



Article Theoretical Investigations of the Hexagonal Germanium Carbonitride

Xinhai Yu^{1,*}, Zhenyang Ma¹, Suriguge² and Peng Wang¹

- ¹ College of Airworthiness, Civil Aviation University of China, Tianjin 300300, China; zym_airworthiness@163.com (Z.M.); pw_airworthiness@163.com (P.W.)
- ² Department of Mechanical and Electrical Engineering, Hetao College, Bayannur Inner Mongolia 015000, China; suriguge87@163.com
- * Correspondence: xhyu@cauc.edu.cn; Tel.: +86-022-24-092-288

Received: 21 March 2018; Accepted: 19 April 2018; Published: 24 April 2018



Abstract: The structural, mechanical, elastic anisotropic, and electronic properties of hexagonal germanium carbonitride (*h*-GeCN) are systematically investigated using the first-principle calculations method with the ultrasoft pseudopotential scheme in the frame of generalized gradient approximation in the present work. The *h*-GeCN are mechanically and dynamically stable, as proved by the elastic constants and phonon spectra, respectively. The *h*-GeCN is brittle because the ratio B/G and Poisson's ratio v of the *h*-GeCN are less than 1.75 and 0.26, respectively. For *h*-GeCN, from brittleness to ductility, the transformation pressures are 5.56 GPa and 5.63 GPa for B/G and Poisson's ratio v, respectively. The *h*-GeCN exhibits the greater elastic anisotropy in Young's modulus and the sound velocities. In addition, the calculated band structure of *h*-GeCN reveals that there is no band gap for *h*-GeCN with the HSE06 hybrid functional, so the *h*-GeCN is metallic.

Keywords: GeCN; mechanical properties; elastic anisotropy properties; electronic properties

1. Introduction

Ternary compounds have attracted more and more attention, such as B–C–N [1–3], B–C–O [4–7] superhard materials, and Si-Ge-N [8,9], Si-C-N [10-14], Ge-C-N [15,16], and so on. Si-Ge-N is an alloy of silicon nitride and germanium nitride. The structural, elastic anisotropic, and electronic properties of *m*-Si₂GeN₄ and *m*-SiGe₂N₄ were investigated using density functional theory calculations by Ma et al. [8], where m-Si₂GeN₄ and m-SiGe₂N₄ are alloys of m-Si₃N₄ and m-Ge₃N₄. They found that the m-Si_xGe_{3-x}N₄ (x = 0, 1, 2, 3) series exhibit larger anisotropy and that the anisotropy of *m*-SiGe₂N₄ is largest among the *m*-Si_xGe_{3-x}N₄ (x = 0, 1, 2, 3). The calculated band structures show that both *m*-Si₂GeN₄ and *m*-SiGe₂N₄ are direct semiconductors with band gaps of 4.76 eV and 4.81 eV, respectively. Very recently, the structural, mechanical, anisotropic, electronic, and thermal properties of t-Si₂GeN₄ and t-SiGe₂N₄ in the tetragonal phase were systematically investigated by Han et al. [9]. They found that both t-Si₂GeN₄ and t-SiGe₂N₄ demonstrate brittleness, and that t-Si₂GeN₄ and t-SiGe₂N₄ exhibit larger elastic anisotropy than that of c-Si₂GeN₄ and c-SiGe₂N₄ characterized by Young's modulus, Poisson's ratio, the percentage of elastic anisotropy for shear modulus A_G , the percentage of elastic anisotropy for bulk modulus A_B , and the universal anisotropic index A^U . The electronic structures of *t*-Si₂GeN₄ and *t*-SiGe₂N₄ are both wide-bandgap semiconductor materials, with band gaps of 3.94 eV and 3.83 eV using the HSE06 hybrid functional, respectively. In addition, the effects of temperature and pressure on the Debye temperature, thermal expansion coefficient, heat capacity, and Grüneisen parameters were discussed in detail utilizing the quasi-harmonic Debye model. In addition, other III-V group compounds have also been studied extensively, including

three-dimensional materials [17–22] and some low-dimensional materials, such as few-layer *h*-AlN [23], buckled honeycomb XBi and XBi₃ (X = B, Al, Ga, and In) sheets [24], and buckled III-Bi sheets [25].

Ternary Si–C–N compounds represent a series of novel functional materials that have attracted considerable research interest because of their excellent chemical and physical properties, which include good creep properties, high hardness, excellent oxidation resistance, and thermal shock resistance over a broad temperature range [26]. Several Si–C–N compounds with different chemical compositions have been synthesized through various approaches [27–36]. Very recently, Cui et al. [26] performed an extensive structural search of SiCN compounds using the crystal structure analysis by particle swarm optimization (CALYPSO) algorithm [37]. They revealed that the novel tetragonal SiCN (*t*-SiCN) was more energetically stable than the *c*-SiCN proposed 40 years ago [38], and two high-pressure phases of orthorhombic SiCN (*o*-SiCN) and hexagonal SiCN (*h*-SiCN) were also predicted in the work. The *h*-SiCN and *o*-SiCN were able to be quenched at ambient conditions and exist in metastable phases. The hardnesses of *t*-SiCN, *o*-SiCN, and *h*-SiCN were calculated to be 41.5, 30.0, and 30.2 GPa, respectively [26].

Recently, Xing et al. [15] first investigated the structural, mechanical, electronic, and thermodynamic properties of the tetragonal structure germanium carbonitride (*t*-GeCN) using the density function theory with the ultrasoft pseudopotential scheme in the frame of the generalized gradient approximation and the local density approximation. The elastic constants and phonon spectra have confirmed that *t*-GeCN is mechanically stable and that *t*-GeCN is dynamically stable. The electronic structure of *t*-GeCN shows that it is an indirect semiconductor with band gap of 0.63 eV. The anisotropy studies show that *t*-GeCN exhibits a larger anisotropy in its Poisson's ratio, Young's modulus, shear modulus, sound velocities, and universal elastic anisotropy index. However, the elastic anisotropy of *t*-GeCN in Young's modulus is slightly smaller than that of hexagonal germanium carbonitride (*h*-GeCN).

The structure of *h*-GeCN is based on that of *h*-SiCN [26], with a germanium atom substituting for the silicon atom. The structural, mechanical, elastic anisotropy, and electronic properties of *h*-GeCN have not yet been studied. In this work, we have systematically investigated the physical properties of *h*-GeCN.

2. Theoretical Methods

The total energy calculations were performed using density functional theory (DFT) with the Perdew–Burke–Ernzerhof (PBE) exchange correlation in the framework of the generalized gradient approximation (GGA) [39] as implemented in the Cambridge Serial Total Energy Package (CASTEP) plane wave code [40]. The interactions between the ionic core and valence electrons were described by the ultrasoft pseudopotential [41], and the $2s^22p^2$, $2s^22p^3$, and $4s^24p^2$ were considered as valence electrons for C, N, and Ge, respectively. The equilibrium crystal structures were achieved by utilizing geometry optimization in the Broyden–Fletcher–Goldfarb–Shanno (BFGS) [42] minimization scheme. The plane wave basis set was truncated with a cutoff energy of 500 eV, and the Brillouin zone integration was generated using Monkhorst–Pack *k*-point meshes [43] with a high-quality grid of 0.025 Å⁻¹ (8 × 15 × 9) for total energy and elastic constants calculations, respectively. The elastic constants were calculated by the strain–stress method, which has been successfully utilized previously [44,45]. The bulk modulus, shear modulus, Young's modulus, and Poisson's ratio were estimated via Voigt–Reuss–Hill approximation [46–48]. The HSE06 hybrid functional [49] was used for the calculation of the electronic structures of *h*-GeCN.

3. Results and Discussion

3.1. Structural Properties

The calculated lattice parameters of *h*-GeCN and *t*-GeCN, together with other theoretical results, are all listed in Table 1. For *t*-GeCN, the lattice parameters *a* and *c* of the GGA deviate from the corresponding

previous values [15] by 0.07% and 0.13%, while lattice parameters a and c of the LDA (local density approximation) deviate from the corresponding previous values [15] by 0.26% and 1.47%; that is to say, the values of the GGA deviate from other previous values less than do those of the LDA. Therefore, in this paper, all the results are based on the GGA. In addition, the lattice parameters are a = 3.621 Å in this work, and a = 3.622 Å [2] with GGA for c-BN, while the lattice parameters are a = 3.582 Å in this work, and a = 3.576 Å [2] with LDA for c-BN; the experimental value of c-BN is 3.620 Å, so the result of GGA is very close to the experiment value for *c*-BN. Therefore, in this work, all the results are based on the crystal structure from GGA. The crystal structure of the hexagonal representation and rhombohedral representation for h-GeCN are shown in Figure 1. The red, black, and blue spheres represent Ge, C, and N atoms, respectively. For *h*-GeCN, the bond lengths of the C–N, C–Ge, and N–Ge bonds are 1.362 Å, 2.091 Å, and 2.180 Å, respectively. The C–N and C–Ge bond lengths are slightly greater than the N–Si (1.895 Å) and C–Si (1.875 Å) bond lengths in *h*-SiCN, while the C–N bond length is slightly smaller than that of C–N (1.373 Å) in *h*-SiCN. Compared with *t*-GeCN, the C–N bond length in *h*-GeCN is slightly smaller than that (1.445 Å) in t-GeCN, while the C–Ge and N–Ge bond lengths in h-GeCN are slightly greater than the C–Ge (2.015 Å) and N–Ge (1.884 Å) bond lengths in t-GeCN. In addition, there are C–C bond lengths (1.619 Å) in t-GeCN. The lattice constants and conventional cell volumes of h-GeCN and t-GeCN are shown in Figure 2. From Figure 2a, the compression along the lattice constants' a-axis and *c*-axis for *h*-GeCN is slightly larger than that of *t*-GeCN when the pressure increases. In addition, it is clear that the compression of *h*-GeCN is slightly larger than that of *t*-GeCN; that is to say, the bulk modulus of *t*-GeCN is slightly larger than that of *h*-GeCN.

Table 1. The calculated lattice parameters (in Å), cell volume (in Å³), and density (in g/cm^3) of hexagonal germanium carbonitride (*h*-GeCN) and tetragonal structure germanium carbonitride (*t*-GeCN).

| Material | Methods | Pressure | а | С | V | ρ |
|----------|---------|----------|-------|--------|---------|-------|
| h-GeCN | GGA | 0 | 3.165 | 10.701 | 92.819 | 5.292 |
| | | 5 | 3.122 | 10.627 | 89.687 | 5.477 |
| | | 10 | 3.087 | 10.570 | 87.207 | 5.633 |
| | | 15 | 3.056 | 10.523 | 85.100 | 5.772 |
| | | 20 | 3.029 | 10.480 | 83.270 | 5.899 |
| t-GeCN | GGA | 0 | 4.326 | 7.046 | 131.861 | 4.967 |
| | | 0^{1} | 4.323 | 7.037 | 131.490 | 4.981 |
| | | 0 | 4.216 | 6.993 | 124.298 | 5.269 |
| | LDA | 0 1 | 4.205 | 6.892 | 121.836 | 5.376 |

¹ Ref. [15]. GGA, generalized gradient approximation. LDA, local density approximation.

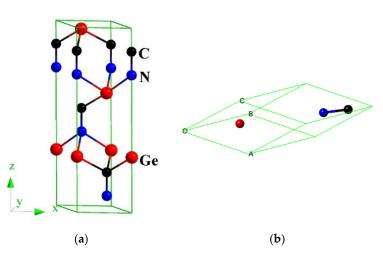


Figure 1. The crystal structures of *h*-GeCN: hexagonal representation (**a**) and rhombohedral representation (**b**).

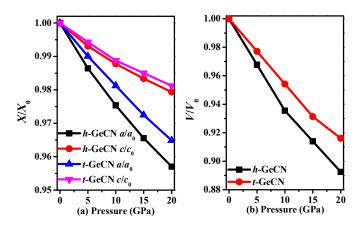


Figure 2. The lattice constants X/X_0 (**a**) and primitive cell volume V/V_0 (**b**) as functions of pressure for *h*-GeCN and *t*-GeCN.

3.2. Stability and Mechanical Properties

The stability of *h*-GeCN can be characterized by phonon spectra and Born stability conditions. The phonon spectra of *h*-GeCN are displayed in Figure 3. The phonon spectra show that all the lattice vibrations in the Brillouin region are positive, indicating that the *h*-GeCN is dynamically stable. The elastic constants of *h*-GeCN under different pressures are listed in Table 2. The criteria for mechanical stability of hexagonal symmetry are [50] $C_{44} > 0$, $C_{11}^2 > C_{12}^2$, and $(C_{11} + 2C_{12})C_{33} > 2 C_{12}^2$. From Table 2, we note that all the elastic constants of *h*-GeCN under different pressures satisfy the Born stability conditions of hexagonal symmetry.

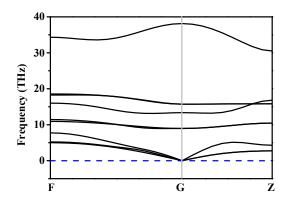


Figure 3. The phonon spectra for *h*-GeCN.

Table 2. The calculated elastic constants (in GPa) of *h*-GeCN and *t*-GeCN.

| Material | Pressure | <i>C</i> ₁₁ | <i>C</i> ₁₂ | <i>C</i> ₁₃ | <i>C</i> ₁₄ | <i>C</i> ₂₄ | C ₃₃ | C ₄₄ | C ₅₆ | <i>C</i> ₆₆ |
|-------------------------|----------|------------------------|------------------------|------------------------|------------------------|------------------------|-----------------|-----------------|-----------------|------------------------|
| | 0 | 232 | 77 | 21 | 4 | -4 | 512 | 42 | 4 | 78 |
| | 5 | 282 | 95 | 48 | -7 | 7 | 586 | 60 | -7 | 93 |
| h-GeCN | 10 | 329 | 106 | 68 | -20 | 20 | 644 | 65 | -20 | 112 |
| | 15 | 361 | 126 | 91 | -27 | 27 | 694 | 68 | -27 | 117 |
| | 20 | 380 | 158 | 109 | -33 | 33 | 741 | 59 | -33 | 111 |
| t-GeCN | 0 1 | 263 | 143 | 94 | - | - | 492 | 151 | - | 167 |
| ¹ Ref. [15]. | | | | | | | | | | |

The elastic moduli of *h*-GeCN under different pressures are listed in Table 3. According to our previous prediction, the bulk modulus of *t*-GeCN (183 GPa) is indeed larger than that of *h*-GeCN

(130 GPa). Similarly, the shear modulus and the Young's modulus are the same as the bulk modulus. However, the bulk modulus, shear modulus, and Young's modulus of *h*-GeCN are all slightly larger than those of *m*-Ge₃N₄ [8]. A kind of material showing brittleness or ductility is usually characterized by two physical quantities: B/G and Poisson's ratio v. A larger B/G [51] value (B/G > 1.75) and a larger v (v > 0.26) [52] for a solid represent a ductile state, while a smaller B/G value and a smaller v usually mean that the solid is brittle. The B/G and Poisson's ratio v of *h*-GeCN are also presented in Table 3. From Table 3, with increasing pressure, both B/G and Poisson's ratio v increase. At ambient pressure, B/G = 1.69 and v = 0.25 of *h*-GeCN, indicating that the *h*-GeCN exhibits brittleness. As the pressure increases, the *h*-GeCN changes from brittle to ductile. From brittleness to ductility, we note that the transformation pressures of *h*-GeCN are 5.56 GPa and 5.63 GPa for B/G and Poisson's ratio v, respectively.

| Material | Pressure | В | G | B/G | Ε | υ | A^{U} |
|----------|----------|-----|-----|-------|-----|-------|---------|
| h-GeCN | 0 | 130 | 77 | 1.688 | 193 | 0.253 | 2.112 |
| | 5 | 165 | 95 | 1.737 | 239 | 0.258 | 1.463 |
| | 10 | 193 | 104 | 1.856 | 264 | 0.272 | 1.722 |
| | 15 | 219 | 107 | 2.047 | 276 | 0.290 | 1.950 |
| | 20 | 244 | 98 | 2.490 | 259 | 0.323 | 3.021 |
| t-GeCN | 0 | 183 | 129 | 1.42 | 313 | 0.210 | 0.925 |

Table 3. The calculated elastic modulus (in GPa) of *h*-GeCN.

The Debye temperature (Θ_D) is a fundamental physical property and correlates with many physical properties of solids, such as specific heat and the thermal coefficient [53]. $\Theta_D = (h/k_B)[(3n/4\pi)(N_A\rho/M)]^{1/3}v_m$, where *h* is Planck's constant; k_B is Boltzmann's constant; N_A is Avogadro's number; *n* is the number of atoms in the molecule; *M* is molecular weight; ρ is the density; and v_m is the mean sound velocity, $v_m = [(2/v_s^3 + 1/v_p^3)/3]^{-1/3}$. The v_l and v_t are the longitudinal and transverse sound velocities, respectively, which can be obtained from Navier's equation [54]: $v_p = [(B + 4G/3)/\rho]^{1/2}$, $v_s = (G/\rho)^{1/2}$. The calculated Debye temperature and sound velocity of *h*-GeCN under different pressures are listed in Table 4. At ambient pressure, the Debye temperature of *h*-GeCN is 506 K—smaller than that of *t*-GeCN (756 K). The Debye temperature of *h*-GeCN increases with increasing pressure except for the situation under 20 GPa. The changes of almost all of the sound velocity v_p increases with increasing pressure increases to 20 GPa. The Debye temperature, except for v_p . The sound velocity v_p increases with increasing pressure increases to 20 GPa. The Debye temperature of *h*-GeCN shows different behavior at 20 GPa because the elastic constants and elastic moduli of *h*-GeCN shows different behaviors at 20 GPa.

Table 4. The density (ρ in g/cm³), sound velocity (v_l , v_t , v_m , in m/s), and Debye temperature (Θ_D in K) of *h*-GeCN under pressure.

| Materials | Pressure | v_p | v_s | v_m | Θ_D |
|-----------|----------|-------|-------|-------|------------|
| h-GeCN | 0 | 6631 | 3814 | 4236 | 506 |
| | 5 | 7297 | 4165 | 4628 | 559 |
| | 10 | 7673 | 4297 | 4783 | 583 |
| | 15 | 7916 | 4306 | 4803 | 590 |
| | 20 | 7970 | 4076 | 4566 | 565 |

3.3. Elastic Anisotropy Properties

The sound velocities are determined by the symmetry of the crystal and the propagation direction. The pure transverse and longitudinal modes can only be found in [100] and [001] directions in a hexagonal crystal; the sound propagating modes in other directions are the quasi-transverse or quasi-longitudinal waves. In the primary directions, the sound velocities in a hexagonal crystal can be expressed by

$$[100]: [100]v_l = \sqrt{(C_{11} - C_{12})/2\rho}, [010]v_{t1} = \sqrt{C_{11}/\rho}, [001]v_{t2} = \sqrt{C_{44}/\rho}$$

$$[001]: [001]v_l = \sqrt{C_{33}/\rho}, [100]v_{t1} = \sqrt{C_{44}/\rho}, [010]v_{t2} = \sqrt{C_{44}/\rho}$$

$$(1)$$

where v_{t1} and v_{t2} refer to the first transverse mode and the second transverse mode, respectively. The calculated sound velocities along the primary directions are listed in Table 5. For *h*-GeCN, in the [001] propagation direction with polarization direction [001], the longitudinal sound velocities v_l have the greatest sound velocity at 0 GPa. The smallest sound velocities result along the [001] propagation direction, with polarization directions $[100]v_{t1}$, $[010]v_{t2}$; and along the [100] propagation direction, with polarization directions $[100]v_{t1}$, $[010]v_{t2}$; and along the [100] propagation direction, with polarization directions, the sound velocities have different values. In a sense, the *h*-GeCN is anisotropic. In addition, the sound waves along different propagation directions increase with increasing pressure from 0 GPa to 15 GPa, while the [001] propagation direction with polarization directions [100] v_{t1} , [010] v_{t2} , and the [100] propagation direction with polarization direction [100] v_l decrease from 15 GPa to 20 GPa. Because the difference between C_{11} and C_{12} is smaller, C_{44} decreases from 15 GPa to 20 GPa.

| Material | Р | [100] | | | [001] | | | |
|----------|----|---------------------|---------------|---------------|------------|---------------|---------------|--|
| | | [100]v _l | $[010]v_{t1}$ | $[001]v_{t2}$ | $[001]v_l$ | $[100]v_{t1}$ | $[010]v_{t2}$ | |
| h-GeCN | 0 | 3827 | 4682 | 1992 | 6955 | 1992 | 1992 | |
| | 5 | 4132 | 5074 | 2340 | 7314 | 2340 | 2340 | |
| | 10 | 4449 | 5404 | 2402 | 7561 | 2402 | 2402 | |
| | 15 | 4512 | 5592 | 2427 | 7754 | 2427 | 2427 | |
| | 20 | 4338 | 5675 | 2236 | 7925 | 2236 | 2236 | |

Table 5. The sound velocities along different directions of *h*-GeCN at different pressures.

The Young's modulus of the h-GeCN also exhibits anisotropy. The directional dependence of Young's modulus for h-GeCN and two-dimensional (2D) representations of Young's modulus in the (001), (010), (100), and (111) planes for h-GeCN are illustrated in Figure 4a,b, respectively. From Figure 4a, the shape of the three-dimensional representations of the Young's modulus for *h*-GeCN is similar to a gyroscope with the middle width and the two ends sharp. The two-dimensional representations of Young's modulus for h-GeCN are unfolding figures that cut along the (001), (010), (100), and (111) planes, where black, red, blue, and cyan lines represent the (001), (010), (100), and (111) planes, respectively. The figure obtained along the (001) plane is a circle, and the two figures along the (010) and (100) planes are the same, for a gyroscope plane shape, while the (111) plane is an irregular figure. What is more interesting is that the maximum value (509 GPa) of Young's modulus for h-GeCN occurred at the Z-axis, but the minimum value (130 GPa) of the Young's modulus for h-GeCN occurred at θ = 0.87, φ = 5.08 (more details see [55–57]). Regardless of the three-dimensional figure of the Young's modulus and the ratio of the maximum to the minimum ($E_{\text{max}}/E_{\text{min}} = 509/130 = 3.92$), it is more than that of *t*-GeCN ($E_{\text{max}}/E_{\text{min}} = 2.49$) [15], so the *h*-GeCN has larger anisotropy. In the (010) and (100) planes, the ratio $E_{\text{max}}/E_{\text{min}} = 509/130 = 3.92$; this is the largest ratio of elastic anisotropy in the Young's modulus among these planes. In the (001) plane, the maximal and minimal values of Young's modulus are both 206 GPa, so the ratio of Young's modulus in the (001) plane is $E_{\text{max}}/E_{\text{min}} = 206/206 = 1.00$; therefore, the Young's modulus exhibits isotropy in the (001) plane, and it is the smallest elastic anisotropy in Young's modulus among these planes.

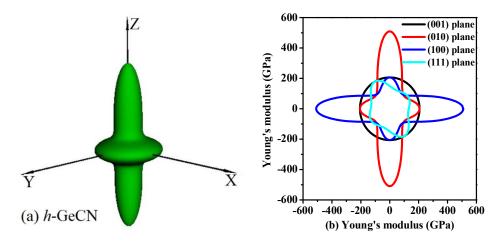


Figure 4. The surface construction of Young's modulus for *h*-GeCN (**a**); and the 2D representation of Young's modulus for *h*-GeCN in the (001) plane, (010) plane, (100) plane, and (111) plane (**b**). The black, red, blue, and cyan lines represent the Young's modulus of *t*-Si₃N₄, *t*-Si₂GeN₄, *t*-SiGe₂N₄, and *t*-Ge₃N₄, respectively. All units are in GPa.

3.4. Electronic Properties

It is well known that the electronic structure determines the fundamental physical and chemical properties of materials [35]. The electronic structures of *h*-GeCN (using the rhombohedral cell) under 0 GPa and 20 GPa are shown in Figure 5. From Figure 5, we can see that *h*-GeCN exhibits metallicity. From 0 GPa to 20 GPa, the electronic structures of *h*-GeCN (using the rhombohedral cell) are almost unchanged. The Fermi energy level is the highest level of electrons full of electrons in a solid energy band when the temperature is absolute zero. The Fermi level of *h*-GeCN at 0 GPa is 3.50 eV, while the Fermi level of *h*-GeCN is 4.30 eV when the pressure is under 20 GPa.

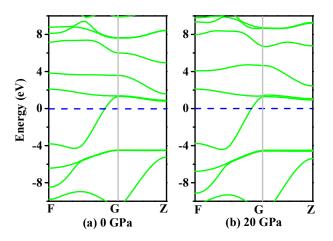


Figure 5. The band structures of *h*-GeCN at 0 GPa (a) and 20 GPa (b).

4. Conclusions

In this work, the structural, elastic, elastic anisotropic, and electronic properties of *h*-GeCN in the *R*3*m* space group were investigated utilizing first-principle calculations. The mechanical and dynamical stability of *h*-GeCN were proved by elastic constants and phonon spectra. The ratio B/G and Poisson's ratio v of the *h*-GeCN are less than 1.75 and 0.26, respectively, both of which indicate that the *h*-GeCN is brittle. For *h*-GeCN, from brittleness to ductility, the transformation pressures are 5.56 GPa and 5.63 GPa for B/G and Poisson's ratio v, respectively. At ambient pressure, the Debye temperature of *h*-GeCN is 506 K—smaller than that of *t*-GeCN. The calculated Young's modulus along

all directions and in the primary planes, and the sound velocities along the primary directions of *h*-GeCN, exhibit greater elastic anisotropy. A three-dimensional figure of the Young's modulus was presented, and the ratio of the maximum to the minimum ($E_{\text{max}}/E_{\text{min}} = 509/130 = 3.92$) is greater than that of *t*-GeCN ($E_{\text{max}}/E_{\text{min}} = 2.49$). In addition, the band structure reveals that the *h*-GeCN is metallic.

Author Contributions: Xinhai Yu and Zhenyang Ma designed the project, Xinhai Yu, Zhenyang Ma, Suriguge, Yanxing Song, and Peng Wang performed the calculations; Xinhai Yu, Zhenyang Ma, Suriguge, and Peng Wang determined the results; Xinhai Yu wrote the manuscript.

Acknowledgments: The project was supported by the National Natural Science Foundation of China (No. 61601468), the Fundamental Research Funds for the Central Universities (No. 3122017087) and the Scientific Research Starting Foundation of Civil Aviation University of China (No. 2016QD01X). Yanxing Song (School of Microelectronics, Xidian University) is thanked for allowing us to use the CASTEP code in Materials Studio.

Conflicts of Interest: The authors declare no conflict of interest.

References

- 1. Fan, Q.Y.; Wei, Q.; Chai, C.C.; Zhang, M.G.; Yan, H.Y.; Zhang, Z.X.; Zhang, J.Q.; Zhang, D.Y. Elastic and electronic properties of Imm2- and I 4 m2-BCN. *Comput. Mater. Sci.* **2015**, *97*, 6–13. [CrossRef]
- Fan, Q.Y.; Wei, Q.; Chai, C.C.; Yan, H.Y.; Zhang, M.G.; Lin, Z.Z.; Zhang, Z.X.; Zhang, J.Q.; Zhang, D.Y. Structural, mechanical, and electronic properties of P3m1-BCN. J. Phys. Chem. Solids 2015, 79, 89–96. [CrossRef]
- 3. Zhang, X.X.; Wang, Y.C.; Lv, J.; Zhu, C.Y.; Li, Q.; Zhang, M.; Li, Q.; Ma, Y. First-principles structural design of superhard materials. *J. Chem. Phys.* **2013**, *138*, 114101. [CrossRef] [PubMed]
- 4. Liu, C.; Zhao, Z.S.; Luo, K.; Hu, M.; Ma, M.D.; He, J.L. Superhard orthorhombic phase of B₂CO compound. *Diam. Relat. Mater.* **2017**, *73*, 87–92. [CrossRef]
- 5. Wang, S.N.; Oganov, A.R.; Qian, G.R.; Zhu, Q.; Dong, H.F.; Dong, X.; Mahdi, D.E. Novel superhard B–C–O phases predicted from first principles. *Phys. Chem. Chem. Phys.* **2016**, *18*, 1859–1863. [CrossRef] [PubMed]
- 6. Qiao, L.P.; Zhao, J. Two B-C-O compounds: structural, mechanical anisotropy and electronic properties under pressure. *Materials* **2017**, *10*, 1413. [CrossRef] [PubMed]
- 7. Zhang, M.G.; Yan, H.Y.; Zheng, B.B.; Wei, Q. Influences of carbon concentration on crystal structures and ideal strengths of B₂C_xO compounds in the B-C-O system. *Sci. Rep.* **2015**, *5*, 15481. [CrossRef] [PubMed]
- 8. Ma, Z.Y.; Yan, F.; Wang, S.X.; Jia, Q.Q.; Yu, X.H.; Shi, C.L. Mechanical, elastic, anisotropy, and electronic properties of monoclinic phase of *m*-SixGe_{3-x}N₄. *Chin. Phys. B* **2017**, *26*, 126105. [CrossRef]
- 9. Han, C.X.; Chai, C.C.; Fan, Q.Y.; Yang, J.H.; Yang, Y.T. Structural, electronic, and thermodynamic properties of tetragonal t-Si_xGe_{3-x}N₄. *Materials* **2018**, *11*, 397. [CrossRef] [PubMed]
- Miao, N.X.; Pu, C.Y.; He, C.Z.; Zhang, F.W.; Lu, C.; Lu, Z.W.; Zhou, D.W. Mechanical and thermodynamic properties of the monoclinic and orthorhombic phases of SiC₂N₄ under high pressure from first principles. *Chin. Phys. B* 2014, 23, 127101. [CrossRef]
- 11. Wang, H.B.; Li, Q.; Wang, H.; Liu, H.Y.; Cui, T.; Ma, Y.M. Design of superhard ternary compounds under high pressure: SiC₂N₄ and Si₂CN₄. *J. Phys. Chem. C* **2010**, *114*, 8609. [CrossRef]
- 12. Du, H.J.; Guo, L.C.; Li, D.C.; Yu, D.L.; He, J.L. First-principles calculations of phase transition and stability of Si₂CN₄ under high pressure. *Chin. Phys. Lett.* **2009**, *26*, 016403. [CrossRef]
- 13. Hart, J.N.; Claeyssens, F.; Allan, N.L.; May, P.W. Carbon nitride: Ab initio investigation of carbon-rich phases. *Phys. Rev. B* 2009, *80*, 174111. [CrossRef]
- 14. Zhang, X.Y.; Chen, Z.W.; Du, H.J.; Yang, C.; Ma, M.Z.; He, J.L.; Tian, Y.J.; Liu, R.P. Structural, elastic, and thermal properties of Laves phase ZrV₂ under pressure. *J. Appl. Phys.* **2008**, *103*, 083533. [CrossRef]
- 15. Xing, M.J.; Li, B.H.; Yu, Z.T.; Chen, Q. Structural, elastic and thermodynamic properties of the tetragonal structure of germanium Carbonitride. *J. Phys. Chem. Solids* **2016**, *91*, 106–113. [CrossRef]
- Liu, X.L.; Kang, J.Y. Structure prediction of ternary Ge-C-N System. In Proceedings of the 2004 13th International Conference on Semiconducting and Insulating Materials (SIMC-XIII-2004), Beijing, China, 20–25 September 2004; pp. 210–213.
- 17. Yang, R.K.; Zhu, C.S.; Wei, Q.; Du, Z. A first-principles study of the properties of four predicted novel phases of AlN. J. Phys. Chem. Solids 2017, 104, 68–78. [CrossRef]

- 18. Liu, C.; Hu, M.; Luo, K.; Cui, L.; Yu, D.L.; Zhao, Z.S.; He, J.L. Novel high-pressure phases of AlN: A first-principles study. *Comput. Mater. Sci.* 2016, *117*, 496–501. [CrossRef]
- Yang, R.K.; Zhu, C.S.; Wei, Q.; Zhang, D.Y. First-principles study on phases of AlP. Solid State Commun. 2017, 267, 23–28. [CrossRef]
- 20. Xu, L.F.; Bu, W. Mechanical and thermodynamic properties of AlX (X = N, P, As) compounds. *Int. J. Mod. Phys. B* 2017, *31*, 1750167. [CrossRef]
- 21. Liu, C.; Ma, M.D.; Yuan, X.H.; Sun, H.; Ying, P.; Xu, B.; Zhao, Z.S.; He, J.L. Metastable phases, phase transformation and properties of AlAs based on first-principle study. *Comput. Mater. Sci.* 2017, *128*, 337–342. [CrossRef]
- 22. Mujica, A.; Rubio, A.; Muñoz, A.; Needs, R.J. High-pressure phases of group-IV, III–V, and II–VI compounds. *Rev. Mod. Phys.* **2003**, *75*, 863–912. [CrossRef]
- Dos Santos, R.B.; de Brito Mota, F.; Rivelino, R.; Kakanakova-Georgieva, A.; Gueorguiev, G.K. Van der Waals stacks of few-layer h-AlN with graphene: An ab initio study of structural, interaction and electronic properties. *Nanotechnology* 2016, 27, 145601. [CrossRef] [PubMed]
- 24. Freitas, R.R.Q.; de Brito Mota, F.; Rivelino, R.; de Castilho, C.M.C.; Kakanakova-Georgieva, A.; Gueorguiev, G.K. Spin-orbit-induced gap modification in buckled honeycomb XBi and XBi₃ (X = B, Al, Ga, and In) sheets. *J. Phys.: Condens. Matter* **2015**, *27*, 485306. [CrossRef] [PubMed]
- Freitas, R.R.Q.; de Brito Mota, F.; Rivelino, R.; de Castilho, C.M.C.; Kakanakova-Georgieva, A.; Gueorguiev, G.K. Tuning band inversion symmetry of buckled III-Bi sheets by halogenation. *Nanotechnology* 2016, 27, 055704. [CrossRef] [PubMed]
- 26. Cui, L.; Wang, Q.Q.; Xu, B.; Yu, D.L.; Liu, Z.Y.; Tian, Y.J.; He, J.L. Prediction of novel SiCN compounds: First-principles calculations. *J. Phys. Chem. C* 2013, *117*, 21943–21948. [CrossRef]
- 27. Badzian, A. Stability of silicon carbonitride phases. J. Am. Ceram. Soc. 2002, 85, 16–20. [CrossRef]
- 28. Raj, R.; An, L.; Shah, S.; Riedel, R.; Fasel, C.; Kleebe, H.J. Oxidation kinetics of an amorphous silicon carbonitride ceramic. *J. Am. Ceram. Soc.* **2001**, *84*, 1803–1810. [CrossRef]
- He, X.M.; Taylor, T.N.; Lillard, R.S.; Walter, K.C.; Nastasi, M. Bonding structure and properties of ion enhanced reactive magnetron sputtered silicon carbonitride films. *J. Phys. Condens. Matter* 2000, *12*, L591. [CrossRef]
- Xiao, X.C.; Li, Y.W.; Song, L.X.; Peng, X.F.; Hu, X.F. Structural analysis and microstructural observation of SiC_xN_y films prepared by reactive sputtering of SiC in N₂ and Ar. *Appl. Surf. Sci.* 2000, 156, 155–160. [CrossRef]
- 31. Chen, L.C.; Chen, K.H.; Wei, S.L.; Kichambare, P.D.; Wu, J.J.; Lu, T.R.; Kuo, C.T. Crystalline SiCN: A hard material rivals to cubic BN. *Thin Solid Films* **1999**, *355*, 112–116. [CrossRef]
- 32. Bendeddouche, A.; Berjoan, R.; Bêche, E.; Hillel, R. Hardness and stiffness of amorphous SiC_xN_y chemical vapor deposited coatings. *Surf. Coat. Technol.* **1999**, *111*, 184–190. [CrossRef]
- 33. An, L.; Riedel, R.; Konetschny, C.; Kleebe, H.J.; Raj, R. Newtonian viscosity of amorphous silicon carbonitride at high temperature. *J. Am. Ceram. Soc.* **1998**, *81*, 1349–1352. [CrossRef]
- 34. Chen, L.C.; Chen, C.K.; Wei, S.L.; Bhusari, D.M.; Chen, K.H.; Chen, Y.F.; Jong, Y.C.; Huang, Y.S. Crystalline silicon carbon nitride: A wide band gap semiconductor. *Appl. Phys. Lett.* **1998**, *72*, 2463–2465. [CrossRef]
- 35. Machorro, R.; Samano, E.C.; Soto, G.; Cota, L. SiC_xN_y thin films alloys prepared by pulsed excimer laser deposition. *Appl. Surf. Sci.* **1998**, 127, 564–568. [CrossRef]
- 36. Riedel, R.; Greiner, A.; Miehe, G.; Dressler, W.; Fuess, H.; Bill, J.; Aldinger, F. The First crystalline solids in the ternary Si-C-N system. *Angew. Chem., Int. Ed.* **1997**, *36*, 603–606. [CrossRef]
- 37. Wang, Y.C.; Lv, J.; Zhu, L.; Ma, Y.M. Crystal structure prediction via particle swarm optimization. *Phys. Rev. B* **2010**, *82*, 094116. [CrossRef]
- Inorganic Crystal Structure Database. ICSD Collection Code 28391. Available online: http://icsd.fizkarlsruhe.de (accessed on 19 April 2018).
- Perdew, J.P.; Burke, K.; Ernzerhof, M. Generalized gradient approximation made simple. *Phys. Rev. Lett.* 1996, 77, 3865. [CrossRef] [PubMed]
- 40. Clark, S.J.; Segall, M.D.; Pickard, C.J.; Hasnip, P.J.; Probert, M.I.J.; Refson, K.; Payne, M.C. First principles methods using CASTEP. Z. Kristallogr.-Cryst. Mater. 2005, 220, 567–570. [CrossRef]
- 41. Vanderbilt, D. Soft self-consistent pseudopotentials in a generalized eigenvalue formalism. *Phys. Rev. B* **1990**, 41, 7892R–7895R. [CrossRef]

- 42. Shanno, D.F.; Kettler, P.C. Optimal conditioning of quasi-newton methods. *Math. Comput.* **1970**, 24, 657–664. [CrossRef]
- 43. Monkhorst, H.J.; Pack, J.D. Special points for brillouin-zone integrations. *Phys. Rev. B* 1976, 13, 5188. [CrossRef]
- 44. Fan, Q.Y.; Wei, Q.; Yan, H.Y.; Zhang, M.J.; Zhang, Z.X.; Zhang, J.Q.; Zhang, D.Y. Elastic and electronic properties of Pbca-BN: First-principles calculations. *Comput. Mater. Sci.* **2014**, *85*, 80–87. [CrossRef]
- 45. Xing, M.J.; Li, B.H.; Yu, Z.T.; Chen, Q. C2/*m*-carbon: structural, mechanical, and electronic properties. *J. Mater. Sci.* **2015**, *50*, 7104–7114. [CrossRef]
- 46. Voigt, W. Lehrburch der Kristallphysik; Teubner, B.G., Ed.; Johnson Reprint Corp.: Leipzig, Germany, 1928.
- 47. Reuss, A. Berechnung der Fließgrenze von Mischkristallen auf Grund der Plastizitätsbedingung für Einkristalle. Z. Angew. Math. Mech. 1929, 9, 49–58. [CrossRef]
- 48. Hill, R. The elastic behavior of a crystalline aggregate. Proc. Phys. Soc. 1952, 65, 349–354. [CrossRef]
- 49. Krukau, A.V.; Vydrov, O.A.; Izmaylov, A.F.; Scuseria, G.E. Influence of the exchange screening parameter on the performance of screened hybrid functionals. *J. Chem. Phys.* **2006**, *125*, 224106. [CrossRef] [PubMed]
- 50. Wu, Z.J.; Zhao, E.J.; Xiang, H.P.; Hao, X.F.; Liu, X.J.; Meng, J. Crystal structures and elastic properties of superhard IrN₂ and IrN₃ from first principles. *Phys. Rev. B* **2007**, *76*, 054115. [CrossRef]
- 51. Pugh, S.F. Relations between the elastic moduli and the plastic properties of polycrystalline pure metals. *Philos. Mag.* **1954**, *45*, 823. [CrossRef]
- 52. Lewandowski, J.J.; Wang, W.H.; Greer, A.L. Intrinsic plasticity or brittleness of metallic glasses. *Philos. Mag. Lett.* **2005**, *85*, 77. [CrossRef]
- 53. Anderson, O.L. A simplified method for calculating the debye temperature from elastic constants. *J. Phys. Chem. Solids* **1963**, 24, 909–917. [CrossRef]
- 54. Panda, K.B.; Ravi, K.S. Determination of elastic constants of titanium diboride (TiB₂) from first principles using FLAPW implementation of the density functional theory. *Comput. Mater. Sci.* **2006**, *35*, 134–150. [CrossRef]
- 55. Marmier, A.; Lethbridge, Z.A.D.; Walton, R.I.; Smith, C.W.; Parker, S.C.; Evans, K.E. ElAM: A computer program for the analysis and representation of anisotropic elastic properties. *Comput. Phys. Commun.* **2010**, *181*, 2102–2115. [CrossRef]
- 56. Ma, Z.Y.; Han, Z.; Yu, X.H.; Wang, D.Y.; Tian, Y. *Pnma*-BN: Another boron nitride polymorph with interesting physical properties. *Nanomaterials* **2017**, *7*, 3. [CrossRef] [PubMed]
- 57. Fan, Q.Y.; Chai, C.C.; Wei, Q.; Yang, Y.T. The mechanical and electronic properties of carbon-rich silicon carbide. *Materials* **2016**, *9*, 333. [CrossRef] [PubMed]



© 2018 by the authors. Licensee MDPI, Basel, Switzerland. This article is an open access article distributed under the terms and conditions of the Creative Commons Attribution (CC BY) license (http://creativecommons.org/licenses/by/4.0/).