



Article Lemneolemnanes A–D, Four Uncommon Sesquiterpenoids from the Soft Coral Lemnalia sp.

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Abstract: Four undescribed sesquiterpenoids, lemneolemnanes A–D (1–4), have been isolated from the marine soft coral *Lemnalia* sp. The absolute configurations of the stereogenic carbons of 1–4 were determined by single-crystal X-ray crystallographic analysis. Compounds 1 and 2 are epimers at C-3 and have an unusual skeleton with a formyl group on C-6. Compound 3 possesses an uncommonly rearranged carbon skeleton, while 4 has a 6/5/5 tricyclic system. Compound 1 showed significant anti-Alzheimer's disease (AD) activity in a humanized *Caenorhabditis elegans* AD pathological model.

Keywords: soft coral; Lemnalia sp.; sesquiterpenoids; anti-Alzheimer's disease activity



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1. Introduction

Soft corals belonging to the genus *Lemnalia* have long been considered as a rich source of various terpenes [1], encompassing sesquiterpenes [2], diterpenes [3,4], and diterpenoid glycosides [5,6]. Sesquiterpenes, in particular, represent a prominent class of metabolites derived from *Lemnalia*, exhibiting diverse carbon skeletons, including ylangane-type [7], eremophilane-type [8], nardosinane-type [9–11], and neolemnane-type [12], among others. Typically, neolemnane sesquiterpenes have a 6/8 bicyclic skeleton, with the potential for producing rearranged skeletons [8,13]. In addition, these sesquiterpenoids with distinct skeletons exhibited a myriad of bioactivities, such as cytotoxic [7,14], anti-inflammatory [15,16], antimicrobial [17], antiviral [10], and neuroprotective activities [18]. In summary, the investigation of sesquiterpenes in soft corals of the genus *Lemnalia* is of significant interest due to their structural diversity and diverse bioactivities.

In our ongoing quest for novel bioactive metabolites from soft corals, our investigation of the *Lemnalia* sp., which was collected from the Xisha Islands, has led to the isolation of four unreported sesquiterpenoids, lemneolemnanes A–D (**1**–**4**), and three that have been previously reported: 4-acetoxy-2,8-neolemnadien-5-one (**5**) [19], paralemnolin G (**6**) [20], laevinone A (**7**) [14] (Figure 1). Compounds **1** and **2** are epimers at C-3, featuring an unusual carbon skeleton with a formyl group on C-6. Compound **3** possesses a rearranged 6/6 carbon skeleton, while **4** exhibits a rare skeleton with a 6/5/5 tricyclic system. Compound **4** is the second compound with the 6/5/5 tricyclic skeleton [13]. Herein, we report the isolation, structural elucidation, plausible biosynthetic pathways, and anti-AD activity of these compounds.



Figure 1. Structures of 1–7.

2. Results and Discussion

Compound **1** was isolated as colourless crystals. Its optical rotation value was $[\alpha]^{25}_{D}$ + 57.8 (c 1.0, MeOH). The IR absorption peaks were found at 3415, 2808, 1728, and 1600 cm^{-1} , indicating the presence of hydroxyl, aldehyde, ester carbonyl groups, and olefin, respectively. The molecular formula $C_{18}H_{26}O_4$ was determined by HRESIMS m/z 307.1902 $[M + H]^+$ (calcd for $C_{18}H_{27}O_4$, 307.1904), implying six degrees of unsaturation. The ¹³C NMR spectrum (Table 1) displayed the presence of eighteen carbon signals which, in combination with the HSQC spectrum, can be classified as one aldehyde carbonyl ($\delta_{\rm C}$ 194.0), one ester carbonyl ($\delta_{\rm C}$ 170.8), two protonated sp² ($\delta_{\rm C}$ 149.7 and 129.0), two non-protonated sp² ($\delta_{\rm C}$ 142.8 and 139.4), one oxyquaternary sp³ ($\delta_{\rm C}$ 76.8), one quaternary sp³ ($\delta_{\rm C}$ 41.1), one oxymethine sp³ (δ_C 73.8), one methine sp³ (δ_C 34.4), four methylene sp³ (δ_C 43.6, 32.2, 27.3, and 26.3), and four methyl ($\delta_{\rm C}$ 26.2, 22.8, 21.2, and 16.6) carbons. The ¹H NMR spectrum (Table 1), in conjunction with the HSQC spectrum, displayed a singlet of the aldehydric proton at δ_H 9.35/ δ_C 194.0, two olefinic protons at δ_H 6.21 (d, J = 5.7, 1.9 Hz)/ δ_C 149.7 and $\delta_{\rm H}$ 5.59 (dd, J = 5.7, 2.2 Hz)/ $\delta_{\rm C}$ 129.0, one oxymethine at $\delta_{\rm H}$ 6.51 (d, J = 5.7 Hz)/ $\delta_{\rm C}$ 73.8, one methine at $\delta_{\rm H}$ 2.47, m/ $\delta_{\rm C}$ 34.4, four pairs of methylene protons at $\delta_{\rm H}$ 3.07 (d, J =18.9 Hz) and 2.93 (d, $J = 18.9 \text{ Hz})/\delta_{\text{C}}$ 32.2, δ_{H} 2.14, m and 2.04, m/ δ_{C} 26.3, δ_{H} 1.87 (d, J = 15.8 Hz) and 1.42 (d, *J* = 15.8 Hz)/ $\delta_{\rm C}$ 43.6, and $\delta_{\rm H}$ 1.49, m/ $\delta_{\rm C}$ 27.3, one methyl doublet at $\delta_{\rm H}$ 0.98 (d, J = 6.7 Hz)/ $\delta_{\rm C}$ 16.6, and three methyl singlets at $\delta_{\rm H}$ 2.15/ $\delta_{\rm C}$ 21.2, $\delta_{\rm H}$ 1.14/ $\delta_{\rm C}$ 26.2, and $\delta_{\rm H}$ $0.87/\delta_{\rm C}$ 22.8.

The planar structure of **1** was elucidated by extensive analysis of ${}^{1}H{-}^{1}H$ COSY and HMBC spectra (Figure 2). The ${}^{1}H{-}^{1}H$ COSY correlations revealed the presence of two spin systems from H-4 to H-5 and H-9 to H₂-10, H₂-10 to H₂-11, H₂-11 to H-12, and H-12 to H₃-15, while the HMBC spectrum showed correlations from H₃-13 to C-1, C-2, C-8, and C-12; H₃-14 to C-2, C-3, and C-4; H-16 to C-5, C-6, and C-7; and H-9 to C-1, C-7, C-10, and C-11, suggesting the presence of the 6/8 neolemnane sesquiterpene scaffold. The HMBC correlations from H-4 to C-17 and H₃-18 to C-17 indicated the existence of an acetoxyl group on C-4, while the correlations from H-16 to C-5, C-6, and C-7 revealed the presence of a formyl substituent on C-6. The presence of a formyl group on a sesquiterpene skeleton is not common.

The relative configuration of **1** was established by NOESY correlations (Figure 2). In the NOESY spectrum, H-5 displayed a correlation to H-16, confirming the *E*-configuration of the $\Delta^{5,6}$ double bond. NOESY cross-peaks from H₃-13 to H₃-15, H-2a, H₃-15 to H-2a, and H₃-14 to H-2a revealed that the three methyl groups were on the same face. The absence of a NOESY correlation from H-4 to H₃-14 suggested that H-4 and H₃-14 were on the opposite side. Thus, the relative configurations at C-1, C-3, C-4, and C-12 in **1** were established as $1R^*$, $3R^*$, $4R^*$, and $12S^*$. Finally, a suitable crystal of **1** was obtained for X-ray analysis using

a diffractometer equipped with Cu K α radiation. The Ortep diagram of 1 (Figure 3) shows the absolute configurations of C-1, C-3, C-4, and C-12 1*R*, 3*R*, 4*R*, and 12*S*, which were ascertained by a flack parameter of 0.2 (7) (Table S4).

NO	1		2		
	δ_{C} Type	$\delta_{ m H}$ (J in Hz)	δ_{C} Type	$\delta_{ m H}$ (J in Hz)	
1	41.1, C		40.7, C		
2a 2b	43.6, CH ₂	1.87 d (15.8) 1.42 d (15.8)	46.2, CH ₂	1.87 d (15.1) 1.61 d (15.1)	
3	76.8 <i>,</i> C		77.0 <i>,</i> C		
4	73.8, CH	6.51 d (5.6)	73.7, CH	6.34 d (5.6)	
5	149.7, CH	6.21 dt (5.6; 1.9)	150.2, CH	6.30 dt (5.6; 1.9)	
6	142.8, C		142.6, C		
7a 7b	32.2, CH ₂	3.07 d (18.9) 2.93 d (18.9)	32.6, CH ₂	3.03 m 3.03 m	
8	139.4, C		139.1, C		
9	129.0, CH	5.59 dd (5.7; 2.2)	129.4, CH	5.62 d (3.7)	
10a 10b	26.3, CH ₂	2.14 m 204 m	26.0, CH ₂	2.13 m 203 m	
11a 11b	27.3, CH ₂	1.49 m 1.49 m	27.3, CH ₂	1.52 m 1.52 m	
12	34.4, CH	2.47 m	34.2, CH	2.00 m	
13	22.8, CH ₃	0.87 s	22.9, CH ₃	0.92 s	
14	26.2, CH ₃	1.14 s	27.2, CH ₃	1.44 s	
15	16.6, CH ₃	0.98 d (6.7)	16.5, CH ₃	0.96 d (6.7)	
16	194.0, CH	9.35 s	194.1, C	9.38 s	
17	170.8, C		170.4, C		
18	21.2, CH ₃	2.15 s	21.1, CH ₃	2.16 s	

Table 1. 1 H NMR (500 MHz) and 13 C NMR (125 MHz) data for 1–2 in CDCl₃.



Figure 2. Key HMBC correlations (red arrows), ¹H–¹H COSY correlations (thick black line), and NOESY correlations (dotted arrow) of **1–4**.



Figure 3. ORTEP diagrams of 1-4 (displacement ellipsoids are drawn at the 50% probability level).

Compound **2** was obtained as colourless crystals with an optical rotation, $[\alpha]^{25}_D$ +1.6 (*c* 1.0, MeOH). The molecular formula of **2** was determined as $C_{18}H_{26}O_4$ by HRESIMS *m/z* 307.1898 [M + H]⁺ (calcd for $C_{18}H_{27}O_4$, 307.1904), which is the same as that of **1**. The ¹H and ¹³C NMR spectra (Table 1) of **2** closely resembled those of **1**, with some chemical shift changes around the chiral C-3, suggesting that **2** could be a stereoisomer of **1**. Furthermore, the HMBC correlations and ¹H–¹H COSY cross-peaks (Figure 2) confirmed that the planar structure of **2** was the same as that of **1**.

In the NOESY spectrum (Figure 2), H-5 displayed a correlation to H-16, consistent with the *E*-configuration of the $\Delta^{5,6}$ double bond. The NOESY correlations from H₃-13 to H-2a, H₂-11, H₃-15 to H-2a, H₂-11, revealed that the two methyl groups were on the same face. In addition, the NOESY correlations from H-4 to H₃-14 and H-12 to H₃-14 indicated that H-4, H₃-14, and H-12 were on the same side. Thus, the relative configuration at C-3 in **2** was different from that in **1**, and the relative configurations at C-1, C-3, C-4, and C-12 in **2** were established as $1R^*$, $3S^*$, $4R^*$, and $12S^*$. Ultimately, the absolute configurations at C-1, C-3, C-4 and C-12 in **2** were established as 1R, 3S, 4R, and 12S [Flack parameter of 0.02(6)] (Table S5) by the single-crystal X-ray diffraction experiment (Figure 3). Therefore, **1** and **2** are epimers at C-3.

Compound **3** was isolated as colourless crystals and its molecular formula $C_{18}H_{26}O_5$ was determined by the HRESIMS m/z 345.1670 [M + Na]⁺ (calcd for $C_{18}H_{26}O_5Na$, 345.1672), requiring six degrees of unsaturation. Compound **3** was levorotatory, with $[\alpha]^{25}D - 10.8$ (c 1.0, MeOH). The ¹³C NMR spectrum (Table 2) displayed the presence of eighteen carbon signals which, in combination with the HSQC spectrum, can be classified as two ester carbonyl (δ_C 173.5 and 170.3), one protonated sp² (δ_C 132.5), one non-protonated sp² (δ_C 128.7), one oxyquaternary sp³ (δ_C 63.1), one quaternary sp³ (δ_C 37.6), two oxymethine sp³ (δ_C 72.1 and 65.0), two methine sp³ (δ_C 43.7 and 39.5), three methylene sp³ (δ_C 32.6, 26.4, and 24.6), one methoxy (δ_C 51.7) and four methyl (δ_C 20.9, 18.8, 16.0, and 15.4) carbons. The ¹H NMR spectrum (Table 2), in conjunction with the HSQC spectrum, displayed a singlet of the olefinic proton at δ_H 5.58/ δ_C 149.7, two oxymethine at δ_H 5.68 (d, J = 5.3 Hz)/ δ_C 72.1 and δ_H 3.15, m/ δ_C 65.0, two methine at δ_H 2.24, m/ δ_C 43.7 and δ_H 1.06, m/ δ_C 39.5, three pairs of methylene protons at δ_H 2.60, m/ δ_C 32.6, δ_H 2.03, m and 1.74, m/ δ_C 26.4 and δ_H 1.30, m and 1.08, m/ δ_C 24.6, one methyl doublet at δ_H 0.80 (d, J = 6.8 Hz)/ δ_C 15.4, and four methyl singlets at δ_H 3.65/ δ_C 51.7, δ_H 1.99/ δ_C 20.9, δ_H 1.66/ δ_C 18.8, and δ_H 0.92/ δ_C 16.0.

The ¹H–¹H COSY correlations (Figure 2) from H-4 to H-5, H-5 to H-6, and H-9 to H_2 -10, H_2 -10 to H_2 -11, H_2 -11 to H-12, and H-12 to H_3 -15 revealed the presence of two spin systems, and together with the HMBC correlations (Figure 2) from H_3 -13 to C-1, C-2, C-8, and C-12; H_3 -14 to C-2, C-3, and C-4; and H-9 to C-5 and C-8, enabling the construction of an adecalin scaffold. In addition, the HMBC correlations from H-4 to C-17 and H_3 -18 to C-17 indicated the existence of an acetoxy group on C-4. The presence of a carbomethoxy on C-7 was supported by the HMBC correlations from H-6 to C-7 and H_3 -16 to C-7. The established planar structure of **3** has the same decalin scaffold as other nardosinane-type

sesquiterpenoids, but differs from previously reported nardosinanes by the presence of the 2-ethoxy-2-methyl-2-oxoethyl group.

	3						
NO	$\delta_{\rm C}$ ^a Type	$\delta_{ m H}$ ^b (J in Hz)	¹ H– ¹ H COSY	НМВС	NOESY		
1	37.6, C						
2	132.5, CH	5.58 s		C-1, C-4, C-8, C-12, C-13, C-14	H ₃ -14		
3	128.7, C						
4	72.1 <i>,</i> CH	5.68 d (5.3)	H-5	C-2, C-3, C-5, C-6, C-17	H-5		
5	43.7, CH	2.24 m	H-4, H-6	C-1, C-3, C-4, C-6, C-7, C-8, C-9	H-4, H-9		
6a 6b	32.6, CH ₂	2.60 m 2.60 m	H-5	C-4, C-5, C-7, C-8			
7	173.5, C						
8	63.1, C						
9	65.0 <i>,</i> CH	3.15 m	H-10a	C-5, C-8, C-10, C-11	H-5, H-10a		
10a 10b	26.4, CH ₂	1.74 m 2.03 m	H-9, H ₂ -11	C-8, C-9, C-11, C-12	H-9, H-12		
11a 11b	24.6, CH ₂	1.30 m 1.08 m	H ₂ -10, H-12	C-1, C-9, C-10, C-12, C-15			
12	39.5 <i>,</i> CH	1.06 m	H-11, H ₃ -15	C-1, C-2, C-11, C-13, C-15	H-10a		
13	16.0, CH ₃	0.92 s		C-1, C-2, C-8, C-12	H ₃ -15		
14	18.8, CH ₃	1.66 s		C-2, C-3, C-4	H-2		
15	15.4, CH ₃	0.80 d (6.8)	H-12	C-1, C-11, C-12	H ₃ -13		
16	51.7, CH ₃	3.65 s		C-7			
17	170.3, C						
18	20.9, CH ₃	1.99 s		C-17			

Table 2. 1D and 2D NMR data for 3 in CDCl₃.

^a Recorded at 125 MHz. ^b Recorded at 500 MHz.

The relative configurations of **3** were partially established by the NOESY spectrum (Figure 2). The NOESY correlations from H-2 to H₃-14 indicated the Z-configuration of the $\Delta^{2,3}$ double bond. The correlation from H₃-13 to H₃-15 in the NOESY spectrum indicated that Me-13 and Me-15 are cofacial. And the NOESY correlations from H-5 to H-4, H-9, and H-9 to H-10a, and H-12 to H-10a, showed that H-4, H-5, H-9, and H-12 are on the same side. Only the relative configuration of C-8 remained undetermined by the NOESY correlations. Since **3** could be obtained as a suitable crystal, the absolute configurations at C-1, C-4, C-5, C-8, C-9, and C-12 in **3** were determined as 1*S*, 4*R*, 5*S*, 8*S*, 9*R*, and 12*S* by a single-crystal X-ray diffraction experiment. Figure 3 shows the Ortep diagram of **3** with a Flack parameter of 0.04 (7) (Table S6).

Compound 4 was obtained as colourless crystals and is dextrorotatory, $[\alpha]^{25}_{D}$ +181.9 (c 1.0, MeOH). The molecular formula of **3** was established as $C_{15}H_{20}O_3$ by HRESIMS m/z249.1485 $[M + H]^+$ (calcd for C₁₅H₂₁O₃, 249.1485), requiring six degrees of unsaturation. The ¹³C NMR spectrum (Table 3) displayed the presence of fifteen carbon signals which, in combination with the HSQC spectrum, can be classified as two ketone carbonyl (δ_{C} 217.8 and 211.0), one protonated sp² (δ_{C} 141.0), one non-protonated sp² (δ_{C} 137.6), one oxyquaternary sp³ (δ_C 93.6), two quaternary sp³ (δ_C 57.6 and 51.6), one methine sp³ (δ_C 39.4), four methylene sp³ (δ_{C} 43.9, 42.9, 33.7, and 30.3) and three methyl (δ_{C} 16.2, 12.1, and 11.0) carbons. The ¹H NMR spectrum (Table 3), in conjunction with the HSQC spectrum, displayed one olefinic proton at δ_H 5.73 (d, $J = 1.7 \text{ Hz})/\delta_C$ 141.0, one methine at δ_H 2.28, m/ δ_C 39.4, four pairs of methylene protons at $\delta_{\rm H}$ 2.73 (d, J = 16.1 Hz) and 2.10, m/ $\delta_{\rm C}$ 42.9, $\delta_{\rm H}$ 2.36, m and 2.22, m/ δ_C 33.7, δ_H 2.22, m and 2.13, m/ δ_C 43.9 and δ_H 2.00, m and 1.71, m/ δ_C 30.3, two methyl doublets at $\delta_{\rm H}$ 1.61 (d, J = 1.6 Hz)/ $\delta_{\rm C}$ 12.1 and $\delta_{\rm H}$ 0.95 (d, J = 6.8 Hz)/ $\delta_{\rm C}$ 16.2, one methyl singlet at $\delta_{\rm H}$ 1.04/ $\delta_{\rm C}$ 11.0, and a singlet of the hydroxyl proton $\delta_{\rm H}$ 3.20. The ¹H NMR signal at $\delta_{\rm H}$ 5.73 (d, d, J = 1.7 Hz), along with the carbon signals at $\delta_{\rm C}$ 141.0, 137.6, 217.8, and 211.0 indicated the presence of one trisubstituted double bond and two ketone groups, respectively. Thus, the remaining three degrees of unsaturation indicated the existence of a tricyclic system in 4.

C13.		
4		
H- ¹ H COSY	HMBC	NOESY
	C-1, C-3, C-4, C-8, C-13, C-14	H ₃ -13, H ₃ -15

C-5, C-7, C-8

C-1, C-4, C-5, C-6, C-8, C-9

C-1, C-4, C-7, C-8, C-10, C-11

C-1, C-10, C-12, C-15

C-1, C-2, C-10, C-11, C-13, C-15

C-1, C-2, C-8, C-12

C-2, C-3, C-4

C-1, C-11, C-12

C-4, C-5, C-8

Table 3. 1D and 2D NMR data for 4 in CDCl₃.

 $\delta_{\rm H}$ ^b (*J* in Hz)

5.73 d (1.7)

2.36 m

2.22 m

2.00 m

1.71 m

2.73 d (16.1)

2.10 m

2.22 m

2.13 m

2.28 m

 $1.04 \mathrm{s}$

1.61 d (1.6)

0.95 d (6.8)

3.20 s

1

H-6

H-7

H-12

H-11, H-15

H-12

 $\delta_{\rm C}$ ^a Type

51.6, C

141.0, CH

137.6, C

93.6. C

217.8, C

33.7, CH₂

30.3, CH₂

57.6, C

42.9, CH₂

211.0, C

43.9, CH2

39.4, CH

11.0, CH₃

12.1, CH₃

16.2, CH₃

NO.

1

2

3

4

5

6a

6b

7a

7b

8 9a

9b

10

11a

11b

12

13

14

15

OH

^a Recorded at 125 MHz. ^b Recorded at 500 MHz.

Firstly, the ¹H–¹H COSY correlations (Figure 2) from H₂-11 to H-12 and from H-12 to H₃-15 and the HMBC correlations (Figure 2) from H₃-13 to C-1, C-2, C-8 and C-12; H₃-14 to C-2, C-3, and C-4; and H-9 to C-1, C-8, C-10, and C-11 constructed a 6/5 ring system. Furthermore, the ¹H–¹H COSY correlations from H-6 to H-7, in combination with the HMBC correlations from H-6 to C-5 and C-8 and 4-OH to C-4, C-5, and C-8 indicated the presence of a cyclopentanone ring fused with the cyclohexene ring. Consequently, the planar structure of **4** was elucidated as a 6/5/5 ring system.

The NOESY spectrum (Figure 2) showed correlations from H₃-13 to H₃-15, suggesting that H₃-13 and H₃-15 are on the same face of the molecule. Moreover, the NOESY correlation between H₃-13 and H₂-7 revealed that H₃-13 and H₂-7 are on the same face. The absolute configurations of C-1, C-4, C-8, and C-12 in 4 were established as 1*S*, 4*R*, 8*S*, and 12*S* which was confirmed by a single-crystal X-ray diffraction experiment (Figure 3) with the Flack parameter of -0.1(2) (Table S7).

Based on the structural features of 1-4, we postulated that these compounds were originated from the same precursor 5 (Scheme 1). First, the methylation of 5 led to the formation of intermediate 1a. Then, 1a was converted to the intermediate 1c by the reduction of the C-5 carbonyl of 1a to the 5-hydroxyl group in 1b, followed by dehydration of **1b** to give a double bond in **1c**. The hydroxylation of Me-16 of **1c** led to the formation of a primary alcohol in 1d which, after oxidation, gave an aldehyde functionality at C-6 in 1e. Finally, hydroxylation of the C2/C3 double bond in 1e led to the formation of 1 and 2. This was due to the fact that C-3 could form a more stable carbocation (tertiary) than C-2, and nucleophilic addition would occur via the SN1 mechanism, resulting in the formation of two isomers with opposite configurations. Compound **3** originated from the intermediate 3a, a hydroxylation product of 5 at C-7. Compound 3a was oxidized at C-7 to the form 3b, which was converted to 3c by the epoxidation of C-8/C-9. The dehydrogenation of 3c at C-4 was cyclized at C-7/C-4 to form 3d. Then, 3d was converted to the lactone product 3e via Baeyer–Villiger oxidation. Compound 3e was converted to 3f by hydrolysis, followed by the dehydration and reduction of 3f to form the intermediate 3g. Finally, the methylation of **3g** led to the formation of **3**, while in the proposed biosynthetic pathway of compound **4**, the precursor **4a** was the hydroxylation product of **5** at C-10. Then, **4a** was converted to the intermediate 4b by the oxidation of C-10. The dehydrogenation of 4b at C-4 induced subsequent cyclization of C-4/C-8 which led to the formation of the intermediate 4c, which was converted to the form 4 by the deacetylation of C-4.

H-7, H₃-15 H₃-14

H₃-13



Scheme 1. Proposed biosynthetic pathways of 1-4.

Compounds **1–4** were first screened for cytotoxic and anti-inflammatory properties. However, none of the compounds showed significant anti-inflammatory activity in CuSO₄-induced transgenic fluorescent zebrafish, or cytotoxic activity. Furthermore, **1–4** were evaluated for anti-AD activity using the transgenic *Caenorhabditis elegans* model. Memantine was used as a positive control due to its ability to reduce amyloid β (A β) peptide levels in human neuroblastoma cells, as well as to inhibit A β oligomer-induced synaptic loss [21], which effectively alleviates AD-like symptoms in worms [22]. Surprisingly, at a concentration of 100 μ M/L, **1** significantly alleviated AD-like symptoms in the worm model (p < 0.05) (Figure 4). This result suggests that **1** has the potential to be a candidate for the development of an anti-AD agent based on its observed biological activity.



Figure 4. Anti-Alzheimer's disease (AD) activity of **1** (worms were treated with 100 μ M/L of **1**, 100 μ M/L memantine was used as a positive control, and 0.1% dimethyl sulfoxide was used as a negative control).

3. Materials and Methods

3.1. General Experimental Procedures

Melting points were measured on a microscopic melting point apparatus (Shanghai Zhuoguang Technology Company Limited, Shanghai, China). Optical rotations were measured on a Jasco P-1020 digital polarimeter (Jasco, Tokyo, Japan). The UV spectra were recorded on a Beckman DU640 spectrophotometer (Beckman Ltd., Shanghai, China). CD spectra were obtained on a Jasco J-810 spectropolarimeter (Jasco, Tokyo, Japan). IR (KBr) spectra were taken on a Nicolet NEXUS 470 spectrophotometer in KBr discs (Thermo Scientific, Beijing, China). NMR spectra were recorded using the Agilent 500 MHz (Agilent, Beijing, China), (¹H, 500 MHz; ¹³C, 125 MHz) or the Bruker AVANCE NEO 400 (¹H, 400 MHz; ¹³C, 100 MHz), (Bruker, Faellanden, Switzerland). The 7.26 ppm and 77.16 ppm resonances of CDCl₃ were used as internal references for the ¹H and ¹³C NMR spectra, respectively. HRESIMS spectra were measured on Micromass Q-Tof Ultima GLOBAL GAA076LC mass spectrometers (Autospec-Ultima-TOF, Waters, Shanghai, China). The crystallographic data were obtained on a Bruker APEX-II CCD diffractometer (Bruker, Beijing, China) equipped with graphite-monochromatized Cu K α radiation. Semi-preparative HPLC was performed using a Waters 1525 pump (Waters, Singapore) equipped with a 2998 photodiode array detector and a SilGreen C₁₈ column (SilGreen, 10×250 mm, 5μ m). Silica gel (200-300 mesh and 300-400 mesh), (Qingdao Marine Chemical Factory, Qingdao, China) was used for column chromatography. Chiral HPLC analysis and resolution were conducted on a chiral analytical column (Daicel, Shanghai, China).

3.2. Animal Material

The soft coral *Lemnalia* sp. was collected from Xisha Island (Yagong Island) of the South China Sea in 2014, and was frozen immediately after collection. The specimen was identified by Prof Ping-Jyun Sung, Institute of Marine Biotechnology, National Museum of Marine Biology & Aquarium, Pingtung 944, Taiwan. The voucher specimen (No. xs-yg-12) was deposited at the State Key Laboratory of Marine Drugs, Ocean University of China, People's Republic of China.

3.3. Extraction and Isolation

A frozen specimen of Lemnalia sp. (7.20 kg, wet weight) was thawed, homogenized, and then exhaustively extracted with CH₃OH six times (3 days each time) at room temperature. The combined solutions were concentrated in vacuo and were subsequently desalted by redissolving in CH_3OH to yield a residue (175.18 g). The crude extract was subjected to silica gel vacuum column chromatography and eluted with a gradient of petroleum ether/acetone (100:1 to 1:1, v/v) and subsequently eluted with a gradient of CH₂Cl₂/MeOH (10:1 to 1:1, v/v) to obtain sixteen fractions (Frs.1–16). Each fraction was detected by TLC. Fr.2 was subjected to silica gel vacuum column chromatography and eluted with petroleum ether/acetone, from 100:1 to 1:1, v/v, to give three subfractions Sfrs.2.1–2.3. Sfr.2.2 was purified by reversed-phase HPLC (OD-H, MeOH/H₂O = 80/20, flow rate = 1.5 mL/min, I = 210 nm) to afford 5 (28.8 mg, t_R = 36 min). Fr.4 was subjected to silica gel vacuum column chromatography and eluted with petroleum ether/acetone, from 80:1 to 1:1, v/v, to give seven subfractions Sfrs.4.1–4.7. Sfr.4.5 was purified by reversed-phase HPLC (OD-H, MeOH/H₂O = 70/30, flow rate = 1.5 mL/min, I = 210 nm) to afford mixtures of **3** and **6** (24.6 mg, $t_R = 42$ min). The mixture of **3** was purified with a chiral column (Daicel Chiral pack IC, n-hexane/isopropanol = 60:40, flow rate = 1.0 mL/min, I = 210 nm) to give 3 (11.9 mg, $t_R = 36$ min). Fr.7 was subjected to silica gel vacuum column chromatography and eluted with petroleum ether/acetone, from 50:1 to 1:1, v/v, to give five subfractions Sfrs.7.1–7.5. Sfr.7.2 was purified by reversed-phase HPLC (OD-H, MeCN/H₂O = 50/50, flow rate = 1.5 mL/min, I = 320 nm) to afford **1** (7.1 mg, t_R = 72 min). Fr.9 was subjected to silica gel vacuum column chromatography and eluted with petroleum ether/acetone, from 30:1 to 1:1, v/v, to give four subfractions Sfrs.9.1–9.4. Sfr.9.4 was purified by reversed-phase HPLC (OD-H, MeOH/H₂O = 60/40, flow rate = 1.5 mL/min, I = 210 nm) to afford 7

(19.7 mg, $t_R = 45$ min). Fr.11 was subjected to silica gel vacuum column chromatography and eluted with petroleum ether/acetone, from 20:1 to 1:1, v/v, to give six subfractions Sfrs.11.1–11.6. Sfr.11.1 was purified by reversed-phase HPLC (OD-H, MeCN/H₂O = 30/70, flow rate = 1.5 mL/min, I = 210 nm) to afford 4 (20.7 mg, $t_R = 57$ min). Sfr.11.2 was purified by reversed-phase HPLC (OD-H, MeCN/H₂O = 50/50, flow rate = 1.5 mL/min, I = 320 nm) to afford 2 (7.3 mg, $t_R = 45$ min).

Lemneolemnane A (1): colourless crystals, mp 157.9–159.5 °C; $[\alpha]^{25}_{D}$ +57.8 (*c* 1.0, MeOH); CD (*c* 1.0, MeOH) = $\Delta \varepsilon 197$ + 51.50, $\Delta \varepsilon 226$ –50.33, $\Delta \varepsilon 260$ + 17.29; UV (MeOH) λ_{max} (log ε) = 196 (1.45) nm, λ_{max} (log ε) = 213(0.91) nm, λ_{max} (log ε) = 230(1.42) nm; IR (KBr) ν_{max} = 3415, 2808, 1728, 1600 cm⁻¹; HRESIMS *m*/*z* 307.1902 [M + H]⁺ (calcd for C₁₈H₂₇O₄, 307.1904). For ¹H NMR and ¹³C NMR data, see Table 1.

Lemneolemnane B (2): colourless crystals, mp 186.5–188.3 °C; $[\alpha]^{25}_{\rm D}$ +1.6 (*c* 1.0, MeOH); CD (*c* 0.5, MeOH) = $\Delta \varepsilon 199 + 70.30$, $\Delta \varepsilon 225-83.68$, $\Delta \varepsilon 255 + 17.48$; UV (MeOH) $\lambda_{\rm max}$ (log ε) = 196 (1.62) nm, $\lambda_{\rm max}$ (log ε) = 212(1.16) nm, $\lambda_{\rm max}$ (log ε) = 226(1.66) nm; IR (KBr) $\nu_{\rm max}$ = 3431, 2831, 1726, 1599 cm⁻¹; HRESIMS *m*/*z* 307.1898 [M + H]⁺ (calcd for C₁₈H₂₇O₄, 307.1904). For ¹H NMR and ¹³C NMR data, see Table 1.

Lemneolemnane C (3): colourless crystals, mp 135.6–137.9 °C; $[\alpha]^{25}_{D}$ –10.8 (*c* 1.0, MeOH); CD (*c* 0.5, MeOH) = $\Delta \epsilon 201 + 18.60$; UV (MeOH) λ_{max} (log ϵ) = 195 (1.99) nm, λ_{max} (log ϵ) = 193(0.96) nm; IR (KBr) ν_{max} = 1641 cm⁻¹; HRESIMS *m*/*z* 323.1854 [M + H]⁺ (calcd for C₁₈H₂₇O₅, 323.1853) and 345.1670 [M + Na]⁺ (calcd. for C₁₈H₂₆O₅Na, 345.1672). For ¹H NMR and ¹³C NMR data, see Table 2.

Lemneolemnane D (4): colourless crystals, mp 176.3–177.8 °C; $[\alpha]^{25}_{D}$ +181.9 (*c* 1.0, MeOH); CD (*c* 0.5, MeOH) = $\Delta \epsilon 198 + 43.31$, $\Delta \epsilon 231-35.02$, $\Delta \epsilon 309 + 72.61$; UV (MeOH) λ_{max} (log ϵ) = 193 (1.15) nm, λ_{max} (log ϵ) = 208 (0.78) nm, λ_{max} (log ϵ) = 218 (0.80) nm; IR (KBr) ν_{max} = 3415, 1716 cm⁻¹; HRESIMS *m*/*z* 249.1485 [M + H]⁺ (calcd for C₁₅H₂₁O₃, 249.1485) and 266.1751 [M + NH₄]⁺ (calcd. for C₁₅H₂₄O₃ N, 266.1751). For ¹H NMR and ¹³C NMR data, see Table 3.

3.4. Anti-Alzheimer's Disease Activity

The transgenic *Caenorhabditis elegans* strain CL4176 with genotype smg-1ts (myo-3/A β_{1-42} long 3'-untranslated region (UTR)) was obtained from the *Caenorhabditis* Genetics Center (CGC) (University of Minnesota, Minneapolis, MN, USA). Worms were routinely cultured on nematode growth medium (NGM) plates for 72 h to grow to the L3 stage at 16 °C, using *Escherichia coli* OP50 as the standard food resource. Throughout the bioactivity assays, 60~80 worm eggs were placed on NGM containing 100 μ M/L of the test compounds, 100 μ M/L memantine as a positive control, and 0.1% dimethyl sulfoxide solvent as a negative control. When the worms grew into L3 larvae, they were then transferred into another incubator which was set at 25 °C for another 32 h. Paralysed worms were observed and counted under a dissecting microscope every 2 h until all worms were paralysed. The anti-AD activity of the tested compound was indicated by the percentage of individual paralyzed worms in the tested worm population throughout the whole experiment in contrast with the negative control group. For the anti-AD activity assay, a log-rank survival test was used to compare the significance among treatments. *p* values at 0.05 or lower were considered statistically significant.

3.5. X-ray Crystallographic Analysis

X-ray crystallographic data were collected on a Bruker APEX-II CCD diffractometer. The crystal was kept at 150.0 K or 293.0 K during data collection. Using Olex2, the structures were solved with the SHELXT structure solution program using intrinsic phasing and refined with the SHELXL refinement package using least squares minimization.

4. Conclusions

Soft corals belonging to the genus *Lemnalia* have been recognized as a rich source of various bioactive secondary metabolites. Our continuing chemical investigation of the

Xisha soft coral *Lemnalia* sp. resulted in the isolation of four unreported sesquiterpenoids, lemneolemnanes A–D (1–4), together with three previously described sesquiterpenoids **5**–7. Compounds **1** and **2** are epimers at C-3 that have an unusual skeleton with a formyl group on C-6. Based on the proposed biosynthetic pathways, **1** and **2** could be artifacts originating from a hydration of the C-2 and C-3 double bond of the octacyclic ring of the intermediate after formylation. Compound **3** features a decalin scaffold and a 2-ethoxy-2-methyl-2-oxoethyl group, while **4** possesses a 6/5/5 tricyclic system. Based on the structural characterization of lemneolemnanes A–D (**1**–4), we hypothesized their possible biosynthetic pathways, originating from **5**. Compound **1** showed significant anti-Alzheimer's disease (AD) activity in a *Caenorhabditis elegans* model. The above research enriches chemical libraries of the soft corals *Lemnalia* sp. and provides some references in further exploration of potential biological activity. Meanwhile, the discovery of these rare molecular structures will provide new ideas.

Supplementary Materials: The following supporting information can be downloaded at: https://www.mdpi.com/article/10.3390/md22040145/s1, Tables S1–S4: NMR data of 1–4; Tables S5–S8 and Figures S1–S4: X-ray crystallographic analyses of 1–4; Figures S5–S42: NMR spectra of 1–7.

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