Supplementary Materials:

Understanding the Photo- and Electro-carboxylation of Methyl Benzophenone with Carbon Dioxide

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Equipments.

Photoreactions were carried out under a LED lamp (customized by Shanghai Guige, 365 nm) irradiation. ¹H and ¹³C NMR spectra were recorded on a JEOL (100 MHz) or Bruker (400 MHz) in DMSO-d₆. Proton chemical shifts were referenced to the residual proton signal of the solvent at 7.26 ppm (CHCl₃) and 2.50 ppm (DMSO-d₆). High-resolution mass spectra were recorded on a Waters (Acquity UPLC/XEVO G2-XS QTOF) spectrometer. IR measurements were performed on a FTIR (Nicolet is10) spectrometer fitted with a Pike Technologies MIRacle Single Reflection ATR adapter. Liquid chromatograph-mass spectra were recorded on a Thermo Scientific Exactive (MSQ PLUS/U3000) spectrometer. Conversion yield is measured by HPLC (ACQUITY Arc, Waters).

Supplementary data.

^L COOH ¹**H NMR** (100 MHz, DMSO-d₆): δ = 12.30 (s, 1H), 7.68 (dd, *J* = 27.3, 7.5 Hz, 3H), 7.53 (t, *J* = 7.8 Hz, 3H), 7.48 – 7.20 (m, 3H), 3.80 (s, 2H); ¹³**C NMR** (100 MHz, DMSO-d₆): δ = 197.94, 172.76, 138.51, 137.77, 135.02, 133.64, 132.55, 131.28, 130.49, 129.85, 128.99, 126.90, 39.94, 38.78, 21.61; **HRMS** (ESI+) *m/z*: calcd. For C1₅H1₃O₃ (M+H)⁺ 241.0859, found 241.0851.





¹H NMR (400MHz, DMSO-d₆): δ =13.20 (s, 1H), 7.53 – 7.42 (m, 2H), 7.42 – 7.26 (m, 3H), 7.24 – 7.11 (m, 2H), 7.06 (td, *J* = 7.1, 2.2 Hz, 1H), 6.85 (d, *J* = 1.3 Hz, 1H), 6.14 (s, 1H), 2.19 (s, 3H); ¹³C NMR (100 MHz, DMSO-d₆): δ =175.75, 143.10, 142.40, 138.15, 132.21, 128.27, 128.20, 128.06, 127.82, 127.48, 125.35, 81.64, 56.43, 40.05, 21.01; HRMS(ESI+) *m*/*z*: calcd. For C₁₅H₁₄O₃ (M+H)+ 243.1021, found 243.1011.





¹H NMR (400 MHz, DMSO-d₆): δ = 13.26 (s, 1H), 7.46 – 7.35 (m, 2H), 7.35 – 7.24 (m, 3H), 7.23 – 7.12 (m, 3H), 7.12 – 7.06 (m, 1H), 6.26 (s, 1H), 2.27 (s, 3H); ¹³C NMR (100 MHz, DMSO-d₆): δ = 175.40, 175.30, 144.36, 141.30, 136.89, 128.76, 128.15, 128.11, 127.70, 127.64, 127.56, 124.82, 80.64, 39.99, 21.14; HRMS(ESI+) *m*/*z*: calcd. For C₁₅H₁₄O₃ (M+H)⁺ 243.1021, found 243.1017.



HO COOH Ph

¹**H NMR** (400 MHz, DMSO-d₆): δ = 13.08 (s, 1H), 7.40 – 7.20 (m, 7H), 7.07-7.17 (d, *J* = 7.9 Hz, 2H), 6.22 (s, 1H), 2.28 (s, 3H); ¹³**C NMR** (100 MHz, DMSO-d₆): δ =175.39, 144.36, 141.30, 136.89, 128.75, 128.15, 127.70, 127.63, 127.56, 80.54, 40.00, 21.15; **HRMS**(ESI+) *m*/*z*: calcd. For C₁₅H₁₄O₃ (M+H)⁺243.1021, found 243.1023.







Supplementary figures and tables.

Figure S1. Photocarboxylation of *o-*, *m-* and *p-*methyl benzophenone with CO₂ in different solvent.



Figure S2. Electrocarboxylation of *o-*, *m-* and *p*-methyl benzophenone with CO₂ in different solvent.



Figure S3. Cyclic voltammogram of a) *o*-methyl benzophenone and b) *o*-acylphenylacetic acid on glassy carbon electrode under argon (black trace) or with CO₂ bubbling (red trace). The reaction solution consists of supporting electrolyte (0.1 M of Bu₄NBr) and substrate (0.08 mM) in 20 mL of **DMSO** by using Ag/AgBr as the reference electrode and Al as the counter electrode.

Quantum chemical computational details.

All calculations in this work were performed using Gaussian 09 program package [1]. Full geometry optimizations in DMSO solvent were performed to locate all the stationary points, using the M062X [2] method with the 6-311+G(d) [3] basis, namely M062X/6-311+G(d) at 298.25 K. The self-consistent reaction field (SCRF) method based on the universal solvation model SMD was adopted to evaluate the effect of the solvent [4]. The intrinsic reaction coordinate (IRC) path was traced to check the energy profiles connecting each transition state to two associated minima of the proposed mechanism [5]. The solvation free energies of the proton $\Delta G_{sol}(H^+)$ [6] is defined analogously as followed:

$$\Delta G_{\rm sol}(\rm H^+) = G(\rm S-\rm H)^+_{\rm sol} - G(\rm S_{\rm sol}) - G(\rm H^+_{\rm gas})$$

Wherein, S represents the solvent molecule, DMSO in this work; The value of $G(H_{gas}^+)$ was referenced from the literatures [7,8].

Unless specified, the Gibbs free energies at 298.25 K were used in the discussion.



Structure of 2-(2-(carboxymethyl)-phenyl)-2-hydroxy-2-phenylacetic acid.

Energies optimized at M062X/6-311+G(d) level

	E	Н	G	
CO ₂	-188.565	-188.561557	-188.58582	
Im1	-992.745	-992.725734	-992.809	
Im2	-992.838	-992.818457	-992.904	
Im3	-992.845	-992.825355	-992.896	
Ts1	-992.840	-992.821303	-992.889	
Im4	-992.848	-992.830033	-992.894	
Im5-tri	-992.889	-992.869853	-992.938	
Im5	-992.951	-992.933257	-992.996	
Im6	-992.955	-992.936065	-992.987	
Ts2	-992.955	-992.937251	-992.983	
Im7	-992.995	-992.977578	-993.023	
Im8	-993.459	-993.0211386	-993.079	
Ts3	-993.447	-993.0084776	-993.067	
Im9	-993.470	-993.0309856	-993.093	
Im10	-993.902	-993.0431033	-993.131	

Cartesian coordinates of structures optimized at M062X/6-311+G(d) level

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References

- Frisch, M.J.; Trucks, G.W.; Schlegel, H.B.; Scuseria, G.E.; Robb, M.A.; Cheeseman, J.R.; Scalmani, G.; Barone, V.; Mennucci, B.; Petersson, G.A.; Nakatsuji, H.; Caricato, M.; Li, X.; Hratchian, H.P.; Izmaylov, A. F.; Bloino, J.; Zheng, G.; Sonnenberg, J.L.; Hada, M.; Ehara, M.; Toyota, K.; Fukuda, R.; Hasegawa, J.; Ishida, M.; Nakajima, T.; Honda, Y.; Kitao, O.; Nakai, H.; Vreven, T.; Montgomery, J.A.; Peralta, J.J. E.; Ogliaro, F.; Bearpark, M.; Heyd, J.J.; Brothers, E.; Kudin, K.N.; Taroverov, V.N.; Keith, T.; Kobayashi, R.; Normand, J.; Raghavachari, K.; Rendell, A.; Burant, J.C.; Iyengar, S.S.; Tomasi, J.; Cossi, M.; Rega, N.; Millam, J.M.; Klene, M.; Knox, J.E.; Cross, J.B.; Bakken, V.; Adamo, C.; Jaramillo, J.; Gomperts, R.; Stratmann, R.E.; Yazyev, O.; Austin, A.J.; Cammi, R.; Pomelli, C.; Ochterski, J.W.; Martin, R.L.; Morokuma, K.; Zakrzewski, V.G.; Voth, G.A.; Salvador, P.; Dannenberg, J.J.; Dapprich, S.; Daniels, A.D.; Farkas, O.; Foresman, J.B.; Ortiz, J.V.; Cioslowski, J.; Fox, D.J., Gaussian 09 (Revision D.01), I. Gaussian, Wallingford, CT, **2013**.
- 2. Zhao, Y.; Truhlar, D.G. Density functionals with broad applicability in chemistry. *Accounts Chem. Res.* **2008**, *41*, 157-167.
- Krishnan, R.; Binkley, J.S.; Seeger, R.; Pople, J.A. Self-Consistent Molcular-orbital Methods .20. Basis Set for Correlated Wave-functions. J. Chem. Phys. 1980, 72, 650-654.
- 4. Marenich, A.V.; Cramer, C.J.; Truhlar, D.G. Universal solvation model based on solute electron density and on a continuum model of the solvent defined by the bulk dielectric constant and atomic surface tensions. *J. Phys. Chem. B* **2009**, *113*, 6378–6396.
- Gonzalez, C.; Schlegel, H.B. An Improved Algorithm for Reaction-path Following. J. Chem. Phys. 1989, 90, 2154-2161.
- 6. Markovic, Z.; Tosovic, J.; Milenkovic, D.; Markovic, S. Revisiting the solvation enthalpies and free energies of the proton and electron in various solvents. *Comput. Theor. Chem.* **2016**, *1077*, 11-17.
- 7. Fifen, J.J.; Dhaouadi, Z.; Nsangou, M. Revision of the Thermodynamics of the Proton in Gas Phase. *J. Phys. Chem. A* **2014**, *118*, 11090-11097.
- Fifen, J.J. Thermodynamics of the Electron Revisited and Generalized. J. Chem. Theory Comput. 2013, 9, 3165-3169.