Short Note

**N<sup>1</sup>-Benzyliodene-N<sup>2</sup>-(2-((2-(benzylideneamino)ethyl)amino)ethyl)amino)ethyl)ethane-1,2-diamine**

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**Abstract:** A tetraethylene pentamine-diamine (L<sup>4</sup>), the biggest compound in the family of dibenzylated diimine-polyamines (L<sup>1</sup>–L<sup>4</sup>) has been synthesized by classical Schiff-base reaction between benzaldehyde and the diamine tetraethylenepentamine, and the structure was confirmed by elemental analysis, ESI-MS spectrometry and by IR and <sup>1</sup>H-NMR spectroscopy.

**Keywords:** imine compounds; amine compounds; polyamines; dibenzylated

Improved understanding of the role of polyamines in metabolism [1,2], and the differences in polyamine biology between normal cells and tumor cells [3], have increased current interest in this type of compounds in the field of drug development [4,5]. The activity of polyamines is very much dependent on their charge and the charge density they display at physiological pH [6].

During the last ten years, some of us have been involved in the studies of many different water-soluble bis-chromophoric polyamines as fluorescent chemosensors [7–10]. However, more recently studies in new active MALDI-TOF-MS matrices reveals that the introduction of imine groups into the polyamine chain increases the energy absorbed in the UV region, and consequently, the potential application as a MALDI matrix increase [11,12].
Following the method previously reported by Bernardo et al. for polyamine systems [13], in this paper we describe the synthesis and characterization of the tetraethylene pentamine-diamine (L4), derived from benzaldehyde and the diamine tetraethylenepentamine. The broader applicability of this method was demonstrated by the synthesis of a few related compounds (L1–L3) [14] (See scheme 1).

**Scheme 1.** Schematic representation of compounds L1–L4.

![Scheme 1](image)

Experimental

A solution of benzaldehyde (0.129 g, 1.225 mmol) in absolute ethanol (20 mL) was added dropwise to a refluxing solution of tetraethylenepentamine (0.115 g, 0.612 mmol) in the same solvent (15 mL). The resulting solution was gently refluxed with magnetic stirring for 4 h. The colour changed from colourless to yellow. The solution was concentrated under vacuum to 1/3 of its volume. Diethyl ether was added to the solution and then cooled at 0 °C during 24 h. The yellow crystals formed were filtered off and dried under vacuum. At room temperature the crystals were not stable and a yellow oil was obtained.

L4: N1-Benzylidene-N2-(2-((2-((2-(benzylideneamino)ethyl)amino)ethyl)amino)ethyl)ethane-1,2-diamine

Yield: 125 mg (56%).

ESI-MS: m/z (rel.int%): 366.26 (100) ([M+H]+).

1H-NMR (CDCl3): δ = 8.3 (s, 2H, N=C–H); 7.5–7.7 (m, 4H, C-Har); 7.4–7.1 (m, 6H, C-Har); 3.8–3.2 (m, 4H, CH2); 2.9–2.1 (m, 12H, CH2) ppm.

IR (cm⁻¹): 1658 (C=N, Imine), 1589, 1492 (C=C, Ar).

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References and Notes


14. The smaller parent compounds derived from 1,2-ethanediamine (L₁), diethylenetriamine (L₂), and triethylenetetramine (L₃) were obtained by a similar methodology, using 0.038, 0.063 and 0.089 g of diamine, respectively. Compound L₁: *N₁,N₂*-Dibenzylideneethane-1,2-diamine; Yield: 121 mg (84%); ESI-MS: *m/z* (rel. int%): 237.13 (100) ([M+H]+); ¹H NMR (CDCl₃): δ = 8.1 (s, 2H, N=C–H); 7.8 (m, 4H, C-Har); 7.2 (m, 6H, C-Har); 3.8 (s, 4H, CH₂) ppm; IR (cm⁻¹): 1647 (C=N, Imine), 1599, 1498 (C=C, Ar); Elemental analysis: Calcd for C₁₆H₁₆N₂: C, 81.32; H, 6.82; N, 11.85. Found: C, 80.87; H, 7.02; N,12.05. Compound L₂: *N₁*-Benzylidene-*N₂*-(2-(benzylideneamino)-ethyl)ethane-1,2-diamine; Yield: 103 mg (71%); ESI-MS: *m/z* (rel. int%): 279.17 (100) ([M+H]+); ¹H-NMR (CDCl₃): δ = 8.2 (s, 2H, N=C–H); 7.8–7.6 (m, 4H, C-Har); 7.4–7.2 (m, 6H, C-Har); 3.8 (m, 4H, CH₂); 2.9 (m, 4H, CH₂) ppm; IR (cm⁻¹): 1649 (C=N, Imine), 1586, 1491 (C=C, Ar); Elemental analysis: Calcd for C₁₈H₂₁N₃: C, 77.38; H, 7.58; N, 15.04. Found: C, 77.16; H, 8.03; N, 15.34. Compound L₃: *N₁*,*N¹*’-(Ethane-1,2-diyl)bis(*N₂*-benzylideneethane-1,2-diamine); Yield: 132 mg (89%); ESI-MS: *m/z* (rel. int%): 323.22 (100) ([M+H]+); ¹H-NMR (CDCl₃): δ = 8.1 (s, 2H, N=C–H); 7.7–7.5 (m, 4H, C-Har); 7.4–7.1 (m, 6H, C-Har); 3.7–3.4 (m, 2H, CH₂); 2.9–2.1 (m, 8H, CH₂) ppm; IR (cm⁻¹): 1656 (C=N, Imine), 1576, 1499 (C=C, Ar); Elemental analysis: Calcd for C₂₀H₂₆N₄: C, 74.50; H, 8.13; N, 17.38. Found: C, 74.78; H, 8.16; N, 17.49.

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