## Supporting Information: Quantitative Assessment of Tetrel Bonding Utilizing Vibrational Spectroscopy

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Figure S1: Schematic representation of monomers (36-63) with atomic charges from the natural population analysis calculated at CCSD(T)/aug-cc-pVTZ level of theory. Colors are used to correlate charges to specific atoms.



Figure S2: Noncovalent interactions (NCIs) plot of complexes **1-35** calculated at gradient isosurfaces s = 0.5 au., where green indicates weak attractive, blue indicates strong attractive, and red surface indicates repulsive noncovalent interactions.



Figure S3: Selected molecular orbitals of the T-acceptors. Calculated at  $\rm HF/6-31G(d)$  level of theory.



Figure S4: CCSD(T)/aug-cc-pVTZ electron difference density distributions  $\Delta \rho(\mathbf{r})$  given for complexes **1-35**.  $\Delta \rho(\mathbf{r})$  is plotted for an electron density surface with a constant density value of 0.001 a.u. Blue regions indicate an increase in the electron density, red regions indicate a density decrease relative to the superimposed density of the monomer.



Figure S5: Combination of donor and acceptor NBO orbitals involved in the electron delocalization of selected complexes. Delocalization energies are given in kcal/mol

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		CCSD(T	)/aug-cc-pVT	Z	CCSD(T	)/aug-cc-pV52	
#	Complex (symm.)	$E_{int}$	$E_{int}(CP)$	BSSE	$E_{int}$	$E_{int}(CP)$	BSSI
C - Tet	rel bond						
1	FCH <sub>3</sub> …FH (C <sub>3v</sub> )	-1.51	-1.30	0.21	-1.36	-1.33	0.0
2	$FCH_3 \cdots OH_2 (C_s)$	-2.11	-1.89	0.22	-1.96	-1.93	0.0
3	$FCH_3 \cdots NH_3 (C_{3v})$	-2.27	-2.07	0.20	-2.15	-2.12	0.0
4	$ClCH_3 \cdots NH_3 (C_{3v})$	-2.09	-1.90	0.19	-2.00	-1.96	0.0
5	$\operatorname{BrCH}_3 \cdots \operatorname{NH}_3 (C_{3v})$	-2.02	-1.81	0.21	-1.91	-1.88	0.0
6	$HOCH_3 \cdots NH_3 (C_s)$	-1.39	-1.23	0.16	-1.27	-1.25	0.0
7	$CF_4 \cdots NH_3 (C_{3v})$	-1.69	-1.30	0.39	-1.38	-1.32	0.0
Si - Tet	trel bond						
8	FSiH <sub>3</sub> ···FH (C <sub>s</sub> )	-2.34	-1.91	0.43	-2.08	-2.02	0.0
9	$FSiH_3 \cdots OH_2$ (C <sub>s</sub> )	-4.55	-3.96	0.59	-4.31	-4.24	0.0
10	$FSiH_3 \cdots NH_3 (C_{3v})$	-8.91	-8.06	0.85	-8.81	-8.72	0.0
11	$ClSiH_3 \cdots NH_3 (C_{3v})$	-8.16	-7.44	0.72	-8.19	-8.09	0.1
12	$BrSiH_3 \cdots NH_3 (C_{3v})$	-8.35	-7.58	0.77	-8.39	-8.30	0.0
13	$HOS_{1}H_{3}\cdots NH_{3}(C_{s})$	-4.81	-4.29	0.52	-4.70	-4.64	0.0
14	$\operatorname{SiH}_4 \cdots \operatorname{NH}_3 (C_{3v})$	-2.41	-2.12	0.29	-2.32	-2.29	0.0
15	$S_1F_2H_2\cdots NH_3(C_s)$	-11.73	-10.47	1.26	-11.46	-11.32	0.1
16a 16b	$SiF_3H\cdots NH_3(C_s)$	-19.43	-17.34	1.89	-19.06	-18.84	0.2
17	$SIF_3 \Pi \cdots N \Pi_3 (C_{3v})$	-27.00	-20.00	2.17	-27.29	-27.01	0.2
 Ge - Te	etrel bond						
uc - 10							
18	$FGeH_3 \cdots NH_3 (C_{3v})$	-9.17	-8.58	0.59	-9.04	-8.97	0.0
19	$CIGeH_3 \cdots NH_3 (C_{3v})$	-7.29	-6.82	0.47	-7.28	-7.20	0.0
20	$\operatorname{BrGeH}_{3} \cdots \operatorname{NH}_{3} (C_{3v})$	-7.08	-6.60	0.48	-7.09	-7.01	0.0
21	$HOGeH_3 \cdots NH_3 (C_s)$	-5.08	-4.68	0.40	-4.97	-4.92	0.0
22	$GeH_4 \cdots NH_3 (C_{3v})$	-2.08	-1.88	0.20	-2.01	-1.99	0.0
Double	bond - Tetrel bond						
23	$\rm CO_2 \cdots NH_3~(C_s)$	-3.21	-2.95	0.26	-3.09	-3.04	0.0
<b>24</b>	$SCO \cdots NH_3 (C_s)$	-1.99	-1.71	0.28	-1.79	-1.75	0.0
25	$CF_2O\cdots NH_3$ (C <sub>s</sub> )	-5.81	-5.09	0.72	-5.35	-5.25	0.1
26a	$CF_2S\cdots NH_3$ (C <sub>s</sub> )	-4.02	-3.34	0.68	-3.53	-3.44	0.0
26b	$CF_2S\cdots NH_3$ (C <sub>s</sub> )	-22.68	-19.85	2.83	-23.73	-23.27	0.4
27	$SiF_2O\cdots NH_3$ (C <sub>s</sub> )	-52.10	-50.12	1.98	-53.34	-53.07	0.2
Anionio	c - Tetrel bond						
28	$\mathrm{CH_3}^+ \cdots \mathrm{NH_3} (\mathrm{C_{3v}})$	-135.20	-133.96	1.24	-135.92	-135.74	0.1
29	$\text{FNH}_3^+ \cdots \text{NH}_3 (C_{3v})$	-23.56	-23.20	0.36	-23.49	-23.43	0.0
30	$FCH_3 \cdots Cl^- (C_{3v})$	-10.16	-9.73	0.43	-10.11	-9.98	0.1
31	$FSiH_3 \cdots Cl^- (C_{3v})$	-32.76	-31.52	1.24	-33.42	-33.14	0.2
32	$FGeH_3 \cdots Cl^- (C_{3v})$	-36.81	-35.79	1.02	-37.18	-36.93	0.2
33	$CO_2 \cdots Cl^- (C_s)$	-8.89	-8.43	0.46	-8.89	-8.77	0.1
34	$SCO\cdots Cl^{-}(C_s)$	-5.88	-5.48	0.40	-5.85	-5.75	0.1
35	$CF_2S\cdots Cl^-$ (C <sub>s</sub> )	-49.44	-46.46	2.98	-50.61	-50.02	0.5

\*Interaction energies  $(E_{int})$ , counterpoise (CP) corrected interaction energies and basis set superposition error calculated at CCSD(T)/aug-cc-pVTZ and DLPNO-CCSD(T)/aug-cc-pV5Z levels. All energies were obtained using CCSD(T)/aug-cc-pVTZ geometries. Values are given in kcal/mol.

## Table S2: Deviation from DLPNO-CCSD(T)/aug-cc-pV5Z interaction energies $^{\ast}$

#	Complex (symm.)	(a)	(b)	(c)	
C - Tet	rel bond				
1 2 3 4 5 6 7	$\begin{array}{l} \operatorname{FCH}_3\cdots\operatorname{FH}(\operatorname{C}_{3\mathrm{v}})\\ \operatorname{FCH}_3\cdots\operatorname{OH}_2(\operatorname{C}_s)\\ \operatorname{FCH}_3\cdots\operatorname{NH}_3(\operatorname{C}_{3\mathrm{v}})\\ \operatorname{ClCH}_3\cdots\operatorname{NH}_3(\operatorname{C}_{3\mathrm{v}})\\ \operatorname{BrCH}_3\cdots\operatorname{NH}_3(\operatorname{C}_{3\mathrm{v}})\\ \operatorname{BrCH}_3\cdots\operatorname{NH}_3(\operatorname{C}_{3\mathrm{v}})\\ \operatorname{HOCH}_3\cdots\operatorname{NH}_3(\operatorname{C}_{3\mathrm{v}})\\ \end{array}$	-0.03 -0.04 -0.05 -0.06 -0.07 -0.02 -0.02	$\begin{array}{c} 0.18 \\ 0.18 \\ 0.15 \\ 0.13 \\ 0.14 \\ 0.14 \\ 0.37 \end{array}$	$\begin{array}{c} 0.15 \\ 0.15 \\ 0.12 \\ 0.09 \\ 0.11 \\ 0.12 \\ 0.31 \end{array}$	
Si - Tet	rel bond				
8 9 10 11 12 13 14 15 16a 16b 17	$\begin{array}{l} {\rm FSiH}_{3}\cdots {\rm FH}\;({\rm C}_{\rm s})\\ {\rm FSiH}_{3}\cdots {\rm OH}_{2}\;({\rm C}_{\rm s})\\ {\rm FSiH}_{3}\cdots {\rm NH}_{3}\;({\rm C}_{3\rm v})\\ {\rm ClSiH}_{3}\cdots {\rm NH}_{3}\;({\rm C}_{3\rm v})\\ {\rm BrSiH}_{3}\cdots {\rm NH}_{3}\;({\rm C}_{3\rm v})\\ {\rm HOSiH}_{3}\cdots {\rm NH}_{3}\;({\rm C}_{3\rm v})\\ {\rm SiH}_{4}\cdots {\rm NH}_{3}\;({\rm C}_{3\rm v})\\ {\rm SiF}_{2}{\rm H}_{2}\cdots {\rm NH}_{3}\;({\rm C}_{{\rm s}})\\ {\rm SiF}_{3}{\rm H}\cdots {\rm NH}_{3}\;({\rm C}_{{\rm s}})\\ {\rm SiF}_{3}{\rm H}\cdots {\rm NH}_{3}\;({\rm C}_{{\rm s}})\\ {\rm SiF}_{4}\cdots {\rm SiF}_{4}\;({\rm C}_{{\rm s}})\\ {\rm SiF}_$	$\begin{array}{c} -0.11\\ -0.28\\ -0.66\\ -0.65\\ -0.72\\ -0.35\\ -0.17\\ -0.85\\ -1.30\\ -1.65\\ -1.75\end{array}$	$\begin{array}{c} 0.32\\ 0.31\\ 0.19\\ 0.07\\ 0.05\\ 0.17\\ 0.12\\ 0.41\\ 0.59\\ 0.52\\ 0.80\\ \end{array}$	$\begin{array}{c} 0.26\\ 0.24\\ 0.10\\ -0.03\\ -0.04\\ 0.11\\ 0.09\\ 0.27\\ 0.37\\ 0.24\\ 0.50\\ \end{array}$	
Ge - Te	etrel bond				
18 19 20 21 22	$\begin{array}{l} \operatorname{FGeH}_{3}\cdots\operatorname{NH}_{3}\left(\operatorname{C}_{3v}\right)\\ \operatorname{ClGeH}_{3}\cdots\operatorname{NH}_{3}\left(\operatorname{C}_{3v}\right)\\ \operatorname{BrGeH}_{3}\cdots\operatorname{NH}_{3}\left(\operatorname{C}_{3v}\right)\\ \operatorname{HOGeH}_{3}\cdots\operatorname{NH}_{3}\left(\operatorname{C}_{s}\right)\\ \operatorname{GeH}_{4}\cdots\operatorname{NH}_{3}\left(\operatorname{C}_{3v}\right) \end{array}$	-0.39 -0.38 -0.41 -0.24 -0.11	$0.20 \\ 0.09 \\ 0.07 \\ 0.16 \\ 0.09$	$0.13 \\ 0.01 \\ -0.01 \\ 0.11 \\ 0.07$	
Double	bond - Tetrel bond				
23 24 25 26a 26b 27	$\begin{array}{c} {\rm CO}_2 {\cdots} {\rm NH}_3 \ ({\rm C}_{\rm s}) \\ {\rm SCO} {\cdots} {\rm NH}_3 \ ({\rm C}_{\rm s}) \\ {\rm CF}_2 {\rm O} {\cdots} {\rm NH}_3 \ ({\rm C}_{\rm s}) \\ {\rm CF}_2 {\rm S} {\cdots} {\rm NH}_3 \ ({\rm C}_{\rm s}) \\ {\rm CF}_2 {\rm S} {\cdots} {\rm NH}_3 \ ({\rm C}_{\rm s}) \\ {\rm SiF}_2 {\rm O} {\cdots} {\rm NH}_3 \ ({\rm C}_{\rm s}) \\ \end{array}$	-0.09 -0.04 -0.16 0.10 -3.42 -2.95	0.17 0.24 0.56 -0.58 -0.59 -0.97	0.12 0.20 0.46 -0.49 -1.05 -1.24	
Anionic - Tetrel bond					
28 29 30 31 32 33 34 35	$\begin{array}{c} {\rm CH_3}^+ \cdots {\rm NH_3} \ ({\rm C}_{3v}) \\ {\rm FNH_3}^+ \cdots {\rm NH_3} \ ({\rm C}_{3v}) \\ {\rm FCH_3} \cdots {\rm CI}^- \ ({\rm C}_{3v}) \\ {\rm FSiH_3} \cdots {\rm CI}^- \ ({\rm C}_{3v}) \\ {\rm FGeH_3} \cdots {\rm CI}^- \ ({\rm C}_{3v}) \\ {\rm FGeH_3} \cdots {\rm CI}^- \ ({\rm C}_{3v}) \\ {\rm SCO} \cdots {\rm CI}^- \ ({\rm C}_{s}) \\ {\rm SCO} \cdots {\rm CI}^- \ ({\rm C}_{s}) \\ {\rm CF}_2 {\rm S} \cdots {\rm CI}^- \ ({\rm C}_{s}) \\ \end{array}$	-1.78 -0.23 -0.25 -1.62 -1.14 -0.34 -0.27 -3.56 0.71	-0.54 0.13 0.18 -0.38 -0.12 0.12 0.12 0.13 -0.58	-0.72 0.07 0.05 -0.66 -0.37 0.00 0.03 -1.17 0.28	
		0.11	0.20	0.20	

\*Difference between DLPNO-CCSD(T)/aug-cc-pV5Z  $E_{int}(CP)$  and CCSD(T)/aug-cc-pVTZ  $E_{int}(CP)$  (a) and CCSD(T)/aug-cc-pVTZ  $E_{int}$  (b). Difference between DLPNO-CCSD(T)/aug-cc-pVTZ  $E_{int}$  and CCSD(T)/aug-cc-pVTZ  $E_{int}$  (c). Energies and mean absolute deviation (MAD) given in kcal/mol.

Table S3: Atomic Cartesian coordinates (in Å) of complexes 1-35 optimized at the CCSD(T)/aug-cc-pVTZ level of theory.

1	$FCH_3 \cdots F_2$	H, $C_{3v}$	
С	-0.64760	0.00000	0.00000
F	-2.03974	0.00000	0.00000
Η	-0.29694	0.51589	-0.89355
Η	-0.29694	-1.03178	0.00000
Η	-0.29694	0.51589	0.89355
F	2.32388	0.00000	0.00000
Η	3.24545	0.00000	0.00000

<b>2</b>	$\mathrm{FCH}_3\mathrm{\cdots}\mathrm{OH}_2,\mathrm{C}_s$				
С	0.58503	-0.01958	0.00000		
F	1.97844	0.01163	0.00000		
Η	0.25798	-1.05889	0.00000		
Η	0.22171	0.48606	-0.89362		
Η	0.22171	0.48606	0.89362		
0	-2.44987	0.00395	0.00000		
Η	-3.06002	-0.73963	0.00000		
Η	-3.02146	0.77758	0.00000		

3	$\mathrm{FCH}_3 \mathrm{\cdots} \mathrm{N}$	$H_3, C_{3v}$	
С	-0.59814	0.00000	0.00000
F	-1.99271	0.00000	0.00000
Η	-0.24618	-0.51535	-0.89263
Н	-0.24618	1.03070	0.00000

Η	-0.24618	-0.51535	0.89263
Ν	2.61945	0.00000	0.00000
Η	3.00978	0.46868	0.81177
Η	3.00978	0.46868	-0.81177
Н	3.00978	-0.93735	0.00000

4	$\mathrm{ClCH}_3\cdots$	$\mathrm{NH}_3, \mathrm{C}_{3v}$	
С	0.06943	0.00000	0.00000
$\operatorname{Cl}$	-1.72827	0.00000	0.00000
Η	0.41195	-0.51492	-0.89188
Η	0.41195	1.02985	0.00000
Η	0.41195	-0.51492	0.89188
Ν	3.35811	0.00000	0.00000
Η	3.74840	0.46869	0.81179
Η	3.74840	0.46869	-0.81179
Η	3.74840	-0.93738	0.00000

5	$\operatorname{BrCH}_3\cdots$	$\mathrm{NH}_3, \mathrm{C}_{3v}$	
С	0.86226	0.00000	0.00000
Br	-1.09071	0.00000	0.00000
Η	1.19569	-0.51600	-0.89374
Η	1.19569	1.03200	0.00000
Η	1.19569	-0.51600	0.89374
Ν	4.16614	0.00000	0.00000
Η	4.55633	0.46871	0.81183
Н	4.55633	0.46871	-0.81183

6	$(OH)CH_3$	$\cdots$ NH <sub>3</sub> , C <sub>s</sub>	
Η	-2.49864	0.81659	0.00000
Ο	-2.13732	-0.07379	0.00000
С	-0.71198	0.03232	0.00000
Η	-0.32011	-0.98345	0.00000
Η	-0.33794	0.54691	0.88947
Η	-0.33794	0.54691	-0.88947
Ν	2.65001	0.01052	0.00000
Η	3.28000	0.80658	0.00000
Η	2.89631	-0.54671	-0.81216
Н	2.89631	-0.54671	0.81216

7	$\mathrm{CF}_4 {\cdots} \mathrm{NH}$	$C_3, C_{3v}$	
С	-0.55900	0.00000	0.00000
F	-1.88668	0.00000	0.00000
F	-0.13050	-0.62387	-1.08058
F	-0.13050	1.24774	0.00000
F	-0.13050	-0.62387	1.08058
Ν	2.86699	0.00000	0.00000
Η	3.25551	0.46896	0.81227
Η	3.25551	0.46896	-0.81227
Η	3.25551	-0.93793	0.00000

 $\mathbf{8} \qquad \mathrm{FSiH}_3 \cdots \mathrm{FH}, \, \mathrm{C}_s$ 

Si	0.43320	0.01974	0.00000
F	2.04988	-0.02546	0.00000
Н	-0.03966	-1.38225	0.00000
Н	0.00693	0.72855	-1.22281
Н	0.00693	0.72855	1.22281
F	-2.53084	0.03461	0.00000
Η	-2.93319	-0.79531	0.00000

9	$FSiH_3\cdots C$	$\mathrm{OH}_2,\mathrm{C}_s$	
Si	0.01186	0.31211	0.00000
F	-0.00996	1.93530	0.00000
Η	1.43780	-0.07630	0.00000
Н	-0.68509	-0.11005	-1.23076
Η	-0.68509	-0.11005	1.23076
0	-0.05586	-2.46085	0.00000
Η	0.33866	-2.89719	-0.76172
Н	0.33866	-2.89719	0.76172

10	$FSiH_3\cdots NH_3, C_{3v}$		
Si	-0.20928	0.00000	0.00000
F	-1.84579	0.00000	0.00000
Η	0.12338	-0.72003	-1.24713
Η	0.12338	1.44006	0.00000
Η	0.12338	-0.72003	1.24713
Ν	2.31409	0.00000	0.00000
Η	2.69382	0.47094	0.81569

Η	2.69382	0.47094	-0.81569
Η	2.69382	-0.94188	0.00000

11	$ClSiH_3 \cdots NH_3, C_{3v}$		
Si	0.33660	0.00000	0.00000
Cl	-1.78021	0.00000	0.00000
Н	0.66936	-0.71839	-1.24428
Н	0.66936	1.43677	0.00000
Н	0.66936	-0.71839	1.24428
Ν	2.91649	0.00000	0.00000
Н	3.29796	0.47068	0.81525
Η	3.29796	0.47068	-0.81525
Н	3.29796	-0.94136	0.00000

12	$BrSiH_3 \cdots NH_3, C_{3v}$		
Si	1.06281	0.00000	0.00000
$\operatorname{Br}$	-1.22723	0.00000	0.00000
Η	1.38429	-0.71948	-1.24615
Η	1.38429	1.43893	0.00000
Η	1.38429	-0.71948	1.24615
Ν	3.62836	0.00000	0.00000
Η	4.00960	0.47077	0.81539
Η	4.00960	0.47077	-0.81539
Η	4.00960	-0.94154	0.00000

**13** (HO)SiH<sub>3</sub>···NH<sub>3</sub>, 
$$C_s$$

Η	-2.48714	0.75291	0.00000
Ο	-2.00799	-0.07791	0.00000
Si	-0.33120	0.02073	0.00000
Η	0.07573	-1.39753	0.00000
Η	0.10142	0.74724	1.21938
Η	0.10142	0.74724	-1.21938
Ν	2.49327	-0.00992	0.00000
Η	2.89338	0.92319	0.00000
Η	2.86755	-0.48701	-0.81436
Η	2.86755	-0.48701	0.81436

14	$\operatorname{SiH}_4\cdots\operatorname{NH}_3, \operatorname{C}_{3v}$		
Si	-1.13339	0.00000	0.00000
Η	-2.62311	0.00000	0.00000
Η	-0.67520	-0.70437	-1.22000
Η	-0.67520	1.40873	0.00000
Η	-0.67520	-0.70437	1.22000
Ν	2.06884	0.00000	0.00000
Η	2.45533	0.46945	0.81310
Η	2.45533	0.46945	-0.81310
Η	2.45533	-0.93887	0.00000

15	$SiF_2H_2\cdots$	$\rm NH_3,  C_s$	
F	-0.12376	1.22628	0.00000
Si	0.25601	-0.33778	0.00000
F	1.87344	-0.28456	0.00000

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Η	-0.05846	-1.00506	1.26959
Н	-0.05846	-1.00506	-1.26959
Ν	-2.29934	-0.38588	0.00000
Н	-2.54738	0.59923	0.00000
Н	-2.73886	-0.80171	-0.81548
Н	-2.73886	-0.80171	0.81548

16a	$SiF_3H\cdots NH_3, C_s$			
Η	1.75218	-0.00856	0.00000	
Si	0.29667	0.17654	0.00000	
F	0.24053	1.79282	0.00000	
F	-0.52830	-0.08799	-1.35509	
F	-0.52830	-0.08799	1.35509	
Ν	0.32143	-2.02817	0.00000	
Η	-0.63431	-2.37362	0.00000	
Η	0.78216	-2.40878	0.82139	
Н	0.78216	-2.40878	-0.82139	

16b	$\mathrm{HSiF}_3\cdots\mathrm{NH}_3,\mathrm{C}_{3v}$		
Si	0.47664	0.00000	0.00000
Н	1.95077	0.00000	0.00000
F	0.23699	0.80167	-1.38853
F	0.23699	-1.60333	0.00000
F	0.23699	0.80167	1.38853
Ν	-1.62736	0.00000	0.00000
Н	-1.99087	-0.47405	0.82109

Η	-1.99087	-0.47405	-0.82109
Η	-1.99087	0.94811	0.00000

17	$\operatorname{SiF}_4\cdots\operatorname{NH}_3, \operatorname{C}_{3v}$		
Si	-0.14293	0.00000	0.00000
F	-1.75198	0.00000	0.00000
F	0.05853	-0.79697	-1.38040
F	0.05853	1.59395	0.00000
F	0.05853	-0.79697	1.38040
Ν	1.92903	0.00000	0.00000
Η	2.29379	0.47403	0.82104
Η	2.29379	0.47403	-0.82104
Н	2.29379	-0.94806	0.00000

18	$FGeH_3 \cdots NH_3, C_{3v}$		
Ge	-0.10836	0.00000	0.00000
F	-1.92384	0.00000	0.00000
Η	0.18762	-0.75421	-1.30632
Η	0.18762	1.50841	0.00000
Η	0.18762	-0.75421	1.30632
Ν	2.51587	0.00000	0.00000
Η	2.89814	0.47046	0.81486
Η	2.89814	0.47046	-0.81486
Η	2.89814	-0.94092	0.00000

 $\mathbf{19} \quad \mathrm{ClGeH}_3\mathrm{\cdots}\mathrm{NH}_3,\,\mathrm{C}_{3v}$ 

Ge	0.21998	0.00000	0.00000
$\operatorname{Cl}$	-1.99578	0.00000	0.00000
Η	0.56785	-0.74722	-1.29422
Η	0.56785	1.49443	0.00000
Η	0.56785	-0.74722	1.29422
Ν	2.97467	0.00000	0.00000
Η	3.35953	0.46997	0.81402
Η	3.35953	0.46997	-0.81402
Η	3.35953	-0.93995	0.00000

20	$BrGeH_3 \cdots NH_3, C_{3v}$		
Ge	0.79891	0.00000	0.00000
$\operatorname{Br}$	-1.57624	0.00000	0.00000
Η	1.14870	-0.74687	-1.29360
Η	1.14870	1.49372	0.00000
Η	1.14870	-0.74687	1.29360
Ν	3.56508	0.00000	0.00000
Η	3.94994	0.46998	0.81403
Η	3.94994	0.46998	-0.81403
Н	3.94994	-0.93997	0.00000

<b>21</b>	$(\mathrm{HO})\mathrm{GeH}_{3}\mathrm{\cdots}\mathrm{NH}_{3},\mathrm{C}_{s}$		
Η	-2.43370	0.76387	0.00000
Ο	-2.00116	-0.09434	0.00000
Ge	-0.18595	0.01398	0.00000
Н	0.20520	-1.47028	0.00000

Η	0.22916	0.77228	1.27946
Η	0.22916	0.77228	-1.27946
Ν	2.72363	-0.01997	0.00000
Η	3.13198	0.90970	0.00000
Η	3.09702	-0.49920	-0.81368
Н	3.09702	-0.49920	0.81368

22	$GeH_4 \cdots NH_3, C_{3v}$		
Ge	-0.60690	0.00000	0.00000
Η	-2.15735	0.00000	0.00000
Η	-0.12312	-0.73139	-1.26680
Η	-0.12312	1.46278	0.00000
Η	-0.12312	-0.73139	1.26680
Ν	2.71572	0.00000	0.00000
Η	3.10263	0.46934	0.81291
Η	3.10263	0.46934	-0.81291
Н	3.10263	-0.93867	0.00000

<b>23</b>	$CO_2 \cdots NH_3, C_s$		
Η	-2.46031	0.93617	0.00000
Ν	-2.10155	-0.01349	0.00000
$\mathbf{C}$	0.82078	0.00415	0.00000
Ο	0.83040	1.17131	0.00000
Ο	0.86401	-1.16202	0.00000
Н	-2.50221	-0.47280	-0.81190
Н	-2.50221	-0.47280	0.81190

<b>24</b>	$SCO \cdots NH_3, C_s$		
Η	-2.74176	0.25897	0.00000
Ν	-2.51252	-0.73021	0.00000
С	0.40035	0.61529	0.00000
Ο	-0.30293	1.54197	0.00000
$\mathbf{S}$	1.37539	-0.61935	0.00000
Η	-2.97013	-1.13168	-0.81234
Η	-2.97013	-1.13168	0.81234

<b>25</b>	$\mathrm{CF}_2\mathrm{O}{\cdots}\mathrm{NH}_3,\mathrm{C}_s$		
Ν	-2.13545	0.02830	0.00000
С	0.54866	0.15032	0.00000
0	0.63058	1.32554	0.00000
F	0.54839	-0.62376	-1.06358
F	0.54839	-0.62376	1.06358
Η	-2.68397	-0.82584	0.00000
Η	-2.43069	0.56117	-0.81244
Н	-2.43069	0.56117	0.81244

**26a**  $CF_2S\cdots NH_3$ ,  $C_s$ Ν -2.24144 1.00020 0.000000.28875С -0.411640.00000  $\mathbf{S}$ 1.337000.806780.00000 F -0.18946 -1.01830-1.05969F -0.18946 -1.01830 1.05969

Η	-3.16833	0.58588	0.00000
Н	-2.19908	1.60796	-0.81245
Η	-2.19908	1.60796	0.81245

26b	$CF_2S\cdots NH_3, C_s$		
Ν	-0.81775	1.37793	0.00000
С	-0.18921	-0.07947	0.00000
$\mathbf{S}$	1.51010	0.00257	0.00000
F	-0.83680	-0.61886	-1.08325
F	-0.83680	-0.61886	1.08325
Η	-1.83974	1.36073	0.00000
Н	-0.45152	1.84535	-0.82816
Н	-0.45152	1.84535	0.82816

27	$SiF_2O\cdots NH_3, C_s$		
Ν	-0.42376	1.66094	0.00000
Si	0.20327	-0.15088	0.00000
0	1.73129	-0.09429	0.00000
F	-0.68147	-0.62119	-1.25214
F	-0.68147	-0.62119	1.25214
Η	-1.43724	1.75497	0.00000
Η	-0.05066	2.13614	-0.81948
Н	-0.05066	2.13614	0.81948

<b>28</b>	$\operatorname{CH_3}^+ \cdots \operatorname{NH_3}, \operatorname{C}_{3v}$		
Ν	0.70513	0.00000	0.00000

С	-0.80599	0.00000	0.00000
Н	-1.14296	0.51675	-0.89504
Н	-1.14296	-1.03351	0.00000
Н	-1.14296	0.51675	0.89504
Н	1.07613	-0.47653	0.82539
Н	1.07613	-0.47653	-0.82539
Η	1.07613	0.95307	0.00000

29	$\text{FNH}_3^+ \cdots \text{NH}_3,  \mathcal{C}_{3v}$		
Ν	-0.39134	0.00000	0.00000
F	-1.76513	0.00000	0.00000
Η	-0.05710	-0.48443	-0.83905
Η	-0.05710	0.96885	0.00000
Η	-0.05710	-0.48443	0.83905
Ν	2.22808	0.00000	0.00000
Η	2.64176	0.46550	0.80625
Η	2.64176	0.46550	-0.80625
Н	2.64176	-0.93099	0.00000

30	$FCH_3 \cdots C$	$l^-, C_{3v}$	
$\mathbf{C}$	-1.23588	0.00000	0.00000
F	-2.65510	0.00000	0.00000
Η	-0.88447	-0.51407	-0.89040
Η	-0.88447	1.02814	0.00000
Η	-0.88447	-0.51407	0.89040
Cl	1.94309	0.00000	0.00000

31	FSiH <sub>3</sub> …C	$\mathrm{Cl}^-, \mathrm{C}_{3v}$	
Si	-0.65514	0.00000	0.00000
F	-2.35776	0.00000	0.00000
Η	-0.50131	-0.73709	-1.27667
Η	-0.50131	1.47418	0.00000
Η	-0.50131	-0.73709	1.27667
Cl	1.84845	0.00000	0.00000

<b>32</b>	$\mathrm{FGeH}_3\cdots\mathrm{Cl}^-,\ \mathrm{C}_{3v}$		
Ge	-0.41308	0.00000	0.00000
F	-2.30517	0.00000	0.00000
Η	-0.31390	-0.76825	-1.33064
Η	-0.31390	1.53649	0.00000
Η	-0.31390	-0.76825	1.33064
Cl	2.15274	0.00000	0.00000

33	$CO_2 \cdots Cl^-$	$$ , $C_s$	
$\operatorname{Cl}$	0.00000	0.00000	1.66499
С	0.00000	0.00000	-1.25540
Ο	0.00000	1.16587	-1.34912
0	0.00000	-1.16587	-1.34912

<b>34</b>	$SCO···Cl^-, C_s$			
Cl	2.05468	0.23409	0.00000	
С	-0.96955	-0.62147	0.00000	

0	-0.65121	-1.73711	0.00000
$\mathbf{S}$	-1.55759	0.84626	0.00000

- **35**  $CF_2S\cdots Cl^-, C_s$
- Cl-1.492190.665900.00000C0.15369-0.278570.00000S1.570570.704910.00000F0.00320-1.11800-1.07951
- $F = 0.00320 \quad -1.11800 \quad 1.07951$