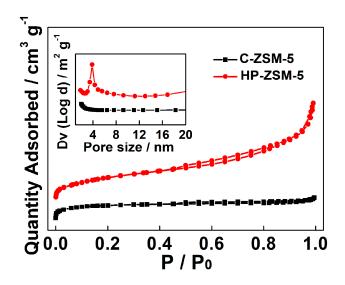


Figure S2. N₂ physisorption isotherms of the samples and the Barrett-Joyner-Halenda adsorption pore size distributions of the catalysts (inside).

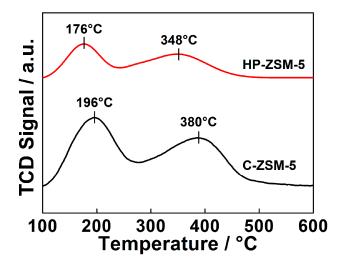
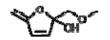


Figure S3. Temperature-programmed desorption of ammonia profiles of the samples.

The identification of by-products

Compound 6: 2-(methoxymethyl)-5-methylene-2,5-dihydrofuran-2-ol



M+H (HRMS)=143.0697

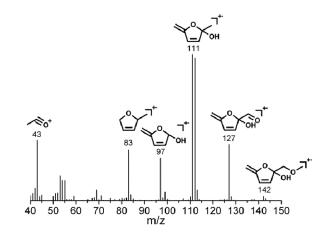
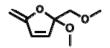


Figure S4. Mass spectrum corresponding to the peak of compound 6 in Figure 3.

 $Compound \ 7: 2, 5-dihydro-2-methoxy-2-(methoxymethyl)-5-methylene furan$



M+H (HRMS)=157.0853

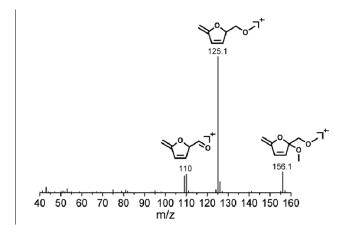
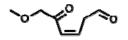


Figure S5. Mass spectrum corresponding to the peak of compound 7 in Figure 3.

Compound 3: (Z)-6-methoxy-5-oxohex-3-enal



M+H (HRMS)=143.0697

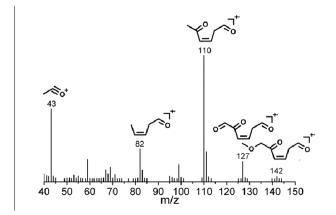


Figure S6. Mass spectrum corresponding to the peak of compound 3 in Figure 3.