



# Article Meirols A–C: Bioactive Catecholic Compounds from the Marine-Derived Fungus *Meira* sp. 1210CH-42

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**Abstract:** Three new catecholic compounds, named meirols A–C (**2**–4), and one known analog, argovin (**1**), were isolated from the marine-derived fungus *Meira* sp. 1210CH-42. Their structures were determined by extensive analysis of 1D, 2D NMR, and HR-ESIMS spectroscopic data. Their absolute configurations were elucidated based on ECD calculations. All the compounds exhibited strong antioxidant capabilities with EC<sub>50</sub> values ranging from 6.01 to 7.47  $\mu$ M (ascorbic acid, EC<sub>50</sub> = 7.81  $\mu$ M), as demonstrated by DPPH radical scavenging activity assays. In the  $\alpha$ -glucosidase inhibition assay, **1** and **2** showed potent in vitro inhibitory activity with IC<sub>50</sub> values of 184.50 and 199.70  $\mu$ M, respectively (acarbose, IC<sub>50</sub> = 301.93  $\mu$ M). Although none of the isolated compounds exhibited cytotoxicity against one normal and six solid cancer cell lines, **1** exhibited moderate cytotoxicity against the NALM6 and RPMI-8402 blood cancer cell lines with GI<sub>50</sub> values of 9.48 and 21.00  $\mu$ M, respectively. Compound **2** also demonstrated weak cytotoxicity against the NALM6 blood cancer cell line with a GI<sub>50</sub> value of 29.40  $\mu$ M.

**Keywords:** marine fungus; natural product; *Meira* sp.; indanone; catechol; meirol; DPPH radical scavenging;  $\alpha$ -glucosidase inhibitor; cytotoxicity

# 1. Introduction

Small-molecule natural products, such as catechol and  $\beta$ -lactam, are widely utilized in the drug discovery process in various ways [1,2]. Because of their low weight, smallmolecule drugs can traverse cell membranes, interact with proteins and enzymes within cells, and disrupt specific processes. Alternatively, a biologically active natural product may inspire the discovery of clinically useful drug agents by offering insights into types of structural features that may prove valuable. Catechols, benzene derivatives containing a 3,4-dihydroxyphenyl group, have recently been discovered as diverse derivatives in microorganisms, plants, insects, and marine organisms [3]. Their ubiquity can be attributed to their rich redox chemistry, ability to cross-link through complex and irreversible oxidation mechanisms, excellent chelating properties, and the various ways in which vicinal hydroxyl groups interact with surfaces of significantly diverse chemical and physical characteristics [4,5]. Aligned with their structural variety, catecholamines have been reported to exhibit a broad spectrum of biological activities, encompassing antiparasitic, antibacterial, metal-chelating, anti-inflammatory, immune-modulation, wound-healing, antioxidant,



Citation: Lee, M.A.; Kang, J.S.; Yang, J.-W.; Lee, H.-S.; Heo, C.-S.; Park, S.J.; Shin, H.J. Meirols A–C: Bioactive Catecholic Compounds from the Marine-Derived Fungus *Meira* sp. 1210CH-42. *Mar. Drugs* **2024**, *22*, 87. https://doi.org/10.3390/md22020087

Academic Editor: Tim Bugni

Received: 23 January 2024 Revised: 6 February 2024 Accepted: 13 February 2024 Published: 14 February 2024



**Copyright:** © 2024 by the authors. Licensee MDPI, Basel, Switzerland. This article is an open access article distributed under the terms and conditions of the Creative Commons Attribution (CC BY) license (https:// creativecommons.org/licenses/by/ 4.0/). neuroprotective, nephroprotective, and metabolic regulation activities [3]. Hence, novel catechol compounds derived from natural sources will serve as a promising foundation for broadening their potential biological and medical applications.

The novel basidiomycetous fungus *Meira* was first isolated from citrus leaves and mite cadavers in 2003 and assigned to *M. geulakonigii* and *M. argovae* [6]. Six species in the genus *Meira* have been reported to date, namely, *M. argovae*, *M. geulakonigii* (Boekhout et al. 2003), *M. nashicola* (Yasuda et al. 2006), *M. miltonrushii* (Rush et al. 2013), *M. siamensis* (Limtong et al. 2017), and *M. nicotianae* (Yichen et al. 2018) [6–9]. These species have been isolated from plant tissues, fruits, and mites. These *Meira* species exhibit promising antagonistic activities against a wide spectrum of fungal and bacterial plant diseases [10]. However, only one compound, argovin (1), has been reported from the genus *Meira* [11]. Thus, information about the active constituents responsible for the various biological activities of *Meira* sp. is still limited.

In 2023, the first marine-derived *Meira* sp. was described in our previous report as *Meira* sp. 1210CH-42 [12]. We identified new thiolactones and steroids from the *Meira* sp. strain, which exhibited  $\alpha$ -glucosidase inhibitory activities [12]. In our ongoing efforts to find new bioactive small compounds, we have paid attention to the fungus *Meira* because the genus has not been studied well, and the strain was the first marine-derived species showing good bioactivities in our preliminary screening. Therefore, a mass culture was conducted to isolate bioactive secondary metabolites from the strain. As a result, we discovered novel bioactive catecholic compounds from the culture extract of the fungus *Meira* sp. 1210CH-42. Herein, we describe the structure determination of catecholic compounds (1–4), along with their antioxidant,  $\alpha$ -glucosidase inhibitory, cytotoxic, tyrosinase inhibitory, and antimicrobial activities.

## 2. Results and Discussion

## 2.1. Structure Elucidation of New Compounds

Compound **1** was isolated as a white amorphous powder, and its molecular formula was determined to be C<sub>9</sub>H<sub>8</sub>O<sub>3</sub> based on an HR-ESIMS analysis at m/z 165.0552 [M + H]<sup>+</sup> (calcd for C<sub>9</sub>H<sub>9</sub>O<sub>3</sub><sup>+</sup>, 165.0552), with six degrees of unsaturation. The <sup>1</sup>H and <sup>13</sup>C NMR data of **1** are summarized in Tables 1 and 2. The <sup>1</sup>H NMR spectrum of **1** in CD<sub>3</sub>OD revealed two olefinic protons ( $\delta_H$  7.15 and 6.84) and two methylene protons ( $\delta_H$  3.01 and 2.63). The <sup>13</sup>C NMR and HSQC spectra showed the presence of nine signals, including one carbonyl carbon ( $\delta_C$  209.1), two oxygen-bearing sp<sup>2</sup> ( $\delta_C$  153.1 and 143.1), two non-protonated sp<sup>2</sup> ( $\delta_C$  144.7 and 130.8), two protonated sp<sup>2</sup> ( $\delta_C$  117.2 and 116.9), and two methylene ( $\delta_C$  37.6 and 23.3) carbons. The structure of **1** was identified as argovin (4,5-dihydroxyindan-1-one) by an analysis of its NMR and HRMS data and a comparison of its spectroscopic data with those previously reported in the literature (Figure 1) [11].

Compound **2** was purified as a white amorphous powder. The molecular formula of **2** was determined to be C<sub>8</sub>H<sub>7</sub>NO<sub>3</sub> by HR-ESIMS analysis at m/z 188.0324 [M + Na]<sup>+</sup> (calcd for C<sub>8</sub>H<sub>7</sub>NO<sub>3</sub>Na<sup>+</sup>, 188.0324), which was determined to possess six degrees of unsaturation. The <sup>1</sup>H NMR spectrum of **2** displayed the signals for two olefinic protons ( $\delta_{\rm H}$  7.19 and 6.90) and two methylene protons ( $\delta_{\rm H}$  4.34, overlapped) (Table 1). The <sup>13</sup>C NMR and HSQC spectra exhibited one carbonyl ( $\delta_{\rm C}$  174.5), two oxygen-bearing sp<sup>2</sup> ( $\delta_{\rm C}$  150.4 and 141.6), two non-protonated sp<sup>2</sup> ( $\delta_{\rm C}$  132.4 and 125.2), two protonated sp<sup>2</sup> ( $\delta_{\rm C}$  116.9 and 116.5), and one methylene ( $\delta_{\rm C}$  44.5) carbons (Table 2). The <sup>1</sup>H-<sup>1</sup>H COSY correlation of H-6 ( $\delta_{\rm H}$  6.90)/H-7 ( $\delta_{\rm H}$  7.16) revealed a pair of ortho-coupled aromatic protons at  $\delta_{\rm H}$  6.90, d (*J* = 8.0 Hz,  $\delta_{\rm C}$  116.9)/ $\delta_{\rm H}$  7.16, d (*J* = 8.0 Hz,  $\delta_{\rm C}$  116.5). Furthermore, the HMBC correlations, from H-3 ( $\delta_{\rm H}$  4.34) to C-1 ( $\delta_{\rm C}$  174.5)/C-3a ( $\delta_{\rm C}$  132.4)/C-4 ( $\delta_{\rm C}$  141.6)/C-7a ( $\delta_{\rm C}$  125.2); from H-6 ( $\delta_{\rm H}$  6.90) to C-4 ( $\delta_{\rm C}$  141.6)/C-7a ( $\delta_{\rm C}$  125.2); and from H-7 ( $\delta_{\rm H}$  7.16) to C-1 ( $\delta_{\rm C}$  174.5)/C-3a ( $\delta_{\rm C}$  132.4), s) with a carbonyl carbon C-1 ( $\delta_{\rm C}$  174.5)/C-3a ( $\delta_{\rm C}$  132.4), s) with a carbonyl carbon C-1 ( $\delta_{\rm C}$  174.5)/C-3a ( $\delta_{\rm H}$  4.34, s) with a carbonyl carbon C-1 ( $\delta_{\rm C}$  174.5)/C-3a ( $\delta_{\rm H}$  4.34, s) with a carbonyl carbon C-1 ( $\delta_{\rm C}$  174.5)/C-3a ( $\delta_{\rm H}$  4.34, s) with a carbonyl carbon C-1 ( $\delta_{\rm C}$  174.5)/C-3a ( $\delta_{\rm H}$  4.34, s) with a carbonyl carbon C-1 ( $\delta_{\rm C}$  174.5)/C-3a ( $\delta_{\rm H}$  4.34, s) with a carbonyl carbon C-1 ( $\delta_{\rm C}$  174.5)/C-3a ( $\delta_{\rm H}$  4.34, s) with a carbonyl carbon C-1 ( $\delta_{\rm C}$  174.5)/C-3a ( $\delta_{\rm H}$  4.34, s) with a carbonyl carbon C-1 ( $\delta_{\rm C}$  174.5)/C-3a ( $\delta_{\rm H}$  4.34, s) with a carbonyl carbon C-1 ( $\delta_{\rm C}$  174.5)/C-3a ( $\delta_{\rm H}$  4.34, s) with a carbonyl carbon C-1 ( $\delta_{\rm C}$  174.5)/C-3a ( $\delta_{\rm H}$  4.34, s) with

the structure of **2** was elucidated as a previously unreported catecholic compound, 4,5dihydroxyisoindolin-1-one, and **2** was named meirol A.

Compound **3** was obtained as a purple amorphous powder. The molecular formula of **3** was analyzed for  $C_9H_8O_4$  on the basis of its parent ion in the HR-ESIMS analysis at m/z 203.0321 [M + Na]<sup>+</sup> (calcd for C<sub>9</sub>H<sub>8</sub>O<sub>4</sub>Na<sup>+</sup>, 203.0320), which required six degrees of unsaturation. The <sup>1</sup>H NMR spectrum of **3** displayed signals of two olefinic protons ( $\delta_{\rm H}$  7.14 and 6.91), one oxygenated methine proton ( $\delta_H$  5.48), and two methylene protons ( $\delta_H$  3.02 and 2.47) (Table 1). In total, nine carbon resonances were observed in the  $^{13}$ C NMR spectrum of **3**. These resonances were assigned to one carbonyl ( $\delta_C$  205.6), two oxygen-bearing sp<sup>2</sup> ( $\delta_{\rm C}$  154.2 and 130.0), two non-protonated sp<sup>2</sup> ( $\delta_{\rm C}$  144.1 and 143.5), two protonated sp<sup>2</sup> ( $\delta_{\rm C}$ 118.6 and 116.7), one oxymethine ( $\delta_{\rm C}$  66.9), and one methylene ( $\delta_{\rm C}$  48.4) carbons, with the assistance of the HSQC spectrum (Table 2). Analysis of the <sup>1</sup>H-<sup>1</sup>H COSY spectrum led to the construction of an ortho-coupled (J = 8.1) isolated two-proton spin system illustrating homonuclear coupling correlation between H-6 ( $\delta_H$  6.91, d, J = 8.1) and H-7 ( $\delta_H$  7.14, d, J = 8.1). The HMBC correlations from H-6 ( $\delta_{\rm H}$  6.91) to C-4 ( $\delta_{\rm C}$  130.0)/C-5 ( $\delta_{\rm C}$  154.2)/C-7a  $(\delta_{\rm C}$  144.1) and from H-7 ( $\delta_{\rm H}$  7.14) to C-5 ( $\delta_{\rm C}$  154.2)/C-7a ( $\delta_{\rm C}$  144.1) confirmed the position of the spin system within an aromatic ring fragment. Additionally, the HMBC correlations from H-2 ( $\delta_{\rm H}$  3.02 and 2.47) to C-1 ( $\delta_{\rm C}$  205.6)/C-3 ( $\delta_{\rm C}$  66.9)/C-7a ( $\delta_{\rm C}$  144.1); from H-3  $(\delta_{\rm H} 5.48)$  to C-1 ( $\delta_{\rm C} 205.6$ )/C-3a ( $\delta_{\rm C} 143.5$ )/C-4 ( $\delta_{\rm C} 130.0$ ); and from H-7 ( $\delta_{\rm H} 7.14$ ) to C-1  $(\delta_{C}$  205.6) allowed for the assembly of a 1-indanone ring system. Furthermore, the presence of a hydroxy group located at C-3 ( $\delta_{\rm C}$  66.9) was elucidated through the HMBC correlations from H-3 ( $\delta_{\rm H}$  5.48) to C-1 ( $\delta_{\rm C}$  205.6)/C-3a ( $\delta_{\rm C}$  143.5)/C-4 ( $\delta_{\rm C}$  130.0). Therefore, the planar structure of **3** was elucidated, as shown in Figure 2. To determine the absolute configuration of 3, the theoretical electronic circular dichroism (ECD) spectra of 3 and its enantiomer were calculated (Figure 2). The experimental ECD spectrum of **3** showed good agreement with the calculated ECD spectrum of (3R)-3 (Figure 3). Thus, the absolute configuration of **3** was determined as 3*R*, and the new indanone derivative, **3**, was given the name meirol B.

Compound 4 was isolated as a light purple amorphous powder. The molecular formula of 4 was determined to be identical to that of 3 ( $C_9H_8O_4$ ), as evidenced by the ion at m/z 203.0322 [M + Na]<sup>+</sup> (calcd for C<sub>9</sub>H<sub>8</sub>O<sub>4</sub>Na<sup>+</sup>, 203.0320) from the HR-ESIMS analysis, implying six degrees of unsaturation. However, its <sup>1</sup>H and <sup>13</sup>C NMR spectra showed different chemical shifts compared with **3**. The <sup>1</sup>H NMR spectrum of **4** (Table 1) showed the signals for two olefinic protons ( $\delta_H$  7.17 and 6.87), one oxygenated methine proton ( $\delta_{\rm H}$  4.41), and two methylene protons ( $\delta_{\rm H}$  3.51 and 2.70). The <sup>13</sup>C NMR and HSQC spectra of 4 exhibited the presence of nine carbon resonances, including one carbonyl ( $\delta_{C}$  207.0), two oxygen-bearing sp<sup>2</sup> ( $\delta_{C}$  153.8 and 128.1), two non-protonated sp<sup>2</sup> ( $\delta_{C}$  143.0 and 139.8), two protonated sp<sup>2</sup> ( $\delta_{C}$  118.0 and 117.3), one oxymethine ( $\delta_{C}$  75.0), and one methylene ( $\delta_C$  32.2) carbons (Table 2). The <sup>1</sup>H-<sup>1</sup>H COSY correlation of H-2 ( $\delta_H$  4.41)/H-3 ( $\delta_H$  3.51 and 2.70) and H-6 ( $\delta_H$  6.87)/H-7 ( $\delta_H$  7.17) and the strong HMBC correlations from H-2 ( $\delta_H$  4.41) to C-1 ( $\delta_C$  207.0)/C-3a ( $\delta_C$  139.8) and from H-3 ( $\delta_H$  3.51 and 2.70) to C-1 ( $\delta_C$  207.0)/C-2 ( $\delta_C$  75.0)/C-3a ( $\delta_C$  139.8)/C-4 ( $\delta_C$  128.1)/C-7a ( $\delta_C$  143.0) indicated that a hydroxy group was located at C-2 ( $\delta_C$  75.0), and two oxygenated quaternary carbons were located at C-4 ( $\delta_C$  128.1) and C-5 ( $\delta_C$  153.8). In addition, the presence of a 1-indanone system was supported by the <sup>1</sup>H-<sup>1</sup>H COSY correlation between H-6 ( $\delta_{\rm H}$  6.87, d, J = 8.1) and H-7 ( $\delta_H$  7.17, d, J = 8.1), as well as the HMBC correlations from H-6 ( $\delta_H$  6.87) to C-4  $(\delta_{\rm C} \ 128.1)/{\rm C}$ -5 ( $\delta_{\rm C} \ 153.8$ )/C-7a ( $\delta_{\rm C} \ 143.0$ ) and from H-7 ( $\delta_{\rm H} \ 7.17$ ) to C-1 ( $\delta_{\rm C} \ 207.0$ )/C-3a  $(\delta_C 139.8)/C-5$  ( $\delta_C 153.8$ ). Based on these data analyses, the planar structure of 4 was determined to be a new positional isomer of **3** (Figure 2). The absolute configuration of 4 was defined by comparing its calculated and experimental ECD spectra. As shown in Figure 3, the calculated ECD spectrum of 2S-4 closely matched the experimental spectrum, confirming the 2S absolute configuration for 4. Therefore, the structure of the new natural product, 4, was clearly determined, and 4 was named meirol C.



Figure 1. Structures of 1–4 from the marine fungus strain *Meira* sp. 1210CH-42.

1 2 3 4 Position  $\delta_{\rm H}$ , Mult (J in Hz)  $\delta_{\rm H}$ , Mult (J in Hz)  $\delta_{\rm H}$ , Mult (J in Hz)  $\delta_{\rm H}$ , Mult (J in Hz) 2.47, d (18.6) 2 2.63, t (5.6) 4.41, dd (7.7, 4.5) 3.02, dd (18.6, 6.6) 2.70, dd (16.7, 4.5) 3 3.01, t (5.6) 4.34, s 5.48, d (6.6) 3.51, dd (16.7, 7.7) 6 6.84, d (8.0) 6.90, d (8.0) 6.91, d (8.0) 6.87, d (8.1) 7 7.15, d (8.0) 7.16, d (8.0) 7.17, d (8.1) 7.14, d (8.0)

**Table 1.** <sup>1</sup>H NMR data of 1-4 (600 MHz for <sup>1</sup>H in CD<sub>3</sub>OD).

**Table 2.** <sup>13</sup>C NMR data of 1–4 (150 MHz for <sup>13</sup>C, in CD<sub>3</sub>OD).

Position	1	2	3	4
	δ <sub>C</sub> , Type			
1	209.1, C	174.5 <i>,</i> C	205.6, C	207.0, C
2	37.6, CH <sub>2</sub>	NH	48.4, CH <sub>2</sub>	75.0, CH
3	23.3, CH <sub>2</sub>	44.5, CH <sub>2</sub>	66.9, CH	32.2, CH <sub>2</sub>
3a	144.7, C	132.4, C	143.5, C	139.8, C
4	143.1, C	141.6, C	130.0, C	128.1, C
5	153.1 <i>,</i> C	150.4 C	154.2, C	153.8, C
6	116.9, CH	116.9, CH	118.6, CH	117.3, CH
7	117.2, CH	116.5, CH	116.7, CH	118.0, CH
7a	130.8, C	125.2, C	144.1 <i>,</i> C	143.0, C



Figure 2. Key <sup>1</sup>H-<sup>1</sup>H COSY and HMBC correlations of 1–4.



Figure 3. Experimental and calculated ECD spectra of 3 and 4.

#### 2.2. Bioactivity Evaluation of Compounds

The antioxidant activities of 1–4 were assessed using the 1,1-diphenyl-2-picrylhydrazyl (DPPH) radical scavenging assay. As indicated in Table 3, 1–4 exhibited considerable free radical scavenging activities with EC<sub>50</sub> values ranging from 6.01  $\pm$  0.07 to 7.47  $\pm$  0.13 µM, showing better activities than the positive control, ascorbic acid (EC<sub>50</sub> = 7.81  $\pm$  0.25 µM). Also, 1–4 were evaluated for  $\alpha$ -glucosidase inhibitory activities (Table 3). Compounds 1 and 2 exhibited significant inhibitory effects with IC<sub>50</sub> values of 184.50  $\pm$  2.93 and 199.70  $\pm$  1.87 µM, respectively. Meanwhile, 3 showed weaker activity (IC<sub>50</sub> = 367.43  $\pm$  3.01 µM) than the positive control, acarbose (IC<sub>50</sub> = 301.93  $\pm$  3.55 µM). A structure–activity relationship analysis of 1–4 indicated that the hydroxy groups at C-2 or C-3 in diol-indanone impacted their antioxidant properties but did not have a significant impact on their  $\alpha$ -glucosidase inhibitory activity. Furthermore, all the compounds were screened for their cytotoxic activity against six solid and seven blood cancer cell lines (Table 3). Compounds 1 and 2 showed selective cytotoxicity against two out of the seven blood cancer cell lines (HL-60, acute myelogenous

leukemia; Raji, Burkitt's lymphoma; K562, chronic myelogenous leukemia; RPMI-8402, T cell acute lymphocytic leukemia; NALM6, B cell acute lymphocytic leukemia; U266, multiple myeloma; WSU-DLCL2, diffuse large B cell lymphoma). Among the tested compounds, only 1 showed weak cytotoxicity (GI<sub>50</sub> =  $21.00 \pm 0.47 \mu$ M) against the RPMI-8402 cell line. Compounds 1 and 2 exhibited selective cytotoxicity against the NALM6 blood cancer cell line (GI<sub>50</sub> =  $9.47 \pm 0.41$  and  $29.55 \pm 2.27 \mu$ M, respectively), while the other compounds did not demonstrate significant cytotoxicity (GI<sub>50</sub> >  $30 \mu$ M). Additionally, all compounds showed no cytotoxicity against one normal cell (MRC-9) and six solid cancer cell lines (PC-3, prostate; NCI-H23, lung; HCT-15, colon; NUGC-3, stomach; ACHN, renal; MDA-MB-231, breast) even at a concentration of  $30 \mu$ M.

Compound	DPPH Radical Scavenging Activity (EC <sub>50</sub> $\pm$ SD, $\mu$ M) <sup>a</sup>	α-Glucosidase Inhibitory Activity (IC <sub>50</sub> $\pm$ SD, $\mu$ M) <sup>b</sup>	Cytotoxic Activity (GI <sub>50</sub> $\pm$ SD, $\mu$ M) <sup>c</sup>	
			<b>RPMI-8402</b> <sup><i>d</i></sup>	NALM6 <sup>d</sup>
1	$7.47\pm0.13$	$184.50\pm2.93$	$21.00\pm0.47$	$9.47\pm0.41$
2	$6.82\pm0.06$	$199.70\pm1.87$	>30	$29.55\pm2.27$
3	$6.01\pm0.07$	$367.43\pm3.01$	>30	>30
4	$6.20\pm0.12$	>555.00	>30	>30
Ascorbic acid	$7.81 \pm 0.25$	-	_	-
Acarbose	_	$301.93\pm3.55$	_	-
Doxorubicin	-	-	$0.015 \pm 0.13  imes 10^{-2}$	$0.003 \pm 0.06  imes 10^{-2}$

**Table 3.** Antioxidant,  $\alpha$ -glucosidase inhibitory, and cytotoxic activities of 1–4.

<sup>*a*</sup> The half maximal effective concentration ( $\mu$ M); <sup>*b*</sup> the 50% inhibitory concentration ( $\mu$ M); <sup>*c*</sup> the 50% growth inhibitory concentration ( $\mu$ M) values represent the means  $\pm$  standard deviation based on triplicate experiments; <sup>*d*</sup> RPMI-8402, T cell acute lymphocytic leukemia; NALM6, B cell acute lymphocytic leukemia.

Compounds 1–4 did not show significant tyrosinase inhibitory activity at a concentration of 100  $\mu$ M (kojic acid, IC<sub>50</sub> = 41.85  $\mu$ M). Also, 1–4 were evaluated for their antimicrobial properties against three Gram-positive bacteria (*Bacillus subtilis* KCTC 1021, *Micrococcus luteus* KCTC 1915, and *Staphylococcus aureus* KCTC 1927) and three Gram-negative bacteria (*Escherichia coli* KCTC 2441, *Salmonella typhimurium* KCTC 2515, and *Klebsiella pneumonia* KCTC 2690). However, none of the compounds inhibited the growth of Gram-positive and Gram-negative bacteria at a concentration of 32.0  $\mu$ g/mL.

#### 3. Materials and Methods

## 3.1. General Experimental Procedures and Reagents

NMR spectra were acquired with a Bruker AVANCE III 600 spectrometer (Bruker Biospin GmbH, Rheinstetten, Germany) with a 3 mm probe operating at 600 MHz  $(^{1}H)$ and 150 MHz (<sup>13</sup>C). Chemical shifts were expressed in ppm with reference to the solvent peaks ( $\delta_H$  3.31 and  $\delta_C$  49.15 ppm for CD<sub>3</sub>OD). UV spectra were recorded with a Shimadzu UV-1650PC spectrophotometer (Shimadzu Corporation, Kyoto, Japan). IR spectra were obtained on an OPUS FT/IR-ALPHA II spectrophotometer (Bruker OPTIK GmbH & Co. KG, Ettlingen, Germany). Optical rotations were measured with a Rudolph analytical Autopol III S2 polarimeter (Rudolph Research Analytical, Hackettstown, NJ, USA). LR-ESIMS data were obtained with an ISQ EM mass spectrometer. HR-ESIMS data were obtained with a Waters SYNPT G2 Q-TOF mass spectrometer (Waters Corporation, Milford, MA, USA) at the Korea Basic Science Institute (KBSI) in Cheongju, Republic of Korea. ECD spectra were recorded with a JASCO J-1500 circular dichroism spectrometer (JASCO Corporation, Tokyo, Japan) at the Center for Research Facilities, Changwon National University, in Changwon, Republic of Korea. HPLC was performed using a BLS-Class pump (Teledyne SSI, Inc., State College, PA 16803, USA) with a Shodex RI-201H refractive index detector (Shoko Scientific Co., Ltd., Yokohama, Japan). Columns for HPLC were YMC-Triart C<sub>18</sub> (250 mm  $\times$  10 mm, 5  $\mu$ m), YMC-Triart C<sub>8</sub> (250 mm  $\times$  10 mm, 5  $\mu$ m), and YMC-CHIRAL PREP CD PM (250 mm  $\times$  4.6 mm, S-10  $\mu$ m). RP silica gel (YMC-Gel ODS-A, 12 nm, S-75 µm) was used for open-column chromatography. Organic solvents were purchased

as HPLC grade, and ultrapure waters were obtained from the Milipore Mili-Q Direct 8 system. The reagents used in the bioassay were purchased from Sigma-Aldrich and Tokyo Chemical Industry. Cancer cell lines were purchased from the Japanese Cancer Research Resources Bank (JCRB) (NUGC-3, JCRB Cell Bank/Cat. # JCRB0822), the DSMZ-German Collection of Microorganisms and Cell Cultures (RPMI-8402, DSMZ/Cat # ACC 290; WSU-DLCL2, DSMZ/Cat # ACC 575), and the American Type Culture Collection (ATCC) (PC-3, ATCC/Cat. # CRL-1435; MDA-MB-231, ATCC/Cat. # HTB-26; ACHN, ATCC/Cat. # CRL-1611; NCI-H23, ATCC/Cat. # CRL-5800; HCT-15, ATCC/Cat. # CCL-225; HL-60, ATCC/Cat. # CCL-240; Raji, ATCC/Cat # CCL-86; K562, ATCC/Cat # CCL-243; NALM6, ATCC/Cat # CRL-3273; U266, ATCC/Cat # TIB-196).

## 3.2. Fungal Strain and Fermentation

The fungal strain *Meira* sp. 1210CH-42 was obtained from a seawater sample collected at the Chuuk Islands, Federated States of Micronesia, in 2010, as described previously [12]. The strain was identified as *Meira* sp. (GenBank accession number OQ693946) through DNA amplification and by sequencing the ITS region of the rRNA gene, as described earlier [12]. The cultures of strain 1210CH-42 were performed in modified Bennett's broth medium (1% D-glucose, 0.2% tryptone, 0.1% yeast extract, 0.1% beef extract, 0.5% glycerol, sea salt 10 g/L, pH 7.0). A seed culture was prepared from a spore suspension of strain 1210CH-42 by inoculating it into 2 L flasks and incubating it at 28 °C for 7 days in a rotary shaker at 120 rpm. The seed culture was inoculated aseptically into a 100 L fermenter containing 70 L of sterilized culture medium (0.1% v/v). Large-scale fermentation was conducted at 28 °C, 40 rpm, and with an airflow rate of 10 L/min (LPM) for a duration of 21 days before being harvested.

## 3.3. Extraction and Isolation of Compounds 1–4

A culture broth of strain 1210CH-42 (total, 70 L) was harvested via high-speed centrifugation at 60,000 rpm. Subsequently, the supernatant was extracted two times with ethyl acetate (EtOAc, 140 L). The EtOAc layer was evaporated under reduced pressure to yield a crude extract (4.61 g). The crude extract was subjected to ODS open column chromatography (YMC Gel ODS-A, 12 nm, S75 µm) followed by stepwise gradient elution with MeOH/H<sub>2</sub>O (*v*/*v*) (20:80, 40:60, 60:40, 80:20, and 100:0) as an eluent. The 20% MeOH (0.89 g) was subjected to ODS open column chromatography (YMC Gel ODS-A, 12 nm, S75  $\mu$ m) followed by stepwise gradient elution with MeOH/H<sub>2</sub>O (v/v) (5:95, 10:90, 15:85, 20:80, and 100:0) to provide five fractions (F1–F5). Fraction F4 (293 mg) was purified by a semi-preparative reversed-phase HPLC (YMC-Triart  $C_{18}$  column 250 mm  $\times$  10 mm i.d., 5  $\mu$ m; 15% MeOH in H<sub>2</sub>O; flow rate: 2.0 mL/min; detector: RI) to yield **1** (18.1 mg,  $t_{\rm R}$ 62.0 min). Fraction F2 (140 mg) was further purified by a semi-preparative RP HPLC (YMC-Triart C<sub>18</sub> column 250 mm  $\times$  10 mm i.d., 5  $\mu$ m; 5% MeOH in H<sub>2</sub>O; flow rate: 2.0 mL/min; detector: RI) to obtain eighteen subfractions (F2.SF1-F2.SF18). Compound 2 was re-purified from subfraction F2.SF16 by a semi-preparative RP HPLC (YMC-Triart C<sub>8</sub> column 250 mm  $\times$  10 mm i.d., 5  $\mu\text{m};$  5% MeOH in H\_2O; flow rate: 2.0 mL/min; detector: RI) to yield 2 (8.2 mg,  $t_R$  38.0 min). Fraction F3 (266 mg) was purified using a semi-preparative RP HPLC (YMC-Triart C<sub>18</sub> column 250 mm  $\times$  10 mm i.d., 5  $\mu$ m; 10% MeOH in H<sub>2</sub>O; flow rate: 2.0 mL/min; detector: RI) to yield eighteen subfractions (F3.SF1–F3.SF18). Subsequently, compounds 3 and 4 were isolated from subfraction F3.SF4 using an analytical HPLC (YMC-CHIRAL PREP CD PM 250 mm  $\times$  4.6 mm, S-10  $\mu$ M; flow rate: 0.8 mL/min; detector: RI) with an isocratic elution of 5% MeOH in H<sub>2</sub>O to yield 3 (4.3 mg,  $t_{\rm R}$  11.0 min) and 4 (5.2 mg, t<sub>R</sub> 13.0 min).

Argovin (1): White amorphous powder; UV (MeOH)  $\lambda_{max}$  (log  $\varepsilon$ ) 213 (3.71), 236 (3.95), 283 (3.82) nm; IR (MeOH)  $\nu_{max}$  3200, 1651, 1589, 1471, 1299 cm<sup>-1</sup>; <sup>1</sup>H and <sup>13</sup>C NMR data (CD<sub>3</sub>OD), see Tables 1 and 2; HR-ESIMS *m*/*z* 165.0552 [M + H]<sup>+</sup> (calcd. for C<sub>9</sub>H<sub>9</sub>O<sub>3</sub><sup>+</sup>, 165.0552; 187.0371) and [M + Na]<sup>+</sup> (calcd for C<sub>9</sub>H<sub>8</sub>O<sub>3</sub>Na<sup>+</sup>, 187.0371).

Meirol A (2): White amorphous powder; UV (MeOH)  $\lambda_{max}$  (log  $\varepsilon$ ) 219 (3.96), 261 (3.83), 290 (3.39) nm; IR (MeOH)  $\nu_{max}$  3233, 1626, 1448, 1419, 1292 cm<sup>-1</sup>; <sup>1</sup>H and <sup>13</sup>C NMR data (CD<sub>3</sub>OD), see Tables 1 and 2; HR-ESIMS m/z 166.0503 [M + H]<sup>+</sup> (calcd. for C<sub>8</sub>H<sub>8</sub>NO<sub>3</sub><sup>+</sup>, 166.0504; 188.0324) and [M + Na]<sup>+</sup> (calcd for C<sub>8</sub>H<sub>7</sub>NO<sub>3</sub>Na<sup>+</sup>, 188.0324).

Meirol B (3): Purple amorphous powder;  $[\alpha]_D^{25}$  +23.3 (*c* 0.1, MeOH); UV (MeOH)  $\lambda_{max}$  (log  $\varepsilon$ ) 214 (3.81), 235 (3.98), 287 (3.74) nm; ECD (MeOH,  $\lambda$  [nm] ( $\Delta\varepsilon$ ), *c* = 1.58 mM) 309 (-0.83), 282 (+1.44), 271 (+1.10), 237 (+3.00), 209 (0.56); IR (MeOH)  $\nu_{max}$  3270, 1681, 1592, 1488, 1280 cm<sup>-1</sup>; <sup>1</sup>H and <sup>13</sup>C NMR data (CD<sub>3</sub>OD), see Tables 1 and 2; HR-ESIMS *m*/*z* 203.0321 [M + Na]<sup>+</sup> (calcd for C<sub>9</sub>H<sub>8</sub>O<sub>4</sub>Na<sup>+</sup>, 203.0320).

Meirol C (4): Light purple amorphous powder;  $[\alpha]_D^{25}$  +23.3 (*c* 0.1, MeOH); UV (MeOH)  $\lambda_{\text{max}}$  (log ε) 212 (3.72), 237 (3.95), 291 (3.77) nm; ECD (MeOH,  $\lambda$  [nm] ( $\Delta$ ε), *c* = 1.58 mM) 343 (+0.71), 314 (-0.50), 290 (+0.94), 270 (+0.59), 242 (+2.57); IR (MeOH)  $\nu_{\text{max}}$  3256, 1686, 1611, 1286 cm<sup>-1</sup>; <sup>1</sup>H and <sup>13</sup>C NMR data (CD<sub>3</sub>OD), see Tables 1 and 2; HR-ESIMS *m*/*z* 203.0322 [M + Na]<sup>+</sup> (calcd for C<sub>9</sub>H<sub>8</sub>O<sub>4</sub>Na<sup>+</sup>, 203.0320).

# 3.4. Computational Analysis

The initial geometry optimization and conformational searches were generated using Conflex 8 (rev. B, Conflex Corp., Tokyo, Japan). The optimization and calculation for electronic circular dichroism (ECD) were conducted utilizing the Gaussian 16 program (rev. B.01, Gaussian Inc., Wallingford, CT, USA). Conformational searches were executed through MMFF94s force field calculations, with a search limit set at 5 kcal/mol. The conformers were optimized using the ground state method at the CAM-B3LYP/6-31 G+ (d, p) level in MeOH with an IEFPCM model for ECD. The theoretical calculations of ECD spectra were performed using TD-SCF at the CAM-B3LYP /6-31 G+ (d, p). The ECD spectrum was derived by calculating the Boltzmann-weighted sum of conformer spectra. The final ECD spectra were simulated using SpecDis (v. 1.71) with  $\sigma$  values ranging from 0.20 to 0.30 eV. All calculated curves were UV-shifted by +10 to +15 nm to better simulate experimental spectra.

## 3.5. Antioxidant Activity Assay

The DPPH radical scavenging activities of **1–4** were determined by the reported method [13]. In a 96-well plate, 100  $\mu$ L of sample solution (in MeOH) was mixed with 100  $\mu$ L of DPPH solution (0.16 mM in MeOH), shaken several times, and then incubated at room temperature for 30 min. The absorbance at 517 nm was recorded. Ascorbic acid was used as the positive control, and the experiments were performed in triplicate.

## 3.6. *α-Glucosidase Inhibitory Activity Assay*

The  $\alpha$ -glucosidase inhibitory activity was determined by measuring the absorbance increase resulting from the hydrolysis of *p*-nitrophenyl- $\alpha$ -D-glucopyranoside (*p*NPG, TCI) by  $\alpha$ -glucosidase at 405 nm using a microplate reader, according to the reference to previously reported literature [12]. The 130 µL sample solution (in 0.1 mM PBS) with the 30 µL  $\alpha$ -glucosidase solution (0.2 U/mL) was incubated at 37 °C for 10 min. Subsequently, 40 µL of 5 mM *p*NPG was added. The reaction mixture was further incubated at 37 °C for 20 min. The  $\alpha$ -glucosidase inhibitory activity was determined using a microplate reader at 405 nm. The negative control was prepared by substituting PBS buffer for the sample in the same way as the test. Acarbose served as the positive control, and the experiments were conducted in triplicate.

#### 3.7. Cytotoxicity Assay

The cytotoxic activities of **1–4** were measured using the CellTiter-Glo luminescent cell viability assay (Promega, Madison, WI, USA) and conducted by the SRB (sulforhodamine B) assay, as previously described [14,15]. The luminescence signal was quantified using a GloMax-Multi Detection System (Promega, Madison, WI, USA), and GI<sub>50</sub> values were

determined utilizing a relative GI<sub>50</sub> model in GraphPad Prism (GraphPad, San Diego, CA, USA). Doxorubicin was used as the positive control.

## 3.8. Tyrosinase Inhibitory Activity Assay

The tyrosinase inhibitory activity was assessed using L-DOPA and the 96-well microplate method, as previously reported [16]. Briefly, 140  $\mu$ L of 20 mM phosphate buffer (pH 6.8) and 20  $\mu$ L of mushroom tyrosinase (480 U/mL) in the same buffer were added to wells containing 20  $\mu$ L of the test compounds. After incubation for 10 min at 25 °C, 20  $\mu$ L of L-DOPA (0.85 mM) was added to the 200  $\mu$ L reaction system. The incubation was continued for another 20 min at 25 °C; then, the colored end product's absorbance was measured at 475 nm using a microplate reader. Kojic acid was utilized as a positive control in the reference sample experiment. All experiments were performed in triplicate for each compound.

## 3.9. Antibacterial Assay

The antibacterial activities were determined using the 96-well microplate method, as described in a published report, against three Gram-positive and three Gram-negative bacteria [17].

## 4. Conclusions

In summary, four catecholic compounds (1–4), including three new ones (2–4), were identified from *Meira* sp. 1210CH-42. Their structures were elucidated by a detailed analysis of NMR and HRESIMS data. The absolute configurations of **3** and **4** were determined by calculating the ECD spectra of their possible isomers. The antioxidant-DPPH assay showed that all compounds exhibited more significant free radical scavenging activity (EC<sub>50</sub> = 6.01–7.47  $\mu$ M) than ascorbic acid. Additionally, **1** and **2** exhibited moderate  $\alpha$ -glucosidase inhibitory activity (IC<sub>50</sub> = 184.50 and 199.70  $\mu$ M, respectively) and selective cytotoxicity against blood cancer cell lines (RPMI-8402 and NALM6, GI<sub>50</sub> = 9.47–29.40  $\mu$ M). As a result, the catecholic indanones from *Meira* sp. 1210CH-42 could serve as a potential agent for antioxidative,  $\alpha$ -glucosidase inhibitory, and anticancer leads. Further investigation is needed to determine the biological mechanism of compounds from the marine-derived fungus *Meira*.

**Supplementary Materials:** The following are available online at https://www.mdpi.com/article/ 10.3390/md22020087/s1: Figures S1–S6: <sup>1</sup>H, <sup>13</sup>C NMR, HSQC, COSY, HMBC, and HR-ESIMS data of **1**. Figures S7–S30: <sup>1</sup>H, <sup>13</sup>C NMR, HSQC, COSY, HMBC, HR-ESIMS, UV, and IR data of **2–4**. Figures S31–S32 and Tables S1–S4: TDSCF-ECD calculation data of **3–4**.

Author Contributions: Conceptualization, H.J.S.; investigation, M.A.L., J.S.K., J.-W.Y., H.-S.L., C.-S.H. and S.J.P.; resources, M.A.L.; writing—original draft preparation, M.A.L.; writing—review and editing, H.J.S. and M.A.L.; project administration, H.J.S.; funding acquisition, H.J.S. All authors have read and agreed to the published version of the manuscript.

**Funding:** This research was supported by the Korea Institute of Marine Science and Technology Promotion (KIMST) Grant funded by the Ministry of Oceans and Fisheries, Korea (Grant No. 20220027) and by the Korea Institute of Ocean Science and Technology (PEA0211).

Institutional Review Board Statement: Not applicable.

**Data Availability Statement:** The data presented in the article are available in the Supplementary Materials.

Acknowledgments: The authors express gratitude to Jung Hoon Choi, Korea Basic Science Institute, Ochang, Korea, for providing mass data.

Conflicts of Interest: The authors declare no conflicts of interest.

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