

Article

Design, Synthesis of *N*-phenethyl Cinnamide Derivatives and Their Biological Activities for the Treatment of Alzheimer's Disease: Antioxidant, Beta-amyloid Disaggregating and Rescue Effects on Memory Loss

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Abstract: Gx-50 is a bioactive compound for the treatment of Alzheimer's disease (AD) found in Sichuan pepper (*Zanthoxylum bungeanum*). In order to find a stronger anti-AD lead compound, 20 gx-50 (1–20) analogs have been designed and synthesized, and their molecular structures were determined based on nuclear magnetic resonance (NMR) and mass spectrometry (MS) analysis, as well as comparison with literature data. Compounds 1–20 were evaluated for their anti-AD potential by using DPPH radical scavenging assay for considering their anti-oxidant activity, thioflavin T (ThT) fluorescence assay for considering the inhibitory or disaggregate potency of A β , and transgenic *Drosophila* model assay for evaluating their rescue effect on memory loss. Finally, compound 13 was determined as a promising anti-AD candidate.

Keywords: Alzheimer's disease; gx-50; DPPH assay; ThT assay; transgenic *Drosophila* model assay

1. Introduction

Alzheimer's disease (AD) is a neurodegenerative disorder that mainly occurs in the elderly. It is characterized by intelligence decline and memory loss, as well as changes in emotions and personality [1]. There are approximately 36 million AD patients worldwide, and this figure will double every 20 years [2]. Since the 1990s, a great deal of financial support has been invested to explore the molecular pathogenesis of AD, which had provided robust support to develop effective pharmacological treatments [3]. Currently, the clinically used AD drugs mainly include cholinesterase inhibitors, including tacrine, donepezil, rivastigmine, galanthamine, and huperzine A, and the N-methyl-D-aspartate (NMDA) receptor antagonist, that is memantine. These drugs could alleviate cognitive symptoms in this disease. Unfortunately, there is no effective means to cure, or only slow, the progression of this disease [4,5].

The pathogenesis of AD has not yet been fully understood. Therefore, a variety of AD hypotheses have been proposed, such as the amyloid cascade hypothesis, the Tau phosphorylation hypothesis, the neurovascular hypothesis, the oxidative stress hypothesis, and the immune hypothesis [4]. One of

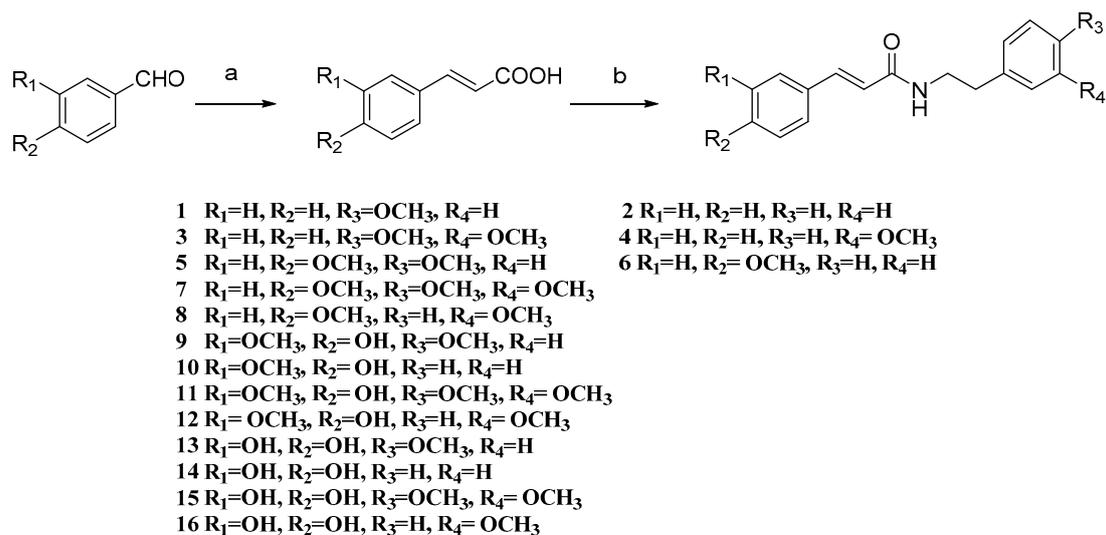
the major pathological hallmarks of AD is the aggregation of amyloid plaques in the brain. Its essence are the fibrils of the amyloid- β peptide ($A\beta$) [6,7]. Oxidative stress also plays an important role in the development and progress of AD, and 1,1-diphenyl-2-picrylhydrazyl (DPPH) assay is one of the most common methods to evaluate the anti-oxidative activity of compounds [8,9]. Moreover, transgenic *Drosophila* that can express AD-related proteins has been successfully used to screen the potential anti-AD compounds [10].

N-[2-(3,4-Dimethoxyphenyl)ethyl]-3-phenyl-acrylamide (gx-50), a potential drug candidate for the treatment of AD, was isolated from Sichuan pepper (*Zanthoxylum bungeanum*). Gx-50 could penetrate the blood brain barrier (BBB) and enter the brain tissue to improve the cognitive function of dementia mice, and meanwhile could reduce the $A\beta$ plaques in brain tissues [11–13]. In order to find a much stronger drug candidate for the treatment of AD, 20 analogues of gx-50 were designed and synthesized in this study. Consequently, the DPPH assay and thioflavin T (ThT) assay were employed to evaluate their anti-oxidative activity and their inhibition and disaggregation on $A\beta$ aggregation. Finally, Pavlov's olfactory memory test was used to evaluate their rescue effects on memory loss as analyzed on the behavior of AD model flies. Herein, we will report the design, synthesis, and anti-AD bioactivity of these gx-50 derivatives.

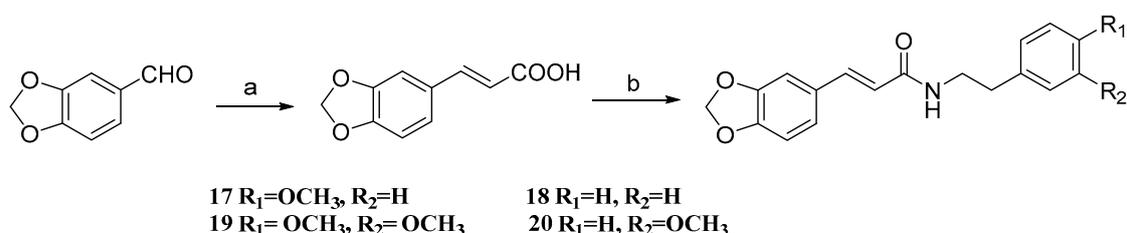
2. Results and Discussion

2.1. Chemistry

The synthesis of the gx-50 derivatives is described in Schemes 1 and 2. All compounds were synthesized by following previously reported methods with little modifications [14]. Specifically, the cinnamic acid was synthesized firstly by the substituted aromatic aldehyde with malonic acid through the Knoevenagel reaction. Then in the presence of 3-(ethyliminomethylideneamino)-*N,N*-dimethylpropan-1-amine hydrochloride (EDCI) and 4-dimethylaminopyridine (DMAP), the cinnamic acids with different substitutions reacted with 4-methoxyphenethylamine, 2-phenylethylamine, 3,4-dimethoxyphenethylamine, and 3-methoxyphenethylamine, respectively. Finally, one molecule of water was removed at room temperature to produce the corresponding target compounds 1–20. Among them, compounds 12, 16, and 20 have never been reported by searching the SciFinder database. Regarding the known compounds, their chemical structures were determined and confirmed by NMR and MS data analysis as well as literature comparison. Some of them were reported to show potent biological activities. For example, compounds 1, 9, and 13 exhibited induction of apoptosis in U-937 cells at 100 μ M [15]. Compound 3 and 11 had an antihyperglycemic effect as inhibition percentage of 20.1% and 30.7% in sucrose-loaded model (SLM) at a dosage of 100 mg/kg-body weight [16,17]. Compound 5 inhibited platelet aggregation with the IC_{50} value of 2.6 μ M [18]. Compound 10 and 11 showed anti-inflammatory activity with 50% NO inhibition concentration as 14.08 μ M and 15.08 μ M, respectively [19,20]. Compound 10 and 14 revealed 5-lipoxygenase inhibitory activity with the IC_{50} value of 0.12 μ M and 1 μ M, respectively [21]. Compound 15 could inhibit tyrosinase as the IC_{50} value of 0.6 mM [22,23]. In this study, all compounds 1–20 were evaluated for their anti-AD bioactivity via DPPH assay, ThT test, and Pavlov's olfactory memory test for the first time.



Scheme 1. Synthesis of compounds 1–16. Reagents and conditions: (a) anhydrous pyridine, piperidine, malonic acid, oil bath, 90 °C, 4 h; (b) cinnamic acid, amine, EDCl, DMAP, DMF, room temperature, overnight.



Scheme 2. Synthesis of compounds 17–20. Reagents and conditions: (a) anhydrous pyridine, piperidine, malonic acid, oil bath, 90 °C, 4 h; (b) cinnamic acid, amine, EDCl, DMAP, DMF, room temperature, overnight.

2.2. Pharmacology

2.2.1. Determination of Anti-Oxidant Activity Based on DPPH Assay

The oxidative damage to neurons is closely related to the pathogenesis of AD [24]. In this article, the anti-oxidative activity of compounds 1–20 was evaluated by DPPH assay with Vitamin C (Vc) as the positive control [25]. The free radical scavenging ability was evaluated based on the UV absorbance change of the solution measured at 517 nm using a microplate reader [26]. As shown in Table 1, the compound concentration and their anti-oxidative activity exhibited a favorable concentration-dependent relationship. In order to discuss their structure-activity relationship, compounds 1–20 were classified into three groups as follows. The first group contains compounds 1–8, which only contained methoxy substituents. These compounds showed weak DPPH scavenging activity. The second group contains compounds 9–16, which had the substituents of hydroxyls and methoxyls. These compounds exhibited comparable DPPH scavenging activity, compared to the positive control. Moreover, compounds 13–16 with two hydroxyls showed much stronger activity than compounds 9–12 with just one hydroxyl substituent. The third group contains compounds 17–20, which contained 3,4-(methylenedioxy)cinnamic acid moiety. These compounds showed the weakest DPPH scavenging activity. In addition, the DPPH scavenging rate of compound 3 (gx-50) was less than 10%, and its derivatives 9–16 exhibited stronger DPPH scavenging ability ranging from 16.69 ± 0.46 to 60.85 ± 0.37 . As discussed above, the phenolic hydroxyl may be a functional group for the anti-oxidant ability of gx-50 derivatives. This summarized structure-activity relationship is consistent with the previously reported conclusion [27].

Table 1. The DPPH assay result for compounds 1–20.

Compound	R ₁	R ₂	R ₃	R ₄	Inhibition Ratio (%)		
					50 μ M	100 μ M	200 μ M
1	H	H	OCH ₃	H	2.97 \pm 1.67	4.35 \pm 2.29	4.93 \pm 2.13
2	H	H	H	H	3.1 \pm 0.30	6.69 \pm 0.37	9.20 \pm 0.67
3	H	H	OCH ₃	OCH ₃	2.92 \pm 1.7	5.03 \pm 2.01	6.18 \pm 0.65
4	H	H	H	OCH ₃	0.99 \pm 0.55	2.91 \pm 1.24	3.13 \pm 0.59
5	H	OCH ₃	OCH ₃	H	4.78 \pm 2.8	9.78 \pm 0.61	11.22 \pm 0.43
6	H	OCH ₃	H	H	1.08 \pm 0.70	5.11 \pm 2.06	12.91 \pm 1.57
7	H	OCH ₃	OCH ₃	OCH ₃	3.61 \pm 0.89	5.58 \pm 0.78	9.58 \pm 1.39
8	H	OCH ₃	H	OCH ₃	0.06 \pm 2.63	2.81 \pm 2.78	3.42 \pm 0.78
9	OCH ₃	OH	OCH ₃	H	30.21 \pm 1.52	36.28 \pm 3.43	46.57 \pm 0.86
10	OCH ₃	OH	H	H	16.69 \pm 0.46	34.06 \pm 2.57	44.07 \pm 2.20
11	OCH ₃	OH	OCH ₃	OCH ₃	17.43 \pm 0.46	40.28 \pm 0.21	50.19 \pm 1.90
12	OCH ₃	OH	H	OCH ₃	28.74 \pm 0.78	46.93 \pm 0.42	51.83 \pm 1.42
13	OH	OH	OCH ₃	H	54.34 \pm 1.44	56.16 \pm 1.78	56.37 \pm 2.86
14	OH	OH	H	H	54.21 \pm 1.21	55.01 \pm 0.35	54.17 \pm 2.62
15	OH	OH	OCH ₃	OCH ₃	57.85 \pm 0.15	59.96 \pm 0.79	58.33 \pm 2.71
16	OH	OH	H	OCH ₃	58.86 \pm 0.43	60.85 \pm 0.37	57.68 \pm 0.54
17	OCH ₃	H	-	-	2.13 \pm 0.32	4.60 \pm 0.33	4.82 \pm 0.36
18	H	H	-	-	0.63 \pm 0.27	1.99 \pm 1.50	2.27 \pm 1.52
19	OCH ₃	OCH ₃	-	-	1.44 \pm 1.68	1.39 \pm 0.85	2.02 \pm 0.52
20	H	OCH ₃	-	-	1.57 \pm 0.13	4.09 \pm 0.31	4.53 \pm 0.83
Vc					41.42 \pm 2.83	72.00 \pm 1.15	78.88 \pm 0.80

Values are expressed as the means \pm SD of at least three independent experiments.

2.2.2. Inhibition of Cu²⁺-Induced A β Aggregation and Disaggregation

The inhibitory activities of compounds 1–20 on copper-mediated A β _{1–42} aggregation and disaggregation were evaluated by using ThT assay [28,29] with resveratrol and curcumin as positive controls. As shown in Table 2, the inhibitory and disaggregate potency of compounds 1–4 were less than 5%. Specifically, the inhibitory and disaggregate potency of compound 3 (gx-50) were 1.73 \pm 2.15% and 2.39 \pm 1.35%, respectively. However, when R₂ becomes methoxyl group, as in compounds 5–8, the inhibitory and disaggregate potency obviously increased except for compound 8. Therefore, R₂ may be an active position that could increase inhibitory or disaggregate potency of the test compounds. The inhibitory and disaggregate potency of compounds 9–12 were less than 10%. Compounds 13–16 exhibited equal or better inhibitory and disaggregate potency than curcumin and resveratrol. Compounds 17–20 had little inhibitory or disaggregate potency. The most active compound was compound 15, with 61.85 \pm 1.70% and 64.44 \pm 0.76%, respectively. As discussed, the catechol part could be the bioactive part for the gx-50 derivatives to inhibit A β aggregation and disaggregate A β aggregate.

Table 2. Thioflavin T (ThT) assay results for compounds 1–20.

Compound	R ₁	R ₂	R ₃	R ₄	A β Aggregation (inhib. %) ^a	Disaggregation (%) ^b
1	H	H	OCH ₃	H	4.83 \pm 8.10	4.40 \pm 2.11
2	H	H	H	H	0.14 \pm 1.49	0.27 \pm 0.94
3	H	H	OCH ₃	OCH ₃	1.73 \pm 2.15	2.39 \pm 1.35
4	H	H	H	OCH ₃	2.28 \pm 3.24	3.85 \pm 2.22
5	H	OCH ₃	OCH ₃	H	14.47 \pm 4.15	11.92 \pm 2.32
6	H	OCH ₃	H	H	12.26 \pm 7.37	11.19 \pm 1.70
7	H	OCH ₃	OCH ₃	OCH ₃	7.87 \pm 2.90	9.04 \pm 1.99
8	H	OCH ₃	H	OCH ₃	3.33 \pm 1.26	4.49 \pm 0.49
9	OCH ₃	OH	OCH ₃	H	2.72 \pm 3.65	1.05 \pm 1.04
10	OCH ₃	OH	H	H	5.94 \pm 3.13	4.16 \pm 1.65
11	OCH ₃	OH	OCH ₃	OCH ₃	4.37 \pm 3.20	2.04 \pm 0.71
12	OCH ₃	OH	H	OCH ₃	2.78 \pm 4.27	0.33 \pm 0.45

Table 2. Cont.

Compound	R ₁	R ₂	R ₃	R ₄	A β Aggregation (inhib. %) ^a	Disaggregation (%) ^b
13	OH	OH	OCH ₃	H	56.43 \pm 1.72	52.76 \pm 0.33
14	OH	OH	H	H	60.68 \pm 1.76	57.28 \pm 3.61
15	OH	OH	OCH ₃	OCH ₃	61.85 \pm 1.70	64.44 \pm 0.76
16	OH	OH	H	OCH ₃	38.47 \pm 2.37	51.51 \pm 1.42
17	OCH ₃	H	–	–	14.34 \pm 2.37	15.50 \pm 0.07
18	H	H	–	–	3.47 \pm 7.57	7.74 \pm 2.00
19	OCH ₃	OCH ₃	–	–	13.06 \pm 4.58	18.68 \pm 1.17
20	H	OCH ₃	–	–	12.93 \pm 3.66	0.61 \pm 0.38
curcumin					48.85 \pm 1.20	43.83 \pm 1.22
resveratrol					60.49 \pm 2.71	51.27 \pm 3.31

^a The inhibition percentage Cu²⁺-induced A β _{1–42} aggregation. ^b The disaggregation percentage of Cu²⁺-induced A β _{1–42} aggregates. Assays were carried out in the presence of 25 μ M inhibitor and 25 μ M A β _{1–42}. Values are expressed as the means \pm SD of at least three independent experiments.

2.2.3. Pavlovian Olfactory Aversive Immediate Memory Test

The Pavlovian olfactory memory experiment was used to observe the rescue effect of compounds on memory loss. As analyzed from the DPPH and ThT assay results, compounds **3** and **13–16** were selected to explore whether they could rescue the memory loss of the *Drosophila* AD model. Compounds **3** and **13–16** were administered at a concentration of 10 μ M. As shown in Figure 1, the PI value of the AD negative control group (P35*H29.3) was 27 (required less than 45). The PI difference between the genetic control group (P35*2U) and the AD negative group was 28 (required bigger than 15). Also, there was a significant difference between the memantine (MEM) 100 μ M group and the AD group ($p < 0.01$). It demonstrated that the model flies had an obvious defect of learning and memory. In addition, the performance index (PI) value of compound **13** was 41.33, and there was a significant difference ($p < 0.05$) with the AD group. However, the PI values of compounds **3** and **14–16** are 39.66, 37.33, 33.16, and 36.83, respectively. It proved that gx-50 (compound **3**) and compounds **14–16** have a weak rescue effect on memory loss for AD flies. However, the rescue effect on memory loss of compound **13** was comparable to that of the positive control at the same concentration. Moreover, compound **13** was also far more effective than gx-50 (compound **3**). Overall, compound **13** should be a promising drug candidate for the treatment of AD. Figure 2 showed the standardization results for Figure 1.

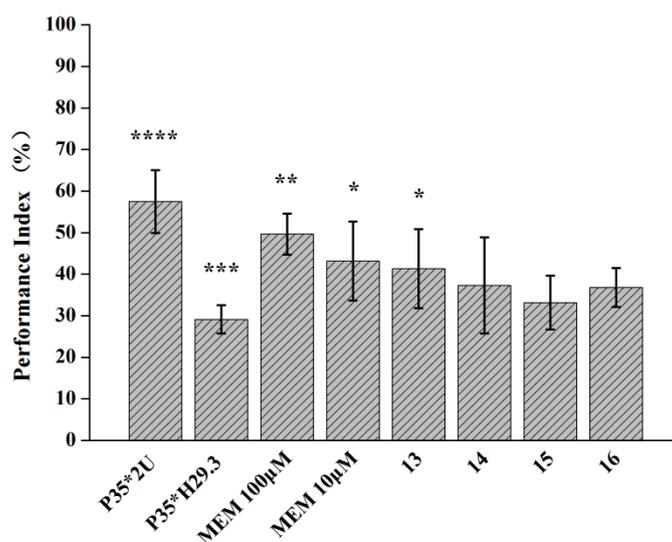


Figure 1. Differential effects of compound **3** and compounds **13–16** on rescuing memory deficit on a *Drosophila* Alzheimer's disease (AD) model at 10 μ M. The data are shown as means \pm SD and analyzed by one-way ANOVA; * $p < 0.05$; ** $p < 0.01$; *** $p < 0.001$; $n \geq 20$. MEM = memantine.

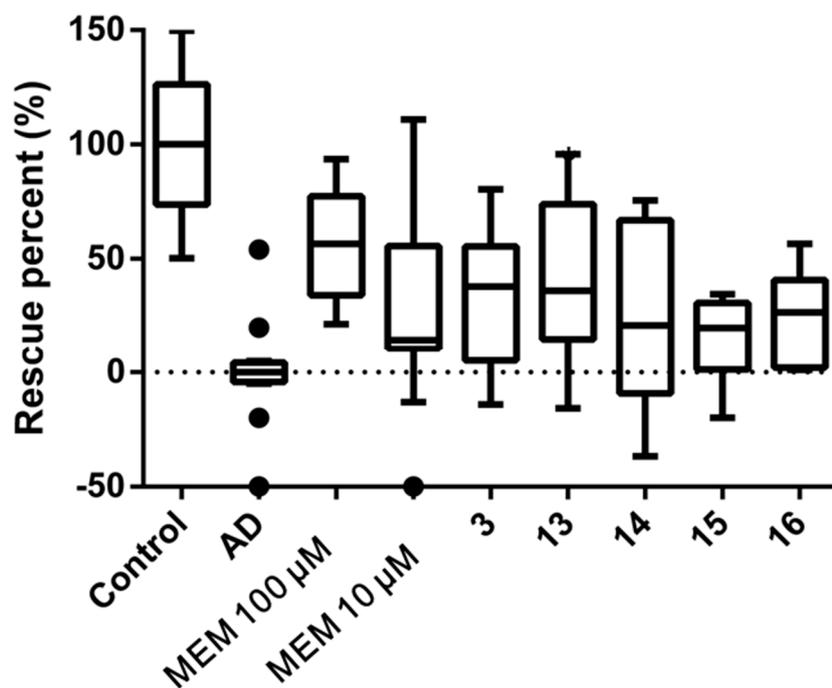


Figure 2. Differential effects of compound 3 and compounds 13–16 on rescuing memory deficit on a *Drosophila* AD model at 10 μ M. The data are shown as means \pm SD and analyzed by Tukey's test; $n \geq 20$.

In conclusion, twenty derivatives of gx-50 have been designed and synthesized. Among them, compound 13 showed the potent anti-oxidative activity, the inhibition and disaggregation on Cu^{2+} -induced $\text{A}\beta_{1-42}$ aggregation, and the rescue effect on memory loss of *Drosophila* AD model. Compared to gx-50 (compound 3), compound 13 would be a much more potent anti-AD candidate for the development of relative drugs.

3. Experimental Section

3.1. General Experimental Procedures

Benzaldehyde, *p*-anisaldehyde, vanillin, and protocatechualdehyde were purchased from Sinopharm Chemical Reagent Co., Ltd. (Shanghai, China). Malonic acid was obtained from Xilong Chemical Co., Ltd. (Shantou, China). The compounds 4-methoxyphenethylamine, 2-phenylethylamine, 3,4-dimethoxyphenethylamine, and 3-methoxyphenethylamine were purchased from Energy Chemical (Shanghai, China). The compound 4-Dimethylaminopyridine was purchased from Macklin (Shanghai, China). The compound 1-(3-dimethylaminopropyl)-3-ethylcarbodiimide hydrochloride was purchased from Accela ChemBio Co. Ltd. (Shanghai, China). Column chromatography was performed with silica gel (200–300 mesh). All other reagents and solvents were analytical grade. NMR spectra were recorded by a Bruker AVANCE III-400 spectrometer (Bruker AXS GmbH, Karlsruhe, Germany) with tetramethylsilane (TMS) as an internal standard. The program MestReNova 9.0 was used to handle the NMR spectra.

3.2. General Procedure for the Preparation of Compounds 1–20

3.2.1. Synthesis of Substituted Cinnamic Acids

The synthesis of substituted cinnamic acids was performed as Reference [30] described, with slight modifications. A mixture of substituted aromatic aldehydes (3 mmol), malonic acid (6 mmol), piperidine (0.3 mmol), was dissolved in pyridine and stirred at 90 $^{\circ}$ C for 4 h. Reaction progress was monitored by

thin layer chromatography (TLC) analysis. When the reaction terminated, the pyridine was removed under vacuum. Then the reaction mixture was poured into ice water and washed by 2N HCl. Finally the precipitate was filtered and washed with hexane three times. After being dried under vacuum, the products were obtained.

3.2.2. Synthesis of Gx-50 Derivatives 1–20

The synthesis of gx-50 derivatives 1–20 was performed as Reference [31] described. EDCI (0.325 g, 1.5 mmol) and DMAP (0.183 g, 1.5 mmol) were added to a solution of a substituted cinnamic acid (1 mmol) in 10 mL of DMF. The reaction solution was stirred in an ice-water bath. After 15 min, phenolic amine (1 mmol) was added and the solution was stirred at room temperature overnight. When the reaction was completed as indicated by TLC analysis, the solution was washed with water and brine. Then the organic phase was dried with anhydrous MgSO_4 . The solvent was then evaporated under vacuum. The crude product was purified by silica gel column chromatography to get the target compounds 1–20 with high yields (95.8%, 96.8%, 97.5%, 96.1%, 96.6%, 94.2%, 95.8%, 97.0%, 95.7%, 95.6%, 98.3%, 96.4%, 98.8%, 97.2%, 98.3%, 96.0%, 98.4%, 97.9%, 96.8%, 98.3%, respectively).

3.2.3. Purity Check of Compounds 1–20 by HPLC/MS Analysis

The purities of gx-50 derivatives 1–20 were analyzed by an HPLC/MS system with UV detection at 205 and 254 nm. Based on the integrity of the peak area, the purity of all compounds were determined over 97%. The HPLC analysis was performed on an Agilent 1200 Infinity system with an INNO C18 column (5 μm , 120 \AA , 4.6 \times 150 mm, Agilent Technologies, Santa Clara, CA, USA). The temperature of column chamber was kept at 25 $^\circ\text{C}$, and the mobile phase at 1.0 mL/min was as follows: ACN in H_2O containing 0.1% HCO_2H (0–30 min: from 0% to 100%).

Compound 12: Colorless amorphous powder; $^1\text{H-NMR}$ (400 MHz, CDCl_3) δ 7.53 (d, $J = 15.2$ Hz, 1H), 7.23 (m, 1H), 7.02 (m, 1H), 6.95 (m, 1H), 6.89 (d, $J = 8.0$ Hz, 1H), 6.79 (m, 3H), 6.19 (d, $J = 15.6$ Hz, 1H), 5.74 (s, 1H), 3.87 (s, 3H), 3.78 (s, 3H), 3.65 (m, 2H), 2.86 (t, $J = 6.8$ Hz, 2H). $^{13}\text{C-NMR}$ (101 MHz, CDCl_3) δ 161.50, 155.10, 142.71, 142.02, 136.35, 135.80, 124.93, 122.55, 117.38, 116.38, 113.38, 110.01, 109.76, 107.14, 104.91, 51.17, 50.46, 35.91, 30.98. ESI/MS calculated for $\text{C}_{19}\text{H}_{21}\text{NNaO}_4$: 350.1363, found: 350.1365 $[\text{M} + \text{H}]^+$.

Compound 16: yellow amorphous powder; $^1\text{H-NMR}$ (400 MHz, $\text{DMSO-}d_6$) δ 9.34 (s, 1H), 9.10 (s, 1H), 8.04 (t, $J = 5.2$ Hz, 1H), 7.21 (m, 2H), 6.93 (s, 1H), 6.77 (m, 5H), 6.32 (d, $J = 15.6$ Hz, 1H), 3.73 (s, 3H), 3.38 (m, 2H), 2.74 (t, $J = 7.2$ Hz, 2H). $^{13}\text{C-NMR}$ (101 MHz, $\text{DMSO-}d_6$) δ 170.55, 164.48, 152.47, 150.73, 146.33, 144.21, 134.53, 131.59, 126.09, 125.58, 123.71, 120.94, 119.44, 118.99, 116.79, 60.10, 40.45, 35.89. ESI/MS calculated for $\text{C}_{18}\text{H}_{19}\text{NNaO}_4$: 336.1206, found: 336.1201 $[\text{M} + \text{Na}]^+$.

Compound 20: Colorless amorphous powder; $^1\text{H-NMR}$ (400 MHz, CDCl_3) δ 7.53 (d, $J = 15.0$ Hz, 1H), 7.23 (m, 1H), 6.97 (m, 2H), 6.80 (m, 4H), 6.14 (d, $J = 15.6$ Hz, 1H), 5.98 (s, 2H), 5.55 (s, 1H), 3.80 (s, 3H), 3.65 (q, $J = 6.6$ Hz, 2H), 2.86 (t, $J = 6.8$ Hz, 2H). $^{13}\text{C-NMR}$ (101 MHz, CDCl_3) δ 165.96, 159.87, 149.05, 148.22, 140.81, 140.52, 129.70, 129.21, 123.83, 121.11, 118.59, 114.49, 111.93, 108.52, 106.35, 101.43, 55.21, 40.65, 35.73. ESI/MS calculated for $\text{C}_{19}\text{H}_{19}\text{NNaO}_4$: 348.1206, found: 348.1204 $[\text{M} + \text{Na}]^+$.

3.3. Biological Assay

3.3.1. DPPH Assay

DPPH was dissolved in ethanol (0.1 mM) daily and stored in the dark. Before usage, its absorbance was calibrated to 0.7 using ethanol by a microplate reader. Ethanol was used as a negative control and Vitamin C (Vc) was used as a positive control. A series of sample solutions with different concentrations in DMSO were prepared, and then 2 μL of each sample solution was removed and added into 96-well plates respectively. Then 198 μL of DPPH solution was added, and each group had 3 parallel holes. The blank group and the positive control group had the same treatment as the sample group.

The 96-well plate was shaken in the SPH-2000 shaking incubator (Xi'an Heb Biotechnology Co., Ltd., Shanghai, China) at 37 °C for 30 min in the dark. Consequently, the absorbance was measured at 517 nm with a RT-6100 microplate reader (Rayto, Shenzhen, China). The inhibition percent of each test compound was calculated based on the equation: DPPH inhibition percent (%) = $[(A_C - A_S)/A_C] \times 100$, where A_C represents the absorbance of the control sample and A_S is the absorbance of the test compound.

3.3.2. ThT Assay

The dried A β_{1-42} peptide powder (purity $\geq 95\%$, Aladdin, Shanghai, China) was dissolved in 1,1,1,3,3,3-hexafluoro-2-propanol (HFIP) at a concentration of 1.0 mg·mL⁻¹. The HFIP/peptide solution was shaken at room temperature for 6 h to gain the monomeric form of A β_{1-42} . Then, the HFIP/peptide solution was dried under a gentle stream of nitrogen gas for 2 h. The dried peptide powder was then dissolved with anhydrous DMSO and stored at -20 °C. For the test of copper-mediated A β_{1-42} inhibition and disaggregation, the procedures were carried out according to the literature [32]. In brief, the above A β_{1-42} stock solution was diluted in 20 μ M HEPES (pH 6.6) with 150 μ M NaCl to the desired final concentration before use. Either 10 μ L of 25 μ M peptide solution and 10 μ L of test compound solution were incubated at 37 °C for 24 h together or the addition of peptide solution and compound solution was accomplished by two steps in an interval of 24 h, it was incubated at 37 °C for 24 h continuously. Then 180 μ L of 50 mM glycine-NaOH buffer (pH 8.0) containing thioflavin T (5 μ M) was added to dilute the above sample solution. After 5 min, the fluorescence intensities were measured (excitation, 450 nm; emission, 485 nm) by a Horiba FluoroMax4 spectrophotometer (HORIBA Scientific, Edison, NJ, USA). The percent inhibition of aggregation was calculated via the expression $(1 - IF_i/IF_c) \times 100\%$, where IF_i and IF_c were the fluorescence intensities for A β_{1-42} in the presence and absence of test compounds after subtracting the background, respectively.

3.4. Pavlov's Olfactory Memory Test

3.4.1. Fly Stocks and Culture

All flies used in this study have been backcrossed for at least five generations, which could make them maintain the genetic background of w¹¹¹⁸(isoCJ1). Therefore, the strain of w¹¹¹⁸ (isoCJ1) ("2U") was an isogenic line used as a control in the experiments. In order to eliminate the potential adverse effects of long-term expression of A β_{42} protein on *Drosophila* strains, only when the flies ("H29.3") that carried the human A β_{42} gene crossed with flies of elav-GAL4^{c155} ("P35") that could start the expression of A β_{42} gene, its offspring could express A β_{42} protein in the nervous system [33–35]. However, offspring flies obtained by crossing P35 and 2U could not express A β_{42} protein. The learning test was conducted using the above flies. All flies were maintained on a 12 h light per day (light from 07:00 to 19:00) at 24 °C and 40–60% relative humidity. Control male flies (elav/Y; +/+) and experimental male flies (elav/Y; UAS-A β_{42} /+) were selected by microscope on the second day after eclosion. The selected flies were reared at 24 °C and 40 \pm 15% relative humidity with standard corn meal food for subsequent drug treatment.

3.4.2. Drug Intervention

All compounds and memantine (purity $\geq 98\%$, Aladdin, Shanghai, China) were dissolved in a vehicle solution, which was composed of 4% sucrose and 1% DMSO. Subsequently, the final concentration of each compound was diluted to 10 μ M with 4% sucrose solution. From the second day to eighth day after eclosion, each group of flies were continuously administered 4 hours per day and then rested in a fresh food environment for 20 h. For the test compound group, flies were fed with 50 μ L of compound solution daily, while control group of flies were fed with the same amount of sugar water. In this experiment, six data points were measured for each group flies, and each final data was the average of olfactory memory of two groups [36–38].

3.4.3. *Drosophila* Olfactory Escape Behavior Test

The task was performed according to a previous reported method with minor modifications [36–38]. Learning and memory tests were conducted at 25 °C and 70% relative humidity in the dark. During the training period, about one hundred flies were loaded into a copper-lined training tube. These flies were successively exposed to two odors, 3-octanol (OCT) and 4-methylcyclohexanol (MCH) for 60 s. Then the fresh air was ventilated for 45 s. Flies would receive an electric shock for 1 minute when they smelt the first odor. To avoid the natural preference of flies for odors, one tube of flies would smell the OCT firstly and receive an electric shock, while the other tube would smell the MCH firstly and receive an electric shock at each data point. The experimental average value of two tubes of the flies was considered as a data point. Then flies were immediately transferred to the choice point of a T-maze and forced to select between the OCT and MCH containers for two minutes. Consequently, the flies in different containers were counted and a performance index (PI) was calculated. If the flies had a 50:50 distribution in the T-maze test, it means the PI = 0 and the flies could not remember the connection of the electric shock and the odor. Additionally, the PI = 100 means that 100% of flies could remember the connection between the odor and the electric shock.

4. Statistical Analysis

All data were expressed as mean \pm standard deviation (SD), and results were subjected to Tukey's test or one-way analysis of variance (ANOVA). $p < 0.05$ was accepted to indicate the significance.

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