

Short Note

Tribenzamidomethyl Hydrazine

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Abstract: A new tribenzamidomethyl hydrazine was synthesized and its IR, ¹H NMR, ¹³C NMR and MS spectroscopic data are presented.

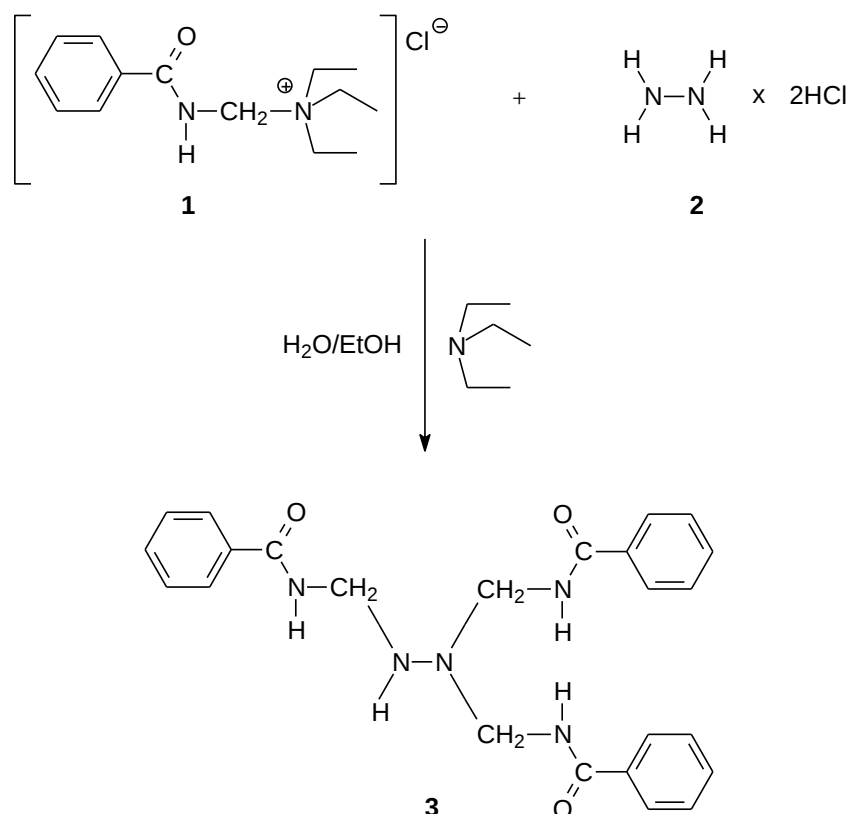
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Many hydrazine derivatives are known to exhibit significant biological activity and several compounds with hydrazine moiety were shown to be effective for treatment of tuberculosis, Parkinson's disease and hypertension [1]. There is evidence that some hydrazines displaying neuroprotective properties can be used as antidepressant drugs [2], while for some trisubstituted hydrazines antimicrobial effect was observed [3]. The hydrazine moiety has been used for modification of peptides as well. For example, novel building blocks for peptide modification have been designed and synthesized to obtain β -sheet tripeptide mimics that comprise the azapeptide modification at one end [4]. Azapeptides, hydrazine-based peptidomimetics, were found to be potent agents against hepatitis [5], AIDS [6] and SARS [7]. In addition, hydrazine derivatives have been used for derivatization of nanostructures [8]. Therefore, the synthesis of hydrazine derivatives is a matter of significant interest from both theoretical and practical perspectives. On the other hand, the promising usefulness of the benzamidomethyl derivatives as biologically active products [9-11] and their implication for pro-drug design [12-15] have been previously reported. Our good results with the benzamidomethylation of amines and hydrazides [16-18] using (benzamidomethyl)triethylammonium chloride (**1**), prompted us to synthesize some new products containing hydrazine and

benzamidomethyl moieties with an aim to obtain new compounds with potential biological activity. In this short note, the synthesis of a novel tribenzamidomethyl hydrazine is reported.

Benzamidomethylation reaction of hydrazine, using **1** as a reagent, was performed smoothly in ethanol/aqueous media, under mild reaction conditions at ambient temperature (Scheme 1). The title compound (**3**) was obtained without using a catalyst. Crystals of the product were easily isolated by simple filtration and stored in a dark place at a temperature below 18 °C.

Scheme 1. Synthetic route to the title compound **3**.



The structure of the newly synthesized compound was confirmed by ^1H NMR, ^{13}C NMR, FTIR spectroscopy and high-resolution mass spectrometry as well.

Synthesis

Materials

All the reagents and solvents were obtained from commercial sources and used without further purification. (Benzamidomethyl)triethylammonium chloride was prepared according to the literature method [19]. Synthetic reaction was monitored by thin-layer chromatography with visualization by UV-light or staining with iodine vapors.

Instrumentation

Melting points were determined on a Reichert heating plate and are uncorrected. NMR spectra were recorded on a Bruker 250 MHz instrument using $\text{DMSO}-d_6$ as solvent and tetramethylsilane as internal

standard. Infrared spectra (KBr pellets) were measured on a Perkin-Elmer System 2000 FT-IR. The ESI mass spectra were recorded on a QToF premiere (Waters) equipped with an Acquity UPLC (Waters)." UPLC – Gradients and solvents: A = water + 0.1% formic acid; B = methanol + 0.1% formic acid; 0 min = 95% A; 2.5 min = 5% A, 6.5 min = 5% A; 6.6 min = 95% A, 8 min = 95% A.

Synthesis of tribenzamidomethyl hydrazine (3)

To a mixture of (benzamidomethyl)triethyl-ammonium chloride (465.1 mg, 1.72 mmol) in 8 mL of water and 0.2 mL of triethylamine, a solution of hydrazine dihydrochloride (59.8 mg, 0.57 mmol) in 8 mL of ethanol was added. Afterwards, the reaction mixture was stirred overnight at room temperature. In this phase, colorless crystals were obtained, which were collected by vacuum filtration. The purification was performed by dissolving the product in ethanol and precipitation with water.

Melting point of pure crystals: 123–125 °C. Typical yield 85%.

FTIR (KBr, cm^{-1}): $\nu(\text{N-H})$ 3294, 3359, Amide I 1642, Amide II 1542;

$^1\text{H-NMR}$ (250 MHz, $\text{DMSO-}d_6$): δ/ppm 8.81 (t, $J = 6.0$ Hz, 1H, $\text{CONHCH}_2\text{NHN}$, overlap), 8.78 (t, $J = 6.0$ Hz, 2H, CONHCH_2NN , overlap) 7.37–7.88 (m, 15H, ArH), 4.51 (t, $J = 5.0$ Hz, 1H, NNH), 4.36 (d, $J = 6.0$ Hz, 4H, CH_2NN , overlap), 4.34 (t, $J = 5.5$ Hz, 2H, CH_2NHN , overlap);

^{13}C NMR (62.89 MHz, CDCl_3): δ/ppm 167.34, 166.41 C=O; 55.68, 58.68 CH_2 ; Ar: 134.41; 131.41, 131.29, 128.28;

HRMS (ESI, pos) m/z : 432.1900 ($\text{M}+\text{H}$) $^+$; 454.1701 ($\text{M}+\text{Na}$) $^+$.

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