

Short Note

Ethyl 3,5-Dimethyl-4-[(4-phenyl-1,3-thiazol-2-yl)carbamoyl]-1*H*-pyrrole-2-carboxylate

Akbar Idhayadhulla, Radhakrishnan Surendra Kumar and Abdul Jamal Abdul Nasser *

P.G & Research Department of Chemistry, Jamal Mohamed College, Tiruchirappalli-620020, Tamil Nadu, India

* Author to whom correspondence should be addressed; E-Mail: jamal_abdulchem@ymail.com.

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Abstract: A new compound, ethyl 3,5-dimethyl-4-[(4-phenyl-1,3-thiazol-2-yl)carbamoyl]-1*H*-pyrrole-2-carboxylate (**3**) was synthesized by the amination method. The synthesized compound (**3**) was characterized by IR, ¹H-NMR, ¹³C-NMR, mass spectral data and elemental analysis.

Keywords: diethyl 3,5-dimethyl-1*H*-pyrrole-2,4-dicarboxylate; 4-phenyl-1,3-thiazol-2-amine; amination reaction

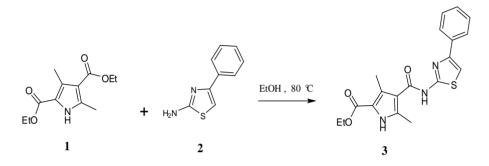
1. Introduction

Thiazoles are an important class of natural and synthetic compounds. Thiazole derivatives display a wide range of biological activities such as anesthetic [1] and anti-inflammatory [2]. In view of their pharmaceutical applications, the synthesis of thiazoles is important. Here, the preparation and characterization of a new thiazole derivative is reported.

2. Results and Discussion

The two educts, diethyl 3,5-dimethyl-1*H*-pyrrole-2,4-dicarboxylate (**1**) and 4-phenyl-1,3-thiazol-2amine (**2**) were synthesized according to previously reported methods [3–6]. The title compound, ethyl 3,5-dimethyl-4-[(4-phenyl-1,3-thiazol-2-yl)carbamoyl]-1*H*-pyrrole-2-carboxylate (**3**) was synthesized by an amination reaction [7,8]. The H-NMR spectrum of compound (**1**) showed a quartet at δ 4.20 and a triplet at δ 1.30, corresponding to -COOC<u>H</u>₂CH₃ and -COOCH₂C<u>H</u>₃ protons in the 2- and 4-position of the pyrrole ring. The H-NMR spectrum (Figure 2) of compound (**3**) showed a singlet at δ 8.11, corresponding to the -CON<u>H</u> proton in the 4-position of the pyrrole ring. A doublet observed at δ 3.87 and δ 2.18 was attributed to the CH₃ protons at 3- and 5-position in the pyrrole ring. A 1D NOE spectrum (Figure 3a and 3b) of compound (**3**) showed interactions between the amide-NH proton and the CH₃ protons at the 3- and 5-position, respectively, thus confirming that the -CONH group is located in 4-position of the pyrrole ring. No signal enhancement was obtained on irradiation of the NH proton at the 1-position of the pyrrole ring. The ¹C-NMR spectrum (Figure 4) of compound (**3**) showed signals at δ 156.9 and δ 166.0, corresponding to the –<u>C</u>OOEt group at the 2-position and the –<u>C</u>ONH group at the 4-position of the pyrrole ring. The mass spectra (EI) of compound (**3**) showed the molecular ion peak at *m/z* 370.43 (M⁺ + 1, 12%), consistent with the assigned molecular formula C₁₉H₁₉N₃O₃S.

Scheme 1. Synthesis of the title compound (3).



Scheme 2. Proposed conformation of compound (3).

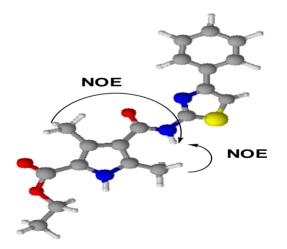
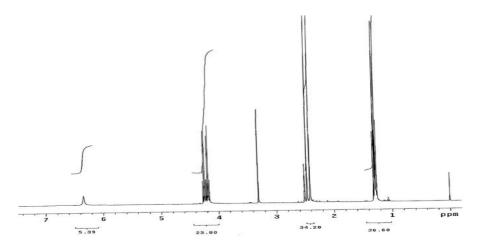


Figure 1. H-NMR spectrum of compound (1) (400 MHz, DMSO-*d*₆).



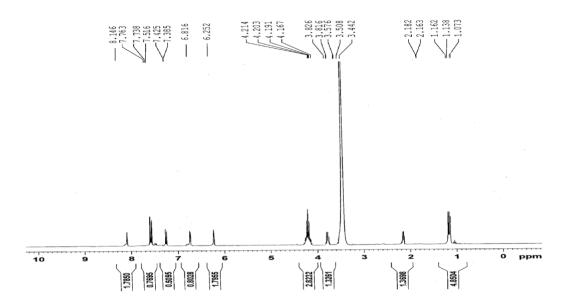
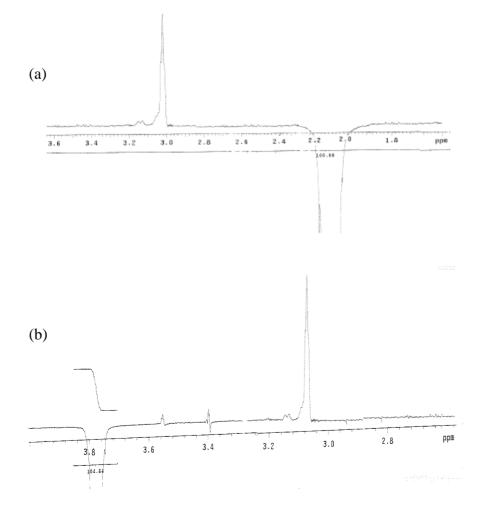


Figure 2. H-NMR spectrum of compound (3) (400 MHz, DMSO-*d*₆).

Figure 3. (a) 1D NOE NMR for CH_3 at 5-position, with irradiation of neighboring CONH proton in compound (3) recorded at 400 MHz in DMSO- d_6 . (b) 1D NOE NMR for CH_3 at 2-position, irradiated with neighboring CONH proton in compound (3) recorded at 400 MHz, in DMSO- d_6 .



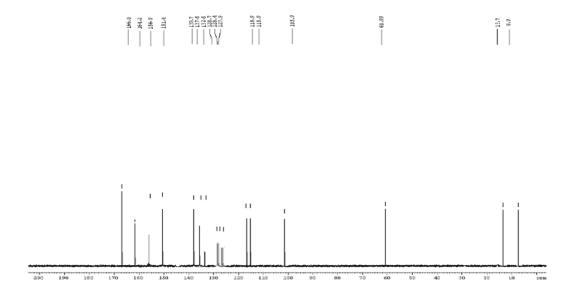
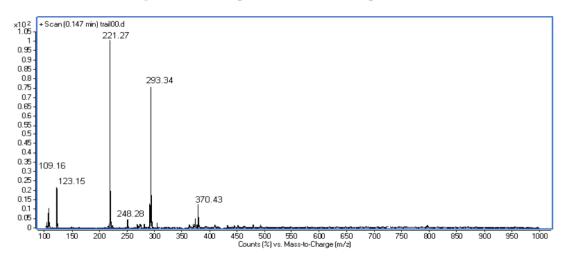


Figure 4. ¹³C-NMR spectrum of compound (3) (100 MHz, DMSO- d_6).

Figure 5. Mass spectrum (EI) of compound (3).



3. Experimental

The melting point was determined in an open capillary tube and it is uncorrected. The IR spectrum was recorded for a KBr pellet on a Shimadzu 8201pc (4000–400 cm⁻¹). The H-NMR and ¹³C-NMR spectra were recorded on a Bruker DRX-400 and a Varian Mercury Plus 400 at 400 MHz (¹H) and 100 MHz (¹³C), respectively. The mass spectrum (EI) was recorded on a Jeol JMS D-300 spectrometer operating at 70 eV. Elemental analysis (C, H, N and S) were carried out using a Varian Elemental Analyzer EL III. The purity of the compound was checked by thin layer chromatography (TLC) with silica gel plates.

3.1. Synthesis of ethyl 3,5-dimethyl-4-[(4-phenyl-1,3-thiazol-2-yl)carbamoyl]-1H-pyrrole-2-carboxylate (**3**)

A mixture of diethyl 3,5-dimethyl-1H-pyrrole-2,4-dicarboxylate (1) (2.39 g, 0.01 mol), 4-phenyl-1,3-thiazol-2-amine (2) (1.78 g, 0.01 mol) and absolute ethanol (30 mL) was heated under

reflux for 48 h. The reaction mixture was cooled and poured into crushed ice. The obtained solid was filtered off and washed with water. The filtered solid was purified by recrystallisation from absolute ethanol to give the title compound (**3**) as a yellow solid (1.20 g, 28%).

Melting point: 147 °C

IR (KBr, cm⁻¹): v 3350 (N-H_{str}), 3050 (C-H_{str} of phenyl ring), 2953 (C-H_{str} of CH₃), 1755 (C=O, ester), 1685 (-HN-C=O), 1626 (C=N), 712 (C-S-C).

¹H-NMR (DMSO- d_6 , 400 MHz): δ 8.11 (s, 1H, NH-CO), 7.32–7.74 (m, 5H, Ph), 6.25 (s, 1H, pyrrole NH), 6.85 (s, 1H, thiazole-H), 4.20 (q, 2H, J = 18.8 Hz, OCH₂CH₃), 3.87 (d, 3H, J = 7.6 Hz, 3-CH₃), 2.18 (d, 3H, J = 4 Hz, 5-CH₃), 1.30 (t, 3H, J = 16.4 Hz, OCH₂CH₃).

¹³C-NMR (DMSO-*d*₆, 100 MHz): δ 166.0 (NHCO), 164.2 (thiazole 2-C), 156.9 (C-<u>C</u>OOEt at 2-position in pyrrole ring), 151.4 (thiazole 4-C), 137.5 (pyrrole 3-C), 139.7 (pyrrole 5-C), 127.9, 128.4, 128.7, 132.1 (phenyl-C), 118.8 (pyrrole 2-C), 115.7 (pyrrole 4-C), 105.0 (thiazole 5-C), 60.2 (O<u>C</u>H₂CH₃), 13.8 (5-CH₃), 8.9 (3-CH₃).

MS (EI): m/z = 370.43 (M⁺ + 1, 12%), 293.34 (75%), 248.28 (5%), 221.27 (100%), 123.15 (22%), 109.16 (10%).

Elemental analysis: Calcd. for $C_{19}H_{19}N_3O_3S$ (MW = 369.45): C, 61.76%; H, 5.18%; N, 11.37%; S, 8.67%. Found: C, 61.71%; H, 5.14%; N, 11.33%; S, 8.64%.

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References and Notes

- 1. Geronikaki, A.; Theophilidis, G. Synthesis of 2-(aminoacetylamino)thiazole derivatives and comparison of their local anaesthetic activity by the method of action potential. *Eur. J. Med. Chem.* **1992**, *27*, 709–716.
- 2. Giridhar, T.; Reddy, R.B.; Prasanna, B.; Chandra Mouli, G.V.P. Amino thiozoles: Part I-Synthesis and pharmacological evalution. *Indian J. Chem.* **2001**, *40B*, 1279–1281.
- 3. Fischer, H. 2,4-Dimethyl-3,5-dicarbethoxypyrrole. Org. Synth. 1935, 15, 17.
- 4. Fabiano, E.; Golding, B.T. On the mechanism of pyrrole formation in the Knorr pyrrole synthesis and by porphobilinogen synthesis. *J. Chem. Soc. Perkin Trans.1* **1991**, *12*, 3371–3375.
- 5. King, L.C.; Hlavacek, R.J. Reaction of ketones with iodine and thiourea. *J. Am. Chem. Soc.* **1950**, 72, 3722–3725.
- 6. Siddiqui, H.L.; Iqbal, A.; Ahmad, S.; Weaver, G.W. Synthesis and spectroscopic studies of new Schiff bases. *Molecules* **2006**, *11*, 206–211.
- 7. Fadda, A.A.; Bondock, S.; Rabie, R.; Etman, H.A. Cyanoacetamide derivatives as synthons in heterocyclic synthesis. *Turk. J. Chem.* **2008**, *32*, 259–286.

8. Refaat, H.M.; Moneer, A.A.; Khalil, O.M. Synthesis and antimicrobial activity of certain novel quinoxalines. *Arch. Pham. Res.* **2004**, *27*, 1093–1098.

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