

## Supplementary Materials: Polycyclic Aromatic Hydrocarbons Adsorption onto Graphene: A DFT and AIMD Study

Bing Li, Pengfei Ou, Yulan Wei, Xu Zhang and Jun Song

**Table S1.** Calculated adsorption energies of polycyclic aromatic hydrocarbons (PAHs) adsorption onto Gr for different initial configurations (refer to the initial configurations (1)–(6) illustrated in Figure 1) by Perdew–Burke–Ernzerhof (PBE)-D3 ( $E_{ad}$ ; in eV).

PAHs	$E_{ad}$					
	Hollow (1)	Top (2)	Top (3)	Bridge (4)	Bridge (5)	Bridge (6)
Nap	-0.598	<b>-0.638</b>	<b>-0.638</b>	-0.635	-0.632	-0.632
Ace	-0.714	<b>-0.762</b>	<b>-0.762</b>	-0.743	-0.757	-0.757
Acp	-0.696	-0.728	<b>-0.731</b>	-0.713	-0.730	-0.730
Flu	-0.779	-0.802	-0.801	-0.802	<b>-0.812</b>	-0.801
Phe	-0.813	-0.866	<b>-0.870</b>	-0.866	-0.862	-0.862
Ant	-0.819	<b>-0.872</b>	<b>-0.872</b>	-0.868	-0.867	-0.862
Flt	-0.916	-0.932	-0.954	-0.932	<b>-0.955</b>	<b>-0.955</b>
Pyr	-0.897	-0.950	-0.960	<b>-0.961</b>	-0.948	-0.950
BaA	-1.030	<b>-1.101</b>	-1.092	-1.093	-1.098	-1.092
Chr	-1.025	<b>-1.099</b>	-1.086	-1.092	-1.086	-1.091
BbF	-1.128	-1.158	-1.178	-1.158	-1.178	<b>-1.182</b>
BkF	-1.135	<b>-1.187</b>	-1.170	-1.166	-1.182	-1.174
BaP	-1.114	-1.189	-1.189	<b>-1.190</b>	-1.178	-1.177
InP	-1.222	<b>-1.269</b>	-1.262	-1.259	-1.254	-1.265
DbA	-1.236	-1.323	-1.323	-1.313	<b>-1.324</b>	-1.309
BeP	-1.190	<b>-1.281</b>	<b>-1.281</b>	-1.267	-1.274	-1.273