

## *Supplementary Materials*

**Table S1.** Calculated enthalpies, entropies, and Gibbs free energies (in Hartree) for optimized equilibrium model structures (H, S, and G, respectively).

Model Structure	H	S	G
Cyclohexane	-40.953141	74.604	-40.988588
MeOH	-23.932299	56.951	-23.959359
N <sub>2</sub>	-19.763962	45.875	-19.785759
O <sub>2</sub>	-31.778169	49.153	-31.801523
He	-2.905283	30.125	-2.919596
Ultem	-339.482655	232.263	-339.593011
Ultem–cyclohexane	-380.450769	269.864	-380.578990
Ultem–MeOH	-363.430548	259.779	-363.553977
Ultem–N <sub>2</sub>	-359.251122	253.892	-359.371754
Ultem–O <sub>2</sub>	-371.264797	255.109	-371.386008
Ultem–He	-342.387718	247.809	-342.505460

**Table S2.** Calculated values of Gibbs free energies of reaction ( $\Delta G$ , in kcal/mol) for various hypothetical supramolecular association processes.

Supramolecular Association Process	$\Delta G$
Cyclohexane + Ultem $\rightarrow$ Ultem–cyclohexane	1.6
MeOH + Ultem $\rightarrow$ Ultem–MeOH	-1.0
N <sub>2</sub> + Ultem $\rightarrow$ Ultem–N <sub>2</sub>	4.4
O <sub>2</sub> + Ultem $\rightarrow$ Ultem–O <sub>2</sub>	5.4
He + Ultem $\rightarrow$ Ultem–He	4.5