

**Solubility Data of the Bioactive Compound Piperine in (Transcutol + Water) Mixtures:
Computational Modeling, Hansen Solubility Parameters and Mixing Thermodynamic
Parameters**

Faiyaz Shakeel, Nazrul Haq and Sultan Alshehri*

Department of Pharmaceutics, College of Pharmacy, King Saud University, P.O. Box 2457,
Riyadh 11451, Saudi Arabia

***Corresponding Author:**

Dr. Sultan Alshehri,
Department of Pharmaceutics, College of Pharmacy,
King Saud University, Riyadh, Saudi Arabia
Email: aslshehri@ksu.edu.sa

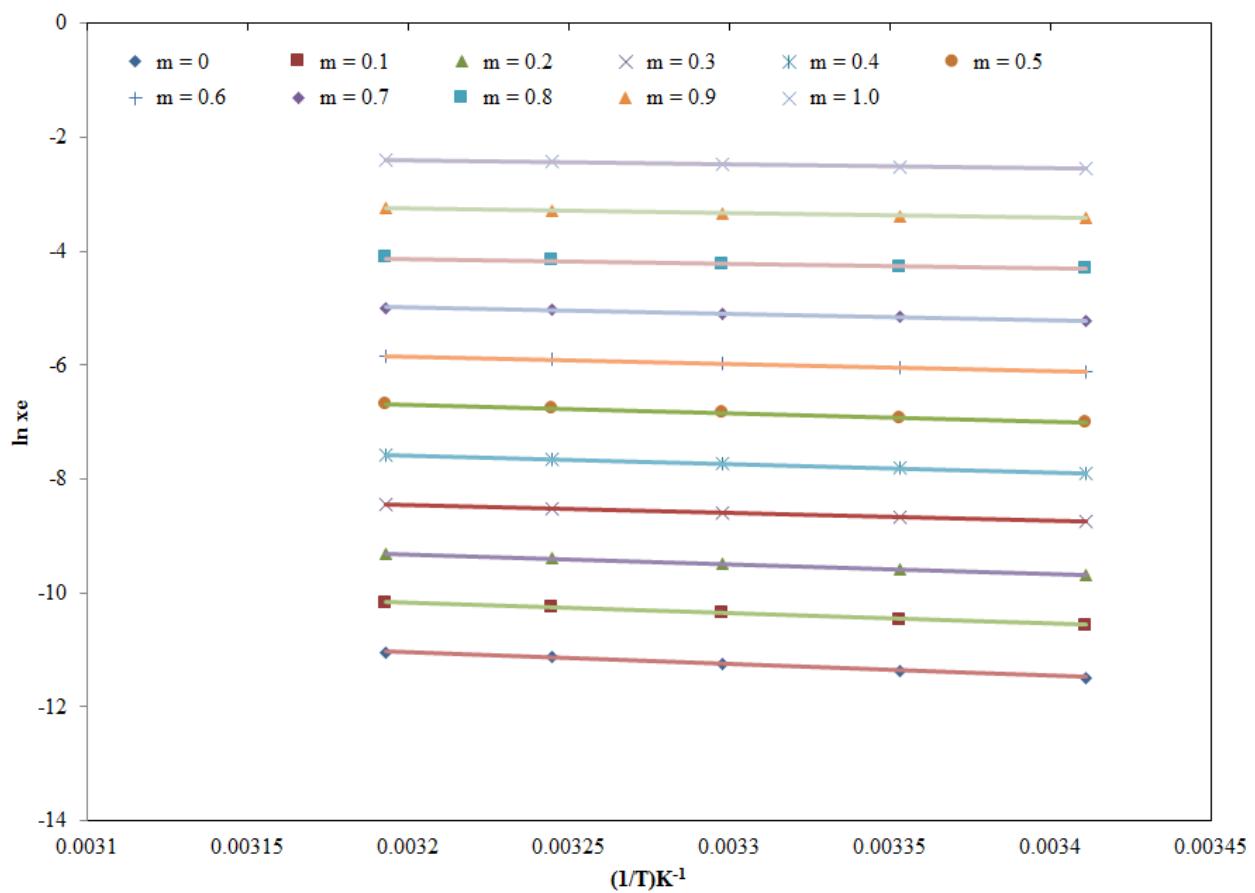


Figure S1. Correlation of experimental solubility values of PPN with Van't Hoff model in different THP + water mixtures at $T = 298.2$ K to 318.2 K; Van't Hoff model solubility values of PPN are represented by solid lines, and experimental solubility values of PPN are represented by the symbols.

Table S1. Hansen solubility parameters ($\delta_{\text{mix}}/\text{MPa}^{1/2}$) for various THP + water mixtures free of PPN at $T = 298.2$ K

m	$\delta_{\text{mix}}/\text{MPa}^{1/2}$
0.1	45.16
0.2	42.52
0.3	39.88
0.4	37.24
0.5	34.60
0.6	31.96
0.7	29.32
0.8	26.68
0.9	24.04

Table S2. The values of mixing enthalpy ($\Delta_{\text{mix}}H/\text{J mol}^{-1}$), mixing entropy ($\Delta_{\text{mix}}S/\text{J mol}^{-1} \text{K}^{-1}$), mixing Gibbs energy ($\Delta_{\text{mix}}G/\text{J mol}^{-1}$) and activity coefficient (γ_i) for PPN dissolution in different THP + water mixtures^a.

T/K	$\Delta_{\text{mix}}H/\text{J mol}^{-1}$	$\Delta_{\text{mix}}G/\text{J mol}^{-1}$	$\Delta_{\text{mix}}S/\text{J mol}^{-1} \text{K}^{-1}$	γ_i
<i>m</i> = 0.0				
298.2	-1170	774	-6.52	4980
303.2	-1200	791	-6.55	5150
308.2	-1220.	811	-6.60	5380
313.2	-1250	830	-6.63	5620
318.2	-1280	854	-6.70	6050
<i>m</i> = 0.1				
298.2	-955	571	-5.12	1995
303.2	-978	587	-5.16	2108
308.2	-1000	603	-5.20	2215
313.2	-1020	620	-5.25	2329
318.2	-1050	641	-5.32	2533
<i>m</i> = 0.2				
298.2	-764	397	-3.89	827
303.2	-784	411	-3.94	875
308.2	-803	424	-3.98	927
313.2	-824	438	-4.02	984
318.2	-847	455	-4.09	1070
<i>m</i> = 0.3				
298.2	-589	242	-2.78	322
303.2	-608	255	-2.84	353
308.2	-626	267	-2.89	380
313.2	-646	281	-2.96	416
318.2	-665	294	-3.01	448
<i>m</i> = 0.4				
298.2	-443	119	-1.88	138
303.2	-457	128	-1.93	148
308.2	-472	137	-1.97	160
313.2	-487	146	-2.02	173
318.2	-503	157	-2.07	189
<i>m</i> = 0.5				
298.2	-316	16	-1.11	56
303.2	-328	23	-1.15	61
308.2	-338	28	-1.19	65
313.2	-351	36	-1.23	71
318.2	-363	43	-1.27	77
<i>m</i> = 0.6				
298.2	-208	-63	-0.48	23
303.2	-217	-58	-0.52	25
308.2	-227	-52	-0.56	27
313.2	-237	-47	-0.60	30
318.2	-248	-41	-0.65	33
<i>m</i> = 0.7				
298.2	-123	-117	-0.02	9
303.2	-131	-114	-0.05	10
308.2	-138	-110	-0.09	11
313.2	-146	-107	-0.12	12
318.2	-155	-102	-0.16	14

$m = 0.8$				
298.2	-57	-147	0.30	3
303.2	-63	-144	0.26	4
308.2	-70	-142	0.23	5
313.2	-75	-140	0.20	5
318.2	-81	-137	0.17	6
$m = 0.9$				
298.2	-14	-149	0.45	1.57
303.2	-18	-147	0.42	1.79
308.2	-22	-146	0.40	2.00
313.2	-26	-145	0.37	2.23
318.2	-30	-144	0.35	2.49
$m = 1.0$				
298.2	8	-111	0.40	0.65
303.2	6	-110	0.38	0.74
308.2	4	-110	0.36	0.83
313.2	1	-110	0.35	0.94
318.2	1	-110	0.34	1.06

^aThe average relative uncertainties are $u(\Delta_{\text{mix}}H) = 0.90$, $u(\Delta_{\text{mix}}G) = 2.04$ and $u(\Delta_{\text{mix}}S) = 1.19$.