<u>Molecules</u> 1998, *3*, M53

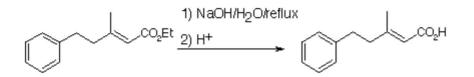
(E)-3-Methyl-5-phenyl-2-pentenoic Acid

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Received: 27 February 1998 / Published: 6 March 1998



The general part of the experimental section [1] has been presented elsewhere. Ethyl (*E*)-3-methyl-5-phenyl-2-pentenoate (3.363 g, 15.4 mmol) was refluxed with sodium hydroxide (2.161 g, 38.5 mmol) in a mixture of water (50 ml) and methanol (10 ml) for 3 hours, cooled and washed with ether (30 ml). The aqueous phase was acidified with concentrated hydrochloric acid to below pH 1. The mixture was extracted with ether (3x30 ml). The combined ether extracts were washed with brine (20 ml), dried (Na₂SO₄), filtered and evaporated under reduced pressure. (*E*)-3-Methyl-5-phenyl-2-pentenoic acid (2.79 g, 95%) was obtained as colourless plates from cyclohexane/light petroleum.

M.p. 58°

Anal. calc. for C₁₂H₁₄O₂ (190.24): C 75.8, H 7.4; found: C 76.2, H 7.3.

UV (ethanol) 305 (208) nm.

IR (CDCl₃) 3500-2800(bs, OH), 3104, 2950, 1694 (s, C=O), 1641, 1260 cm⁻¹.

¹H-NMR (90 MHz, CDCl₃) 2.16 (3H, d, *J* 1.1 Hz, CH₃), 2.43 (2H, m, =C(C*H*₂), 2.73 (2H, m, Ph-C*H*₂), 5.61 (1H, m, =CH), 6.92-7.40 (5H, m, ArH), 9.98 (1H, bs, COOH).

¹³C-NMR (15 MHz, CDCl₃) 19.22 (CH₃), 34.02, 42.85 (CH₂), 115.7 (=CH); 126.2, 128.2 128.5 (ArCH), 140.9 (quat, Cl'), 161.7 (quat, C3), 171.5 (quat, C1).

EI-MS 190(M⁺, 3%), 144(10), 91(100).

Acknowledgment: The authors gratefully acknowledge financial support from the Australian Research Council and The University of Sydney.

References and Notes

1. Moloney, M.G.; Pinhey, J.T.; Stoermer, M.J. "Vinyl Cation Formation by Decomposition of Vinyl-lead Triacetates. The reactions of Vinylmercury and Vinyltin Compounds with Lead Tetraacetate." *J. Chem. Soc. Perkin Trans. 1* **1990**, *10*, 2645.

Sample Availability: No sample available.

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