Electronic Structure and Lithium Diffusion in LiAl₂(OH)₆Cl Studied by First Principle Calculations

Yueping Zhang¹, Xiyue Cheng¹, Chen Wu¹, Jürgen Köhler^{1, 2}, Shuiquan Deng^{1, *}

¹ State Key Laboratory of Structural Chemistry, Fujian Institute of Research on the Structure of Matter (FJIRSM) Chinese Academy of Sciences (CAS) Fuzhou, 350002, China

- ² Max-Planck-Institute for Solid State Research, Heisenbergstr. 1, D-70569 Stuttgart, Germany
- * Correspondence: sdeng@fjirsm.ac. Tel.:+86-0591-6317-3252

Item		Experiment [1]	After relaxation	
Space group		P6 ₃ /mcm (193)	-	
Lattice		a = b = 5.10 Å, c =14.2994 Å;	A = b = 5.158 Å, c = 14.2588	
		$\alpha = \beta = 90^{\circ}; \ \gamma = 120^{\circ}$	Å; α=β=90°; γ=120°	
Wyckoff Position,	Li	2b, (0, 0, 0)	2b, (0, 0, 0)	
Atomic position	Al	4d, (1/3, 2/3, 0)	4d, (1/3, 2/3, 0)	
	Cl	2a, (0, 0, 1/4)	2a, (0, 0, 1/4)	
	0	12k, (0.6350, 0.6350, 0.5672)	12k, (0.6325, 0.6325, 0.5685)	
	Н	12k, (0.6879, 0.6879, 0.6335)	12k, (0.6879, 0.6879, 0.6335)	

Table S1 Crystal data of LiAl₂(OH)₆Cl

Table S2 Calculated geometrical parameters for LiAl₂(OH)₆Cl.

Item	Experimental	Calculated	Item	Experimental	Calculated
	value (Å)	value (Å)		value (°)	value (°)
d _{Li-O}	2.095	2.132	∠0-Li-0	79.4~100.6	79.3~100.7
d _{Al-O}	1.888	1.907	∠0-Al-0	77.5~96.4	77.3~96.1
d _{HCl}	2.304	2.300	∠O-HCl	152.2	153.8
d _{OCl}	3.209	3.208	∠H-O-Li	101.4	99.50



Figure S1 Projection of a) the (2×2×1) layer of the structure of LiAl₂(OH)₆Cl along the [001]



direction; b) layer (1×2×1) of the structure showing only the Al₂(OH)₆ part

Figure S2 Orbital-resolved band structure of LiAl2(OH)6Cl from -0.3 to 0 eV



Figure S3 The arrangement of atom pairs in the structure of of LiAl₂(OH)₆Cl corresponding to the COHP plots (see Fig. 2c) with ① for Cl-H, ② for Cl-O, ③ for H-O ④ for Li-O and ⑤ for Al-O, respectively.

Reference

 Besserguenev, A.V.; Fogg, A.M.; Francis, R.J.; Price, S.J.; O'Hare, D.; Isupov, V.P.; Tolochko, B.P. Chem. Mater. 1997, 9, 241-247.