

Article

Isolation of Scalarane-Type Sesterterpenoids from the Marine Sponge *Dysidea* sp. and Stereochemical Reassignment of 12-*epi*-Phyllactone D/E

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Abstract: The chemical investigation of the marine sponge *Dysidea* sp., which was collected from Bohol province in the Philippines, resulted in the identification of 15 new scalarane-type sesterterpenoids (1–14, 16), together with 15 known compounds. The chemical structures of the new compounds were elucidated based on NMR spectroscopy and HRMS. The structure of 12-*epi*-phyllactone D/E (15) isolated during this study was originally identified in 2007. However, careful inspection of our experimental ¹³C NMR spectrum revealed considerable discrepancies with the reported data at C-9, C-12, C-14, and C-23, leading to the correction of the reported compound to the C-12 epimer of 15, phyllactone D/E. The biological properties of compounds 1–16 were evaluated using the MDA-MB-231 cancer cell line. Compound 7, which bears a pentenone E-ring, exhibits significant cytotoxicity with a GI₅₀ value of 4.21 μM.

Keywords: *Dysidea*; sesterterpenoid; scalarane; marine sponge; marine natural product; anticancer activity; stereochemistry reassignment



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1. Introduction

Sesterterpenoids, such as the ophiobolins, sterigmatocystin, and hippolide A, have attracted a lot of attentions as potent pharmaceutical compounds because of their unique anti-inflammatory activity and cytotoxicity in various cancer cell lines [1]. These compounds are ubiquitous in a broad range of natural sources, from easily accessible terrestrial plants and insects to hard-to-access marine organisms. In marine nature, the scalarane-type scaffolds have emerged as one of the most prevalent structural features of the sesterterpenoids [2]. Since scalarin, a pentacyclic scalarane, was first isolated from the marine sponge *Cacospongia scalaris* in 1972 [3], a number of scalarane-type sesterterpenoids has been isolated from *Dysidea* sp. [4,5], *Phyllospongia* sp. [6–11], *Strepsichordaia* sp. [12], *Cateriospongia* sp. [13,14], *Smenospongia* sp. [15], and *Hyrtios* sp. [16], belonging to the order Dictyoceratida [17].

This family of scalarane derivatives is featured with a trans-fused 6/6/6/6 ring system and can be further categorized into three structural subgroups, namely scalarane, homoscalarane, and bishomoscalarane, based on the presence of single carbon substituents at C-20 and/or C-24 (Figure 1). Among them, bishomoscalarane exhibits an exceptionally broad range of diversity in the carbon framework, arising from two distinctive sites: C-20 and C-24/C-25 (Figure 2). Therefore, cyclopropane or alcohol/esters are frequently found at C-20 adjacent to the A ring [12]. The oxidation of C-24 and C-25 results in the formation of an extra E ring in the form of a lactone or cyclopentenone [18]; 24-oxo-25-norbishomoscalarane has also been identified as another feature of the D ring [8]. In addition, oxidation of the backbone usually occurs at C-3 [6], C-12 [19], and C-16 [12] to produce

hydroxyl or ester substituents. A large group of bishomoscalarane derivatives found in nature is considered to be the outcome of these variations occurring in combinations.

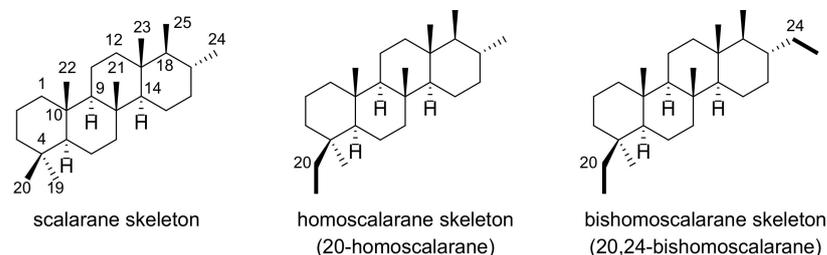


Figure 1. Subtypes of the scalarane skeleton found in marine nature.

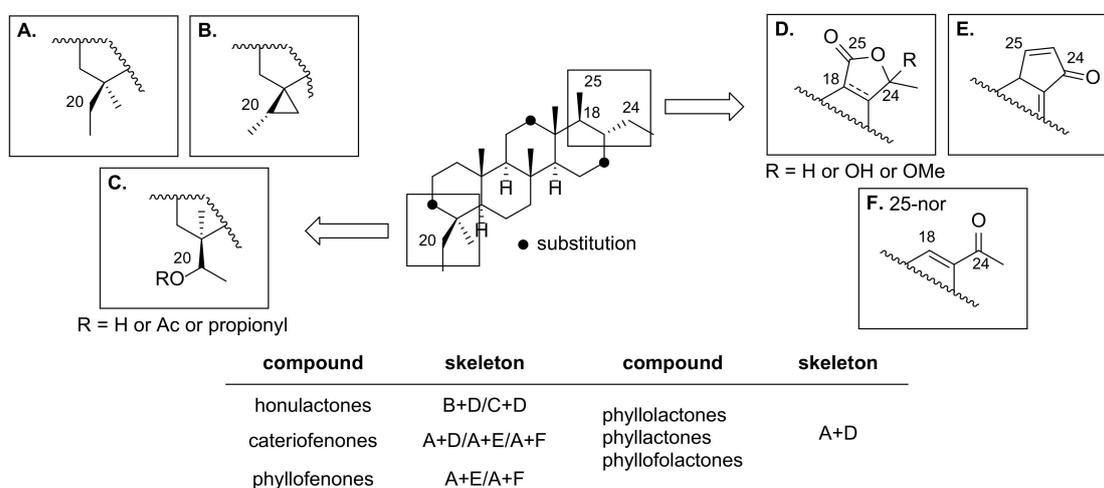


Figure 2. Diversity of bishomoscalarane skeletons frequently found in nature.

The marine sponge *Dysidea sp.* is known to be a rich source of scalaranes, which exhibits useful pharmacological properties, such as anticancer and antimicrobial activities [17,20–24]. In the course of our studies on bioactive natural products from marine organisms, we inspected the chemical components of *Dysidea sp.* collected from the Bohol province in the Philippines. As a result, we identified 15 new scalarane derivatives, including one scalarane, four 20,24-bishomo-25-norscalaranes, and 10 bishomoscalaranes (Figure 3), along with 14 known compounds (Figure S1, Supplementary Materials). In this report, we disclose the structural assignment of the new scalarane sesterterpenoids and their pharmacological properties as anti-cancer agents. In addition, the C-12 configuration of compound 15, which was assigned by Li in 2007 [11], was reinvestigated because of the significant differences observed between the reported and experimental ^{13}C chemical shifts at C-9, C-12, C-14, and C-23.

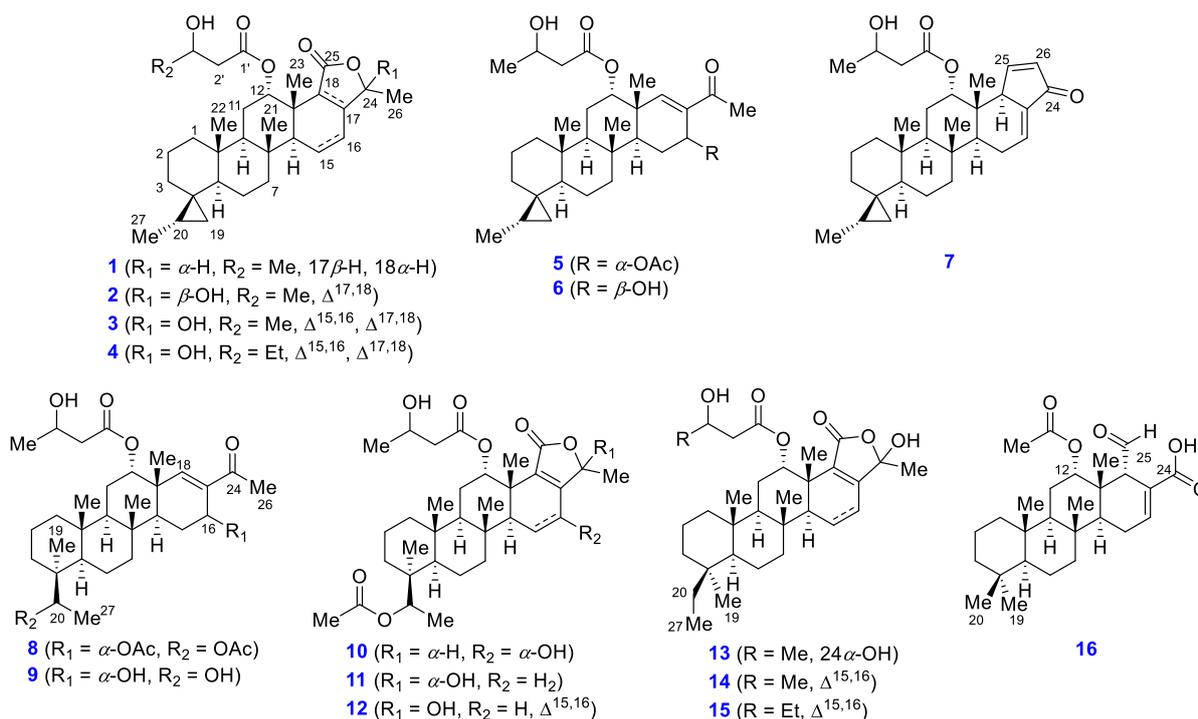


Figure 3. The structures of compounds 1–16 isolated from *Dysidea* sp.

2. Results and Discussion

2.1. Structure Elucidation

Compound **1** was isolated as a colorless oil, and its molecular formula was determined to be $\text{C}_{31}\text{H}_{48}\text{O}_5$ using HRESIMS (m/z $[\text{M} + \text{Na}]^+$ 523.3382, calcd 523.3394), corresponding to eight degrees of unsaturation (DOU). The ^1H NMR spectrum of **1** exhibited three singlet methyl groups at δ_{H} 0.77, 0.82, and 1.04; three doublet methyl groups at δ_{H} 1.07, 1.23, and 1.37; and three oxymethines at δ_{H} 4.03, 4.19, and 5.41. Furthermore, unique upfield signals at δ_{H} 0.57 and -0.49 indicated the presence of a cyclopropane. Analysis of the ^{13}C NMR and HSQC spectra revealed the presence of two ester carbons (δ_{C} 174.8 and 171.7), three oxymethine carbons (δ_{C} 80.4, 74.8, 64.5), 10 methylene carbons, six methine carbons, and six methyl groups. The HMBC data showed notable correlations between the singlet methyl groups and methines from δ_{H} 0.77 to δ_{C} 51.4/50.3, δ_{H} 0.82 to δ_{C} 54.3/51.4, and δ_{H} 1.04 to δ_{C} 54.3, which are known as characteristic correlations occurring from the ring junctions of scalarane-type 6/6/6/6 fused-cyclic systems (Figure 4). Additional HMBC correlations from the doublet methyl group at δ_{H} 1.37 to δ_{C} 80.4/44.9 and from the methine at δ_{H} 2.34 to δ_{C} 174.8/44.9 suggested the existence of a γ -valerolactone moiety. Therefore, our preliminary findings led to the hypothesis that compound **1** possessed a honulactone A-like scaffold (B+D type shown in Figure 2) [12].

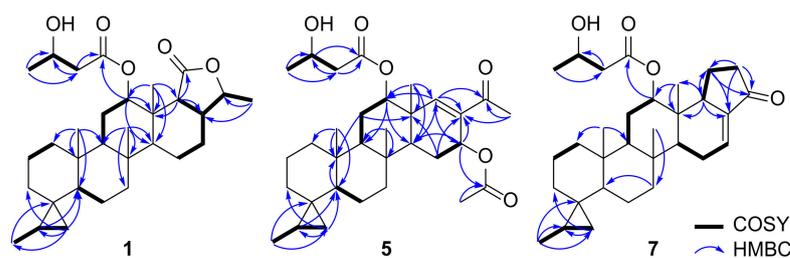


Figure 4. COSY and HMBC correlations observed for compounds **1**, **5**, and **7**.

While the $\Delta^{17,18}$ -olefin in honulactones is considered one of the structural features that forms the unsaturated lactone E-ring, the initially identified γ -valerolactone and DOU suggest the possibility of a saturated terminal lactone in compound **1**. This speculation was confirmed by the ^1H - ^1H COSY cross peak observed for H₂-15–H₂-16–H-17–H-18, as well as HMBC correlations from CH₃-23 (δ_{H} 1.04) to C-18 (δ_{C} 52.5) and from H-18 (δ_{H} 2.34) to C-13 (δ_{C} 38.7). In addition, the cyclopropane moiety inferred from the ^1H NMR data was positioned at C-4 based on the HMBC correlations from H₂-19 (δ_{H} 0.57, and -0.49) to C-3 (δ_{C} 33.2)/C-5 (δ_{C} 50.3) and from CH₃-27 (δ_{H} 1.07) to C-4 (δ_{C} 22.7), and the spin system for CH₃-27–H-20 (δ_{H} 0.72)–H₂-19 (δ_{H} 0.57, -0.49) in the ^1H - ^1H COSY spectrum. Interpretation of the remaining HMBC correlations from CH₃-4' (δ_{H} 1.23) to C-2' (δ_{C} 43.3)/C-3' (δ_{C} 64.5), H₂-2' (δ_{H} 2.49/2.42) to C-1' (δ_{C} 171.7)/C-3', and H-12 (δ_{H} 5.41) to C-1' elucidated the 3-hydroxyl butanoate group at C-12.

The trans-fused cyclic scaffold in **1** was determined from the NOESY cross peaks observed between H-11 β (δ_{H} 1.71) and CH₃-21 (δ_{H} 0.82)/CH₃-22 (δ_{H} 0.77), and CH₃-23 and CH₃-21/H-17 (δ_{H} 1.86) (Figure 5). The NOESY correlations between H-12 and CH₃-23, and H-18 and H-14 (δ_{H} 1.23)/H-24 (δ_{H} 4.03) suggested the β -orientations of H-12 and CH₃-26, respectively. Moreover, the 20S* configuration of CH₃-27 was determined based on the NOESY signals observed between H-19_{cis} (δ_{H} -0.49) and H-3 β (δ_{H} 1.24)/CH₃-27, and H-19_{trans} (δ_{H} 0.57) and H₂-6.

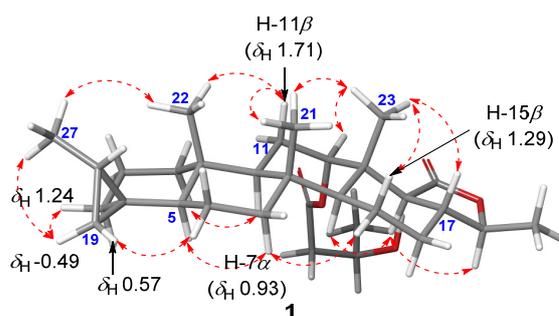


Figure 5. NOESY correlations observed for compound **1**.

Compound **2** was isolated as a colorless oil, and its molecular formula was determined to be C₃₁H₄₆O₆ by HRESIMS (m/z [M + Na]⁺ 537.3167, calcd 537.3187), corresponding to nine degrees of unsaturation. Analysis of the 1D and 2D NMR spectra obtained for **2** indicated a similar carbon framework to **1**, but the higher oxidation state of the lactone in E ring appeared as a major difference. HMBC correlations from CH₃-23 (δ_{H} 1.22) to C-18 (δ_{C} 133.7) and from CH₃-26 (δ_{H} 1.56) to C-17 (δ_{C} 162.9) revealed an α,β -unsaturated lactone in the E ring, which was responsible for the one degree higher DOU than that of **1**. In addition, the ^{13}C chemical shift of C-24 (δ_{C} 104.4) was characteristic of a ketal carbon atom, of which the position was confirmed by HMBC correlations from CH₃-26 to C-24. The β -configuration of OH-24 was determined by the NOESY correlation observed between H-16 α (δ_{H} 2.28) and CH₃-26 (Figure S3, Supplementary Materials).

Compound **3** was isolated as a mixture of two inseparable epimers. The molecular formula of **3** was deduced to be C₃₁H₄₄O₆ by HRESIMS (m/z [M + Na]⁺ 535.3011, calcd 535.3030), corresponding to 10 degrees of unsaturation. An initial inspection of the ^{13}C NMR spectrum revealed that most of the peaks were split into a doublet-like shape, indicating a 1:1 mixture of diastereomers. The 1D and 2D NMR spectra obtained for compound **3** exhibited most of the structural features of **2**, except for one more disubstituted olefin observed at δ_{H} (6.38/6.37)/ δ_{C} (138.84/138.80) and δ_{H} (6.29/6.25)/ δ_{C} (118.6/118.4). The location of the double bond was determined to be $\Delta^{15,16}$ using the consecutive ^1H - ^1H COSY correlations observed for H-14 (δ_{H} 2.69/2.62)–H-15 (δ_{H} 6.38/6.37)–H-16 (δ_{H} 6.29/6.25). The splittings observed in the ^{13}C NMR spectrum were most prominent at CH₃-26 ($\Delta\delta_{\text{C}}$ 1.13 ppm), informing a mixture of C-24 epimers. This phenomenon has often been observed in the case of 24-homosclaranes, which possess both an $\Delta^{15,16}$ -

olefin and 24-hydroxy pentenolide E-ring [25,26]. Since the $\Delta^{15,16}$ -olefin increases the planarity of the D-ring and renders the C-24 stereocenter more isolated, the 24*R** and 24*S** diastereomers exhibit almost identical spectroscopic and chromatographic behaviors to give an inseparable mixture.

Compound **4** was isolated as an inseparable mixture and its molecular formula was determined to be $C_{32}H_{46}O_6$ by HRESIMS (m/z $[M + Na]^+$ 549.3163, calcd 549.3187), indicating 10 degrees of unsaturation. The NMR spectra of **4** were only discriminated from those of **3** by the extra methylene group observed at δ_H 1.50 and δ_C 29.5/29.4, which was also supported by the mass difference of +14. The extra methylene group was observed in the ester side chain located at C-12, which formed a 3-hydroxypentanoate moiety, as supported by the spin system for H₂-2' (δ_H 2.35)–H-3' (δ_H 3.90/3.86)–H₂-4' (δ_H 1.50)–CH₃-5' (δ_H 0.95) in the ¹H-¹H COSY spectrum.

Compound **5** was isolated as a colorless oil. Its molecular formula was determined to be $C_{32}H_{48}O_6$ by HRESIMS (m/z $[M + Na]^+$ 551.3310, calcd 551.3343), corresponding to nine degrees of unsaturation. Our initial analysis of the ¹H NMR spectrum obtained for compound **5** indicated the presence of the scalarane-type scaffold: five singlet methyl groups at δ_H 0.80, 0.84, 1.06, 2.02, and 2.22; two doublet methyl groups at δ_H 1.08 and 1.25; three oxymethines at δ_H 4.19, 5.11, and 5.76; unique cyclopropane signals at δ_H 0.59 and –0.49; and one singlet olefin at δ_H 6.72. The ¹³C NMR and HSQC spectra showed one ketone (δ_C 197.7), two ester carbons (δ_C 172.0, 170.2), one trisubstituted olefin carbon (δ_C 153.2, 135.1), three oxymethine carbons (δ_C 76.8, 65.3, 64.6), eight methylenes, four methines, and seven methyl groups. In addition, the HMBC correlation from the singlet methyl at δ_H 2.22 to δ_C 197.7 suggested the presence of a methyl ketone moiety, instead of the lactone E-ring observed in compounds **1–4**, leading to the conclusion that **5** had a B+F type skeleton, as shown in Figure 2.

Detailed interpretation of the combined spectral data of **5** revealed that the features related to the A-B-C ring system were identical to those of **1–4**. As anticipated, the methyl ketone was positioned at C-17 to form an unsaturated ketone in the D ring on the basis of HMBC correlations between CH₃-26 (δ_H 2.22) and C-17 (δ_C 135.1), and H-18 (δ_H 6.72) and C-17/C-24 (δ_C 197.7) (Figure 4). Moreover, the ¹H-¹H COSY cross peak for H-14 (δ_H 1.76)–H₂-15 (δ_H 1.89, 1.61)–H-16 (δ_H 5.76) and HMBC correlations from δ_H 2.02 (CH₃CO₂–) to δ_C 170.2 (CH₃CO₂–) and from H-16 to δ_C 170.2 positioned an acetate substituent at C-16, of which the relative configuration was assigned to be α -orientation based on the small coupling constants between H₂-15 and H-16 (dd, $J_{H-15-H-16} = 4.3, 1.6$ Hz).

Compound **6** was isolated as an amorphous solid. Its molecular formula was determined to be $C_{30}H_{46}O_5$ by HRESIMS (m/z $[M + Na]^+$ 509.3215, calcd 509.3237), corresponding to eight degrees of unsaturation. The ¹H and ¹³C NMR spectra obtained for compound **6** were almost identical to those of **5**. However, the absence of one ester carbon and the singlet methyl group at δ_H 2.02 suggested deacetylation from **5**, which was further supported by an upfield shift of H-16 (δ_H 4.62). The relative configuration of OH-16 was assigned as β -orientation based on the large coupling constant observed between H-16 and H-15 β (dd, $J_{H-15-H-16} = 9.6, 5.1$ Hz).

Compound **7** was isolated as a yellow oil. Its molecular formula was determined to be $C_{31}H_{44}O_4$ by HRESIMS (m/z $[M + Na]^+$ 503.3113, calcd 503.3132), corresponding to 10 degrees of unsaturation. Preliminary analysis of the ¹H and ¹³C NMR data revealed that the scalarane-type scaffold had a cyclopropane substituent on the A ring. Interpretation of the ¹³C NMR and HSQC spectra exhibited the sp² carbons in the enone systems: three sp² methines at δ_H 7.38/ δ_C 157.5, δ_H 6.34/ δ_C 137.4, and δ_H 6.64/ δ_C 130.4; and one trisubstituted sp² carbon atom at δ_C 136.4. Therefore, HMBC correlations observed from H-25 (δ_H 7.38) to C-17 (δ_C 136.4)/C-18 (δ_C 49.3)/C-24 (δ_C 195.9), H-26 (δ_H 6.34) to C-17/C-24/C-25 (δ_C 157.5), and H-18 (δ_H 3.35) to C-17/C-23 (δ_C 14.5), as well as the ¹H-¹H COSY cross peak for H-18–H-25–H-26, confirmed the presence of a $\Delta^{25,26}$ -cyclopenten-24-one subunit for the E-ring and the trisubstituted double bond at $\Delta^{16,17}$ (Figure 4).

Compound **8** was isolated as a yellow oil, and its molecular formula was determined to be $C_{34}H_{52}O_8$ by HRESIMS (m/z $[M + Na]^+$ 611.3541, calcd 611.3554), corresponding to nine degrees of unsaturation. The 1H NMR spectrum obtained for compound **8** showed similar patterns to that of **5**. However, the upfield peaks observed for the cyclopropane moiety in **5** were substituted by an oxymethine at δ_H 5.35, a methyl singlet at δ_H 1.09, and an acetate at δ_H 2.03, suggesting the C+F type scaffold shown in Figure 2. Therefore, the connectivity of C-27–C-20–C-4–C-19 was determined using the HMBC correlations observed from CH_3 -19 (δ_H 0.99) to C-20 (δ_C 73.2) and from CH_3 -27 (δ_H 1.09) to C-4 (δ_C 39.4)/C-20 (Figure 6). In addition, the acetate at δ_H 2.03 exhibited a HMBC correlation with C-20 to be located at C-20. The relative configuration at C-20 was assigned as 20*R** from the NOESY correlations observed between H-20 (δ_H 5.35) and H-2 β (δ_H 1.47)/ CH_3 -22 (δ_H 0.87), and H-3 β (δ_H 1.67) and CH_3 -27 (Figure 7). Similarly, the configuration of the acetate group at C-16 was assigned as α -orientation based on the small coupling constant observed for H-16 (dd, $J_{H-15-H-16} = 4.3, 1.6$ Hz).

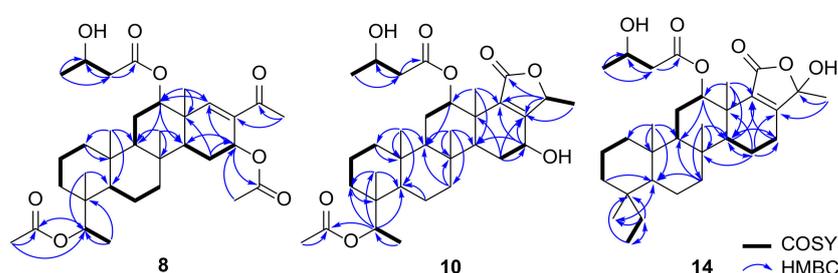


Figure 6. COSY and HMBC correlations observed for compounds **8**, **10**, and **14**.

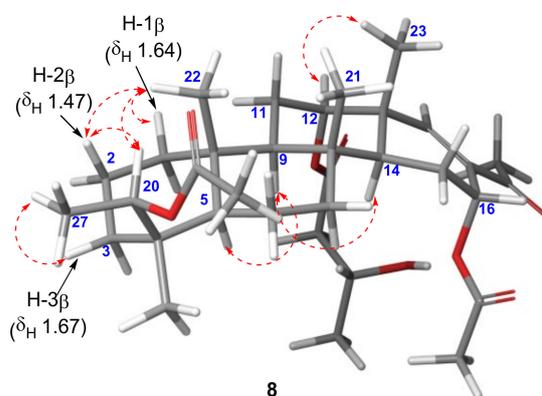


Figure 7. NOESY correlations observed for compound **8**.

Compound **9** was isolated as a colorless oil, and its molecular formula was determined to be $C_{30}H_{48}O_6$ by HRESIMS (m/z $[M + NH_4]^+$ 522.3810, calcd 522.3789) corresponding to seven degrees of unsaturation. Analysis of the 1D and 2D NMR data provided almost identical features to those of **8** to determine the carbon skeleton of compound **9**. In this case, only one ester carbon atom (δ_C 172.2) was observed in the ^{13}C NMR spectrum, and the acetate groups shown in the 1H NMR spectrum of **8** disappeared. This information indicated that compound **9** was the deacetylation product of **8**. Accordingly, the upfield shifts of H-20 (δ_H 4.32) and H-16 (δ_H 4.55) were the major differences, compared to compound **8**.

Compound **10** was isolated as a yellow oil, and its molecular formula was determined to be $C_{33}H_{50}O_8$ by HRESIMS (m/z $[M + Na]^+$ 597.3404, calcd 597.3398), corresponding to nine degrees of unsaturation. Preliminary inspection of the ^{13}C NMR and HSQC data of **10** identified four singlet methyl groups (δ_H 0.87/ δ_C 16.6, δ_H 0.87/ δ_C 16.8, δ_H 0.96/ δ_C 23.3, δ_H 1.13/ δ_C 19.8), three doublet methyl groups (δ_H 1.07/ δ_C 16.0, δ_H 1.18/ δ_C 22.5, δ_H 1.39/ δ_C 18.2), and one acetate group (δ_H 2.03/ δ_C 22.0), indicating a honulactone C-like scaffold (C+D type shown in Figure 2) [12]. A detailed analysis of the 1H NMR spectrum identified

an oxymethine group at δ_{H} 4.44 as a major difference from honulactone C. The location of the oxymethine was determined to be C-16, as indicated by the HMBC correlations from H-16 (δ_{H} 4.44) to C-17 (δ_{C} 162.1)/C-18 (δ_{C} 135.6) and ^1H - ^1H COSY cross peak for H₂-15 (δ_{H} 1.91, 1.84)–H-16 (Figure 6). The configuration of the OH-16 group was assigned as α -orientation based on the small coupling constant observed for H-16 (dd, $J_{\text{H-15-H-16}} = 4.7, 1.4$ Hz), and compound **10** was named as 16 α -hydroxyhonulactone C [12].

Compound **11** was isolated as a yellow oil. Its molecular formula was determined as C₃₃H₅₀O₈ by HRESIMS (m/z [M + Na]⁺ 597.3396, calcd 597.3398), corresponding to nine degrees of unsaturation. The ^1H and ^{13}C NMR data of **11** were almost identical to those of **10**, but a ketal moiety (δ_{C} 104.4) was observed instead of one doublet methyl group and two oxymethines in compound **10**. As shown in compounds **2–4**, the hemiketal functionality in the scalarane-type scaffold usually occurs at C-24 in the E-ring, which was also applicable in this case, as indicated by the HMBC correlations from CH₃-26 (δ_{H} 1.56) to C-17 (δ_{C} 162.9)/C-24 (δ_{C} 104.4). The α -orientation of the hydroxyl group at C-24 was determined by the NOESY correlation between H-16 β (δ_{H} 2.33) and CH₃-26. Thus, compound **11** was named 24 α -hydroxyhonulactone C [12].

Compound **12** was isolated as an inseparable mixture. Its molecular formula was determined as C₃₃H₄₈O₈ by HRESIMS (m/z [M + Na]⁺ 595.3241, calcd 595.3241), corresponding to 10 degrees of unsaturation. Compared to **11**, two more sp² methines at δ_{C} 138.9/138.7 and δ_{H} 6.38, and δ_{C} 118.44/118.35 and δ_{H} 6.28/6.26 were observed in the ^{13}C NMR and HSQC spectra, indicating the presence of a disubstituted double bond. These sp² protons were involved in a spin system for H-14 (δ_{H} 2.66/2.62)–H-15 (δ_{H} 6.38)–H-16 (δ_{H} 6.28/6.26) in the ^1H - ^1H COSY spectrum and used to confirm the presence of the $\Delta^{15,16}$ -olefin, which was further supported by HMBC correlations from H-15 to C-13 (δ_{C} 40.1/40.0)/C-14 (δ_{C} 53.96/53.90)/C-17 (δ_{C} 157.3) and from H-16 to C-14/C-18 (δ_{C} 130.9). As discussed in the cases of **3** and **4**, the presence of the olefin at $\Delta^{15,16}$ and the hemiketal at C-24 rendered compound **12** an inseparable mixture of C-24 epimers.

Compound **13** was isolated as an amorphous solid. Its molecular formula was determined as C₃₁H₄₈O₆ by HRESIMS (m/z [M + Na]⁺ 539.3325, calcd 539.3343), corresponding to eight degrees of unsaturation. Inspection of the ^1H NMR spectrum of **13** revealed most of the structural features of the bishomoscalarane-type skeletons. Precise analysis of the ^{13}C NMR and HSQC data revealed the presence of a triplet methyl group (δ_{H} 0.67/ δ_{C} 8.80) and ketal carbon (δ_{C} 104.4), suggesting the A+D type skeleton shown in Figure 2. While most of the spectral data of **13** were identical to phyllofolactone H, the ketal carbon indicated the oxidation of C-24 to give a 24-hydroxy pentenolide E ring. This insight can be confirmed by the HMBC correlation from CH₃-26 (δ_{H} 1.48) to C-17 (δ_{C} 163.0)/C-24 (δ_{C} 104.4). The configuration of OH-24 was determined to be α -orientation by the NOESY correlation between H-16 β (δ_{H} 2.33) and CH₃-26. Thus, compound **13** was named 24 α -hydroxyphyllofolactone H [19].

Compound **14** was isolated as an inseparable mixture. Its molecular formula was determined as C₃₁H₄₆O₆ by HRESIMS (m/z [M + Na]⁺ 537.3175, calcd 537.3187), corresponding to nine degrees of unsaturation. Similar to compound **3**, the ^{13}C NMR spectrum of **14** showed a 1:1 splitting pattern corresponding to a mixture of two diastereomers. The distinctive spectral features of **14**, differentiated from **13**, were observed as the two sp² methines at δ_{H} 6.40/6.38 and 6.28/6.27, suggesting an unsaturated derivative of **13**. The methines belonged in the ^1H - ^1H COSY correlation for H-14 (δ_{H} 2.68/2.62)–H-15 (δ_{H} 6.40/6.38)–H-16 (δ_{H} 6.28/6.27) to identify the olefin at C-15 (Figure 6). In addition, **14** was determined to be a mixture of C-24 epimers, considering the largest splitting observed at CH₃-26 ($\Delta\delta_{\text{C}}$ 1.13 ppm).

Compound **15** was isolated as an inseparable mixture. Its molecular formula was determined to be C₃₂H₄₈O₆ by HRESIMS (m/z [M + Na]⁺ 551.3366, calcd 551.3343), corresponding to nine degrees of unsaturation. The MS data indicated an additional methylene relative to **14**, which was further supported by the change observed in the coupling pattern of the terminal methyl group of the side chain at C-12 from a doublet to triplet. The ^{13}C

NMR and HSQC data identified the methylene group at δ_{H} 1.51/1.25 and δ_{C} 29.5/29.4, which were involved in the spin system for $\text{H}_2\text{-2}'\text{-H-3}'\text{-H}_2\text{-4}'\text{-CH}_3\text{-5}'$ in the $^1\text{H}\text{-}^1\text{H}$ COSY spectrum to confirm the presence of the 3-hydroxypentanoate side chain. The orientation of the ester at C-12 was assigned as α by the NOESY signal between H-12 (δ_{H} 5.55/5.49) and $\text{CH}_3\text{-23}$ (δ_{H} 1.06/1.05), as well as the small coupling constant observed for H-12 (dd, $J = 2.3, 1.8$ Hz), to identify 12-*epi*-phyllactone D/E.

Interestingly, the identified structure was previously isolated as a mixture of C-24 epimers by Li et al. in 2007 [11], but our experimental ^{13}C NMR data showed some discrepancies with the previously reported data at C-9 (Δ 4.28 ppm), C-11 (Δ 2.7 ppm), C-12 (Δ 2.08 ppm), C-14 (Δ 4.46 ppm), and C-23 (Δ 4.35 ppm) (Figure 8a). In addition, another identification of 12-*epi*-phyllactone D/E was reported by Andersen et al. in 2009 [13]. Although they acquired almost identical experimental NMR data with ours rather than those reported by Li, the isolated compound was estimated to be same as Li's without consideration of the differences in NMR data (Tables S17 and S18, Supplementary Materials). Therefore, we investigated the variations in ^{13}C chemical shifts depending on the orientation of the substituents at C-12.

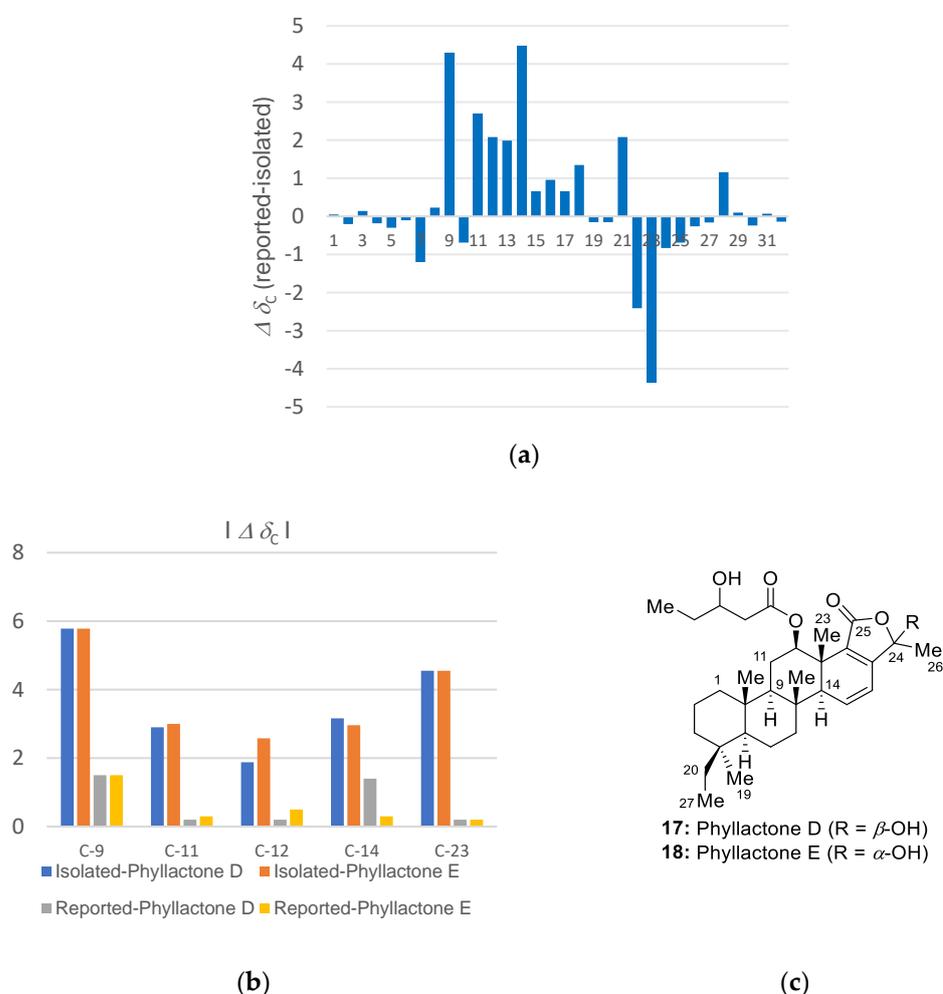


Figure 8. (a) The differences observed between the experimental and reported ^{13}C chemical shifts (Li et al.) of 15 in CDCl_3 . (b) The deviations in the ^{13}C NMR chemical shifts observed for reported (Li et al.) and isolated 15 relative to phyllactone D and phyllactone E. (c) The structures of phyllactone D and E.

Phyllactone D (17) and E (18), the reported 12 β -epimers of 15, were selected for comparison [25]. While C-12 in phyllactones D and E was observed at δ_{C} 75.1 and 75.8,

respectively, the corresponding chemical shifts of the reported and isolated **15** were observed at δ_C 75.3 and 73.2/73.1, respectively. The deviations observed for isolated **15** from phyllactone D/E became more obvious at C-9, C-14, and C-23 (Figure 8b). However, the reported chemical shifts for **15** were better aligned with those of phyllactone D/E. Furthermore, the differences in the ^{13}C NMR chemical shifts observed between isolated **15** and compounds **3**, **4**, **12**, and **14**, which share an identical substructure for the B-E ring system, showed negligible values (< 0.5 ppm) around the C-ring (Table S19, Supplementary Materials). Accordingly, isolated **15** is more likely to be the 12α -epimer. Even though Li determined the 12α -configuration observing the NOESY signal between H-12 and CH₃-23 and $J_{H-12-H-13}$ calculation (3.0, 2.5 Hz), the NMR database suggests that the compound previously reported by Li is presumed to be a mixture of phyllactone D (**17**) and E (**18**).

Compound **16** was isolated as a yellowish oil. Its molecular formula was determined as C₂₇H₄₀O₅ by HRESIMS (m/z [M + Na]⁺ 467.2762, calcd 467.2768), corresponding to eight degrees of unsaturation. The 1H NMR spectrum of **16** revealed five singlet methyl groups at δ_H 0.73, 0.74, 0.79, 0.85, and 0.86; one acetate group at δ_H 1.95; one oxymethine at δ_H 4.80; one olefin at δ_H 7.30; and one aldehyde at δ_H 9.41. The ^{13}C and HSQC NMR spectra showed characteristic peaks for the aldehyde carbon atom at δ_C 196.4, two carbonyl carbons at δ_C 169.6 and 169.6, one trisubstituted olefin at δ_C 145.8 and 124.2, and one oxymethine at δ_C 76.9. The HMBC correlation between the two methyl groups at δ_C 33.3 and 21.4 was identified as a characteristic feature of the 4-dimethyl-sesterterpenoid scaffold (Figure 9). The aldehyde at δ_H 9.41 exhibited a HMBC correlation with C-18 (δ_C 58.7) to be located at C-25. Additional HMBC correlations from H-18 (δ_H 3.07) to C-16 (δ_C 145.8)/C-17 (δ_C 124.2)/C-24 (δ_C 169.6), along with the 1H - 1H COSY cross peak for H-14–H₂-15–H-16, indicated the presence of the acid at C-24 and trisubstituted olefin at C-16. The acetate group (δ_C 21.3/ δ_H 1.95) was positioned at C-12, as indicated by the HMBC correlation from H-12 (δ_H 4.80) to C-1' (δ_C 169.6) and 1H - 1H COSY cross peak for H₂-11–H-12. Thus, the planar structure of **16** was found to be the deacetalization product of scalarin (**19**) [3]. The NOESY correlations between CH₃-23 (δ_H 0.86) and H-12/H-18 determined the configuration of the C-12 acetate and C-18 formyl groups as α .

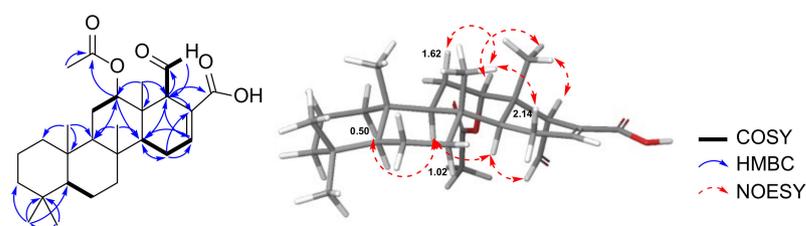


Figure 9. COSY, HMBC, and NOESY correlations observed for compound **16**.

Whereas scalarin (**19**) exists only in its hemiacetal form, the formation of 18 -*epi*-**19** or **19** via the acetalization of **16** was not observed. To rationalize the observed difference in reactivity, 18 -*epi*-**16** was proposed as a plausible precursor of scalarin, and geometrical optimization of **16** and 18 -*epi*-**16** was performed at the B3LYP/6-31G** level of theory. The atomic distance between O-24 to C-25 was calculated to be 3.37 Å for **16** and 2.68 Å for 18 -*epi*-**16** (Figure 10). This result suggests that 18 -*epi*-**16** can undergo acetalization to form scalarin because the β -orientation of C-25 increases its proximity to the acid at C-24. However, the acetalization of the 25α -formyl group in **16** will be restricted due to its remoteness to OH-24 to exist as its aldehyde form.

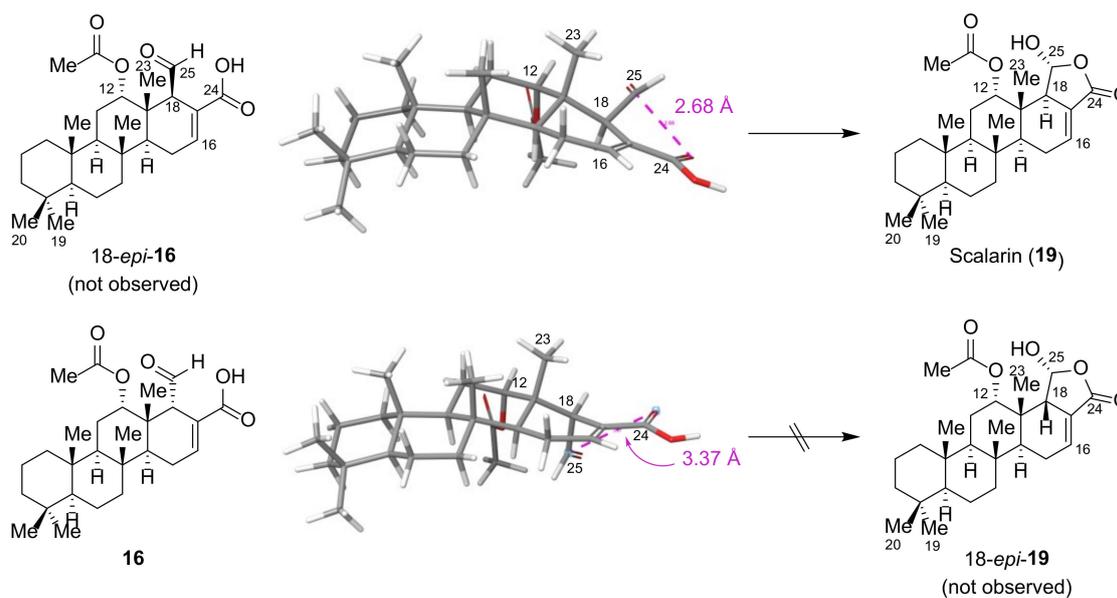


Figure 10. Atomic distances between O-24 and C-25 in compounds **16** and **18-epi-16**.

2.2. Biological Activity

The cytotoxicity of compounds **1–16** against MDA-MB-231 (a human breast cancer cell line) was evaluated to elucidate their potential as anticancer agents. Compounds **1–6**, **8**, **11**, and **13–15** exhibited moderate cytotoxicity with GI_{50} values ranging from 40 to 72 μM . Compounds **9**, **10**, **12**, and **16** were inactive toward the cancer cell line (Table 1). Among the bishomoscalaranes, the highest anticancer activity was exhibited by compound **7**, which has a relatively rare cyclopentenone E-ring (B+E type scaffold in Figure 2), with a GI_{50} value of 4.2 μM .

Table 1. The results of the cytotoxicity tests against MDA-MB-231 (human breast cancer cell line) obtained for compounds **1–16**.

Compound	Cytotoxicity ¹ (MDA-MB-231)							
	1	2	3	4	5	6	7	8
GI_{50} (μM)	69.94	43.38	72.49	54.02	53.58	50.8	4.21	53.55
Compound	9	10	11	12	13	14	15	16
GI_{50} (μM)	>100	>100	71.14	>100	50.71	63.54	40.82	>100

¹ Cisplatin (Sigma-Aldrich, St. Louis, MO) was used as a positive control ($GI_{50} = 1.31 \mu\text{M}$).

The highly diversified structures of the isolated scalaranes provided some information on their structure–activity relationship (SAR). The presence of the $\Delta^{15,16}$ -olefin generally had a detrimental effect that reduced the cytotoxicity in the range of 12–30 μM , as shown by the sets of **2** and **3** (B+D type), **13** and **14** (C+D type), and **13** and **14** (A+D type). Comparing **3** with **4** and **14** with **15**, the homologation of one methylene group at C-4' was beneficial toward increasing the activity to $\sim 20 \mu\text{M}$. A series of compounds **2**, **11**, and **13**, which only differ at the C-4 substituent, indicated the disadvantageous effect of oxidation at C-20 on the anticancer activity. The negative effect of oxidation at C-20 was also observed in the inactive series of compounds **9**, **10**, and **12**.

3. Materials and Methods

3.1. General Experimental Procedures

Specific optical rotations were collected on a Rudolph Research Analytical (Autopol III) polarimeter (Rudolph Research Analytical, Hackettstown, NJ, USA). IR spectra were

measured on a JASCO FT/IR-4100 spectrophotometer (JASCO Corporation, Tokyo, Japan). The 1D and 2D NMR spectra were taken in CDCl₃ using a Bruker 600 MHz spectrometer (Bruker BioSpin GmbH, Rheinstetten, Germany) at 297.1 K. ¹H NMR spectra were collected after 64–128 scans, and ¹³C NMR spectra were collected at a range of 10,000–15,000 scans depending on the sample concentrations. The mixing time for NOESY experiments was set as 0.3 s. Chemical shifts were reported in parts per million relative to CHCl₃ residue (δ_{H} 7.26, δ_{C} 77.1) in CDCl₃. High resolution mass-spectra were obtained on a Sciex X500R Q-TOF spectrometer (Framingham, MA, USA) equipped with an ESI source. MPLC was performed using the TELEDYNE ISCO CombiFlash Companion with the TELEDYNE ISCO RediSep Normal-phase Silica Flash Column (Teledyne ISCO, Lincoln, NE, USA). HPLC was performed on a PrimeLine Binary pump (Analytical Scientific Instruments, Inc., El Sobrante, CA, USA) utilizing silica columns (YMC-Pack Silica, 250 × 10 mm I.D., or 250 × 4.6 mm I.D., 5 μ m; YMC Co. Ltd., Kyoto, Japan), the Shodex RI-101 (Shoko Scientific Co. Ltd., Yokohama, Japan), or the UV-M201.

3.2. Biological Material

The marine sponge used in this study was collected in March 2016 from the Bohol province in the Philippines (N 9°43'31.39" E 124°32'19.86") at a depth of 15 m using scuba diving. The sponge was directly kept frozen at −20 °C until identified as *Dysidea* sp. and chemically analyzed. A voucher sample (163PIL-267) has been stored at the Marine Biotechnology Research Center, Korea Institute of Ocean Science & Technology (KIOST).

3.3. Extraction and Isolation

The lyophilized specimen (wet wt. 1.5 kg) was extracted with MeOH (2.0 L × 3) and CH₂Cl₂ (4.0 L × 2) at room temperature. The combined extracts were concentrated under reduced pressure. The dried residue (89.5 g) was partitioned with *n*-butanol (5.0 L) and water (5.0 L). The *n*-butanol layer was concentrated and further partitioned between *n*-hexane (3.0 L) and 15% aqueous methanol (3.0 L). A portion (12.2 g) of the concentrated 15% aqueous methanol fraction (31.7 g) was subjected to flash column chromatography over C18 (YMC Gel ODS-A, 60 Å, 230 mesh (YMC Co, Ltd., Kyoto, Japan)) with a stepwise gradient solvent system (50%, 60%, 70%, 80%, 90%, and 100% MeOH, acetone, and EtOAc).

The 80% MeOH fraction (612.7 mg) was further separated using MPLC on C18 with a gradient solvent system from 70% MeOH to 100% MeOH over 40 minutes to yield 4 fractions. The third subfraction (250.1 mg) was separated using HPLC (eluent 65% MeOH) to yield **8** (3.9 mg, t_{R} = 38 min), **9** (2.5 mg, t_{R} = 42 min), **10** (2.3 mg, t_{R} = 58 min), honulactone C (9.8 mg), and honulactone D (9.0 mg). The fourth subfractions (175.3 mg) was separated using HPLC (eluent 70% MeOH) to yield **11** (1.8 mg, t_{R} = 28 min), **12** (1.4 mg, t_{R} = 28 min), and honulactone I+J mixture (1.6 mg).

The 100% MeOH fraction (4.22 g) was further separated using MPLC on C18 with a gradient solvent system from 30% MeOH to 100% MeOH over 40 minutes to yield 4 fractions. The second fraction (2.49 g) was directly separated using MPLC on SiO₂ with a gradient solvent system from 70% HX to 100% EtOAc over 80 minutes to yield 8 subfractions (based on TLC analysis). Scalarin (**19**, 213.0 mg) was recrystallized from the second subfraction (572.8 mg) under the HX-EtOAc solvent conditions. The residue (250.0 mg) of the second subfraction was separated using HPLC (HX/acetone = 7/1) to yield **4** (5.2 mg, t_{R} = 54 min), **15** (5.5 mg, t_{R} = 48 min), phyllofolactone H (5.7 mg), and phyllofolactone I (11.5 mg). The third subfraction (295.5 mg) was separated using HPLC (HX/acetone = 7/1) to yield **1** (3.4 mg, t_{R} = 34 min), **3** (7.0 mg, t_{R} = 76 min), **14** (6.0 mg, t_{R} = 66 min), **13** (2.3 mg, t_{R} = 60 min), **16** (4.5 mg, t_{R} = 45 min, honulactone A (21.6 mg), honulactone B (26.2 mg), honulactone E+F mixture (21.4 mg), and phyllofolactone J+K (2.7 mg). The fourth subfraction (380.0 mg) was separated using HPLC (HX/acetone = 5/1) to yield **2** (3.1 mg, t_{R} = 36 min), **5** (1.6 mg, t_{R} = 31 min), **6** (5.6 mg, t_{R} = 32 min), **7** (5.0 mg, t_{R} = 30 min), and phyllofenone C (2.3 mg).

3.4. Assay

Human breast cancer MDA-MB-231 cells were purchased from the American Type Culture Collection (ATCC, Manassas, VA, USA) and were cultured in Dulbecco's modified Eagle medium (DMEM) supplemented with 10% fetal bovine serum (FBS, Gibco, Carlsbad, CA, USA), $1 \times$ antibiotic-antimycotic solution (Thermo Fisher Scientific, Waltham, MA, USA), and 25 mM HEPES (Gibco). Cultures were maintained in a humidified atmosphere of 95% air/5% CO₂ at 37 °C.

Cell viability was determined using a CCK-8 (Cell Counting Kit-8, Dojindo Laboratory, Kumamoto, Japan) assay according to the manufacturer's instructions. MDA-MB-231 cells were seeded at 5×10^3 cells/well into a 96-well plate and then were treated with various concentrations of compounds 1–16. Following treatment for 48 h, the cells were incubated with the CCK-9 solution, and the absorbance was measured at 450 nm using a SpectraMax i3 microplate reader (Molecular Devices, Sunnyvale, CA, USA). GI₅₀ values were calculated from a non-linear regression fit using GraphPad Prism version 9.2.0 (GraphPad Software, La Jolla, CA, USA).

1: colorless oil; $[\alpha]_D^{20} + 20.0$ (c 0.2, CHCl₃); IR (ATR) ν_{\max} 3131, 2954, 2929, 2581, 1770, 1734, 1452, 1381, 1261, 1176, 1027 cm⁻¹; ¹H NMR and ¹³C NMR, see Tables S1 and S2, Supplementary Materials; HRESIMS m/z 523.3382 [M + Na]⁺ (calcd for C₃₁H₄₈O₅Na, 523.3394).

2: colorless oil; $[\alpha]_D^{20} + 40.0$ (c 0.2, CHCl₃); IR (ATR) ν_{\max} 3735, 2954, 2925, 2851, 1731, 1689, 1452, 1374, 1278, 1176, 1014 cm⁻¹; ¹H NMR and ¹³C NMR, see Tables S1 and S2, Supplementary Materials; HRESIMS m/z 537.3167 [M + Na]⁺ (calcd for C₃₁H₄₆O₆Na, 537.3187).

3: colorless oil; $[\alpha]_D^{20} + 45.0$ (c 0.2, CHCl₃); IR (ATR) ν_{\max} 3727, 2957, 2922, 2865, 2848, 1738, 1657, 1458, 1371, 1286, 1621, 1173, 1031 cm⁻¹; ¹H NMR and ¹³C NMR, see Tables S1 and S2, Supplementary Materials; HRESIMS m/z 535.3011 [M + Na]⁺ (calcd for C₃₁H₄₄O₆Na, 535.3030).

4: colorless oil; $[\alpha]_D^{20} + 48.3$ (c 0.2, CHCl₃); IR (ATR) ν_{\max} 3735, 2954, 2922, 2869, 2855, 1731, 1685, 1452, 1374, 1286, 1173, 1021 cm⁻¹; ¹H NMR and ¹³C NMR, see Tables S1 and S2, Supplementary Materials; HRESIMS m/z 549.3163 [M + Na]⁺ (calcd for C₃₂H₄₆O₆Na, 549.3187).

5: colorless oil; $[\alpha]_D^{20} - 20.0$ (c 0.1, CHCl₃); IR (ATR) ν_{\max} 3727, 2961, 2929, 2851, 1734, 1678, 1452, 1367, 1254, 1173, 1027 cm⁻¹; ¹H NMR and ¹³C NMR, see Tables S1 and S2, Supplementary Materials; HRESIMS m/z 551.3310 [M + Na]⁺ (calcd for C₃₂H₄₈O₆Na, 551.3343).

6: amorphous powder; $[\alpha]_D^{20} + 45.0$ (c 0.2, CHCl₃); IR (ATR) ν_{\max} 3735, 2971, 2929, 2865, 1724, 1678, 1657, 1452, 1371, 1296, 1173, 1080, 1027 cm⁻¹; ¹H NMR and ¹³C NMR, see Tables S1 and S2, Supplementary Materials; HRESIMS m/z 509.3215 [M + Na]⁺ (calcd for C₃₀H₄₆O₅Na, 509.3237).

7: colorless oil; $[\alpha]_D^{20} + 33.3$ (c 0.1, CHCl₃); IR (ATR) ν_{\max} 3735, 2957, 2918, 2848, 1727, 1702, 1657, 1458, 1371, 1254, 1176, 1038 cm⁻¹; ¹H NMR and ¹³C NMR, see Tables S1 and S2, Supplementary Materials; HRESIMS m/z 503.3113 [M + Na]⁺ (calcd for C₃₁H₄₄O₄Na, 503.3132).

8: colorless oil; $[\alpha]_D^{20} + 8.3$ (c 0.2, CHCl₃); IR (ATR) ν_{\max} 3727, 2961, 2929, 2851, 1738, 1721, 1671, 1505, 1452, 1374, 1246, 1031 cm⁻¹; ¹H NMR and ¹³C NMR, see Tables S1 and S2, Supplementary Materials; HRESIMS m/z 611.3541 [M + Na]⁺ (calcd for C₃₄H₅₂O₈Na, 611.3554).

9: colorless oil; $[\alpha]_D^{20} - 6.7$ (c 0.1, CHCl₃); IR (ATR) ν_{\max} 2961, 2925, 2851, 1745, 1727, 1505, 1265, 1031 cm⁻¹; ¹H NMR and ¹³C NMR, see Tables S1 and S2, Supplementary Materials; HRESIMS m/z 522.3810 [M + NH₄]⁺ (calcd for C₃₀H₅₂NO₆, 522.3789).

10: colorless oil; $[\alpha]_D^{20} + 73.3$ (c, 0.1, CHCl₃); IR (ATR) ν_{\max} 3477, 3388, 2966, 2923, 2866, 1729, 1457, 1368, 1250 cm⁻¹; ¹H NMR and ¹³C NMR, see Tables S1 and S2, Supplementary Materials; HRESIMS m/z 597.3404 [M + Na]⁺ (calcd for C₃₃H₅₀O₈Na, 597.3398).

11: colorless oil; $[\alpha]_D^{20} + 40.0$ (*c* 0.1, CHCl₃); IR (ATR) ν_{\max} 2965, 2918, 2855, 1731, 1649, 1458, 1374, 1250, 1169, 1035 cm⁻¹; ¹H NMR and ¹³C NMR, see Tables S1 and S2, Supplementary Materials; HRESIMS *m/z* 597.3396 [M + Na]⁺ (calcd for C₃₃H₅₀O₈Na, 597.3398).

12: colorless oil; $[\alpha]_D^{20} + 71.7$ (*c* 0.2, CHCl₃); IR (ATR) ν_{\max} 3392, 2946, 2925, 2858, 1734, 1455, 1367, 1243, 1180 cm⁻¹; ¹H NMR and ¹³C NMR, see Tables S1 and S2, Supplementary Materials; HRESIMS *m/z* 595.3241 [M + Na]⁺ (calcd for C₃₃H₄₈O₈Na, 595.3241).

13: amorphous powder; $[\alpha]_D^{20} + 6.7$ (*c* 0.2, CHCl₃); IR (ATR) ν_{\max} 3727, 2957, 2929, 2848, 1727, 1657, 1455, 1374, 1278, 1176 cm⁻¹; ¹H NMR and ¹³C NMR, see Tables S1 and S2, Supplementary Materials; HRESIMS *m/z* 539.3325 [M + Na]⁺ (calcd for C₃₁H₄₈O₆Na, 539.3343).

14: colorless oil; $[\alpha]_D^{20} + 31.7$ (*c* 0.2, CHCl₃); IR (ATR) ν_{\max} 3717, 2961, 2925, 2872, 1727, 1649, 1458, 1374, 1275, 1257, 1176 cm⁻¹; ¹H NMR and ¹³C NMR, see Tables S1 and S2, Supplementary Materials; HRESIMS *m/z* 537.3175 [M + Na]⁺ (calcd for C₃₁H₄₆O₆Na, 537.3187).

15: colorless oil; $[\alpha]_D^{20} + 30.0$ (*c* 0.2, CHCl₃); IR (ATR) ν_{\max} 3735, 2957, 2925, 2869, 1731, 1448, 1363, 1278, 1176, 1014 cm⁻¹; ¹H NMR and ¹³C NMR, see Tables S1 and S2, Supplementary Materials; HRESIMS *m/z* 551.3366 [M + Na]⁺ (calcd for C₃₂H₄₈O₆Na, 551.3343).

16: colorless oil; $[\alpha]_D^{20} - 70.0$ (*c* 0.2, CHCl₃); IR (ATR) ν_{\max} 3727, 2961, 2922, 2851, 1738, 1646, 1452, 1381, 1225, 1021 cm⁻¹; ¹H NMR and ¹³C NMR, see Tables S1 and S2, Supplementary Materials; HRESIMS *m/z* 467.2762 [M + Na]⁺ (calcd for C₂₇H₄₀O₅Na, 467.2768).

4. Conclusions

A total of 15 novel scalaranes **1–14** and **16**, including 14 bishomoscalaranes and one scalarin derivative, has been isolated from the marine sponge, *Dysidea* sp., and characterized using a combination of 1D and 2D NMR spectroscopy. The isolation and structural identification of compound **15** resulted in the reassignment of the previously characterized 12-*epi*-phyllactone D/E. The actual structure of the reported 12-*epi*-phyllactone D/E was determined to be a mixture of known phyllactones D and E through the precise analysis of the experimental and reported ¹³C chemical shifts. In addition, the effect of the C-18 configuration in **16** on the formation of the hemiacetal E-ring was rationalized by measuring the atomic distances between C-25 and O-24 in **16** and 18-*epi*-**16**. Finally, the evaluation of the anticancer activities of compounds **1–16** against MDA-MB-231 revealed that compound **7** exhibited significant cytotoxicity with a GI₅₀ value of 4.2 μM. Detailed studies to elucidate the biological mechanism of **7** are currently underway in our laboratory.

Supplementary Materials: The following are available online at <https://www.mdpi.com/article/10.3390/md19110627/s1>, I. Experimental procedure; Figure S1: Structures of known compounds isolated from *Dysidea* sp., Tables S1–S16: ¹³C/¹H chemical shifts for **1–16**, Figure S2: Key ¹H-¹H COSY, HMBC correlations of **1–16**, Figure S3: Key NOESY correlations of **1–16**, Tables S17–S19, Figures S4 and S5: Comparison of ¹³C/¹H chemical shifts for 12β epimer (**17**, **18**) and 12α epimer (**15**) reported by Li et al., Andersen et al., and our experiment, II. Computational methods; Supporting Information II; Figures S-1–S-112: ¹H NMR, ¹³C NMR, COSY, HSQC, HMBC, NOESY, and HRMS spectra of **1–16**.

Author Contributions: A.-Y.S. worked on isolation and structure elucidation. A.S. and C.C. performed the biological evaluation. J.L. collected the marine sponge and supervised the whole research work. All authors have read and agreed to the published version of the manuscript.

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Conflicts of Interest: The authors declare no conflict of interest.

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