



Supplementary Materials: Amphiphilic Polypeptides for VEGF siRNA Delivery into Retinal Epithelial Cells

Olga Osipova, Vladimir Sharoyko, Natalia Zashikhina, Natalya Zakharova, Tatiana Tennikova, Arto Urtti and Evgenia Korzhikova-Vlakh

1. Polymer Characterization

1.1. Static and Dynamic Light Scattering of Polymer Solutions

Sample *	SLS			DLS
	<i>dn/dc,</i> cm³/g	M_w	A₂, cm³·mol·g⁻²	Rh-D, nm
KEF1	0.0542	23000	-5.45×10-3	2.2
KEF2	0.0513	17500	-2.04×10-3	2.2
KEF3	0.0574	17400	-1.75×10 ⁻³	2.2
KEI1	0.0537	18800	-2.69×10-3	1.0
KEI2	0.0608	17100	-2.13×10 ⁻³	1.4

Table S1. Data of static (SLS) and dynamic (DLS) light scattering of polymer solutions (DMSO).

* Polymers used in protected forms as P(Lys(Z)-co-Glu(OBzl)-co-Phe) and P(Lys(Z)-co-Glu(OBzl)-co-Ile).

1.2. ¹H NMR Spectroscopy

Poly(L-lysine-co-L-glutamic acid-co-L-isoleucine)



Figure S1. ¹HNMR spectrum of deprotected sample KEI1.

 $Poly(L-\varepsilon-carboxybenzyl-lysine-co-L-\gamma-benzyl-glutamic acid-co-L-phenylalanine)$



Figure S2. ¹HNMR spectrum of protected sample KEF1.

Polymer composition calculation was carried out using following equations:

[**Phe**],%

$$= \left(\frac{(I(Lys(Z))_{6.9-7.4 \text{ ppm}} + I(Glu(OBzl))_{6.9-7.4 \text{ ppm}} + I(P \ e)_{6.9-7.4 \text{ ppm}})/5 - ((I(Lys(Z))_{4.7-5.2 \text{ ppm}} + I(Glu(OBzl))_{4.7-5.2 \text{ ppm}})/2}{(I(Lys(Z))_{6.9-7.4 \text{ ppm}} + I(Glu(OBzl))_{6.9-7.4 \text{ ppm}} + I(P \ e)_{6.9-7.4 \text{ ppm}})/5}\right) \times 100\%$$
(1),

 $\times 100\%$

where [Phe], % - molar fraction of Phe in the copolymer, I(Lys(Z))69-7.4 ppm, I(Glu(OBzl)) 69-7.4 ppm and *I*(*Phe*) 6.9-7.4 ppm. are relative integral areas of 5 aromatic protons of *Phe*, *Z*- and *OBzl*-groups of polymer at 6.9–7.4 ppm, *I*(*Lys*(*Z*))_{4.7-5.2 ppm} and *I*(*Glu*(*OBzl*))_{4.7-5.2 ppm} are relative integral areas of 2 CH₂ protons of *Z*- and *OBzl*-groups of polymer at 4.7–5.2 ppm.

 $Poly(L-\varepsilon-carboxybenzyl-lysine-co-L-\gamma-benzyl-glutamic acid-co-L-isoleucine)$



Figure S3. ¹HNMR spectrum of protected sample KEI2.

Polymer composition calculation was carried out using following equations:

$$[Ile], \% = \left(\frac{(I(Ile)_{0.5-0.9 \text{ ppm}})/6}{(I(Lys(Z))_{6.9-7.5 \text{ ppm}} + (I(Glu(OBzl))_{6.9-7.5 \text{ ppm}}/5)) * 100\%$$
(5)

Where [*Ile*], % - molar fraction of *Ile* in the copolymer, $I(Ile)_{0.5-0.9 \text{ ppm}}$ is relative integral area of 6 CH₃ protons of *Ile* at 0.5–0.9 ppm and $I(Lys(Z))_{6.9-7.5 \text{ ppm}}$ and $I(Glu(OBzl))_{6.9-7.5 \text{ ppm}}$ are relative integral areas of 5 aromatic protons of *Z*- and *OBzl*-groups of polymer at 6.9–7.5 ppm.

2. Determination of Critical Micelle Concentration



Figure S4. Dependences of conductivity on polymer concentration: (a) sample KEF1 and (b) sample KEI1.

3. Dynamic Light Scattering







Figure S5. DLS analysis of polymer nanoparticles prepared from P(Lys-co-Glu-co-Ile), sample KEI2.

4. Release of duplex oligo-dT-dA from polyplex with poly-L-lysine



Figure S6. Release of duplex oligo-dT-dA from the complex with poly-L-lysine:oligonucleotide (ratio 4:1).



5. Cytotoxicity of Nanoparticles

Figure S7. Nanoparticle's cytotoxicity in BEAS-2B and HEK-293 (**a**–**c**) and ARPE-19 cells (**d**) (24 h): (**a**) sample KEF1, (**b**) sample KEI1, (**c**) sample KEI2 and (**d**) sample KEI2.