

Ni oxidation state and ligand saturation impact on the capability of octaazamacrocyclic complexes to bind and reduce CO₂

Supplementary Materials

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Content

Table S1. Total B3LYP/6-311G* energies/enthalpies of different species studied	2
Table S2. Selected B3LYP/6-311G* bond distances (Å) and angles (°) of a) [NiL] and b) [NiLH]	3
Table S3. B3LYP/6-311G* MPA charges of a) [NiL] and b) [NiLH]	4
Table S4. B3LYP/6-311G* QTAIM charges of a) [NiL] and b) [NiLH]	5
Table S5. B3LYP/6-311G* QTAIM BCP characteristics of a) [NiL] and b) [NiLH]	6
Table S6. B3LYP/6-311G* MPA of a) [NiL] and b) [NiLH] localized molecular orbitals on Ni	8
Figure S1. Comparison of geometries of complexes studied without and with CO ₂	10
Figure S2. Frontier orbitals of a) ¹ [NiL] ⁰ and ³ [NiL] ²⁻ ; b) ¹ [NiLH] ⁰ and ² [NiLH] ⁺	11
Figure S3. Frontier orbitals of a) ¹ [NiL-CO ₂] ⁰ and ³ [NiL-CO ₂] ²⁻ ; b) ¹ [NiLH-CO ₂] ⁰ and ² [NiLH-CO ₂] ⁺	13

Table S1. Total B3LYP/6-311G* energies/enthalpies of different [NiL], [NiL-CO₂], [NiLH], [NiLH-CO₂] species studied and CO₂ (all values are in A.U.). [NiLH_x] species denote geometries, where different R and S conformations were tested for the given atom(s) X (for atom labels see Figure 1 in the manuscript).

	E	H		E	H
³ [NiL] ²⁻	-3250.503603	-3250.142369	³ [NiL-CO ₂] ²⁻	-3439.157621	-3438.780621
¹ [NiL] ²⁻	-3250.510626	-3250.118574	¹ [NiL-CO ₂] ²⁻	-3439.157234	-3438.778617
² [NiL] ⁻	-3250.466015	-3250.101912	² [NiL-CO ₂] ⁻	-3439.111696	-3438.730981
¹ [NiL] ⁰	-3250.378540	-3250.011645	¹ [NiL-CO ₂] ⁰	-3439.023880	-3438.640248
² [NiL] ⁺	-3250.186230	-3249.819337	² [NiL-CO ₂] ⁺	-3438.830920	-3438.448238
¹ [NiL] ²⁺	-3249.948182	-3249.580637	¹ [NiL-CO ₂] ²⁺	-3438.592560	-3438.208037
¹ [NiLH] ⁰	-3258.616497	-3258.079627	¹ [NiLH-CO ₂] ⁰	-3447.321361	-3446.765337
² [NiLH] ⁺	-3258.578456	-3258.036079	² [NiLH-CO ₂] ⁺	-3447.22514	-3446.665791
² [NiLH _{Cl}] ⁺	-3258.574194	-3258.031725	² [NiLH _{Cl} -CO ₂] ⁺	-3447.221017	-3446.661914
² [NiLH _{C5}] ⁺	-3258.577090	-3258.034923	² [NiLH _{C5} -CO ₂] ⁺	-3447.224161	-3446.664701
¹ [NiLH] ²⁺	-3258.432714	-3257.886345		-	-
¹ [NiLH _{Cl}] ²⁺	-3258.428449	-3257.881841			
¹ [NiLH _{C3}] ²⁺	-3258.430081	-3257.883263			
¹ [NiLH _{C5}] ²⁺	-3258.430207	-3257.882894			
¹ [NiLH _{ClC3}] ²⁺	-3258.422065	-3257.875684			
CO ₂	-188.643673	-188.628479	¹ [NiL _{fr} -CO ₂] ⁰	-3438.976674	-3438.542170

Table S2a. Selected B3LYP/6-311G* bond distances (Å) and angles (°) of [NiL] complexes (according to the notation in Figure 1).

Distances	³ [NiL] ²⁻	³ [NiL-CO ₂] ²⁻	¹ [NiL] ²⁻	² [NiL] ⁻	¹ [NiL] ⁰	¹ [NiL-CO ₂] ⁰	¹ [NiL _n -CO ₂] ⁰	² [NiL] ⁺	¹ [NiL] ²⁺
Ni-N(1)	2.01	1.99	1.92	1.91	1.90	1.90	1.89	1.91	1.92
Ni-N(3)	1.97	2.05	1.90	1.91	1.91	1.91	1.95	1.91	1.90
Ni-N(5)	1.94	1.95	1.84	1.85	1.83	1.83	1.84	1.83	1.84
Ni-N(7)	1.92	1.95	1.87	1.85	1.86	1.86	1.89	1.86	1.87
Angles									
N(1)NiN(3)	106.70	106.25	102.95	102.37	102.74	102.74	102.96	102.87	102.95
N(3)NiN(5)	80.67	78.96	82.75	82.41	82.10	82.10	81.19	82.42	82.75
N(5)NiN(7)	91.08	91.69	92.60	91.85	92.18	92.18	92.49	92.56	92.60
N(7)NiN(1)	81.38	80.89	81.94	83.30	83.15	83.15	82.68	82.48	81.94
N(1)NiN(5)	168.53	170.39	169.35	171.17	171.67	171.67	175.00	170.92	169.35
N(3)NiN(7)	171.73	160.68	174.97	174.25	174.02	174.02	160.48	174.22	174.97
NiN(1)N(2)	132.28	129.05	132.44	132.89	133.20	133.20	130.18	132.41	132.44
NiN(3)N(4)	114.30	112.99	112.99	113.93	113.94	113.94	113.13	113.57	112.99
NiN(5)N(6)	130.56	129.69	131.03	131.72	131.26	131.26	130.24	131.05	131.03
NiN(7)N(8)	116.17	115.34	117.02	115.97	115.77	115.77	115.14	116.42	117.02
NiN(3)C(3)	124.21	124.91	129.67	129.18	129.26	129.26	129.31	129.48	129.67
NiN(7)C(6)	126.34	126.49	126.41	127.20	127.46	127.46	126.95	126.94	126.41
CO ₂ ^a	-	137.62	-	-	-	179.53	145.24	-	-
NiCO ^{b,c}	-	111.00	-	-	-	87.90	105.56	-	-
N(3)NiC ^b	-	94.83	-	-	-	82.70	99.51	-	-
N(7)NiC ^b	-	102.87	-	-	-	-	99.51	-	-

^a O-C-O angle in CO₂; ^b C denotes the carbon in CO₂; ^c O denotes one of the oxygens in CO₂

Table S2b. Selected B3LYP/6-311G* bond distances (Å) and angles (°) of [NiLH] (according to the notation in Figure 1).

Distances	¹ [NiLH] ⁰	¹ [NiLH-CO ₂] ⁰	² [NiLH] ⁺	² [NiLH-CO ₂] ⁺	¹ [NiLH] ²⁺
Ni-N(1)	2.96	3.08	2.10	2.11	1.95
Ni-N(3)	1.91	2.00	2.06	2.09	1.95
Ni-N(5)	2.77	2.22	2.06	2.06	1.92
Ni-N(7)	1.88	2.01	2.05	2.08	1.96
Angles					
N(1)NiN(3)	114.30	106.85	100.59	101.08	97.28
N(3)NiN(5)	73.47	80.03	84.14	83.21	86.89
N(5)NiN(7)	100.24	101.33	89.57	89.97	89.25
N(7)NiN(1)	71.77	70.55	85.14	83.38	86.50
N(1)NiN(5)	172.00	149.29	172.87	167.37	175.01
N(3)NiN(7)	168.09	170.22	171.30	168.17	175.84
NiN(1)N(2)	107.62	113.70	110.77	109.97	107.80
NiN(3)N(4)	116.42	108.66	103.22	103.93	109.90
NiN(5)N(6)	94.11	119.89	108.97	110.76	111.79
NiN(7)N(8)	119.95	118.78	108.19	108.96	109.69
NiN(3)C(3)	119.55	120.72	113.29	114.18	117.35
NiN(7)C(6)	114.16	116.28	114.85	116.65	114.23
CO ₂ ^a	-	129.12	-	146.63	-
NiCO ^{b,c}	-	118.53	-	105.26	-
N(3)NiC ^b	-	94.55	-	93.88	-
N(7)NiC ^b	-	94.11	-	96.30	-

^a O-C-O angle in CO₂; ^b C denotes the carbon in CO₂; ^c O denotes one of the oxygens in CO₂

Table S3a. B3LYP/6-311G* MPA charges of [NiL] complexes.

Atoms	³ [NiL] ²⁻	³ [NiL-CO ₂] ²⁻	¹ [NiL] ²⁻	² [NiL] ²⁻	¹ [NiL] ⁰	¹ [NiL-CO ₂] ⁰	¹ [NiL _{tr} -CO ₂] ⁰	² [NiL] ⁺	¹ [NiL] ²⁺
Ni	0.730	1.059	0.887	0.958	1.003	1.002	1.102	1.047	1.096
S(1)	-0.022	0.006	-0.010	0.057	0.126	0.127	0.152	0.231	0.339
S(2)	-0.034	0.001	-0.020	0.064	0.146	0.147	0.171	0.219	0.290
N(1)	-0.522	-0.521	-0.556	-0.528	-0.465	-0.464	-0.442	-0.444	-0.388
N(2)	-0.509	-0.516	-0.525	-0.531	-0.528	-0.529	-0.533	-0.477	-0.450
N(3)	-0.478	-0.469	-0.458	-0.482	-0.494	-0.494	-0.48	-0.419	-0.340
N(4)	-0.428	-0.402	-0.436	-0.434	-0.433	-0.432	-0.412	-0.437	-0.439
N(5)	-0.358	-0.345	-0.355	-0.330	-0.300	-0.297	-0.274	-0.241	-0.193
N(6)	-0.323	-0.311	-0.372	-0.291	-0.212	-0.211	-0.196	-0.177	-0.117
N(7)	-0.310	-0.301	-0.279	-0.290	-0.299	-0.296	-0.278	-0.210	-0.122
N(8)	-0.231	-0.216	-0.232	-0.216	-0.203	-0.204	-0.204	-0.189	-0.175
C(1)	0.107	0.114	0.109	0.121	0.150	0.151	0.154	-0.725	-0.714
C(2)	-0.586	-0.610	-0.595	-0.615	-0.627	-0.629	-0.626	-0.636	-0.643
C(3)	0.207	0.252	0.225	0.238	0.257	0.257	0.269	0.279	0.296
C(4)	0.273	0.323	0.278	0.339	0.409	0.410	0.422	0.430	0.452
C(5)	0.228	0.230	0.199	0.259	0.295	0.291	0.283	0.294	0.287
C(6)	0.211	0.215	0.183	0.244	0.289	0.286	0.289	0.306	0.313
C(7)	0.205	0.239	0.175	0.256	0.319	0.317	0.344	0.344	0.363
O(1)		-0.417				-0.260	-0.324		
C		0.109				0.515	0.216		
O(2)		-0.408				-0.260	-0.321		

Table S3b. B3LYP/6-311G* MPA charges of [NiLH]complexes.

Atoms	¹ [NiLH] ⁰	¹ [NiLH-CO ₂] ⁰	² [NiLH] ⁺	² [NiLH-CO ₂] ⁺	¹ [NiLH] ²⁺	CO ₂
Ni	0.038	0.840	0.761	1.148	1.049	
S(1)	0.001	0.033	0.096	0.123	0.166	
S(2)	-0.004	0.056	0.096	0.125	0.177	
N(1)	-0.582	-0.629	-0.591	-0.613	-0.608	
N(2)	-0.419	-0.551	-0.565	-0.579	-0.591	
N(3)	-0.569	-0.408	-0.580	-0.602	-0.604	
N(4)	-0.408	-0.604	-0.616	-0.639	-0.629	
N(5)	-0.453	-0.420	-0.420	-0.416	-0.401	
N(6)	-0.399	-0.427	-0.389	-0.387	-0.358	
N(7)	-0.405	-0.396	-0.378	-0.364	-0.351	
N(8)	-0.613	-0.628	-0.667	-0.665	-0.680	
C(1)	-0.673	-0.650	-0.125	-0.129	-0.151	
C(2)	-0.060	-0.107	-0.423	-0.421	-0.411	
C(3)	-0.450	-0.445	-0.633	-0.631	-0.624	
C(4)	-0.588	-0.639	-0.441	-0.446	-0.427	
C(5)	-0.408	-0.432	-0.776	-0.781	-0.786	
C(6)	-0.780	-0.785	-0.085	-0.088	-0.111	
C(7)	-0.063	-0.061	-0.021	-0.034	-0.059	
O(1)		-0.478		-0.328		-0.254
C		0.033		0.189		0.501
O(2)		-0.482		-0.327		-0.254

Table S4a. B3LYP/6-311G* QTAIM charges of [NiL] complexes.

Atoms	³ [NiL] ²⁻	³ [NiL-CO ₂] ²⁻	¹ [NiL] ²⁻	² [NiL] ⁻	¹ [NiL] ⁰	¹ [NiL-CO ₂] ⁰	¹ [NiL _{tr} -CO ₂] ⁰	² [NiL] ⁺	¹ [NiL] ²⁺
Ni	0.741	0.943	0.89	0.942	0.967	0.965	1.000	0.995	1.029
S(1)	-0.023	-0.027	-0.04	0.026	0.094	0.095	0.120	0.203	0.321
S(2)	-0.011	-0.020	-0.03	0.045	0.118	0.120	0.141	0.192	0.270
N(1)	-0.817	-0.759	-0.74	-0.761	-0.772	-0.770	-0.751	-0.679	-0.541
N(2)	-0.762	-0.750	-0.74	-0.748	-0.763	-0.763	-0.728	-0.750	-0.737
N(3)	-0.692	-0.771	-0.79	-0.747	-0.683	-0.684	-0.663	-0.568	-0.447
N(4)	-0.827	-0.803	-0.79	-0.806	-0.824	-0.822	-0.804	-0.810	-0.795
N(5)	-0.841	-0.726	-0.73	-0.744	-0.759	-0.757	-0.744	-0.710	-0.672
N(6)	-0.816	-0.723	-0.75	-0.721	-0.696	-0.696	-0.681	-0.572	-0.398
N(7)	-0.800	-0.765	-0.80	-0.777	-0.729	-0.728	-0.706	-0.544	-0.390
N(8)	-0.836	-0.779	-0.80	-0.783	-0.773	-0.773	-0.759	-0.744	-0.715
C(1)	0.495	0.442	0.39	0.469	0.557	0.558	0.574	0.502	0.415
C(2)	-0.011	-0.015	-0.01	-0.014	-0.028	-0.028	-0.030	-0.029	-0.030
C(3)	0.596	0.604	0.57	0.601	0.636	0.637	0.649	0.651	0.657
C(4)	0.543	0.582	0.48	0.551	0.630	0.630	0.641	0.633	0.632
C(5)	0.816	0.856	0.81	0.859	0.881	0.881	0.880	0.837	0.781
C(6)	0.837	0.852	0.83	0.868	0.883	0.883	0.883	0.833	0.777
C(7)	0.462	0.423	0.31	0.413	0.536	0.536	0.555	0.481	0.406
O(1)		-1.164			-	-1.081	-1.097		
C		1.658			-	2.149	1.807		
O(2)		-1.163			-	-1.083	-1.093		

Table S4b. B3LYP/6-311G* QTAIM charges of [NiLH] complexes.

Atoms	¹ [NiLH] ⁰	¹ [NiLH-CO ₂] ⁰	² [NiLH] ⁺	² [NiLH-CO ₂] ⁺	¹ [NiLH] ²⁺	CO ₂
Ni	0.151	0.675	0.663	0.890	0.944	
S(1)	-0.115	-0.099	-0.056	-0.034	0.013	
S(2)	-0.117	-0.065	-0.028	0.0002	0.056	
N(1)	-0.676	-0.676	-0.681	-0.689	-0.675	
N(2)	-0.660	-0.656	-0.669	-0.671	-0.665	
N(3)	-0.664	-0.667	-0.678	-0.687	-0.673	
N(4)	-0.681	-0.674	-0.666	-0.666	-0.664	
N(5)	-0.683	-0.688	-0.683	-0.686	-0.657	
N(6)	-0.664	-0.652	-0.655	-0.653	-0.644	
N(7)	-0.658	-0.679	-0.684	-0.695	-0.683	
N(8)	-0.684	-0.664	-0.663	-0.660	-0.650	
C(1)	0.319	0.331	0.315	0.313	0.303	
C(2)	-0.017	-0.027	-0.019	-0.020	-0.020	
C(3)	0.278	0.264	0.303	0.295	0.268	
C(4)	0.594	0.575	0.529	0.515	0.484	
C(5)	0.328	0.316	0.311	0.312	0.311	
C(6)	0.270	0.272	0.273	0.265	0.241	
C(7)	0.585	0.613	0.529	0.520	0.496	
O(1)		-1.214		-1.104		-1.076
C		1.442		1.749		2.153
O(2)		-1.205		-1.100		-1.076

Table S5a. B3LYP/6-311G* QTAIM BCP characteristics of [NiL] species from QTAIM analysis, i.e., A1-BCP-A2 distance d , charge density ρ , Laplacian $\Delta\rho$, ellipticity ε and delocalization index DI of chosen complexes.

	A1	A2	d [Å]	ρ [e/Å ³]	$\Delta\rho$ [e/Å ⁵]	ε	λ_1 [e/Å ⁵]	λ_2 [e/Å ⁵]	λ_3 [e/Å ⁵]	DI
³ [NiL] ²⁻	Ni	N(1)	1.917	0.70	13.20	0.05	-3.09	-2.96	19.25	0.59
	Ni	N(3)	1.869	0.79	14.80	0.03	-3.63	-3.53	21.96	0.64
	Ni	N(5)	1.896	0.73	13.28	0.03	-3.24	-3.15	19.67	0.58
	Ni	N(7)	1.830	0.86	16.32	0.04	-4.09	-3.94	24.35	0.69
³ [NiL-CO ₂] ²⁻	Ni	N(1)	1.991	0.59	9.88	0.08	-2.48	-2.29	14.65	0.48
	Ni	N(3)	2.051	0.52	8.69	0.08	-2.15	-1.99	12.83	0.42
	Ni	N(5)	1.953	0.65	11.20	0.03	-2.76	-2.67	16.63	0.50
	Ni	N(7)	1.957	0.66	11.19	0.02	-2.88	-2.83	16.90	0.52
	Ni	C	2.052	0.63	1.86	0.06	-3.13	-2.96	7.95	0.57
	C	O(1)	1.221	2.68	-4.33	0.02	-23.39	-23.03	42.09	1.26
	C	O(2)	1.222	2.67	-4.32	0.02	-23.31	-22.89	41.88	1.25
¹ [NiL] ²⁻	Ni	N(1)	1.919	0.70	11.29	0.03	-3.03	-2.93	17.25	0.59
	Ni	N(3)	1.917	0.80	12.03	0.04	-3.76	-3.61	19.40	0.65
	Ni	N(5)	1.853	0.86	12.23	0.05	-4.21	-4.00	20.44	0.72
	Ni	N(7)	1.874	0.80	12.03	0.04	-3.76	-3.61	19.40	0.65
² [NiL] ²⁻	Ni	N(1)	1.911	0.71	11.40	0.03	-3.13	-3.04	17.56	0.60
	Ni	N(3)	1.907	0.73	12.18	0.04	-3.25	-3.12	18.55	0.61
	Ni	N(5)	1.851	0.85	12.81	0.03	-4.05	-3.95	20.82	0.70
	Ni	N(7)	1.853	0.84	12.72	0.03	-3.98	-3.86	20.56	0.68
¹ [NiL] ⁰	Ni	N(1)	1.901	0.73	11.66	0.03	-3.25	-3.15	18.06	0.61
	Ni	N(3)	1.910	0.72	12.04	0.02	-3.23	-3.16	18.42	0.60
	Ni	N(5)	1.834	0.88	13.37	0.03	-4.24	-4.12	21.73	0.69
	Ni	N(7)	1.862	0.82	13.01	0.01	-3.82	-3.79	20.62	0.67
¹ [NiL _{fr} -CO ₂] ⁰	Ni	N(1)	1.893	0.79	11.28	0.02	-3.81	-3.75	18.84	0.62
	Ni	N(3)	1.959	0.67	9.99	0.04	-3.06	-2.96	16.01	0.53
	Ni	N(5)	1.840	0.89	12.27	0.06	-4.45	-4.19	20.91	0.67
	Ni	N(7)	1.894	0.77	11.19	0.07	-3.59	-3.36	18.13	0.60
	Ni	C	1.914	0.77	3.16	0.02	-4.01	-3.94	11.12	0.47
	C	O(1)	1.200	2.82	-3.19	0.00	-24.84	-24.74	46.38	1.29
	C	O(2)	1.201	2.81	-3.12	0.01	-24.76	-24.55	46.19	1.28
² [NiL] ⁺	Ni	N(1)	1.913	0.71	11.43	0.02	-3.20	-3.12	17.75	0.58
	Ni	N(3)	1.836	0.88	13.59	0.02	-4.24	-4.14	21.96	0.68
	Ni	N(5)	1.906	0.74	11.92	0.02	-3.33	-3.27	18.52	0.60
	Ni	N(5)	1.864	0.82	13.13	0.01	-3.80	-3.77	20.70	0.66
¹ [NiL] ²⁺	Ni	N(1)	1.919	0.70	11.29	0.03	-3.03	-2.93	17.25	0.59
	Ni	N(3)	1.905	0.80	12.03	0.04	-3.76	-3.61	19.40	0.65
	Ni	N(5)	1.842	0.80	12.03	0.04	-3.76	-3.61	19.40	0.65
	Ni	N(7)	1.872	0.86	12.23	0.05	-4.21	-4.00	20.44	0.72

Table S5b. B3LYP/6-311G* QTAIM BCP characteristics of [NiLH] species and CO₂ from QTAIM analysis, i.e., A1-BCP-A2 distance d , charge density ρ , Laplacian $\Delta\rho$, ellipticity ε and delocalization index DI of chosen complexes.

	A1	A2	d [Å]	ρ [e/Å ³]	$\Delta\rho$ [e/Å ⁵]	ε	λ_1 [e/Å ⁵]	λ_2 [e/Å ⁵]	λ_3 [e/Å ⁵]	DI
¹ [NiLH] ²⁺	Ni	N(1)	1.951	0.67	10.90	0.11	-2.90	-2.60	16.39	0.56
	Ni	N(3)	1.953	0.67	10.70	0.07	-2.91	-2.72	16.40	0.57
	Ni	N(5)	1.915	0.74	11.90	0.08	-3.36	-3.10	17.85	0.61
	Ni	N(7)	1.956	0.67	10.44	0.074	-2.89	-2.69	16.03	0.57
² [NiLH] ⁺	Ni	N(1)	2.096	0.46	8.37	0.08	-1.64	-1.52	11.53	0.41
	Ni	N(3)	2.064	0.50	9.16	0.05	-1.88	-1.79	12.83	0.44
	Ni	N(5)	2.057	0.51	9.34	0.09	-1.89	-1.73	12.97	0.43
	Ni	N(7)	2.055	0.51	9.48	0.08	-1.87	-1.73	13.09	0.45
² [NiLH-CO ₂] ⁺	Ni	N(1)	2.111	0.46	7.51	0.11	-1.80	-1.62	10.94	0.38
	Ni	N(3)	2.086	0.49	8.10	0.08	-1.95	-1.80	11.85	0.41
	Ni	N(5)	2.060	0.52	8.69	0.07	-2.07	-1.94	12.70	0.42
	Ni	N(7)	2.084	0.50	8.20	0.06	-1.87	-1.77	11.84	0.42
	Ni	C	2.079	0.59	1.65	0.05	-2.85	-2.71	7.21	0.50
	C	O(1)	1.204	2.78	-3.35	0.01	-24.54	-24.25	45.44	1.31
	C	O(2)	1.205	2.78	-3.32	0.01	-24.54	-24.20	45.43	1.30
¹ [NiLH] ⁰	Ni	N(1)	2.966	0.10	1.16	0.79	-0.23	-0.13	1.52	0.10
	Ni	N(3)	1.913	0.70	13.40	0.05	-3.10	-2.96	19.46	0.63
	Ni	N(5)	2.769	0.14	1.64	0.38	-0.40	-0.29	2.33	0.13
	Ni	N(7)	1.888	0.74	14.12	0.00	-3.39	-3.38	20.88	0.66
¹ [NiLH-CO ₂] ⁰	Ni	N(1)	3.086	0.08	0.93	0.33	-0.17	-0.13	1.23	0.06
	Ni	N(3)	2.006	0.57	10.16	0.04	-2.30	-2.22	14.68	0.49
	Ni	N(5)	2.219	0.39	5.16	0.17	-1.66	-1.42	8.24	0.30
	Ni	N(7)	2.008	0.57	10.03	0.07	-2.34	-2.20	14.56	0.49
	Ni	C	1.888	0.93	3.50	0.14	-5.21	-4.56	13.27	0.86
	C	O(1)	1.251	2.50	-6.96	0.03	-21.56	-20.87	35.46	1.17
	C	O(2)	1.254	2.48	-6.91	0.03	-21.22	-20.52	34.83	1.17
CO ₂	C	O(1)	1.200	2.82	-3.19	0.00	-24.84	-24.74	46.38	1.29
	C	O(2)	1.201	2.82	-3.19	0.00	-24.84	-24.74	46.38	1.29

Table S6a. B3LYP/6-311G* Mulliken populations of [NiL] species for localized molecular orbitals on the nickel atom for complexes studied.

	s	d _{z2}	d _{xz}	d _{yz}	d _{x²-y²}	d _{xy}	Σd+s
³ [NiL] ²⁻ α	0.23	0.81	-	-	-	-	1.04
	-	-	0.21	0.43	0.24	-	0.85
	-	-	-	0.16	0.73	-	0.99
	-	-	-	0.58	0.33	-	0.91
	-	-	-	-	-	0.87	0.87
³ [NiL] ²⁻ β	0.33	0.75	-	-	-	-	1.08
	-	-	0.62	0.27	-	-	0.89
	-	-	0.24	0.62	-	-	0.86
	-	-	0.07	-	-	0.86	0.93
¹ [NiL] ²⁻	0.21	0.84	-	-	-	-	1.05
	-	-	0.78	0.17	-	-	0.97
	-	-	0.18	0.72	-	-	0.94
	-	-	-	-	-	0.90	0.90
² [NiL] ⁻ α	0.19	0.86	-	-	-	-	1.05
	-	-	0.73	0.23	-	-	0.96
	-	-	0.21	0.71	-	-	0.92
	-	-	-	-	-	0.94	0.94
² [NiL] ²⁻ β	0.19	0.86	-	-	-	-	1.05
	-	-	0.72	0.24	-	-	0.96
	-	-	0.23	0.71	-	-	0.94
	-	-	-	-	-	0.95	0.95
¹ [NiL] ⁰	0.17	0.87	-	-	-	-	1.04
	-	-	0.64	0.33	-	-	0.97
	-	-	0.31	0.62	-	-	0.93
	-	-	-	-	-	0.95	0.95
¹ [NiL] ²⁺	0.13	0.89	-	-	-	-	1.03
	-	-	0.92	-	-	-	0.94
	-	-	0.00	0.92	-	-	0.95
	-	-	-	-	-	0.93	0.96
² [NiL] ⁺ α	0.15	0.88	-	-	-	-	1.03
	-	0.59	0.33	-	-	-	0.93
	-	-	0.35	0.61	-	-	0.97
	-	-	-	-	-	0.95	0.96
² [NiL] ⁺ β	0.15	0.88	-	-	-	-	1.03
	-	-	-	0.94	-	-	0.94
	-	-	0.94	-	-	-	0.94
	-	-	-	-	-	0.95	0.95
¹ [NiL-CO ₂] ⁰	0.17	0.87	-	-	-	-	1.04
	-	-	0.64	0.33	-	-	0.97
	-	-	0.31	0.62	-	-	0.93
	-	-	-	-	-	0.95	0.95
¹ [NiL _{fr} -CO ₂] ⁰	-	-	0.84	0.11	-	-	0.95
	-	-	0.12	0.86	-	-	0.98
	-	-	-	-	-	0.96	0.96
³ [NiL-CO ₂] ²⁻ α	-	0.74	-	-	-	-	0.93
	-	-	0.48	0.4	-	-	0.88
	-	-	0.43	0.54	-	-	0.97
	-	-	-	-	0.92	-	0.92
	-	-	-	-	-	0.93	0.74
³ [NiL-CO ₂] ²⁻ β	-	-	0.51	0.46	-	-	0.97
	-	-	0.44	0.49	-	-	0.93
	-	-	-	-	-	0.94	0.94

Table S6b. B3LYP/6-311G* Mulliken populations of [NiLH] species for localized molecular orbitals on the nickel atom for complexes studied.

	s	d _{z2}	d _{xz}	d _{yz}	d _{x²-y²}	d _{xy}	Σd+s
¹ [NiLH] ²⁺ α	0.16	0.87 0.01	0.01 0.93 0.03 0.01	0.03 0.71 0.23	0.41	0.23 0.73	1.06 0.97 0.97 0.96
¹ [NiLH] ²⁺ β	0.19	0.87	0.94 0.02	0.03 0.22 0.72		0.74 0.22	1.06 0.97 0.96 0.97
² [NiLH] ⁺ α	0.23	0.81 0.03	0.03 0.84 0.09	0.01 0.05 0.65 0.25	0.98 0.01	0.06 0.29 0.60	1.07 1.00 0.96 0.96 0.95
² [NiLH] ⁺ β	0.35	0.74 0.03	0.03 0.81 0.11	0.07 0.58 0.30		0.06 0.36 0.53	1.12 0.96 0.95 0.94
² [NiLH-CO ₂] ⁺ α	0.09	0.01 0.02 0.82	0.46 0.50	0.49 0.46 0.01	1.00	0.94 0.02	1.00 0.98 0.96 0.97 0.93
² [NiLH-CO ₂] ⁺ β			0.49 0.47	0.47 0.49		0.95	0.97 0.96 0.96
¹ [NiLH] ⁰	0.60 0.02 0.02	0.45 0.08 0.24	0.02 0.22 0.66 0.02	0.03 0.54 0.36	0.08 0.02 0.29 0.30	0.68 0.22	1.15 0.91 0.93 0.94 0.94
¹ [NiLH-CO ₂] ⁰	0.16 0.05	0.10 0.07 0.09	0.73 0.15	0.22 0.12 0.55	0.55 0.16	0.04 0.89	1.03 0.96 0.95 0.94

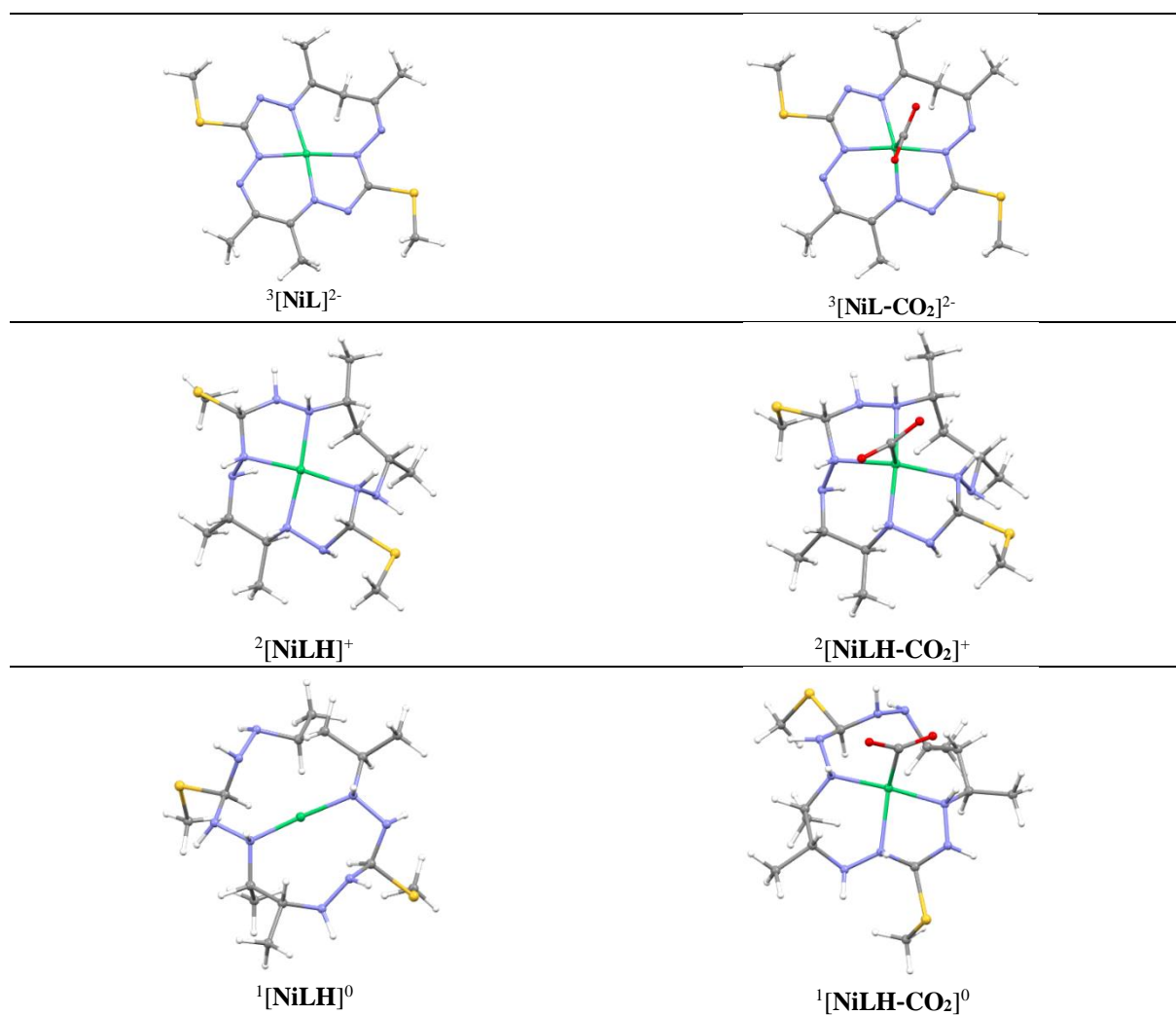


Figure S1. Comparison of geometries of complexes studied without (left column) and with CO₂ (right column).

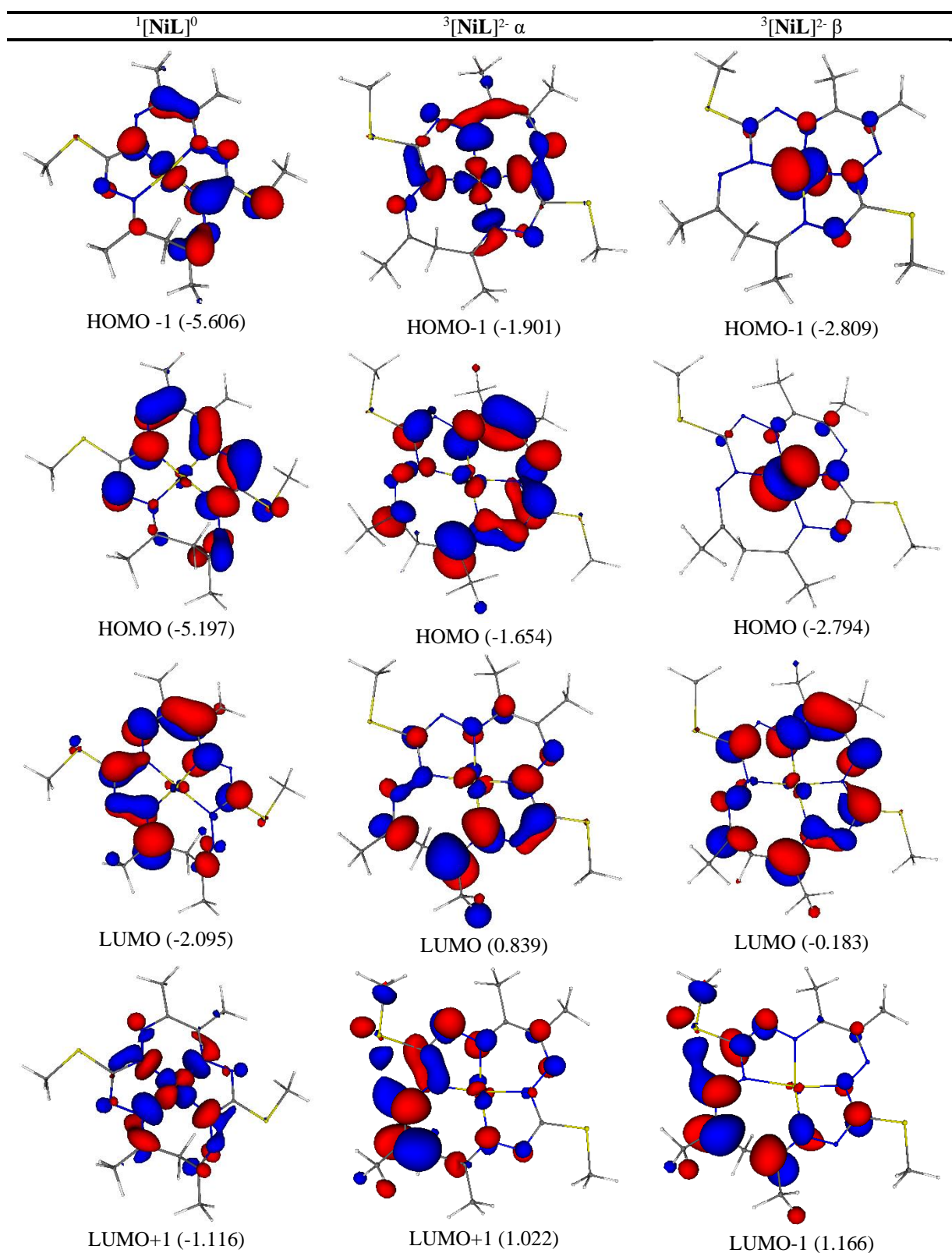


Figure S2a. Frontier orbitals of $^1[\text{NiL}]^0$ and $^3[\text{NiL}]^{2-}$ (eigenvalues are shown in parentheses in eV), isosurface value is 0.04 e bohr⁻³.

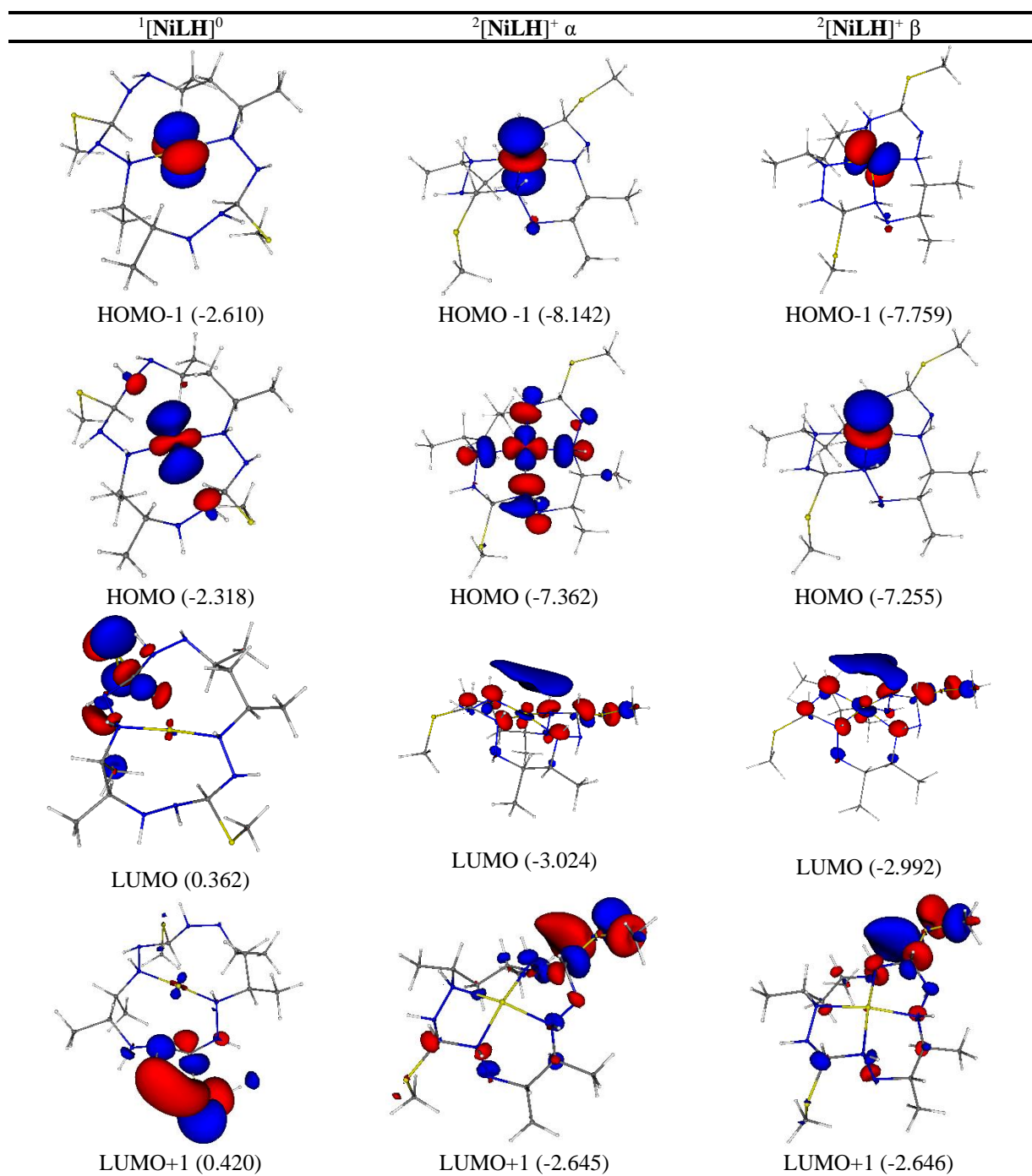


Figure S2b. Frontier orbitals of $^1[\text{NiLH}]^0$ and $^2[\text{NiLH}]^+$ (eigenvalues are shown in parentheses in eV), isosurface value is 0.04 e bohr^{-3} .

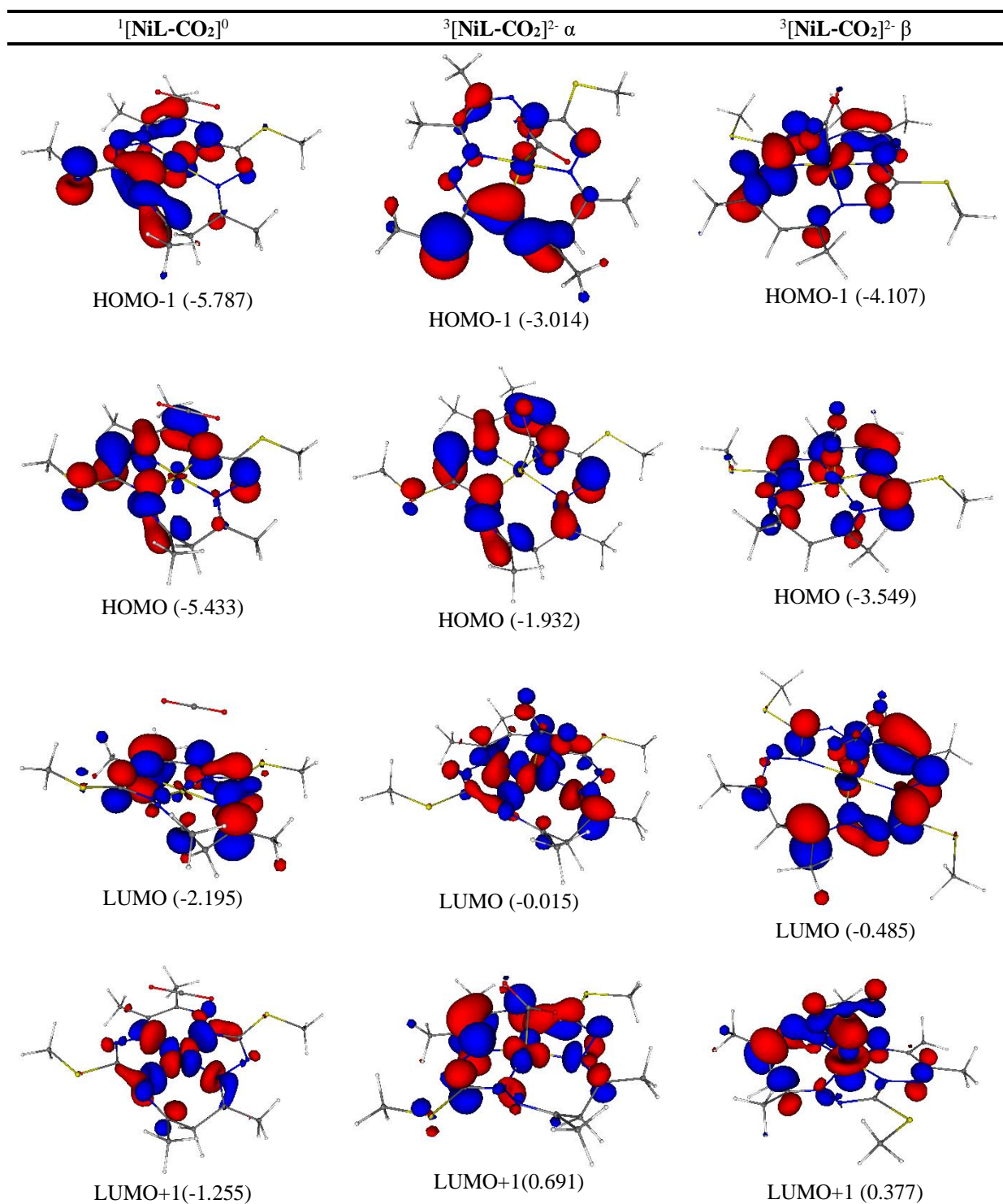


Figure S3a. Frontier orbitals of $^1[\text{NiL-CO}_2]^0$ and $^3[\text{NiL-CO}_2]^{2-}$ (eigenvalues are shown in parentheses in eV), isosurface value is 0.04 e bohr^{-3} .

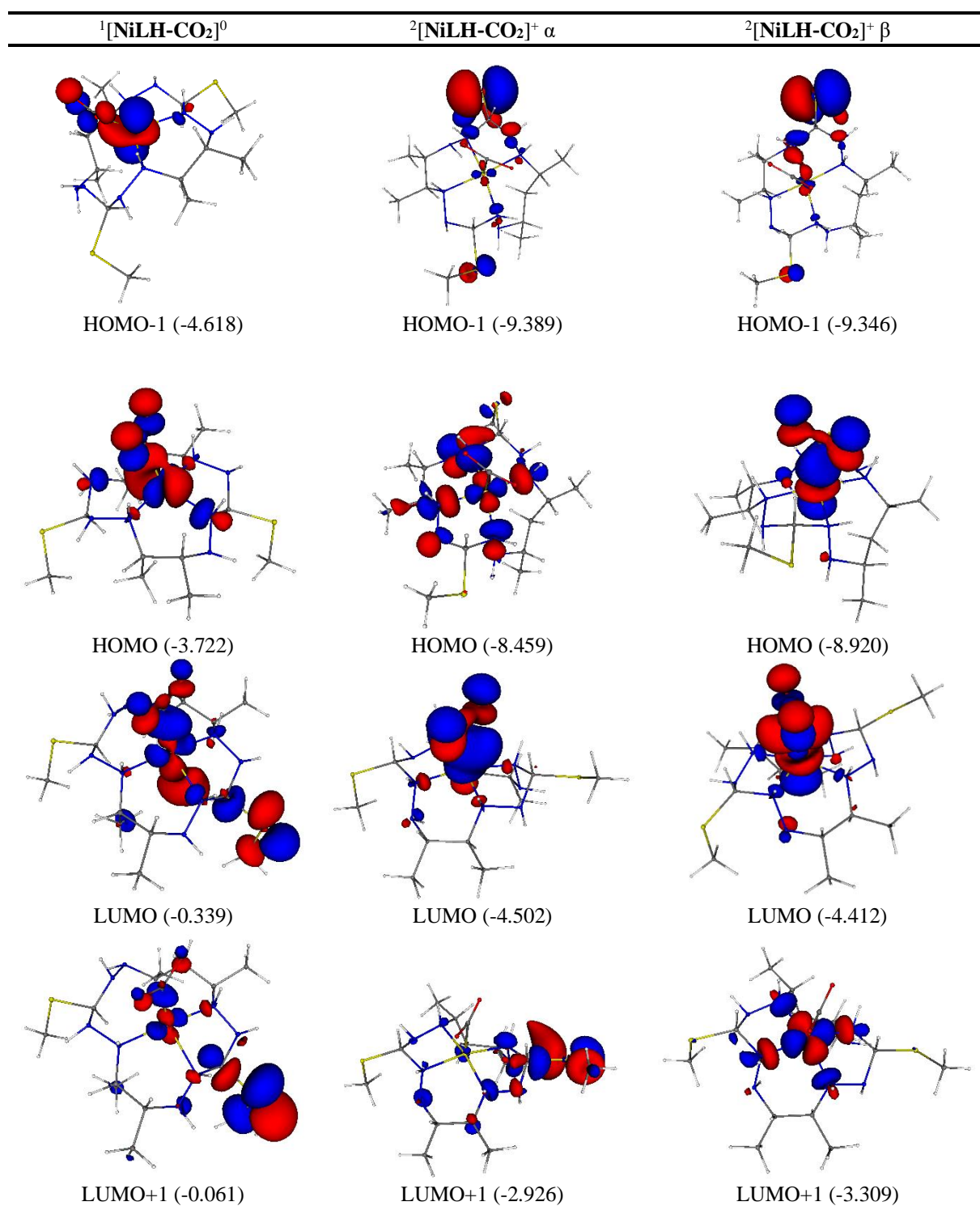


Figure S3b. Frontier orbitals of $^1[\text{NiLH-CO}_2]^0$ and $^2[\text{NiLH-CO}_2]^+$ (eigenvalues are shown in parentheses in eV), isosurface value is 0.04 e bohr⁻³.