

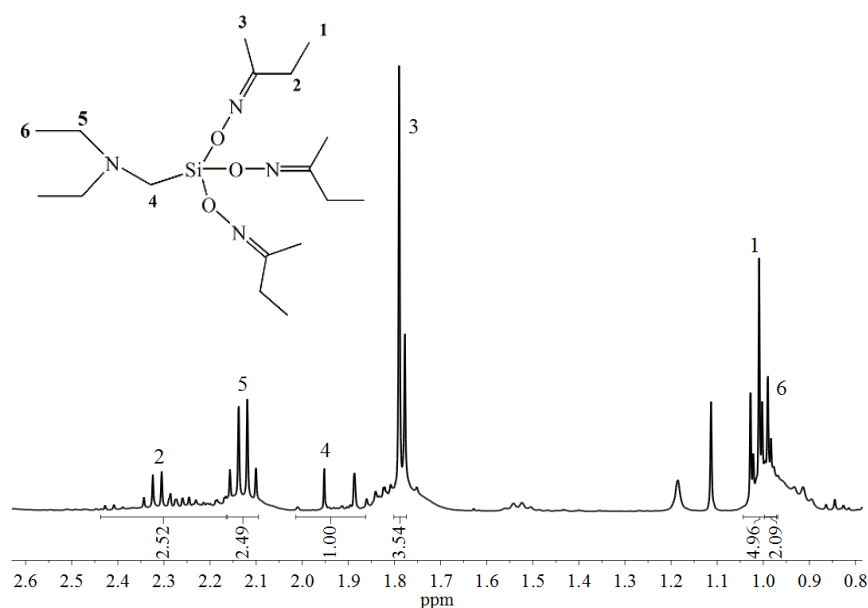
# Impacts of Different Functional Groups on the Kinetic Rates of $\alpha$ -Amine Ketoximesilanes Hydrolysis in the Preparation of Room Temperature Vulcanized Silicone Rubber

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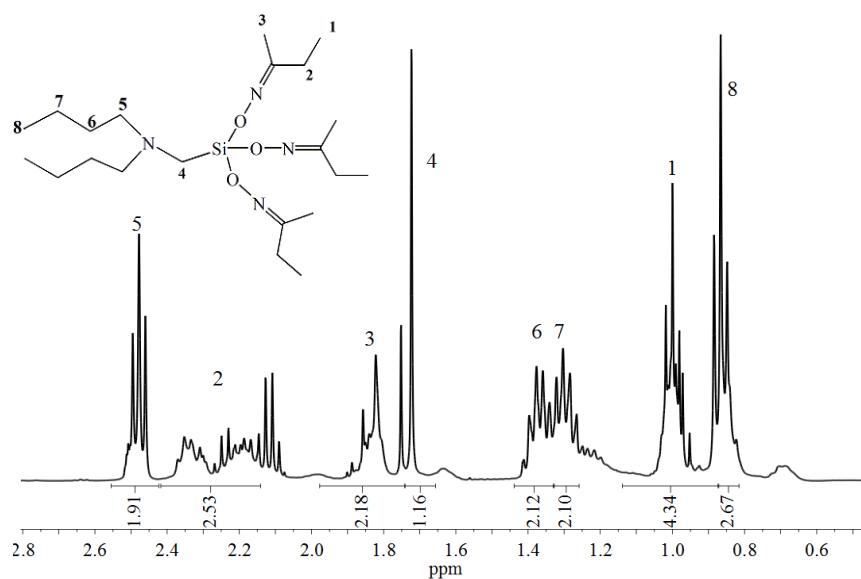
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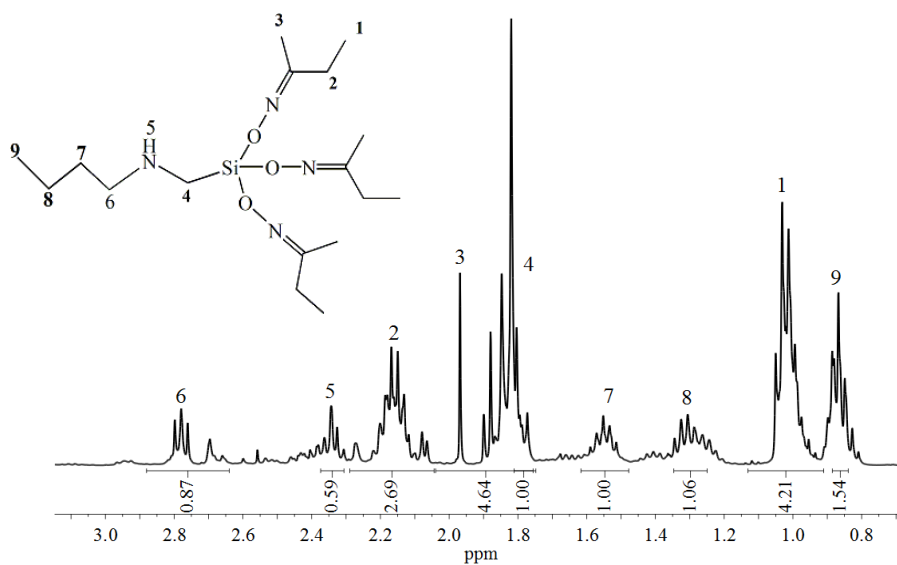
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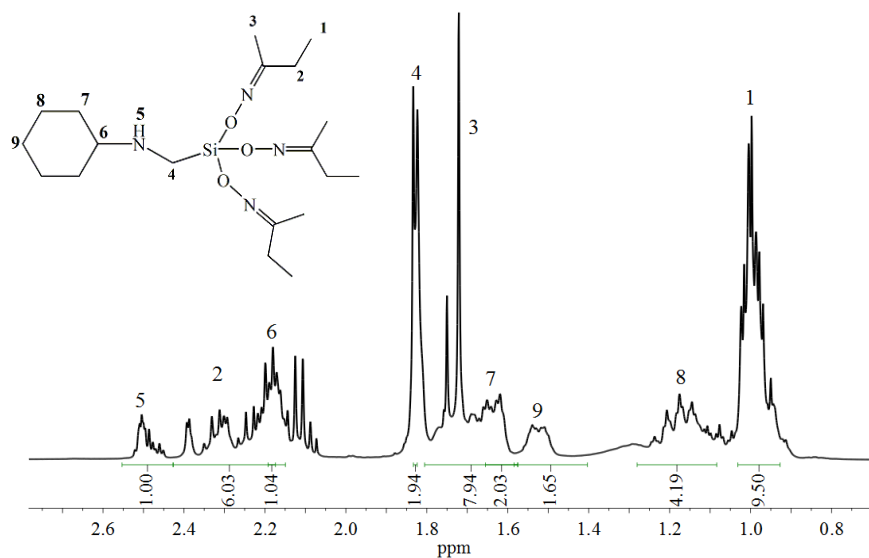
**Figure S1.** The <sup>1</sup>H NMR spectrum of  $\alpha$ -(N,N-diethyl)aminomethyltri(methyleth-ylketoximo)silane (DEMOS).



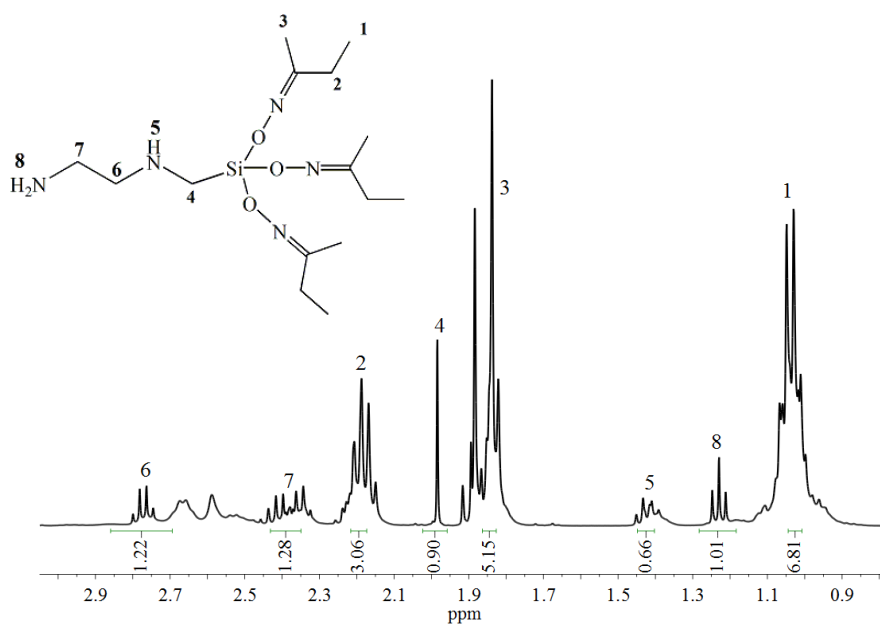
**Figure S2.** The  $^1\text{H}$ NMR spectrum of  $\alpha$ -(N,N-di-n-butyl)aminomethyltri(methylethylketoxime)silane (DBMOS).



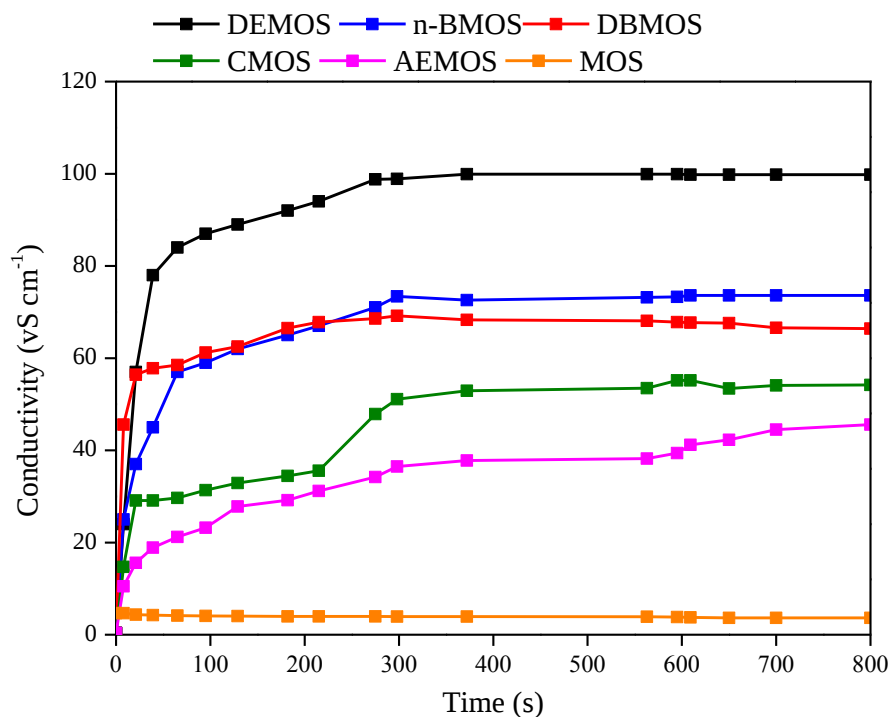
**Figure S3.** The  $^1\text{H}$ NMR spectrum of  $\alpha$ -(N-n-butyl)amino methyltri(methylethylketoxime)silane (n-BMOS).



**Figure S4.** The <sup>1</sup>H NMR spectrum of α-(N-cyclohexyl) aminomethyl tri(methylethylketoxime)silane (CMOS).



**Figure S5.** The <sup>1</sup>H NMR spectrum of α-(β-aminomethyl)aminomethyltri(methylethylketoxime)silane (AEMOS).



**Figure S6.** Conductibility of  $\alpha$ -amine ketoximesilanes in the presences of ethanol and water under different hydrolysis time. The mass ratio of  $\alpha$ -amine ketoximesilanes, water and ethanol was 1:5:13. The experiments were performed at the temperature of 25 °C and the humidity of 40%.

DEMOS,  $\alpha$ -(N,N-diethyl)aminomethyltri(methylethylketoxime)silane;

DBMOS,  $\alpha$ -(N,N-di-n-butyl)aminomethyltri(methylethylketoxime) silane;

n-BMOS,  $\alpha$ -(N-n-butyl)aminomethyl tri(methylethylketoxime) silane;

CMOS,  $\alpha$ -(N-cyclohexyl)aminomethyltri(methylethylketoxime)silane;

AEMOS,  $\alpha$ -( $\beta$ -aminomethyl)aminomethyltri(methylethylketoxime)silane.

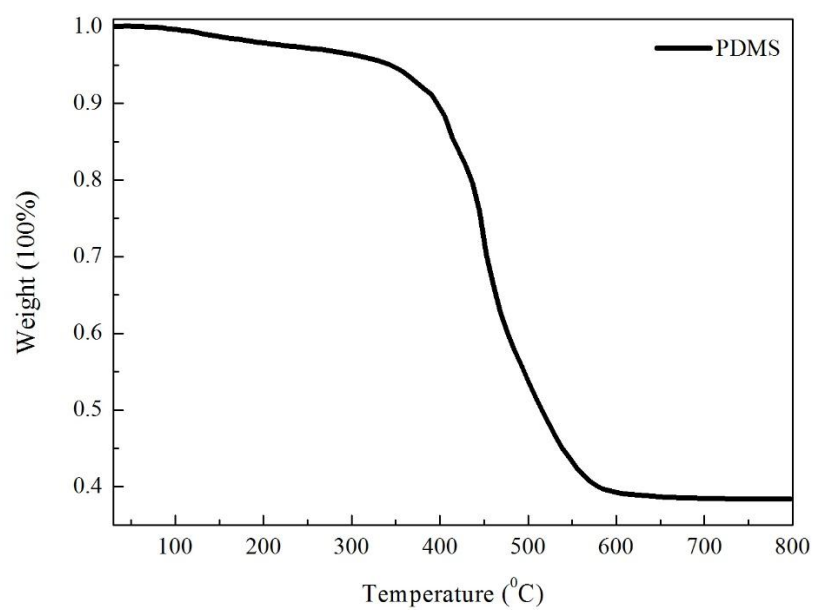


Figure S7. TGA curves of pure PDMS. The degradation temperature (the thermal weight loss consisted 5% of original weight) of pure PDMS was 343 °C. The degradation temperature (the thermal weight loss consisted 5% of original weight) of silicone rubber with CMOS, n-BMOS, DBMOS, DEMOS was 448 °C, 443 °C, 438 °C and 435 °C, respectively.