## Supplementary Materials: The Thermodynamic and Kinetic Properties of 2-Hydroxypyridine/ 2-Pyridone Tautomerization: A Theoretical and Computational Revisit

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Table S	<b>S1.</b> Standard	orientation	of 2-hydro	xypyridine	(2-HPY)	calculated	by u	sing (	CAM-B3L	.YP/
aug-cc-	pvdz level of	theory.								

Contor Number	Atomic Number	A tomia Trues	Coordinates (Angstroms)				
Center Number	Atomic Number	Atomic Type	X	Ŷ	Ζ		
1	6	0	-1.003643	-1.231244	0.000058		
2	6	0	0.904003	0.030899	-0.000008		
3	6	0	0.186126	1.237950	-0.000042		
4	6	0	-1.201170	1.156316	-0.000007		
5	6	0	-1.819476	-0.102518	0.000050		
6	1	0	-1.437416	-2.233442	0.000105		
7	1	0	0.716025	2.188796	-0.000091		
8	1	0	-1.800651	2.067596	-0.000023		
9	1	0	-2.903707	-0.203699	0.000081		
10	7	0	0.338955	-1.173142	0.000033		
11	8	0	2.260559	0.084668	-0.000070		
12	1	0	2.573552	-0.833016	-0.000054		

**Table S2.** Standard orientation of 2-pyridone (2-PY) calculated by using CAM-B3LYP/aug-cc-pvdzlevel of theory.

Contor Number	Atomic Number	A tomia Truno	Coordinates (Angstroms)				
Center Number	Atomic Number	Atomic Type	X	Ŷ	Ζ		
1	6	0	-1.059541	-1.191802	0.000000		
2	6	0	1.064458	0.063913	-0.000001		
3	6	0	0.249231	1.266037	0.000000		
4	6	0	-1.118464	1.203656	0.000000		
5	6	0	-1.807288	-0.048084	0.000000		
6	1	0	-1.493091	-2.190520	0.000001		
7	1	0	0.784351	2.213949	0.000000		
8	1	0	-1.699035	2.128112	0.000001		
9	1	0	-2.892873	-0.101857	0.000001		
10	7	0	0.303930	-1.124328	0.000000		
11	8	0	2.293296	0.004077	0.000000		
12	1	0	0.856392	-1.974323	0.000000		

Conton Number	Atomia Number	A tomia Trues	Coordinates (Angstroms)			
Center Number	Atomic Number	Atomic Type	X	Y	Ζ	
1	6	0	1.231090	-0.557746	-0.000046	
2	6	0	-1.025456	-0.867754	0.000023	
3	6	0	-1.282063	0.507596	0.000055	
4	6	0	-0.190954	1.357671	0.000019	
5	6	0	1.100074	0.821991	-0.000037	
6	1	0	2.217440	-1.024352	-0.000092	
7	1	0	-2.307722	0.869475	0.000103	
8	1	0	-0.341407	2.437261	0.000035	
9	1	0	1.980886	1.460596	-0.000069	
10	7	0	0.187293	-1.395952	-0.000019	
11	1	0	-1.856403	-1.541865	0.000069	

**Table S3.** Standard orientation of pyridine (PY) calculated by using CAM-B3LYP/aug-cc-pvdz level of theory.

Para	meter	C2O11	N10C2	C2C3	N10C1	O11C2C3	N10C2C3	C2C3C4	O11C2C3C4	O11C2N10C1	N10C2C3C4
B51 VD	6-311++G**	1.223	1.411	1.451	1.364	127.0	113.0	121.6	-180.0	180.0	0.0
DOLIF	Aug-cc-pvdz	1.230	1.411	1.452	1.365	126.9	113.2	121.5	-180.0	180.0	0.00
	6-311++G**	1.220	1.399	1.449	1.362	126.6	113.3	121.5	-180.0	180.0	0.00
CAIVI-D5L1F	Aug-cc-pvdz	1.225	1.399	1.450	1.363	126.6	113.5	121.4	-180.0	180.0	0.00
	6-311++G**	1.220	1.401	1.451	1.361	126.7	113.2	121.5	-180.0	180.0	0.00
W D97 AD	Aug-cc-pvdz	1.220	1.397	1.448	1.357	126.7	113.2	121.5	-180.0	180.0	0.00
MOCON	6-311++G**	1.220	1.402	1.453	1.362	126.7	113.1	121.5	-180.0	180.0	0.00
MU62A	Aug-cc-pvdz	1.220	1.398	1.449	1.358	126.7	113.2	121.4	-180.0	180.0	0.00
	6-311++G**	1.221	1.399	1.465	1.375	126.0	113.4	121.5	-180.0	180.0	0.00
CCSD	Aug-cc-pvdz	1.232	1.403	1.463	1.379	126.0	113.6	121.4	-180.0	180.0	0.00
Experi	mental <sup>a</sup>	1.252	1.379	1.437	1.362	124.8	115.0	121.1	-177.6	177.7	2.1
<b>D</b> 21 VD	6-311++G**	1.354	1.326	1.401	1.341	118.5	124.0	117.4	-180.0	180.0	0.0
DOLIF	Aug-cc-pvdz	1.353	1.323	1.397	1.338	118.6	123.9	117.5	-180.0	180.0	0.00
	6-311++G**	1.348	1.319	1.395	1.336	118.5	124.0	117.4	-180.0	180.0	0.00
CAIVI-D5L1F	Aug-cc-pvdz	1.351	1.323	1.399	1.339	118.5	124.1	117.4	-180.0	180.0	0.00
	6-311++G**	1.346	1.322	1.398	1.337	118.5	124.1	117.4	-180.0	180.0	0.00
W D97 AD	Aug-cc-pvdz	1.350	1.326	1.401	1.340	118.4	124.2	117.3	-180.0	180.0	0.00
MOGOV	6-311++G**	1.348	1.321	1.399	1.338	118.3	124.2	117.4	-180.0	180.0	0.00
MU62A	Aug-cc-pvdz	1.351	1.324	1.402	1.340	118.2	124.3	117.3	-180.0	180.0	0.00
	6-311++G**	1.354	1.325	1.407	1.348	118.0	124.5	117.3	-180.0	180.0	0.00
	Aug-cc-pvdz	1.364	1.333	1.413	1.355	118.0	124.6	117.3	-180.0	180.0	0.00
Norm	alized <sup>b</sup>	1.336	1.320	1.41	1.34	118	124.1	117.5	-	-	-

**Table S4.** Some selected bond lengths (Å), bond angles and dihedral angles (degrees) of the optimized structures of 2-pyridone (2-PY) (top 11 lines) and 2-Hydroxypyridine (2-HPY) (bottom 11 lines) which have estimated by using different methods with 6-311++G\*\* and aug-cc-pvdz basis sets. The crystal structure of 2-Pyridone and the microwave-normalized structure of 2-HPY are listed for comparison purposes.

<sup>a</sup> Taken from Reference [2]; <sup>b</sup> Taken from Reference [7].

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Comton Number	A La maine NTerration	A. I	Coordinates (Angstroms)			
Center Number	Atomic Number	Atomic Type	X	Ŷ	Ζ	
1	6	0	0.894017	-1.288327	0.000014	
2	6	0	-0.959359	0.205596	-0.000009	
3	6	0	-0.112054	1.335076	-0.000007	
4	6	0	1.252633	1.090093	0.000005	
5	6	0	1.778009	-0.219808	0.000016	
6	1	0	1.214283	-2.325567	0.000022	
7	1	0	-0.525865	2.336249	-0.000016	

1.939885

2.847723

-2.238063

-1.714105

-0.425270

1.932583

-0.393839

0.037676

-1.235007

-1.050229

0.000007

0.000026

-0.000020

-0.000009

0.000002

Table S5. Standard orientation of the transition state (T.S.) calculated by using B3LYP/6-311++G\*\* level of theory.

Table S6. Imaginary frequencies (negative Signs) Harmonic frequencies (cm<sup>-1</sup>), IR intensities (km/Mole), Reduced masses (amu), force constants (mDyne/Å), and normal coordinates calculated for the transition state (T.S.) by using B3LYP/6-311++G\*\* level of theory.

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0

0

0

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Imaginary Freq.	-1889.5981			
<b>Reduced Masses</b>	1.0705			
<b>Force Constants</b>	2.2521			
<b>IR Intensities</b>	580.9988			
Atom Number	Mass Number	X	Y	Ζ
1	6	0.00	-0.01	0.00
2	6	-0.05	0.01	0.00
3	6	0.01	0.00	0.00
4	6	-0.01	0.00	0.00
5	6	-0.01	-0.01	0.00
6	1	0.00	-0.01	0.00
7	1	0.00	0.00	0.00
8	1	0.00	0.00	0.00
9	1	-0.01	-0.01	0.00
10	8	0.01	0.04	0.00
11	1	0.80	-0.59	0.00
12	7	-0.02	0.00	0.00

Conton Number	A torre in Number	A to main True o	Coordinates (Angstroms)				
Center Number	Atomic Number	Atomic Type	X	Ŷ	Ζ		
1	6	0	2.016618	-1.540155	-0.000186		
2	6	0	3.392801	-1.551071	0.000391		
3	6	0	4.058686	-0.303647	0.000714		
4	6	0	3.337370	0.866023	0.000434		
5	6	0	1.910516	0.837407	-0.000207		
6	1	0	1.443911	-2.460302	-0.000423		
7	1	0	3.938541	-2.483769	0.000607		
8	1	0	5.142238	-0.271514	0.001186		
9	1	0	3.807521	1.838608	0.000673		
10	7	0	1.293419	-0.389741	-0.000491		
11	8	0	1.221139	1.938550	-0.000529		
12	1	0	-0.000043	-0.416342	-0.000651		
13	6	0	-4.058719	-0.303515	0.000683		
14	6	0	-3.337346	0.866106	0.000410		
15	6	0	-1.910460	0.837458	-0.000216		
16	6	0	-2.016735	-1.540155	-0.000202		
17	1	0	-5.142272	-0.271322	0.001144		
18	1	0	-3.807455	1.838713	0.000643		
19	1	0	0.000431	1.905944	-0.001177		
20	8	0	-1.220960	1.938448	-0.000518		
21	7	0	-1.293476	-0.389794	-0.000489		
22	6	0	-3.392914	-1.550992	0.000364		
23	1	0	-3.938712	-2.483656	0.000567		
24	1	0	-1.444093	-2.460346	-0.000441		

**Table S7.** Standard orientation of the transition state (T.S.) of the double proton shifts between a mixed dimer tautomerization of 2-HPY $\leftrightarrow$ 2-PY using B3LYP/6-311++G\*\* level of theory.

**Table S8.** Imaginary frequencies (negative Signs) Harmonic frequencies (cm<sup>-1</sup>), IR intensities (km/Mole), Reduced masses (amu), force constants (mDyne/Å), and normal coordinates calculated for the transition state (T.S.) of the double proton shifts between a mixed dimer tautomerization of 2-HPY $\leftrightarrow$ 2-PY using B3LYP/6-311++G\*\* level of theory.

Imaginary Frequency	-1310.7244			
<b>Reduced Masses</b>	1.0771			
<b>Force Constants</b>	1.0903			
IR Intensity	500.4067			
Atom Number	Mass Number	X	Ŷ	Ζ
1	6	0.00	-0.02	0.00
2	6	-0.01	0.00	0.00
3	6	0.00	0.01	0.00
4	6	0.01	-0.01	0.00
5	6	-0.01	0.04	0.00
6	1	-0.01	0.00	0.00
7	1	-0.01	0.00	0.00
8	1	0.00	0.00	0.00
9	1	-0.01	0.00	0.00
10	7	-0.01	0.00	0.00
11	8	0.01	-0.02	0.00
12	1	0.83	0.00	0.00
13	6	0.00	-0.01	0.00
14	6	0.01	0.01	0.00
15	6	-0.01	-0.04	0.00
16	6	0.00	0.02	0.00
17	1	0.00	0.00	0.00
18	1	-0.01	0.00	0.00
19	1	-0.55	0.00	0.00
20	8	0.01	0.02	0.00
21	7	-0.01	0.00	0.00
22	6	-0.01	0.00	0.00
23	1	-0.01	0.00	0.00
24	1	-0.01	0.00	0.00



**Figure S1.** The structure of the Transition State for the double proton shifts between a mixed dimer tautomerization of 2-HPY $\leftrightarrow$ 2-PY using B3LYP/6-311++G\*\* level of theory. The color scheme is pink: nitrogen; yellow: carbon; blue: hydrogen; and red: oxygen.