

NMR investigation of the supramolecular complex formed by a phenylboronic acid-ferrocene electroactive probe and native or derivatized β -cyclodextrin

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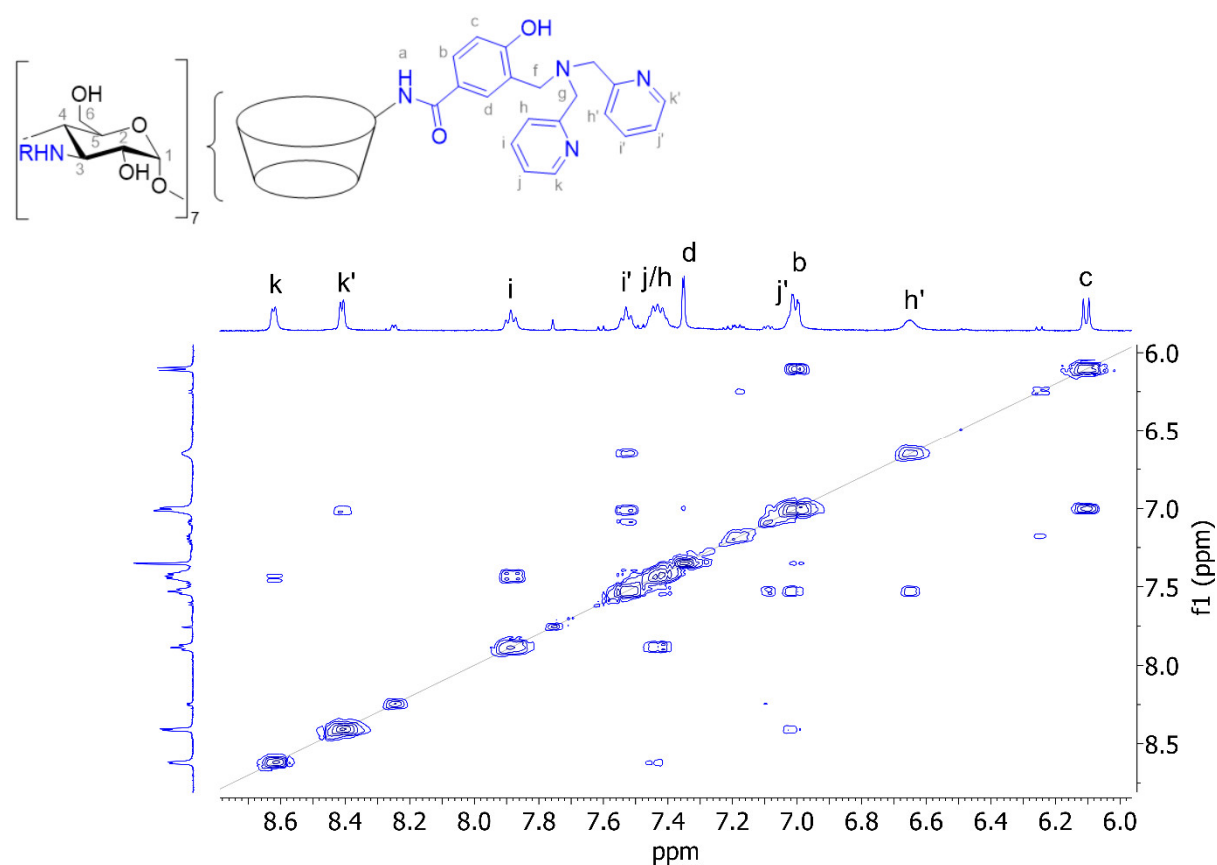


Figure S1. 2D COSY map (500 MHz, 25 °C, D₂O) spectrum of dpa-*p*-HB-β-CD.

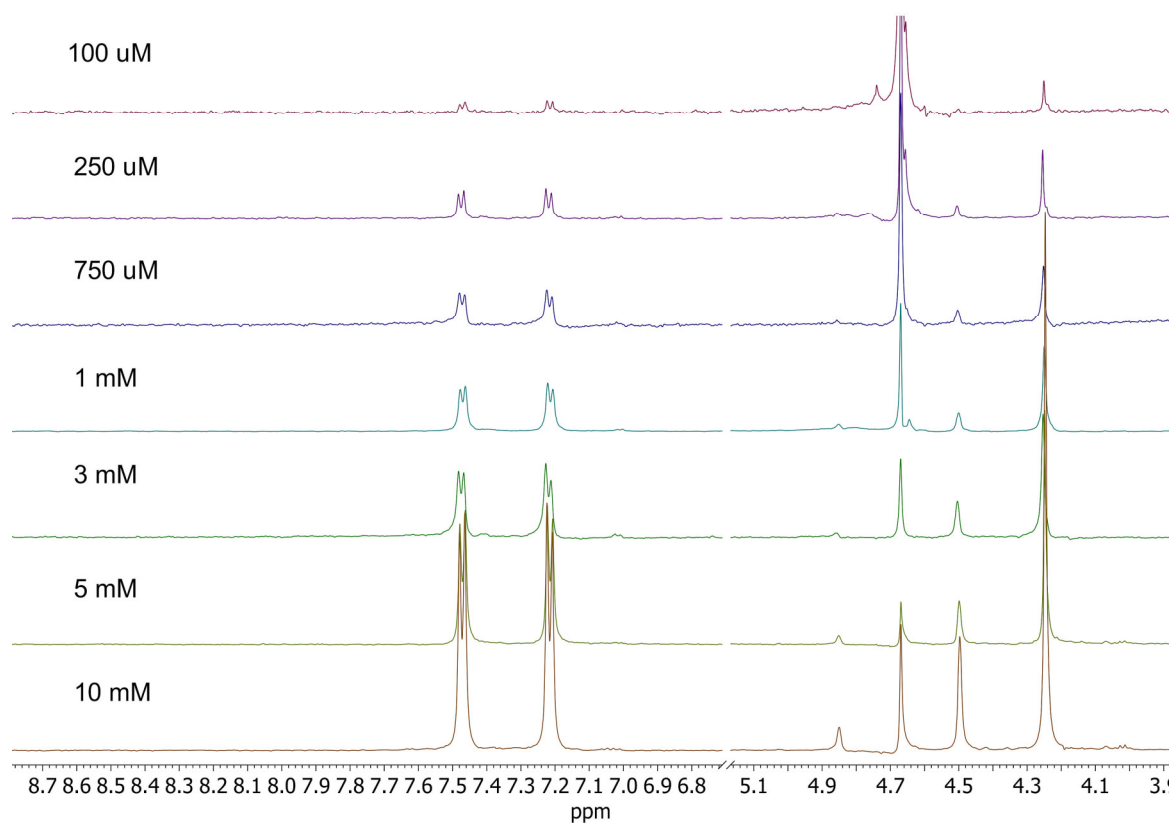


Figure S2. ¹H NMR titration (500 MHz, 25 °C, DMSO-d₆/D₂O (1:10 v/v), [Na₂CO₃] = 10 mM, [4-Fc-PB] probe = 0.1 mM ÷ 10 mM. No proton shifts of 4-Fc-PB have been observed, indicating the absence of any self-inclusion.

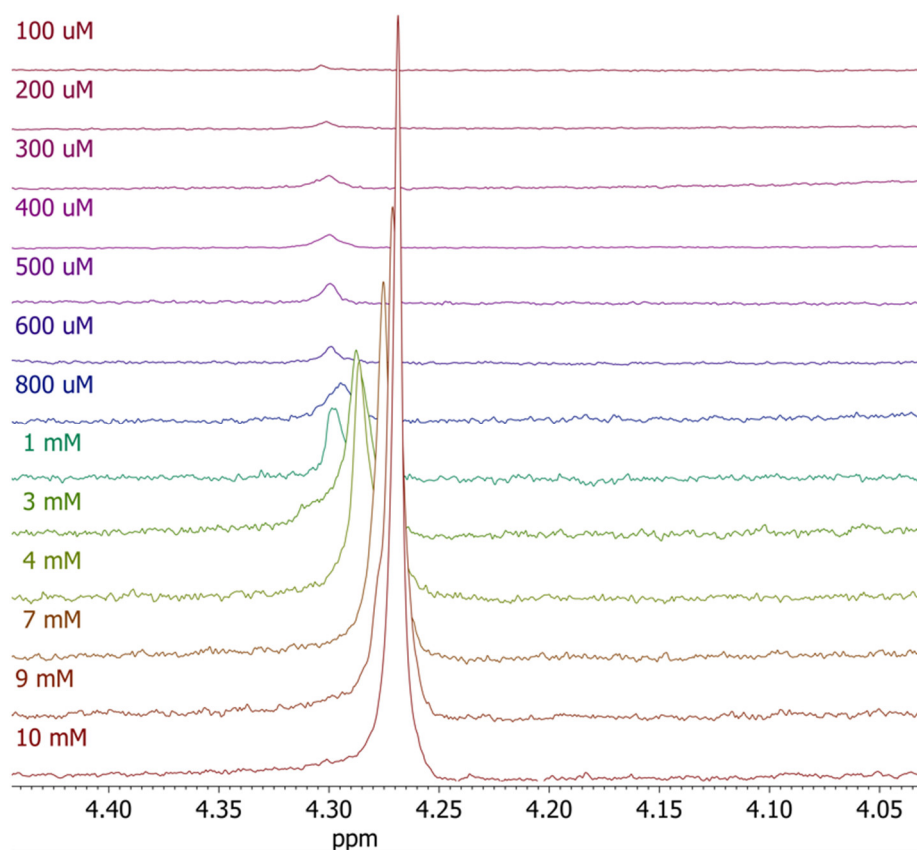


Figure S3. ^1H NMR titration (500 MHz, 25 $^\circ\text{C}$, $\text{DMSO-d}_6/\text{D}_2\text{O}$ (1:10 v/v), of [4-Fc-PB] probe = 0.1mM-10mM, $[\beta\text{-CDs}] = 5 \text{ mM}$, $[\text{Na}_2\text{CO}_3] = 10 \text{ mM}$. Proton shift of H_8 belonging to 4-Fc-PB have been observed indicating an inclusion in $\beta\text{-CDs}$.

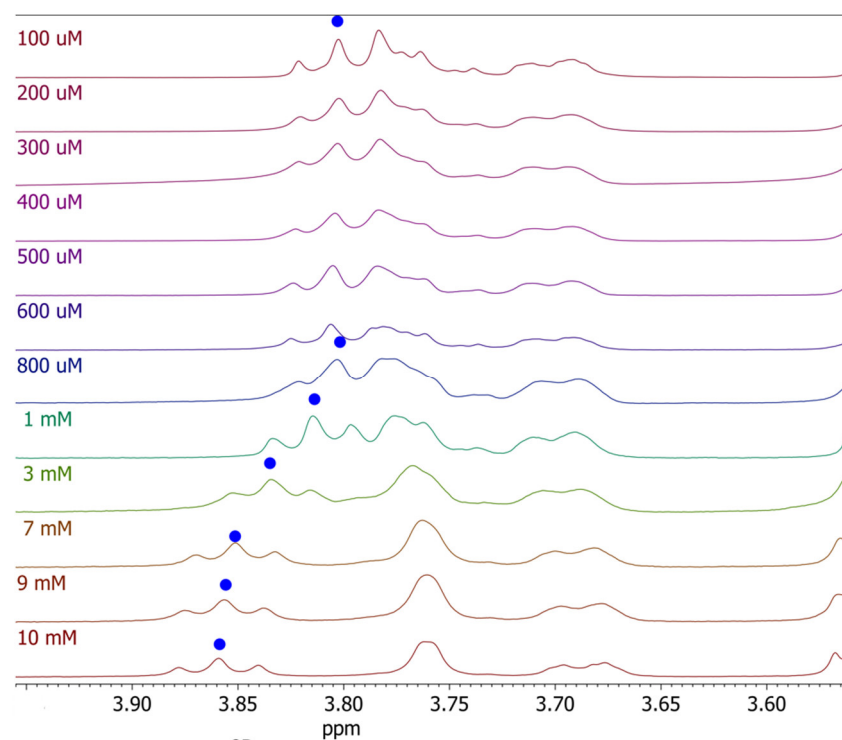


Figure S4. ^1H NMR titration (500 MHz, 25 $^\circ\text{C}$, $\text{DMSO-d}_6/\text{D}_2\text{O}$ (1:10 v/v), of [4-Fc-PB] probe = 0.1mM-10mM, $[\beta\text{-CDs}] = 5 \text{ mM}$, $[\text{Na}_2\text{CO}_3] = 10 \text{ mM}$. Proton shift of H_3 belonging to the internal

cavity of β -CDs have been observed indicating an inclusion of 4-Fc-PB probe in β -CDs.

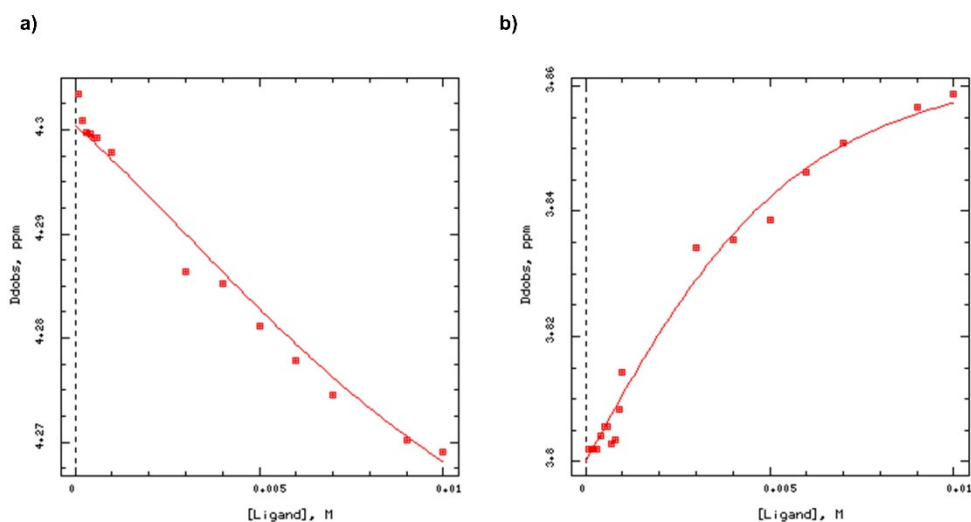


Figure S5. Plot of proton chemical shifts from ^1H NMR spectra, $[4\text{-Fc-PB}] = 0.1 - 10 \text{ mM}$, $[\beta\text{-CDs}] = 5 \text{ mM}$, in $\text{DMSO-}d_6/\text{D}_2\text{O}$ (1:10 v/v), $[\text{Na}_2\text{CO}_3] = 10 \text{ mM}$, 500 MHz, 25 $^\circ\text{C}$. a) H_8 protons shifts of 4-Fc-PB probe; b) H_3 protons shifts of natural β -CDs.

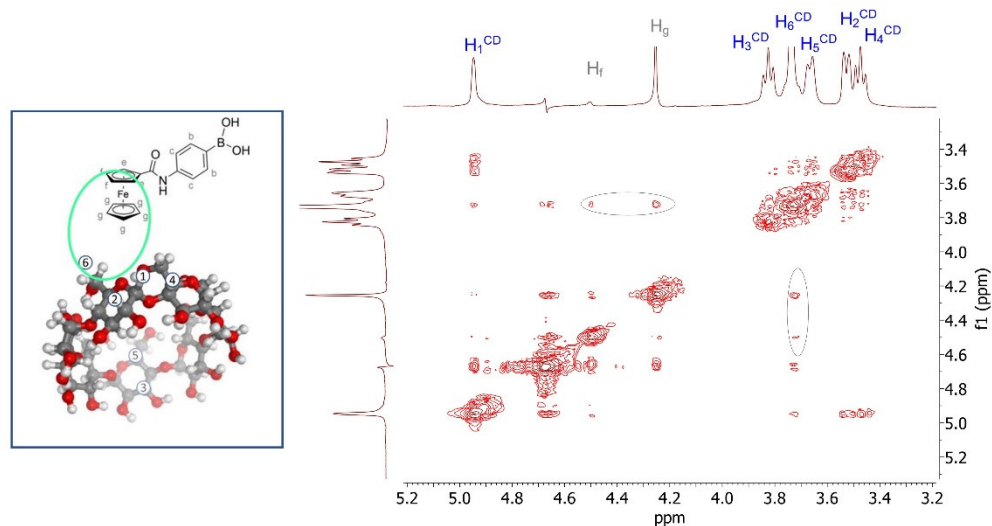


Figure S6. 2D ROESY map (500 MHz) of $[4\text{-Fc-PB}]$ probe = 10 mM and natural $[\beta\text{-CDs}] = 10 \text{ mM}$, $\text{DMSO-}d_6/\text{D}_2\text{O}$ (1:10 v/v), $[\text{Na}_2\text{CO}_3] = 10 \text{ mM}$, mixing time: 0.3 s; scans: 64; spinlock strength: 5.2 KHz.

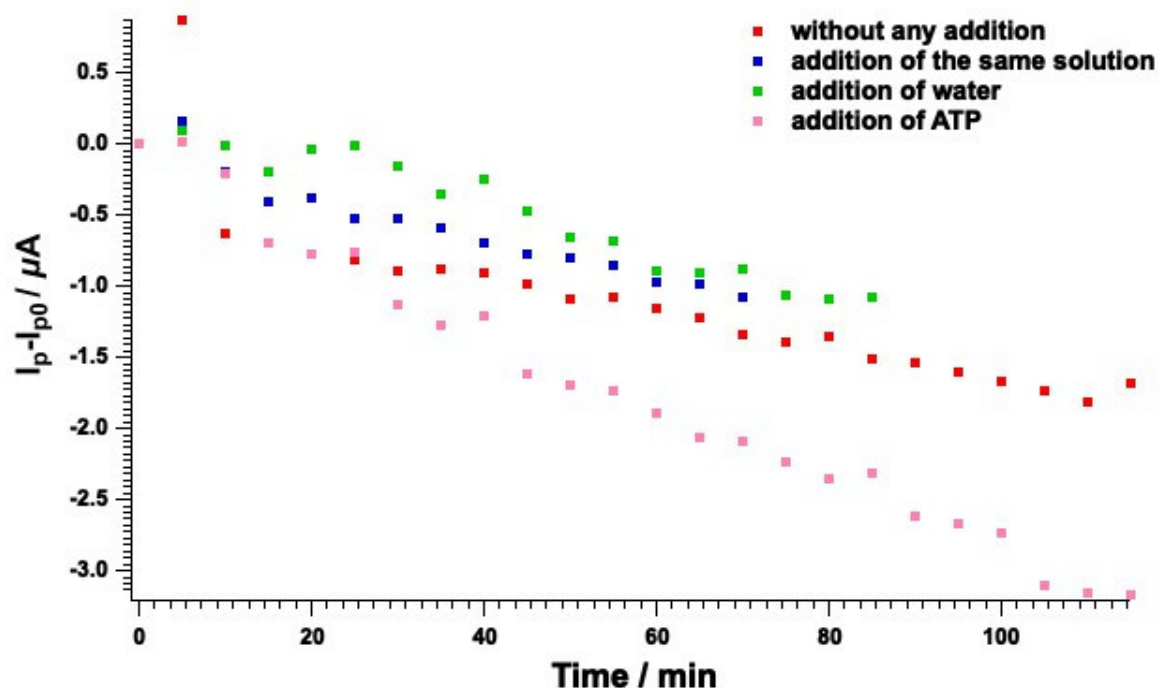


Figure S7. Trend in time (4-Fc-PB reduction peaks) by addition ATP (3 measurements for 0M, 200 μ M, 300 μ M, 400 μ M, 500 μ M, 600 μ M, 750 μ M, 1 mM for a total of 24 measurements) compared to three control experiments (addition of water, addition of the same solution of the electrochemical bath, no solution added). Measurements achieved in methanol: $[\text{NaClO}_4] = 0.2 \text{ M}$ (1:10 in v/v) water solution; $[\text{CHES buffer}] = 1 \text{ mM}$ (pH adjusted to 9.0 with NaOH).

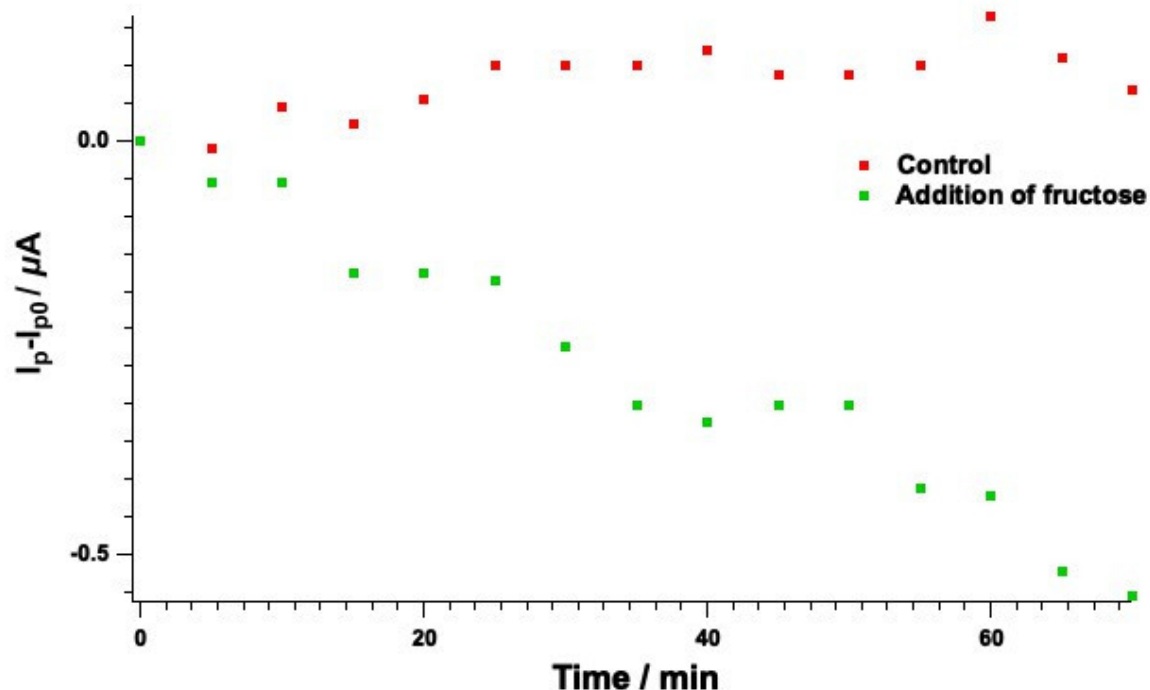


Figure S8. Trend in time (4-Fc-PB reduction peaks) by addition of fructose (3 measurements for 0M, 1 mM, 2 mM, 3 mM, 5 mM, for a total of 15 measurements) compared to the control experiment (addition of water). Measurements achieved in methanol: $[\text{NaClO}_4] = 0.2 \text{ M}$ (1:10 in v/v) water solution; $[\text{NaH}_2\text{PO}_4 \text{ buffer}] = 1 \text{ mM}$ (pH adjusted to 9.0 with NaOH).

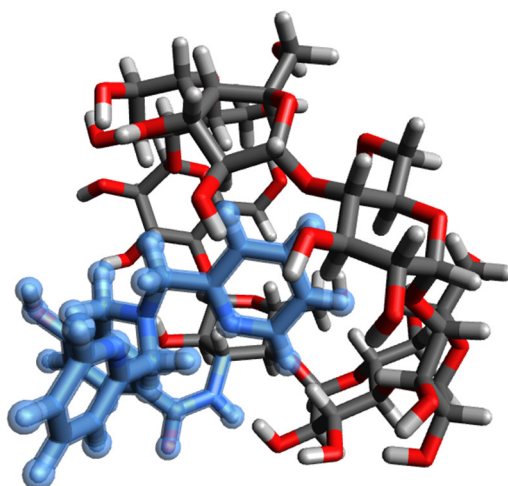


Figure S9. 3D representation of dpa-*p*-HB- β -CD, with dpa group highlighted in light blue colour.

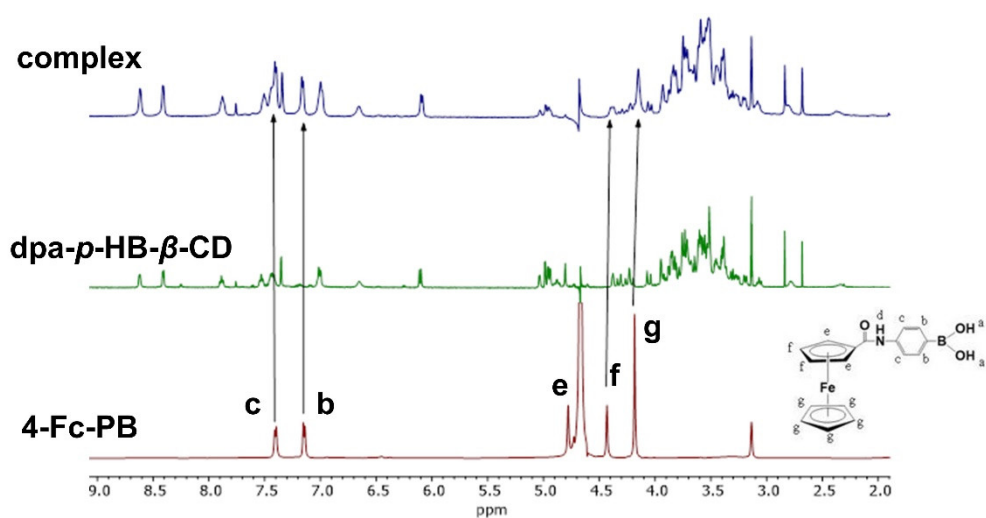


Figure S10. ^1H NMR (500 MHz, $\text{DMSO-d}_6/\text{D}_2\text{O}$ (1:10 v/v), $[\text{Na}_2\text{CO}_3] = 10 \text{ mM}$) spectrum of pure $[\text{4-Fc-PB}] = 5 \text{ mM}$ (low), pure $[\text{dpa-}p\text{-HB-}\beta\text{-CDs}] = 5 \text{ mM}$ (middle) and $[\text{4-Fc-PB}] = 5 \text{ mM}/[\text{Zn}^{2+}\text{dpa-}p\text{-HB-}\beta\text{-CD}] = 5 \text{ mM}$ supramolecular complex (top).

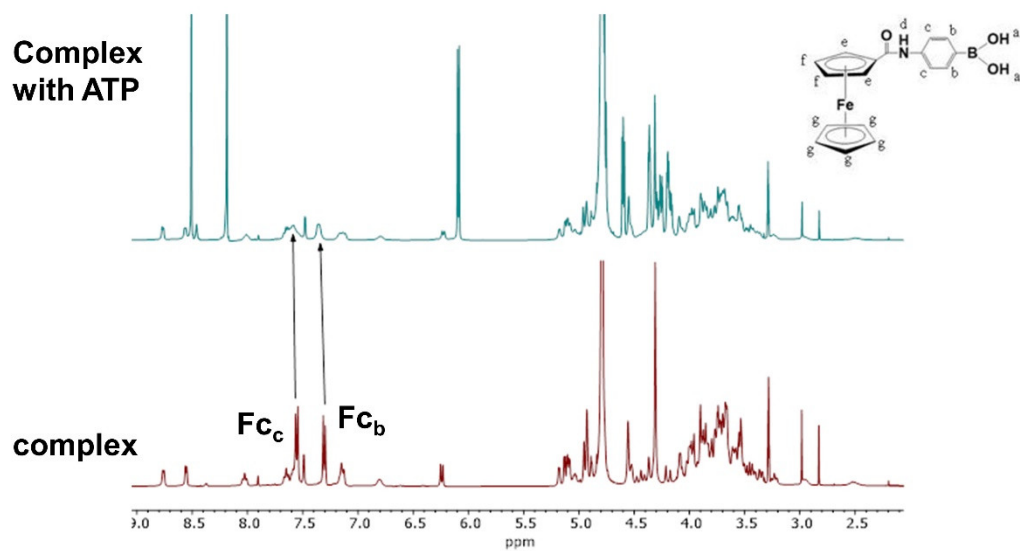


Figure S11. ¹H NMR (500 MHz, DMSO-d₆/D₂O (1:10 v/v), [Na₂CO₃] = 10 mM) spectrum of 4-Fc-PB/dpa-*p*-HB-β-CD supramolecular complex (5:5 mM) (low) and 4-Fc-PB/dpa-*p*-HB-β-CD/ATP (5:5:30 mM) (top).