

Short Note

(1*R*,5*S*)-6-(4-Methyl-2-oxo-2,5-dihydrofuran-3-yl)-3phenyl-4-oxa-2,6-diazabicyclo[3.2.0]hept-2-en-7-one

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Abstract: Efficient large-scale and feasible industrial synthesis of the 1-oxacephem core structure from 6-aminopenicillanic acid (6-APA) has been reported for several decades. Via the industrial synthesis route, a byproduct (compound 9) containing a butenolide unit was purified and characterized by NMR and HRMS in this work. It is worth noting that compound 9 is an entirely new compound. Additionally, a plausible mechanism and effects on the formation of 9 by different Lewis acids were proposed. The discovery of compound 9 could improve the purity of this feasible industrial synthesis and provide considerable cost savings.

Keywords: industrial feasible synthesis; 6-APA; butenolide; 1-oxacephem

1. Introduction

Antibacterial substances are of great importance and necessity in treating infectious diseases caused by pathogenic bacteria [1–3]. Due to its unique antimicrobial activity and novel structure among the synthetic antibiotics, the 1-oxacephem core structure as an important pharmaceutical scaffold has attracted immense interest from medicinal chemists [4–6]. A variety of synthetic compounds prepared from the 1-oxacephem intermediate, including prominent antibiotics such as Flomoxef, Moxalactam, and OCP-9-176 (Figure 1), have a broad spectrum of activity against Gram-positive and Gram-negative aerobic and anaerobic bacteria [7–9].

A feasible industrial route by which to synthesize 1-oxacephem **8** in good yield starting from commercially available 6-aminopenicillanic acid (6-APA) (Figure 2) was reported by Nagata of the Shionogi company [10,11]. In this sophisticated method designed to retain all the carbon atoms, preparing epioxazolinoazetidinones having an unconjugated ester moiety at the β -lactam nitrogen was a breakthrough. However, byproducts of and probable mechanisms in this industrial synthesis of 1-oxacephem **8** have not been systematically explored. In this work, we focused on the byproduct **9** ((1*R*,5*S*)-6-(4-methyl-2-oxo-2,5-dihydrofuran-3-yl)-3-phenyl-4-oxa-2,6-diazabicyclo[3.2.0]hept-2-en-7-one).



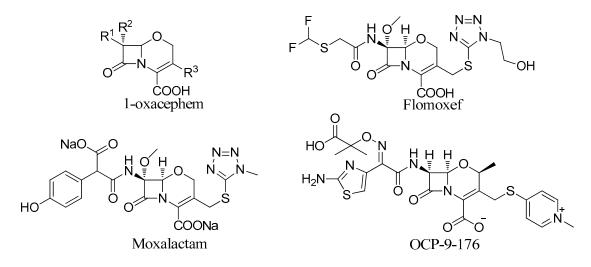


Figure 1. Synthetic 1-oxacephem antibiotics.

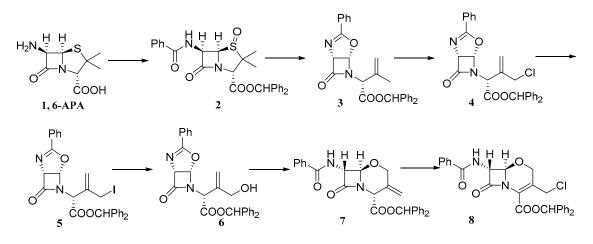


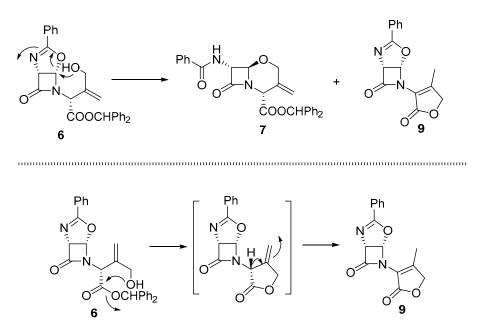
Figure 2. Feasible industrial synthesis of 1-oxacephem 8.

2. Results and Discussion

Intramolecular etherification proceeded from the less-hindered β side with stereoselectivity to furnish a versatile exomethylene intermediate 7 in 79% yield and accompanied by a byproduct 9 in 15% yield. The probable mechanism which afforded the butenolide 9 catalyzed by boron fluoride ethyl ether involved two reactions: (a) an intramolecular transesterification and (b) isomerization of the double bond promoted by a Lewis acid (Scheme 1).

Systematic studies of the reaction conditions to obtain byproduct **9** in highest yield revealed that Lewis acids played key roles (Table 1). When the reaction was catalyzed by $BF_3 \cdot Et_2O$ and $Yb(OTf)_3$, the major product was compound **7** (Table 1, entries 1 and 6) with yields of 90% and 56%, respectively. Our best result was achieved with $BF_3 \cdot Et_2O$ at 25 °C, conditions in which **7** was formed in 90% yield, along with only a small amount of readily separable **9** (Table 1, entry 1). When the Lewis acid was changed to LiCl or ZnCl₂, byproduct **9** was obtained as a dominant product (Table 1, entries 2, 3, 4).

To our surprise, when EtOH was used as the solvent instead of EtOAc (Table 1, entry 3), the yield of byproduct **9** increased to 92%. These results suggested that ethyl alcohol and Lewis acid LiCl were suitable for this transformation to generate the byproduct **9** in an excellent yield.



Scheme 1. The probable mechanism of formation of 9 catalyzed by a Lewis acid ($BF_3 \cdot Et_2O$).

Entry	Lewis Acid a	Temperature	Solvent	Yields of 7 b	Yields of 9 b
1	BF ₃ ·Et ₂ O	25 °C	EtOAc	90%	1%
2	LiCl	25 °C	EtOAc	33%	60%
3	LiCl	25 °C	EtOH	1%	92%
4	ZnCl ₂	25 °C	EtOAc	29%	65%
5	FeCl ₃	25 °C	EtOAc	46%	42%
6	Yb(OTf) ₃	25 °C	EtOAc	56%	36%

Table 1. Screening of the reaction conditions.

^a 1 mol % Lewis acid was used. ^b Isolated yields.

3. Materials and Methods

3.1. General Information

All the reactions were monitored by thin-layer chromatography. The byproducts were purified by column chromatography over silica gel (Qingdao Haiyang Chemical Co., 200–300 mesh, Qingdao, China). Melting points were determined on a Beijing Keyi XT4A apparatus (Beijing synthware glass, Beijing, China). All NMR spectra were recorded with a Bruker DPX 400 MHz spectrometer (Agilent, Santa Clara, CA, USA) with TMS as the internal standard. Chemical shifts are given as δ ppm values relative to TMS. Mass spectra (MS) were recorded on an Esquire 3000 mass spectrometer (Varian, Palo Alto, CA, USA) by electrospray ionization (ESI).

3.2. Synthesis of (1R,5S)-6-(4-Methyl-2-oxo-2,5-dihydrofuran-3-yl)-3-phenyl-4-oxa-2,6-diazabicyclo[3.2.0] hept-2-en-7-one (9)

A solution of LiCl (1 mol %) was added to intermediate **6** (1 eq, 1 g) in EtOH (10 mL) in a round-bottom flask and reacted at room temperature for 7 h. The reaction system was evaporated to give a residue, which was purified by silica gel flash column chromatography (EtOAc/*n*-hexane = 1:7) to afford the product **9**, yield 92%. White solid; m.p. 199.2–200.3 °C; $[\alpha]_D^{25}$ + 18.9° (C 1.05, CHCl₃); ¹H-NMR (400 MHz, CDCl₃) δ 8.02 (d, *J* = 7.4 Hz, 2H), 7.54 (t, *J* = 7.4 Hz, 1H), 7.45 (t, *J* = 7.6 Hz, 2H), 6.81 (d, *J* = 3.3 Hz, 1H), 5.45 (d, *J* = 3.3 Hz, 1H), 4.73 (s, 2H), 2.18 (s, 3H); ¹³C-NMR (100 MHz, CDCl₃) δ

168.64, 167.16, 163.73, 148.48, 132.40, 128.59, 128.50, 126.73, 119.77, 84.70, 82.26, 71.77, 13.17; HRMS (ESI): m/z calcd for C₁₅H₁₂N₂O₄ (M + H)⁺, 285.0875; found, 285.0880.

4. Conclusions

In summary, byproduct **9** was obtained in the industrial synthesis of the 1-oxacephem core structure from 6-aminopenicillanic acid. To the best of our knowledge, this is the first report about the byproduct **9**. We explored the effects on the formation of azetidinone-fused butenolide **9** caused by different Lewis acids and explored its probable mechanism of formation. The study of byproduct **9** is valuable for efficient large-scale and feasible industrial synthesis of the 1-oxacephem core structure.

Supplementary Materials: Supplementary materials are available online.

Author Contributions: D.-J.F. and E.Z. designed and synthesized the compounds. V.P., M.-A.T., L.S. and X.Z. revised the manuscript. D.-J.F. wrote the manuscript and H.-M.L. was responsible for the correspondence of the manuscript. All authors read and approved the final manuscript.

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Conflicts of Interest: The authors declare no conflict of interest.

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