Supporting Information for Fluorescent Thienothiophene-Containing Squaraine Dyes and Threaded Supramolecular Complexes with Tunable Wavelengths between 600-800 nm Wenqi Liu, Hannah H. McGarraugh, and Bradley D. Smith* Department of Chemistry and Biochemistry, 236 Nieuwland Science Hall, University of Notre Dame, Notre Dame, IN 46556, USA <u>*smith.115@nd.edu</u>

- A. Compound Characterization
- B. Titration Data
- C. Photophysical Data
- D. Chemical Stability Data
- E. Fluorescence Quantum Yields

A. Compound Characterization

The peaks assignments below are based on precedent and logic. They have not been confirmed by correlation methods.



¹H NMR (500 MHz, CDCl₃) of **S1**.



 ^{13}C NMR (126 MHz, CDCl_3) of S1.



Chemical Formula: C₃₄H₃₁N₂O₃S* Exact Mass: 547.2050 Molecular Weight: 547.6925 m/z: 547.2050 (100.0%), 548.2083 (68.%), 549.2008 (4.5%), 549.2117 (3.9%), 549.2117 (2.7%), 550.2041 (1.7%) Elemental Analysis: C, 74.56; H, 5.71; N, 5.11; O, 8.76; S, 5.85



HRMS-ESI of S1.



¹H NMR (500 MHz, CDCl₃) of **S1PEG**.



MS-MALDI (DHBA as matrix) of **S1PEG**. A set of peaks around 2363 reflect the polydispersity of the PEG_{45} chains.



¹H NMR (500 MHz, CDCl₃) of **4**.





¹³C NMR (126 MHz, CDCl₃) of **4**.



Chemical Formula: C₆H₃BrS₂ Exact Mass: 217.8860 Molecular Weight: 219.1140 m/z: 217.8860 (100.0%), 219.8839 (97.3%), 219.8818 (9.0%), 221.8797 (8.8%), 218.8893 (6.5%), 220.8873 (6.3%), 218.8853 (1.6%), 220.8833 (1.6%) Elemental Analysis: C, 32.89; H, 1.38; Br, 36.47; S, 29.26



HRMS-ESI of 4.



¹H NMR (500 MHz, Acetone-d₆) of **5**.



¹³C NMR (126 MHz, Acetone-d₆) of **5**.



Exact Mass: 214.0355 Molecular Weight: 214.3205 m/z: 214.0355 (100.0%), 215.0388 (9.7%), 216.0313 (9.0%), 215.0349 (1.6%) Elemental Analysis: C, 50.44; H, 5.64; N, 6.54; O, 7.46; S, 29.92



HRMS-ESI of 5.



HRMS-ESI of 6.





¹H NMR (500 MHz, CDCl₃) of **S2PEG**.



MS-MALDI (DHBA as matrix) of **S2PEG**. A set of peaks around 2421 reflect the polydispersity of the PEG₄₅ chain.



¹H NMR (500 MHz, Acetone-d₆) of **7**.



 ^{13}C NMR (126 MHz, Acetone-d₆) of **7**.







¹H NMR (500 MHz, Acetone-d₆) of 8.



 ^{13}C NMR (126 MHz, Acetone-d_6) of $\boldsymbol{8}.$

Chemical Formula: C₁₃H₁₆NOS₂+ Exact Mass: 266.0668 Molecular Weight: 266.3965 m/z: 266.0668 (100.0%), 267.0701 (14.1%), 268.0626 (9.0%), 267.0662 (1.6%) Elemental Analysis: C, 58.61; H, 6.05; N, 5.26; O, 6.01; S, 24.07



HRMS of 8.



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 $\label{eq:chemical Formula: $C_{30}H_{28}N_2O_4S_4$ \\ Exact Mass: 608.0932$ \\ Molecular Weight: 608.8040$ \\ m/z: 608.0932 (100.0\%), 609.0965 (32.4\%), 610.0890 (18.1\%), 611.0923$ \\ (5.9\%), 610.0999 (5.1\%), 609.0926 (3.2\%), 612.0848 (1.2\%), 610.0959$ \\ (1.0\%)$ \\ Elemental Analysis: C, 59.19; H, 4.64; N, 4.60; O, 10.51; S, 21.06$ \\ \end{tabular}$



HRMS-ESI of S3.



¹H NMR (500 MHz, CDCl₃) of **S3PEG**.



Chemical Formula: C₂₁₂H₃₉₄N₈NaO₉₄S₄+ Exact Mass: 4707.5071 Molecular Weight: 4710.6752

 Exact Mass: 4707.5071

 Molecular Weight: 4710.6752

 m/z: 4709.5138 (100.0%), 4708.5105 (87.6%), 4710.5172 (75.7%), 4707.5071 (38.2%), 4711.5206 (28.4%), 4711.5181

 (19.3%), 4712.5239 (18.1%), 4711.5096 (18.1%), 4710.5147 (16.9%), 4710.5063 (15.8%), 4711.5206 (14.0%),

 4712.5130 (13.7%), 4712.5214 (12.2%), 4709.5114 (7.4%), 4709.5029 (6.9%), 4713.5248 (5.5%), 4713.5163 (5.1%),

 4713.5139 (3.5%), 4714.5197 (3.3%), 4710.5132 (3.2%), 4713.5273 (3.2%), 4712.5105 (3.1%), 4710.5109 (3.0%),

 4709.5099 (2.8%), 4709.5075 (2.6%), 4713.5163 (2.5%), 4711.5166 (2.4%), 4711.5142 (2.2%), 4714.5172 (2.2%),

 4713.5273 (1.7%), 4711.5072 (1.3%), 4713.5054 (1.2%), 4708.5065 (1.2%), 4708.5042 (1.1%), 4712.5021 (1.1%)

 Elemental Analysis: C, 54.05; H, 8.43; N, 2.38; Na, 0.49; O, 31.93; S, 2.72



MS-MALDI (DHBA as matrix) of **S3PEG**. A set of peaks around 4391 reflect the polydispersity of the PEG₄₅ chains.

B. Titration Data



Figure S1. (*left*) Fluorescence (ex: 630 nm, em: 700 nm, slit 3 nm) titration isotherm for incremental addition of **M2** to a solution of **S1PEG** (250 nM) in water. (*right*) Threading kinetic profile generated by mixing equal molar concentration (50 nM each) of **S1PEG** and **M2** in a stopped flow device (ex: 630 nm, em: 700 nm, slit 3 nm). The red lines are computer fits of experimental data to 1:1 binding model or second order kinetic model, respectively.



Figure S2. (*left*) Fluorescence (ex: 723 nm, em: 753 nm, slit 5 nm) titration isotherm for incremental addition of **M2** to a solution of **S2PEG** (250 nM) in water. (*right*) Threading kinetic profilegenerated by mixing equal molar concentration (50 nM each) of **S2PEG** and **M2** in a stopped flow device (ex: 723 nm, em: 753 nm, slit 3 nm). The red lines are computer fits of experimental data to 1:1 binding model or second order kinetic model, respectively.

C. Photostability Data



Figure S3. Photostability of free **S1PEG** (5 μ M, *left*) and **M2** \supset **S1PEG** (5 μ M, *right*) in H₂O with continuous irradiation at 550 nm over 15 h (fluorescence spectrum was collected every 30 min).



Figure S4. Photostability test of free **S2PEG** (2 μ M, *left*) and **M2** \supset **S3PEG** (2 μ M, *right*) in H₂O with continuous irradiation at 650 nm over 15 h (fluorescence spectrum was collected every 30 min). Note: slight increase in fluorescence over time for free **S2PEG** is attributed to slow deaggregation.



Figure S5. Photo stability test of the **S3PEG** (2 μ M, *left*) and **M2** \supset **S3PEG** (2 μ M, *right*) in H₂O with continuous irradiation at 750 nm over 15 h (fluorescence spectrum was collected every 30 min).

D. Chemical Stability Data



Figure S6. Chemical stability test of **S1PEG** and **M2** \supset **S1PEG**. (*left*) Change in fluorescent maxima band for solutions of (A) **M2** \supset **S1PEG** (5.0 µM) or (B) free **S1PEG** (5 µM), in the presence of excess nucleophile Na₂S (5 mM) in water at 20°C. (*right*) Photograph of samples containing, (A) **M2** \supset **S1PEG** (80 µM) or (B) free **S1PEG** (80 µM), after sitting in the presence of excess nucleophile Na₂S (100 mM) in water at 20 °C.



Figure S7. Chemical stability test of **S2PEG** and **M2** \supset **S2PEG** by monitoring the change in fluorescent maxima band for solutions of (A) **M2** \supset **S2PEG** (5.0 µM) or (B) free **S2PEG** (5 µM) over time in the presence of excess nucleophile Na₂S (5 mM) in water at 20 °C.



Figure S8. Chemical stability test of **S3PEG** and **M2** \supset **S3PEG** by monitoring change in fluorescent maxima band for solutions of (A) **M2** \supset **S3PEG** (1.0 µM) or (B) free **S3PEG** (1 µM), in the presence of excess nucleophile Na₂S (1 mM) in water at 20 °C.

E. Fluorescence Quantum Yields

Table S1. Integrated fluorescent and quantum yield of ICG, S3PEG, and M2 \supset S3PEG in H₂O.

	ICG	S3PEG	$\mathbf{M2} \supset \mathbf{S3PEG}$
Integrated fluorescence	2.2×10 ⁶	2.9×10 ⁶	4.4×10 ⁶
Quantum yield %	5.3	7.0	10.6

Table S2. Photophysical properties of squaraines and their complexes with M1 in CHCl₃.

	S1	M1⊃S1	S2	M1⊃S2	S 3	M1⊃S3
Abs (nm)	627	650	692	717	767	783
Em (nm)	651	685	715	749	792	820
logɛ (M ⁻¹ cm ⁻¹)	5.39	4.68	5.48	4.84	5.59	5.02