

Supporting Information (SI)

For

Design and synthesis of a fluorescent probe with a large Stokes shift for detecting thiophenols and its application in water sample and living cells

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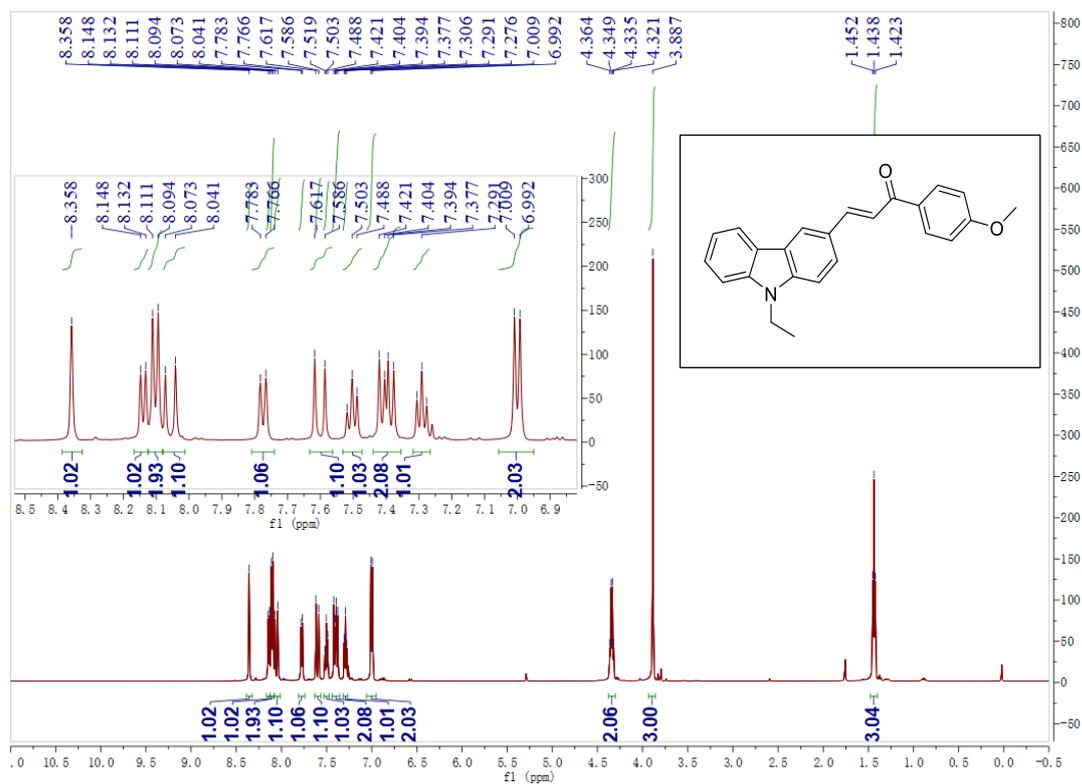
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¹ These authors contributed equally to this work.

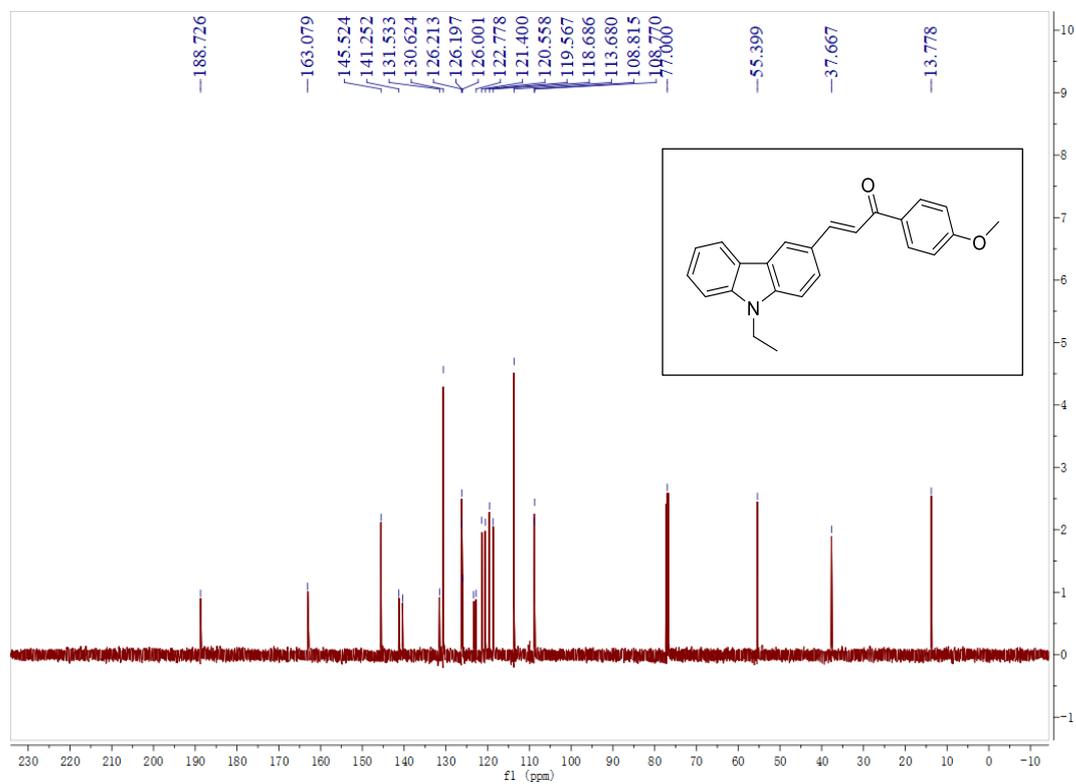
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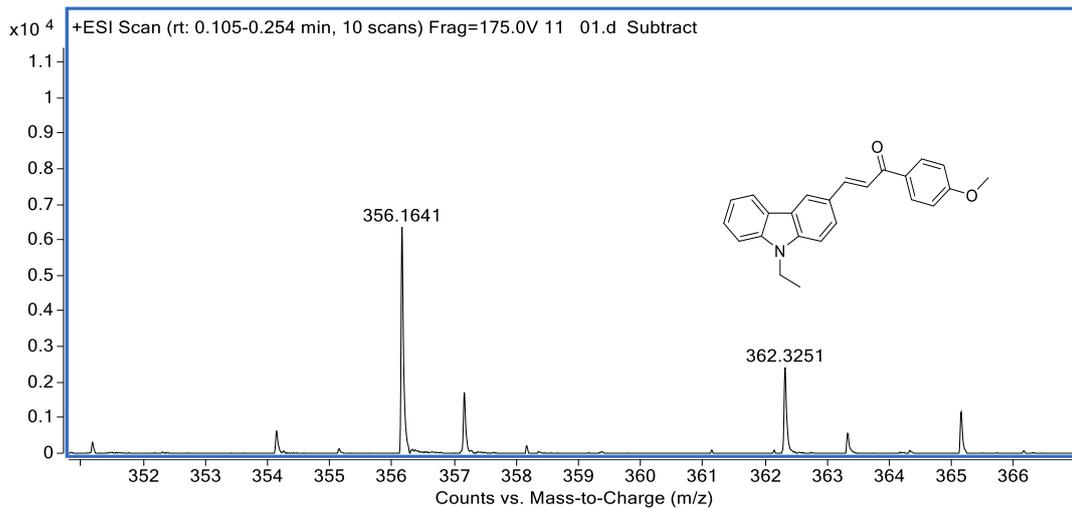
1. Structure characterization



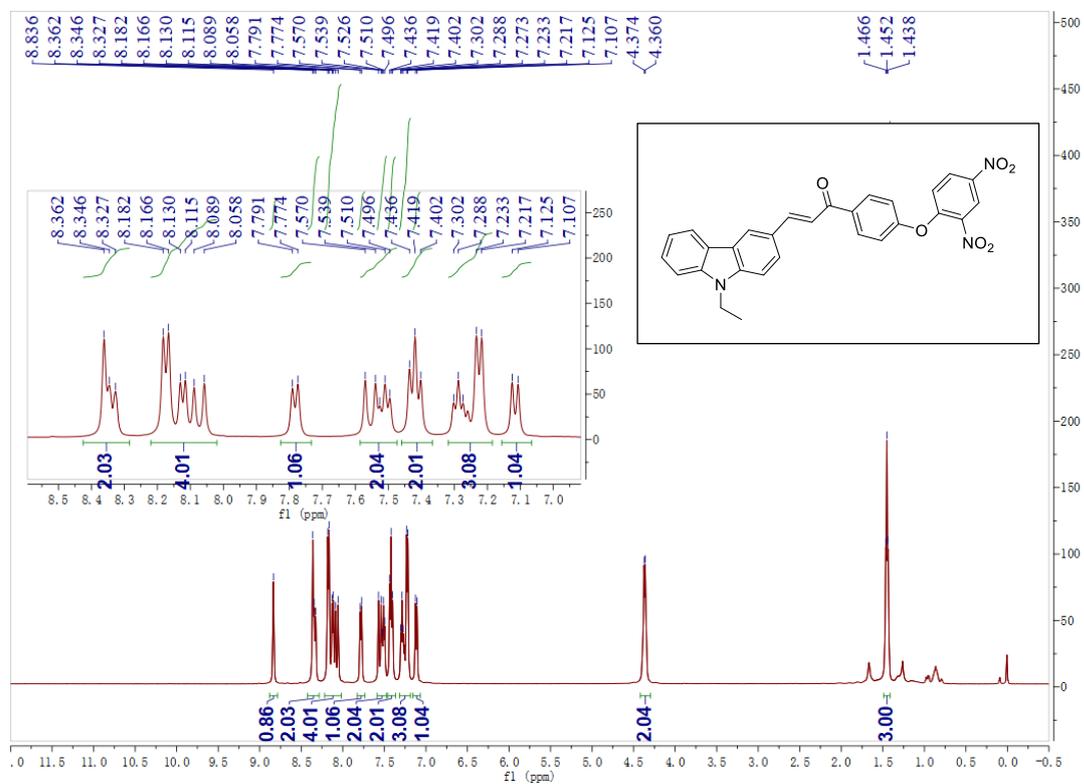
¹H NMR spectrum of KCP-OCH₃



¹³C NMR spectrum of KCP-OCH₃

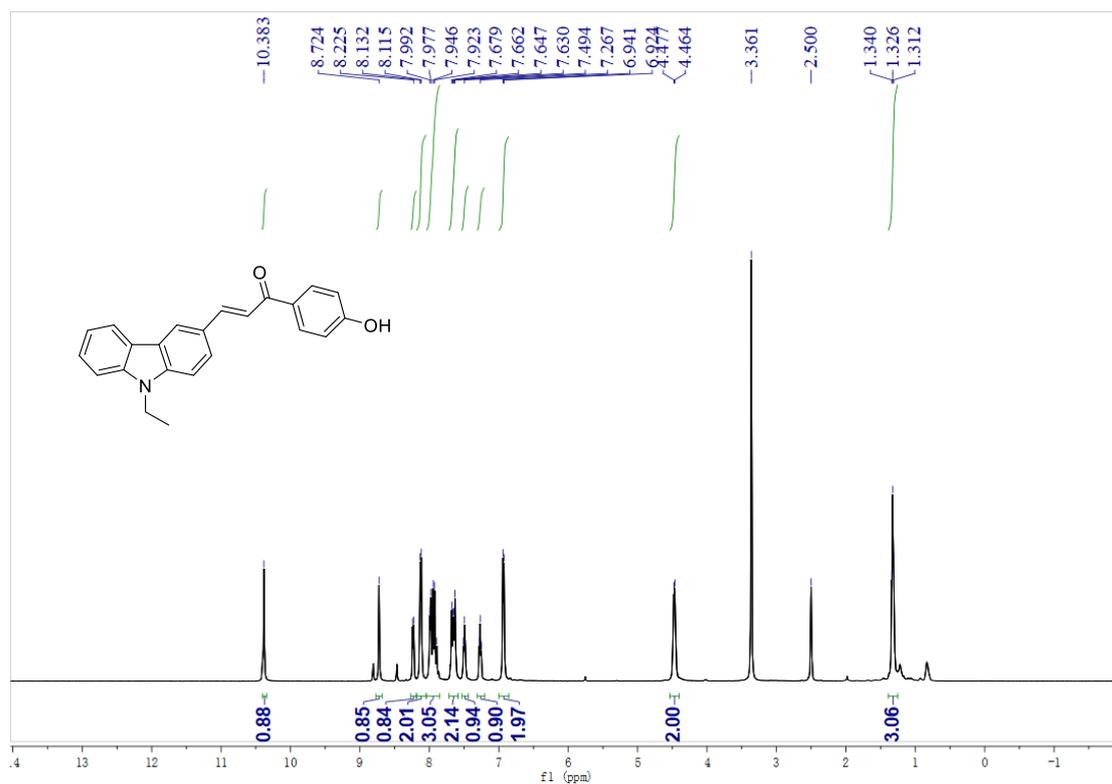


HR-MS spectrum of KCP-OCH₃

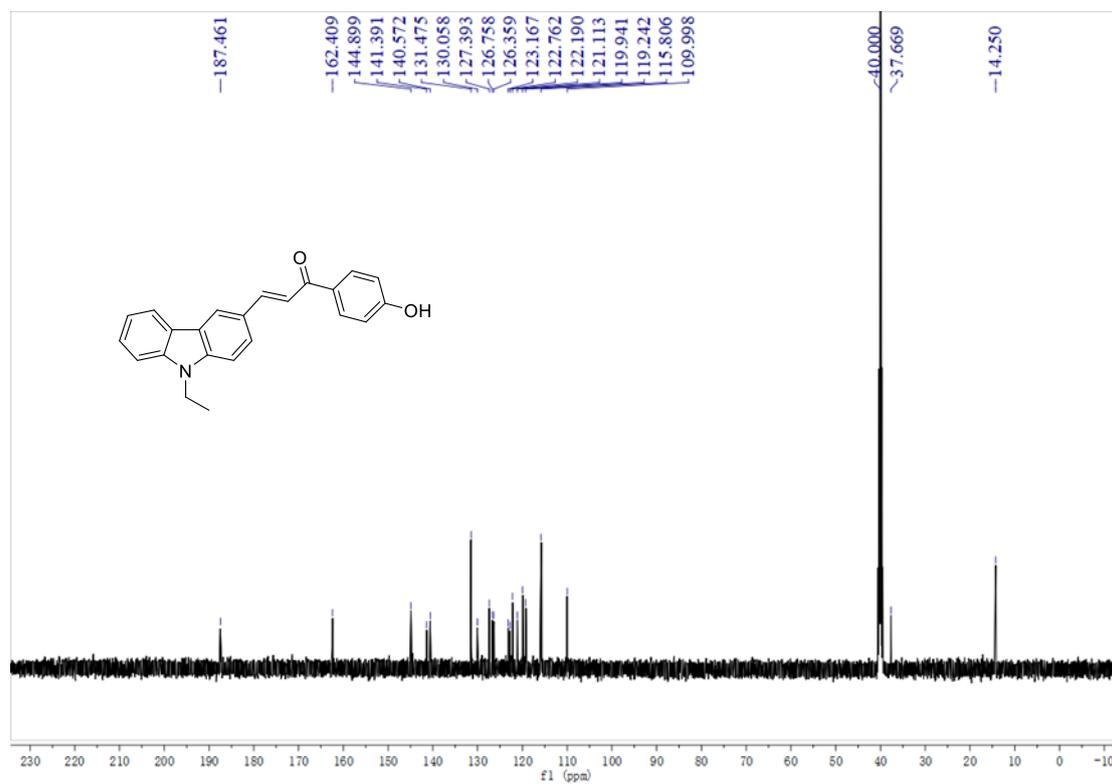


¹H NMR spectrum of probe-KCP

2. Additional spectra

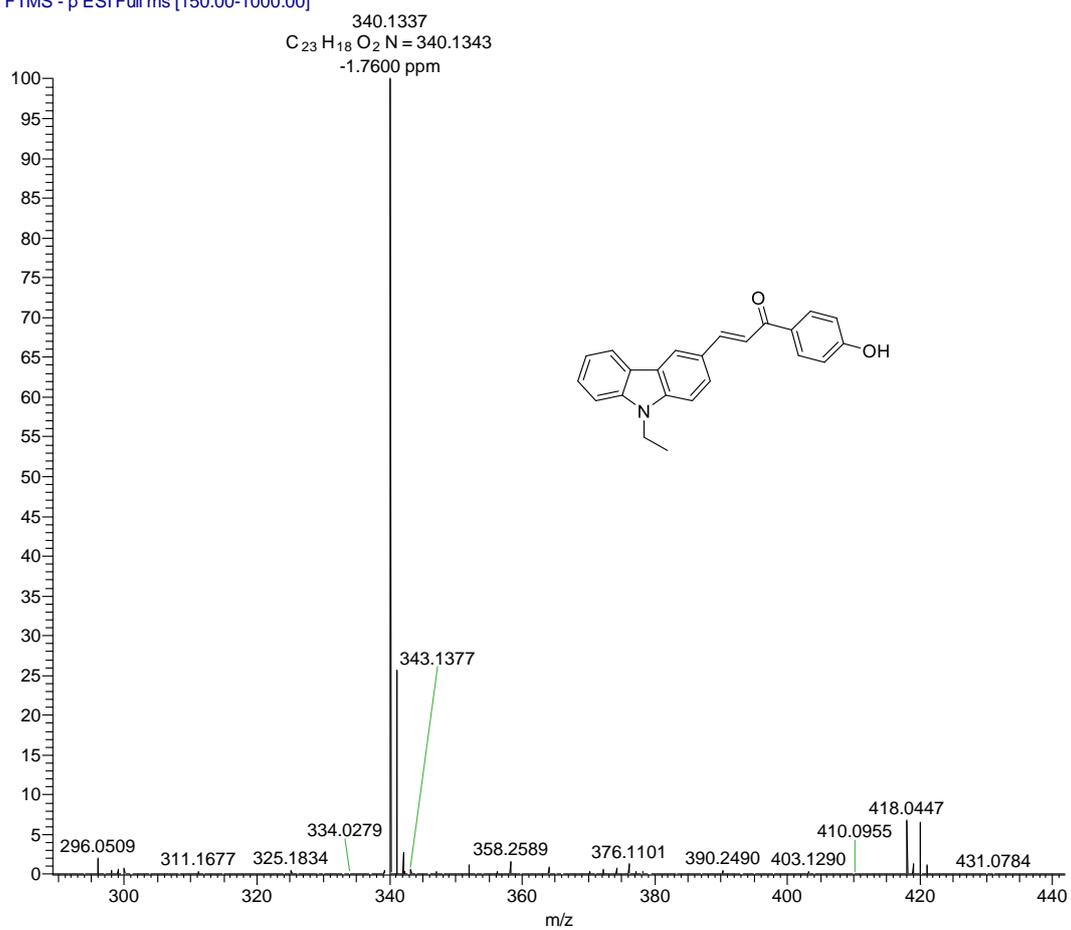


¹H NMR spectrum of probe-OH

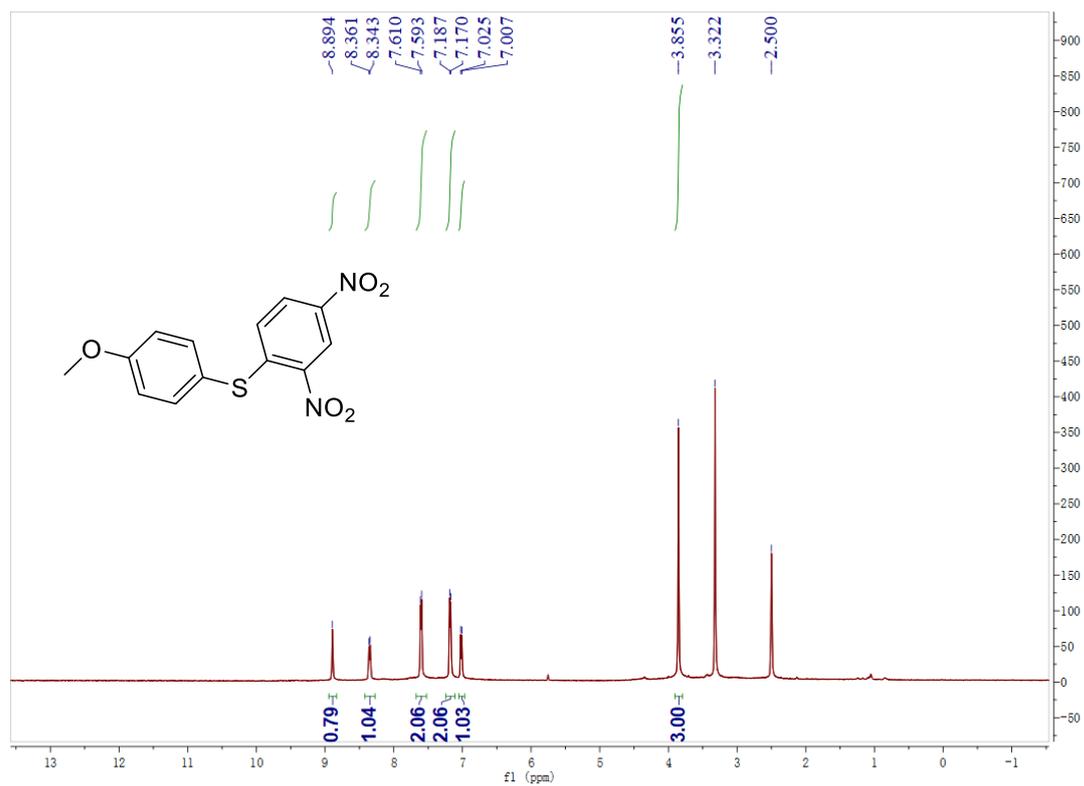


¹³C NMR spectrum of probe-OH

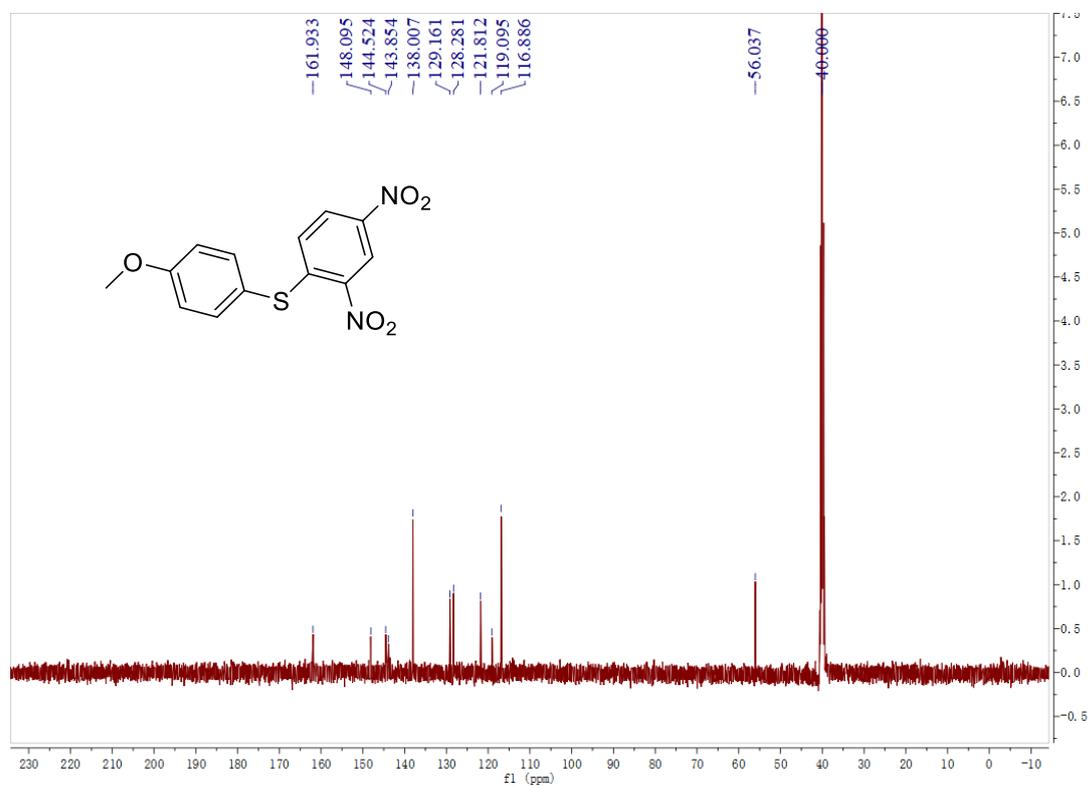
20181203-C2-OH_181203102851 #33-34 RT: 0.36-0.37 AV: 2 NL: 4.18E6
T: FTMS - p ESI Full ms [150.00-1000.00]



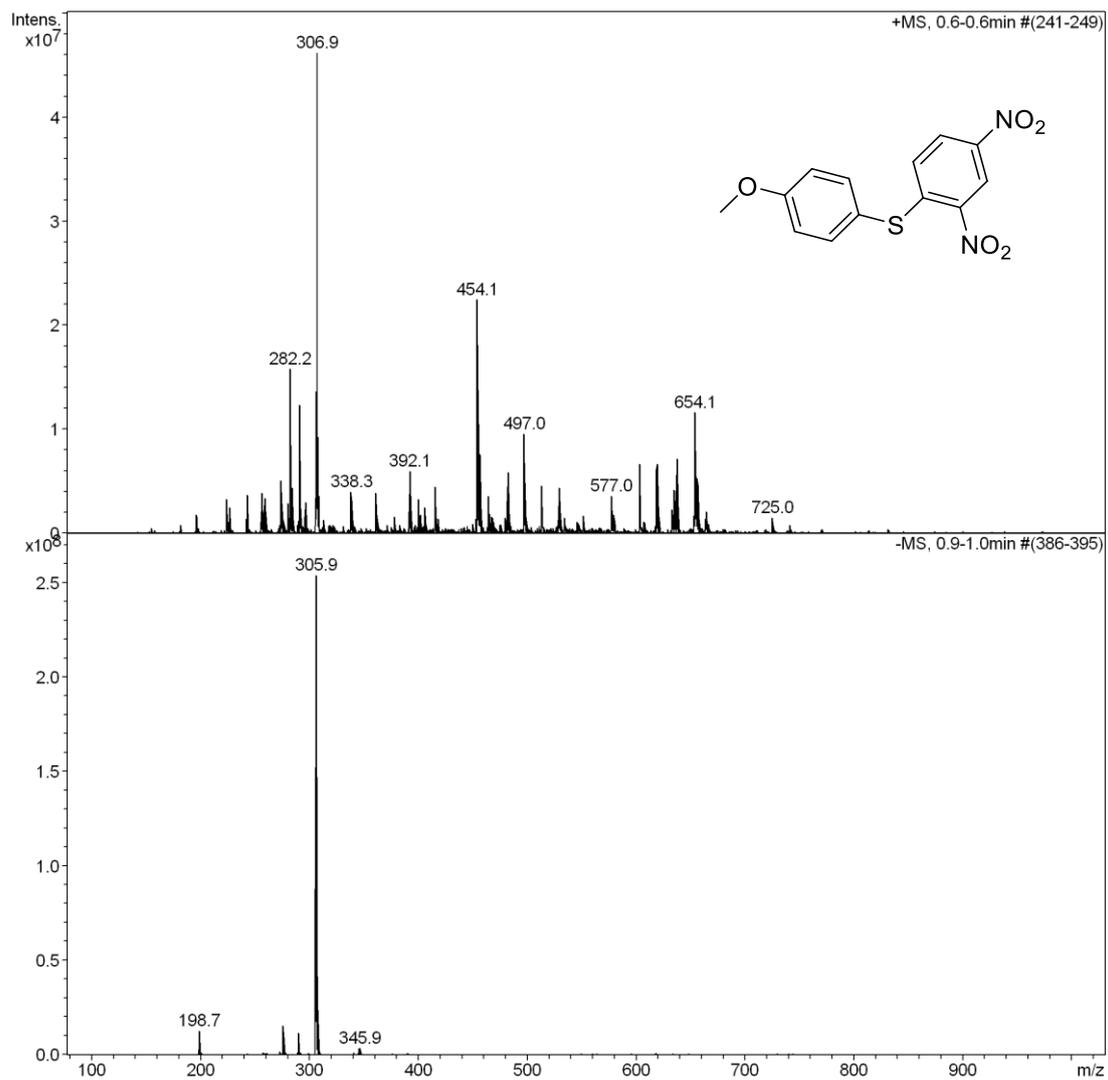
HR-MS spectrum of probe-KCN1



¹H NMR spectrum of S-NO₂



¹³C NMR spectrum of S-NO₂



MS spectrum of S-NO₂

Figure S1. data for investigation of the sensing mechanism.

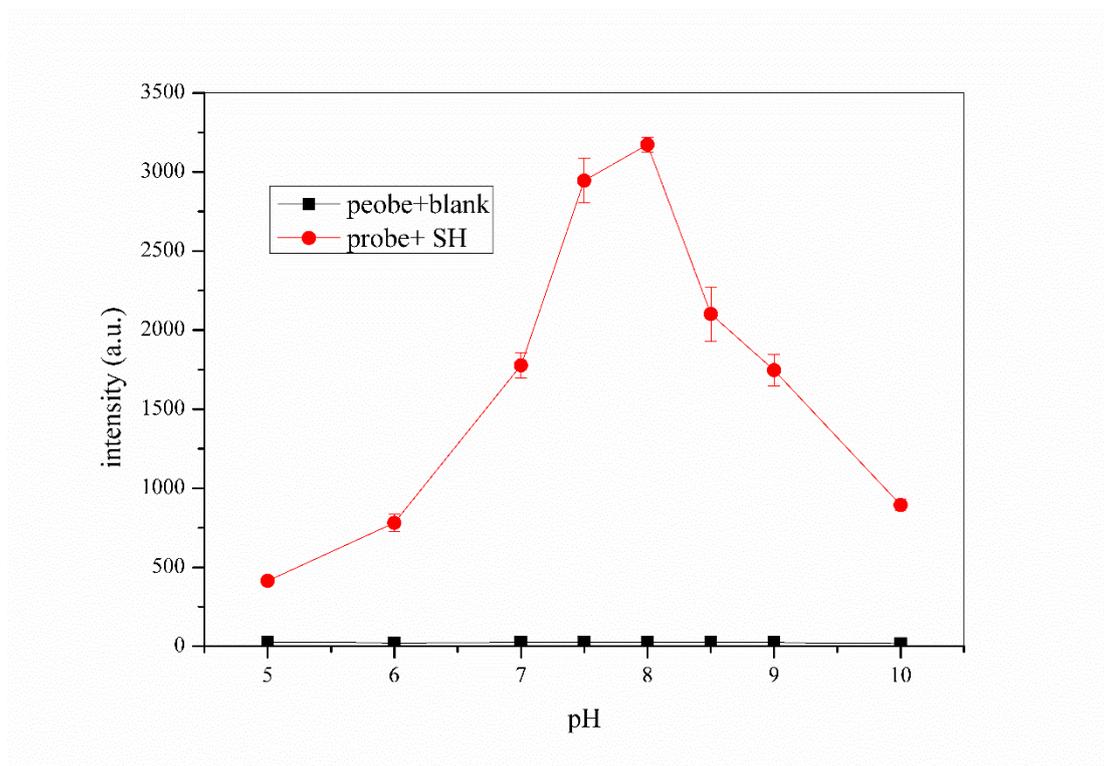


Figure S2. The effect of pH on the fluorescence intensity ($\lambda_{em} = 540$ nm) of probe-KCP (10 μ M, $\lambda_{ex} = 410$ nm) in DMSO/PBS buffer (1:1, v/v, 20 mM) upon addition of 100 μ M 4-Methoxy thiophenol after incubation at 37 $^{\circ}$ C for 20 min.

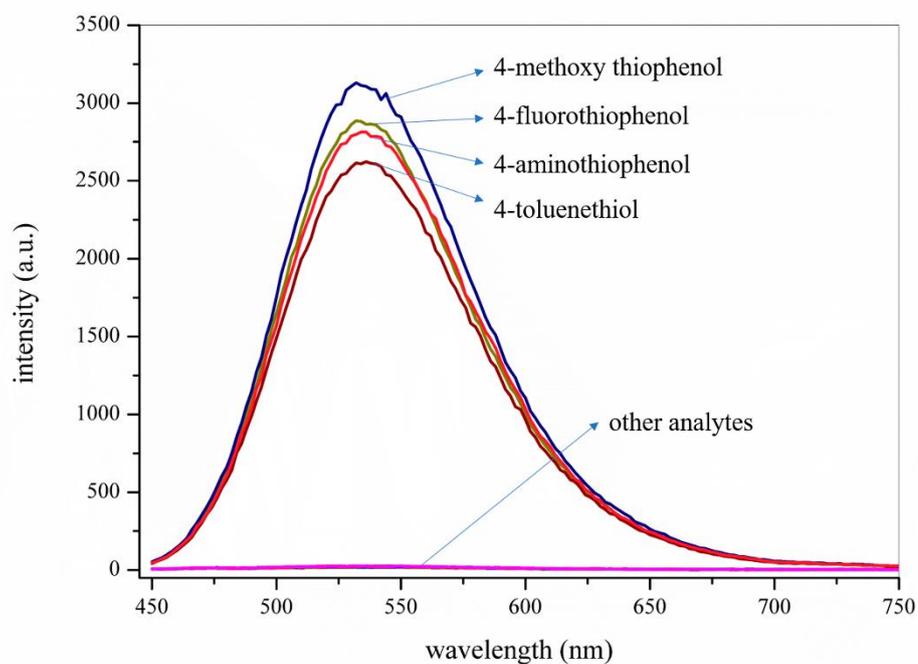


Figure S3. Fluorescence responses of probe-KCP (10 μ M) to thiophenol and other

various analytes (100 μM) in PBS buffer solution (20 mM, pH = 7.4) containing 50 % DMSO.

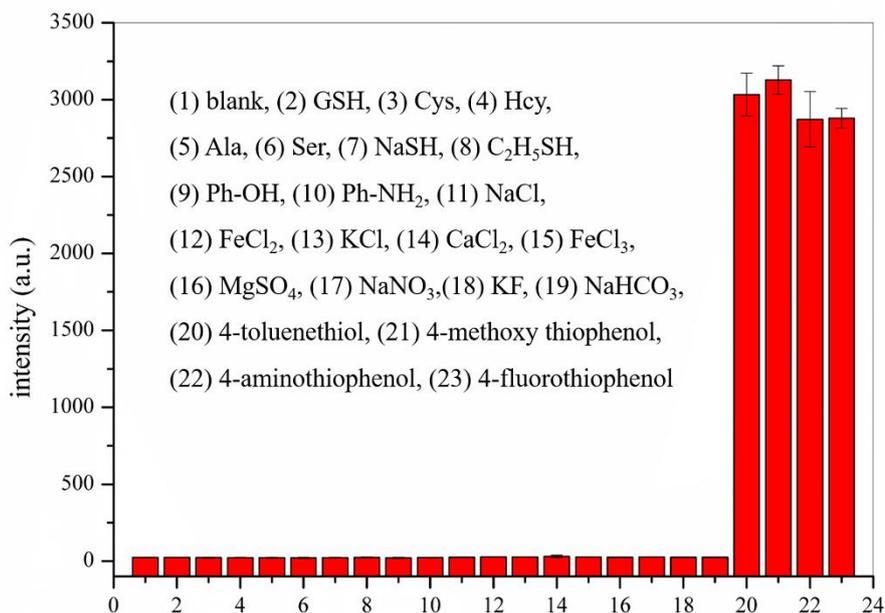


Figure S4. Enhanced fluorescence response at 540 nm of the probe-KCP (10 μM) to thiophenol and other various analytes (100 μM) in PBS buffer solution (20 mM, pH = 7.4) containing 50 % DMSO.

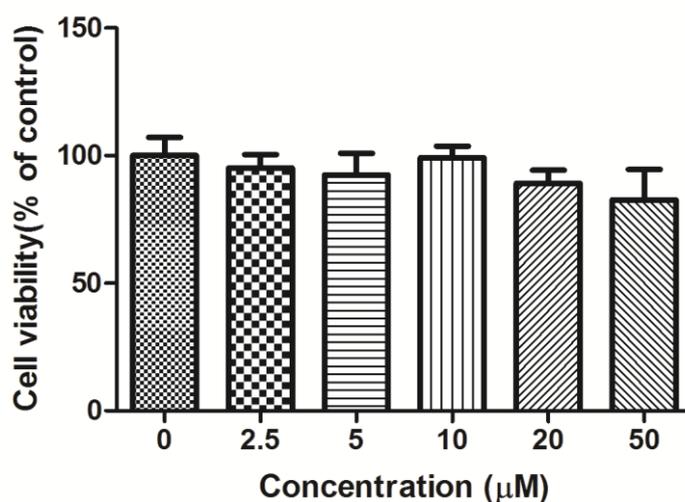


Figure S5. Percentage of viable A549 cells after treatment with different concentrations of the probe-KCP after 24 h using an MTT assay.

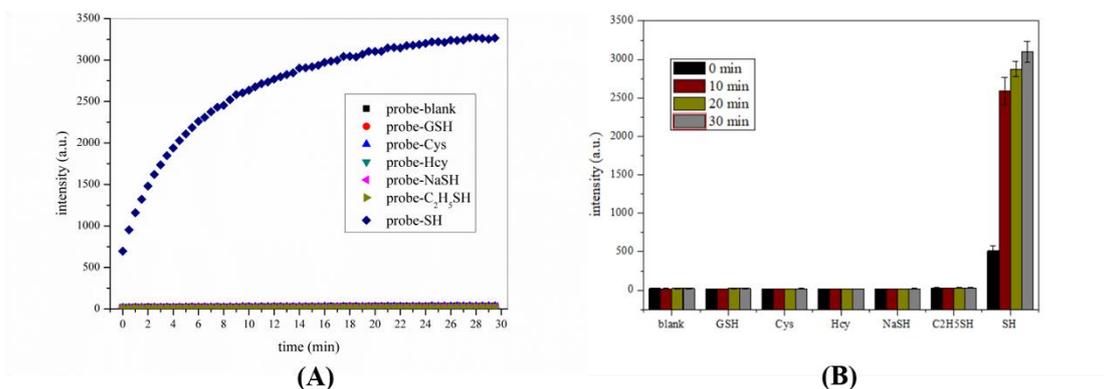


Figure S6. (A) Time-dependent the fluorescence response of probe-KCP (10 μ M) in the absence (blank) and presence of 4-methoxythiophenol, GSH, Cys, Hcy, NaSH or C₂H₅SH (10 equiv) in PBS buffer solution (20 mM, pH = 7.4) containing 50 % DMSO. (B) Time-dependent the fluorescence response of probe-KCP (10 μ M) in the absence (blank) and presence of 4-methoxythiophenol, GSH, Cys, Hcy, NaSH or C₂H₅SH (10 equiv) at 0 min, 10 min, 20 min, 30 min.

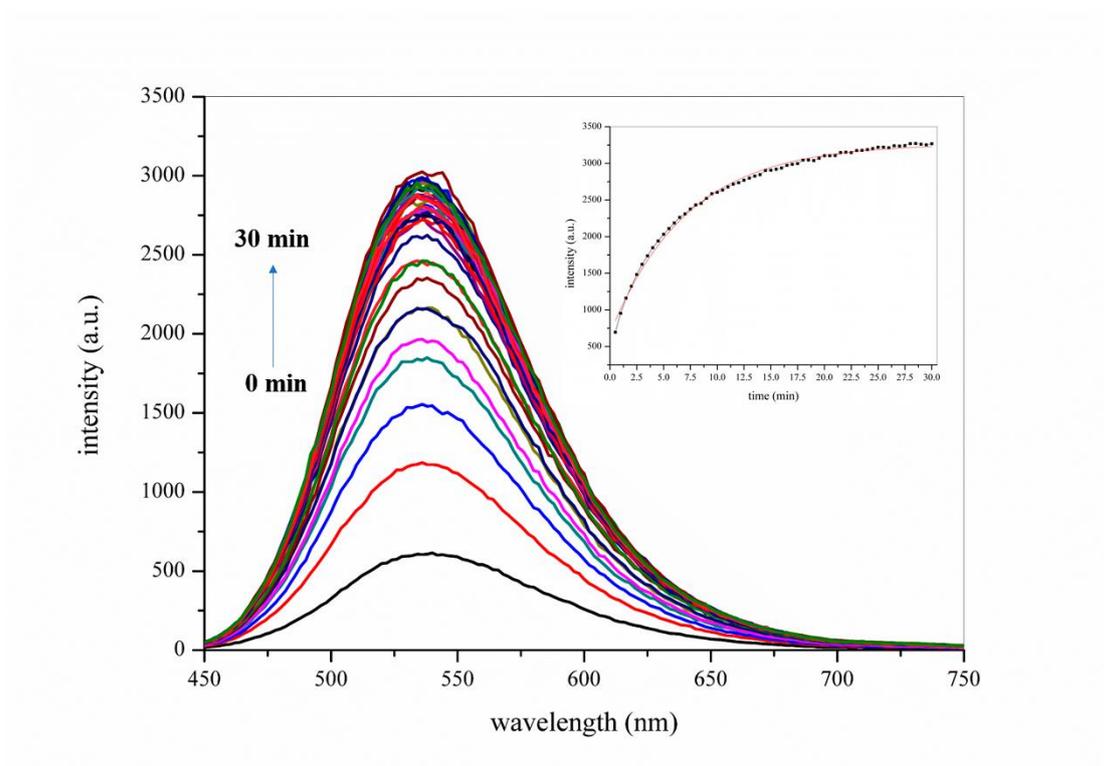


Figure S7. Time-dependent the fluorescence response of probe-KCP (10 μ M) in the presence of 4-Methoxythiophenol (10 equiv) in PBS/ DMSO solution (v:v = 1:1, 20 mM, pH = 7.4).

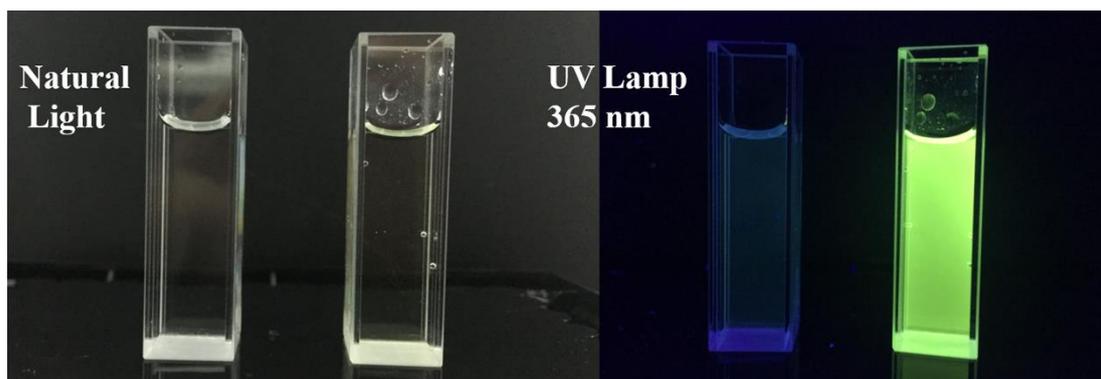


Figure S8. Photograph of probe-KCP solutions (10 μ M) in the presence of 4-methoxythiophenol (10 equiv) under natural light and UV irradiation (365 nm).

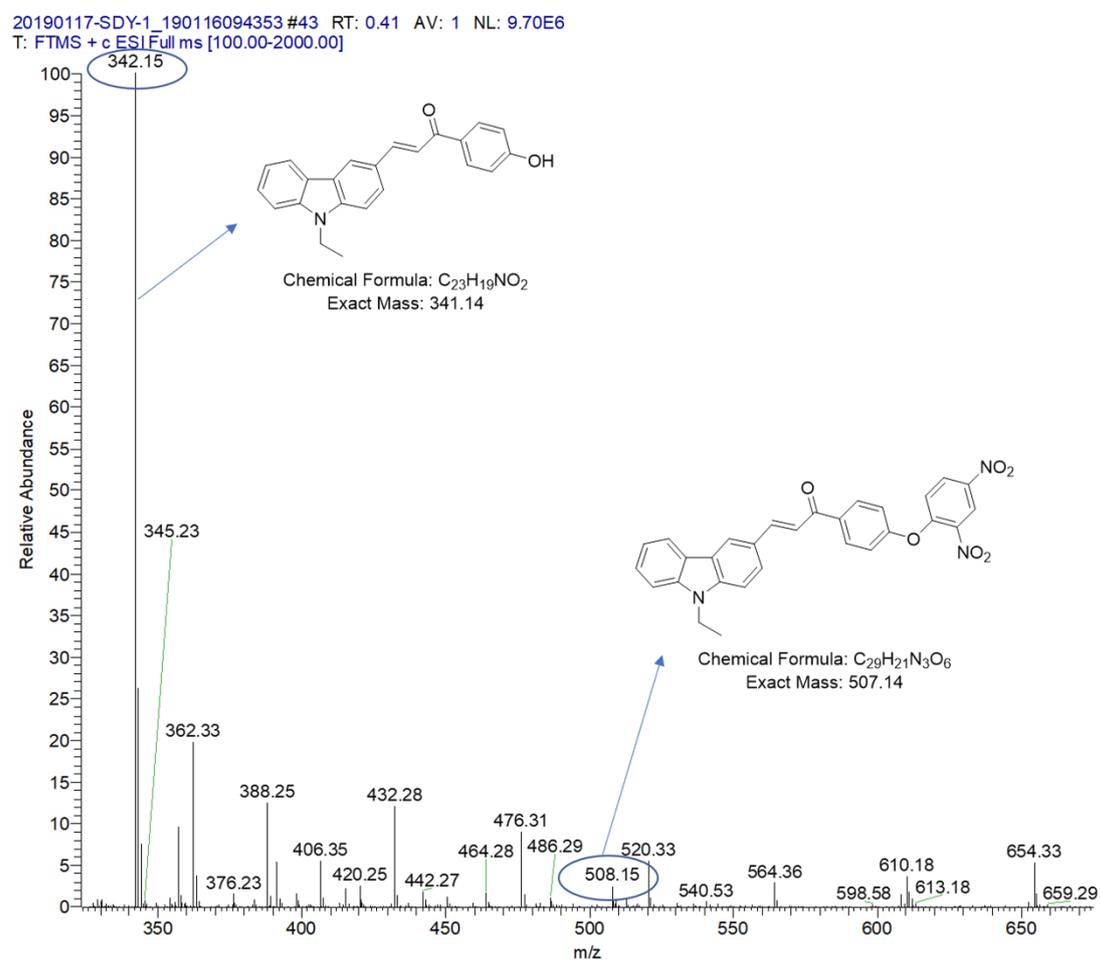


Figure S9. The ESI mass spectrum of probe-KCP in the presence of 4-methoxythiophenol.

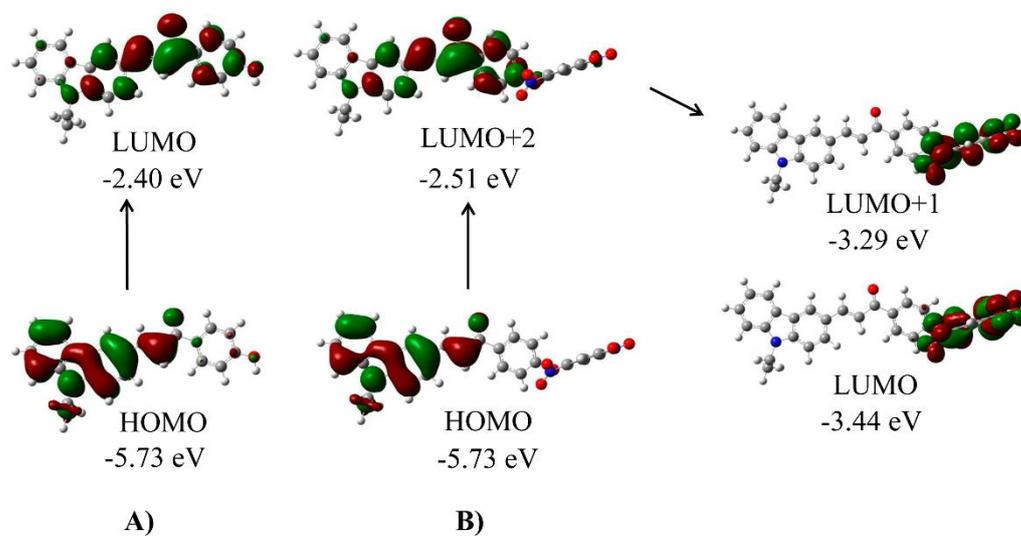


Figure S10. (A) Frontier molecular orbital plots of dye probe-OH in DMSO. (B) Frontier molecular orbital plots of probe-KCP in DMSO. The fluorescence emission of probe-KCN1 moieties is quenched by d-PET.