



Article Chemical Constituents from Euphorbia kansui

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Abstract: In this research, a new triterpenoid, tirucalla-8,24-diene- 3β ,11 β -diol-7-one (1), and eupha-8,24-diene- 3β ,11 β -diol-7-one (2), which was isolated from *Euphorbia kansui* for the first time, together with twelve other known compounds (3–14), were isolated from the ethyl acetate extract of *Euphorbia kansui*. Their structures were elucidated based on High resolution electrospray ionization mass spectrometry (HR-ESI-MS), Infrared Spectroscopy (IR), 1D and 2D Nuclear Magnetic Resonance (NMR) data. Both constituents 1 and 2 exhibited moderate cytotoxicity against colon cancer HCT-116, gastric cancer MKN-45 and breast cancer MCF-7.

Keywords: Euphane and Tirucallane; triterpenes; cytotoxicity; Euphorbia kansui

1. Introduction

The plants of *Euphorbia* contain more than 2000 species spread all over the world, and about 80 species distribute in China [1,2]. The dried root of *Euphorbia kansui* has long been used for the treatment of asthma, edema and ascites in traditional Chinese medicine. The structure type of the compounds in *Euphorbia kansui* are diterpenes, triterpenes, flavonoids, phenolic and acids [3–5]. Among them, diterpenes and triterpenoids are the main compounds in *Euphorbia kansui*, which show a wide range of pharmacological activities, such as antiviral, skin irritating and modulation of multidrug resistance effects [6–9]. A new tirucallane-type triterpene named tirucalla-8,24-diene-3 β ,11 β -diol-7-one (1) was first isolated from natural plants, and an euphane-type triterpene named eupha-8,24-diene-3 β ,11 β -diol-7-one (2) (Figure 1) was isolated from *Euphorbia kansui* for the first time in our present study. The two compounds were identified by 1D and 2D NMR including Heteronuclear Single Quantum Coherence (HSQC), Heteronuclear Multiple-Bond Correlation (HMBC), COrrelation SpectroscopY (COSY), Nuclear Overhauser Effect Spectroscopy (NOESY) and HR-ESI-MS data.

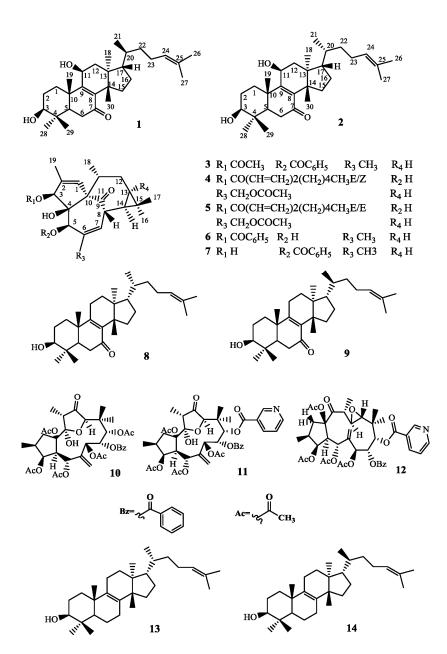


Figure 1. Chemical structures of Compounds 1-14.

2. Results and Discussion

Compound **1** was obtained as a white powder. The molecular formula $C_{30}H_{48}O_3$ was established by HR-ESI-MS (m/z 457.3770 [M + H]⁺, calcd. 457.3682) (Figure 1-1), IR (KBr) ν_{max} 3371, 2977, 2861, 1635, 1456, 1376, 1036, 622 cm⁻¹ (Figure 1-2), UV (MeOH) λ_{max} 253 nm (Figure 1-3) (Figures 1-1, 1-2, 1-3, see the Supplementary Materials); The comparison of ¹H-NMR, ¹³C-NMR (Tables 1 and 2, Figures 1-4 and 1-5, see the Supplementary Materials) and NOESY data (Figure 2) showed that compound **1** and the known compound **2** [10] were semblable in structure, except that a Hydrogen at C-11 of compound **2** was superseded by a hydroxyl group in compound **1**, which was verified by correlations of H-11 (δ_H 4.70, t, J = 8.2 Hz) with C-8 (δ_C 140.46), C-9 (δ_C 161.25) and C-12 (δ_C 42.79) in HMBC spectrum (Figure 3). Compared with ¹H-NMR and ¹³C-NMR of **1** and **2**, we can clearly see the differences between compound **2**: H-C(21) δ_H 0.88, H-C(22) δ_H 1.08–1.13 (m), 1.56–1.62 (m), C-20 δ_C 35.61, C-22 δ_C 35.51) and compound **1**: H-C(21) δ_H 0.94, H-C(22) δ_H 1.03–1.12 (m), C-20 δ_C 36.03, C-22 δ_C 36.24. In the HSQC plot (Figure 1-6, see the Supplementary Materials), δ_H 1.03–1.12 and 1.40–1.47 showed correlations with C(22), C(20) at δ_C 36.24, 36.03 respectively, which indicated compound 1, H-C(22) δ_H 1.03–1.12 (m). The relative configuration of 1 was determined by the ¹H-NMR (Table 1) and NOESY data. ¹H-NMR chemical shift of CH₃-21 (δ_H 0.94, d, *J* = 6.6 Hz) confirmed that compound 1 was classified as the tirucallane series [11–13]. The large coupling constants H-C(3) (δ_H 3.31, *J* = 9.5, 6.4 Hz) obviously indicated that the 3-OH group was in equatorial β-position [10,14]. The NOESY correlations (Figure 2) H-C(3) (δ_H 3.31)/H-C(5) (δ_H 1.65–1.68), H-C(3) (δ_H 3.31)/CH₃-28 (δ_H 1.00), H-C(11) (δ_H 4.70)/CH₃-18 (δ_H 0.71), CH₃-30 (δ_H 1.15)/H-C(17) (δ_H 1.54–1.61), CH₃-29 (δ_H 0.92)/CH₃-19 (δ_H 1.28), CH₃-18 (δ_H 0.71)/CH₃-19, CH₃-30 and CH₃-29 were all in β-orientation. Furthermore, compound 1 showed NOESY correlations between CH₃-18 and H-C(20) (δ_H 1.40–1.47) and CH₃-21, between CH₃-21 and H-12β (δ_H 2.38–2.46). These correlations were consistent with those of tirucallane-type triterpenes [14]. As a result, the structure of compound 1 was identified as tirucalla-8,24-diene-3 β ,11 β -diol-7-one.

Position	Compound 1	Compound 2	
1α	1.56–1.62 (m) 1.52–1.60 (m)		
1β	2.41–2.47 (m) 2.40–2.46 (m)		
2	1.70–1.79 (m)	1.70–1.79 (m)	
3α	3.31 (dd, <i>J</i> = 6.4, 9.6)	3.31 (dd, <i>J</i> = 6.4, 9.6)	
5	1.65–1.68 (m)	1.62–1.68 (m)	
6	2.40–2.47 (m)	2.40–2.48 (m)	
11	4.70 (t, J = 8.2)	4.69 (t, <i>J</i> = 8.2)	
12α	1.76–1.81 (m)	1.74–1.81 (m)	
12β	2.38–2.46 (m)	2.35–2.43 (m)	
15α	2.10–2.12 (m)	2.12–2.19 (m)	
15β	1.42–1.46 (m)	1.40–1.46 (m)	
16α	1.93–1.99 (m)	1.90–1.94 (m)	
16β	1.28–1.33 (m)	1.28–1.32 (m)	
17	1.54–1.61 (m) 1.55–1.60 (m)		
18	0.71 (s)	0.73 (s)	
19	1.28 (s)	1.27 (s)	
20	1.40–1.47 (m)	1.40–1.46 (m)	
21	0.94 (d, <i>J</i> = 6.6)	0.88 (d, J = 6.4)	
22	1.03–1.12 (m) 1.08–1.13 (m) 1.56–1.62 (m		
23	1.83–1.91 (m) 2.01–2.08 (m)	1.84–1.90 (m) 1.98–2.02 (m)	
24	5.09(t, J = 7.2) $5.09(t, J = 7.2)$		
26	1.70 (s) 1.70 (s)		
27	1.62 (s) 1.63 (s)		
28	1.00 (s) 1.01 (s)		
29	0.92 (s)	0.93 (s)	
30	30 1.15 (s) 1.16 (s)		

Table 1. ¹H-NMR data for compounds **1** and **2**.

Record in CDCl ₃ , 400 MHz	for ¹ H, δ in ppm, $J =$ Hz.
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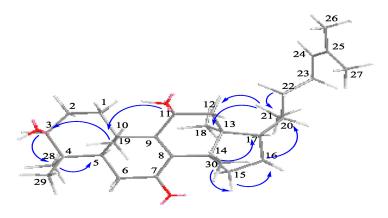
 Table 2. ¹³C-NMR data of compounds 1 and 2.

Position	Compound 1	Compound 2
1	33.68	33.67
2	27.37	27.36
3	78.27	78.26
4	39.07	39.06
5	49.25	49.25
6	35.84	35.84
7	200.10	200.16
8	140.46	140.44
9	161.25	161.30
10	39.58	39.58

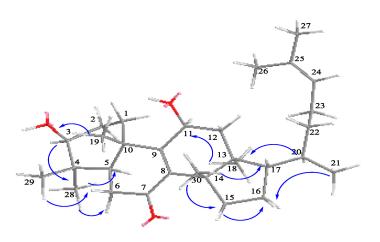
$\begin{array}{cccccccccccccccccccccccccccccccccccc$	Position	Compound 1	Compound 2
1346.1946.1914 47.95 48.02 15 31.88 31.81 16 27.77 27.80 17 49.13 48.69 18 16.06 16.24 19 19.70 19.67 20 36.03 35.61 21 18.68 18.76 22 36.24 35.51 23 24.86 24.82 24 124.98 124.90 25 131.09 131.14 26 25.68 25.73 27 17.62 17.70 28 27.59 27.59 29 15.18 15.19	11	68.11	68.11
$\begin{array}{cccccccccccccccccccccccccccccccccccc$	12	42.79	42.83
$\begin{array}{cccccccccccccccccccccccccccccccccccc$	13	46.19	46.19
$\begin{array}{cccccccccccccccccccccccccccccccccccc$	14	47.95	48.02
$\begin{array}{cccccccccccccccccccccccccccccccccccc$	15	31.88	31.81
$\begin{array}{c ccccccccccccccccccccccccccccccccccc$	16	27.77	27.80
$\begin{array}{cccccccccccccccccccccccccccccccccccc$	17	49.13	48.69
$\begin{array}{cccccccccccccccccccccccccccccccccccc$	18	16.06	16.24
$\begin{array}{cccccccccccccccccccccccccccccccccccc$	19	19.70	19.67
$\begin{array}{cccccccccccccccccccccccccccccccccccc$	20	36.03	35.61
2324.8624.8224124.98124.9025131.09131.142625.6825.732717.6217.702827.5927.592915.1815.19	21	18.68	18.76
24124.98124.9025131.09131.142625.6825.732717.6217.702827.5927.592915.1815.19	22	36.24	35.51
25 131.09 131.14 26 25.68 25.73 27 17.62 17.70 28 27.59 27.59 29 15.18 15.19	23	24.86	24.82
2625.6825.732717.6217.702827.5927.592915.1815.19	24	124.98	124.90
2717.6217.702827.5927.592915.1815.19	25	131.09	131.14
2827.5927.592915.1815.19	26	25.68	25.73
29 15.18 15.19	27	17.62	17.70
	28	27.59	27.59
30 25.62 25.70	29	15.18	15.19
	30	25.62	25.70

Table 2. Cont.

Record in CDCl₃, 100 MHz for ${}^{13}C$, δ in ppm, J = Hz.



Compound 1



Compound ${\bf 2}$

Figure 2. Key NOESY correlations for 1 and 2.

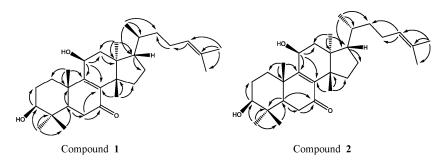


Figure 3. Key HMBC correlations for 1 and 2.

The comparison of ¹H-NMR, ¹³C-NMR (Tables 1 and 2) and NOESY data (Figure 2) showed that compound **2** and the known Compound **1** [10] were semblable in structure, except that a Hydrogen at C-11 of Compound **1** was superseded by a hydroxyl group in compound **2**, which was proven by correlations of H-11 (δ_{H} 4.69, t, J = 8.2 Hz) with C-8 (δ_{C} 140.44), C-9 (δ_{C} 161.30) and C-12 (δ_{C} 42.83) in HMBC spectrum (Figure 3). The large coupling constants H-C(3) (δ_{H} 3.31, J = 9.5, 6.4 Hz) obviously indicated that the 3-OH group was in equatorial β -position [11,12]. The NOESY correlations (Figure 2) H-C(3) (δ_{H} 3.31)/H-C(5) (δ_{H} 1.62–1.68), H-C(3) (δ_{H} 3.31)/CH₃-28 (δ_{H} 1.01), H-C(11) (δ_{H} 4.69)/CH₃-18 (δ_{H} 0.73), CH₃-30 (δ_{H} 1.16)/H-C(17) (δ_{H} 1.55–1.60), CH₃-29 (δ_{H} 0.93)/CH₃-19 (δ_{H} 1.27), CH₃-18 (δ_{H} 0.73)/CH₃-19 (δ_{H} 1.27), showed that H-C(3), H-C(5), H-C(17), and CH₃-28 were all in α -orientation, whereas 11-OH, CH₃-19, CH₃-30 and CH₃-29 were in β -orientation, and no correlations between CH₃-21/CH₃-18 [10,15] indicated that it belonged to the euphane-type triterpenes. The ¹H-NMR chemical shift of CH₃-21 (δ_{H} 0.88, d, J = 6.4 Hz) further confirmed that compound **2** was elucidated to be eupha-8,24-diene-3 β ,11 β -diol-7-one.

All 12 of the known compounds (3–14) were identified according to the spectroscopic data (¹H-NMR, ¹³C-NMR, see the Supplementary Materials) together with the comparsion of those reported, kansuiphorin C (3) [16], 3-O-(2'E,4'Z-decadienoyl)-20-O-acetylingenol (4) [4], 3-O-(2'E,4'E-decadienoyl)-20-O-acetylingenol (5) [4], 3-O-benzoyl-20-deoxyingenol (6) [4], 5-O-benzoyl-20-deoxyingenol (7) [4], kansenone (8) [10], *epi*-kansenone (9) [10], kansuinin A (10) [4,17], kansuinin D (11) [4], kansuinin E (12) [4], euphol (13) [4,18], and tirucallol (14) [4,19].

Compounds 1 and 2 were assessed for their inhibitory effects on HCT-116, MKN-45 and MCF-7 cell lines (Table 3), as well as L-O2 and GES-1 cell lines (Table 4). The results show that compounds 1 and 2 inhibit normal cells (L-O2 and GES-1) less than cancer cells (HCT-116, MKN-45 and MCF-7), and compounds 1 and 2 presented definite anticancer activities with IC₅₀ values of 20.84 ± 1.28 and $33.97 \pm 2.15 \mu$ M against HCT-116 cells, 10.18 ± 1.36 and $14.95 \pm 1.82 \mu$ M against MKN-45 cells, and 10.82 ± 1.18 and $20.11 \pm 2.16 \mu$ M against MCF-7 cells, respectively. kansenone induces apoptosis through both the death receptor and mitochondrial pathways [5], and compounds 1 and 2 were similar with kansenone in structure, excluding that a Hydrogen at C-11 of kansenone was superseded by a hydroxyl group in compounds 1 and 2, thus compounds 1 and 2 may induce apoptosis in the same way as kansenone. Further research will be conducted on the anticancer mechanism of compounds 1 and 2.

Table 3. Cytotoxicity of compounds 1 and 2 against three human cancer cell lines.

Compound	IC ₅₀ (μM)			
r	HCT-116	MKN-45	MCF-7	
1	20.84 ± 1.28	10.18 ± 1.36	10.82 ± 1.18	
2	33.97 ± 2.15	14.95 ± 1.82	20.11 ± 2.16	
Cisplatin	8.465 ± 0.84	6.142 ± 1.12	9.035 ± 0.92	
5-Fu	6.172 ± 2.03	2.624 ± 2.06	1.629 ± 1.42	

Compound	IC ₅₀ (μM)		
r	L-O2	GES-1	
1	56.98 ± 1.74	40.99 ± 0.85	
2	49.89 ± 2.12	40.27 ± 1.28	

Table 4. Cytotoxicity of compounds 1 and 2 against two human normal cell lines.

3. Materials and Methods

3.1. General Experimental Procedures

HPLC: Hanbon NP 7000 (Jiangsu Hanbang Technology Companies, Huaian, China) Serials pump with an NU 3000 Serials UV-Vis detector (Jiangsu Hanbang Technology Companies, Huaian, China), Phecda Si (20×250 mm, 5 µm); Waters 1525 with a 2996 Diode Array Detector (DAD) (Waters, Milford, CT, USA), XBrige-Prep C₁₈ (150×19 mm, 5 µm), (Ultimate XB-C8, 30×150 mm, 5 µm). IR spectra were gained on a Nicole Is5 of Thermo Fisher spectrophotometer (Nicolet Instrument Corporation, Madison, WI, USA). The NMR spectra were measured on Avance 400 spectrometers (Bruker, Karlsruhe, Germany), with TMS as an internal standard. The UV spectra were measured on a Shimadzu UV-2401 UV-Vis spectrophotometer (Shimadzu, Kyoto, Japan). The HR-ESI-MS data were obtained by using a LTQ Orbitrap MS (Thermo Fisher Scientific, San Jose, CA, USA). Column chromatography (CC): silica gel (200–300 mesh, Qingdao Marine Chemical Industry, Qingdao, China).

3.2. Plant Materials

The dried root of *Euphorbia kansui* was collected from Red River valley of Baoji, Shaanxi Province of China, in October 2015 and was identified by Prof. Yu Ping Tang (college of pharmacy, Nanjing University of Chinese Medicine, Nanjing, China). The voucher specimen (20151020) has been deposited in the Herbarium of college of pharmacy, Nanjing University of Chinese Medicine (Nanjing, China).

3.3. Extraction and Isolation

The dried roots of *Euphorbia kansui* (12.2 kg) were extracted twice (each time for 2 h) with 95% EtOH under reflux to give the 95% EtOH extract (871.9 g) by evaporation of the solvent under reduced pressure, and then the 95% EtOH extract was extracted with ethyl acetate (EtOAc) to obtain ethyl acetate extract (530.8 g). Finally, the fraction of EtOAc was subjected to silica gel column chromatography $(14 \times 59 \text{ cm})$ with a gradient elution (Pet and ethyl acetate, 100:1–1:1) to get fractions A-T. Fr. G (2 g) was eluted with Pet:EtOAc (100:20). Compound 1 (40.8 mg) was isolated by HPLC (Pet:EtOAc, 100:38), and further purified by reversed-phase HPLC (MeCN:H₂O, 70:30) flow rate 16 mL/min (t_R 15.342 min). Compound 2 (70.3 mg) was isolated by HPLC (Pet:EtOAc, 100:30), and further purified by reversed-phase HPLC (MeCN:H₂O, 70:30) flow rate 16 mL/min (t_R 16.060 min). Fr. A (18 g) was eluted with Pet:EtOAc (100:3). Compound 13 (6.082 g) was isolated by HPLC (MeCN:H₂O, 95:5) with (Ultimate XB-C8, 30×150 mm, 5 µm) flow rate 16 mL/min (t_R 35.452 min). Compound 14 (782.5 mg) was isolated by HPLC (MeCN:H₂O, 95:5) with (Ultimate XB-C8, 30×150 mm, 5 µm) flow rate 16 mL/min (t_R 33.645 min). Fr. B (3 g) was eluted with Pet:EtOAc (100:5). Compound 6 (320 mg) was isolated by HPLC (Pet:EtOAc, 100:10) flow rate 16 mL/min (t_R 24.003 min). Compound 7 (46 mg) was isolated by HPLC (Pet:EtOAc, 100:10) flow rate 16 mL/min (t_R 22.209 min). Fr. C (5 g) was eluted with Pet:EtOAc (100:6). Compound 3 (920 mg) was isolated by HPLC (Pet:EtOAc, 100:11) flow rate 16 mL/min (t_R 24.128 min). Fr. D (2 g) was eluted with Pet:EtOAc (100:8). Compound 8 (161 mg) was isolated by HPLC (Pet:EtOAc, 100:15) flow rate 16 mL/min (t_R 23.218 min). Compound 9 (28 mg) was isolated by HPLC (Pet:EtOAc, 100:15) flow rate 16 mL/min (t_R 20.674 min). Fr. E (2 g) was eluted with Pet:EtOAc (100:13). Compound 4 (80 mg) was isolated by HPLC (Pet:EtOAc, 100:20) flow rate 16 mL/min (t_R 21.097 min). Compound 5 (40 mg) was isolated by HPLC (Pet:EtOAc, 100:20) flow

rate 16 mL/min (t_R 24.298 min). Fr. H (10 g) was eluted with Pet:EtOAc (100:55). Compound **10** (186 mg) was isolated by HPLC (Pet:EtOAc, 100:40), and further purified by reversed-phase HPLC (MeCN:H₂O, 70:30) flow rate 16 mL/min (t_R 10.502 min). Compound **11** (60 mg) was isolated by HPLC (Pet:EtOAc, 100:40), and further purified by reversed-phase HPLC (MeCN:H₂O, 70:30) flow rate 16 mL/min (t_R 9.702 min). Compound **12** (80 mg) was isolated by HPLC (Pet:EtOAc, 100:40), and further purified by reversed-phase HPLC (MeCN:H₂O, 70:30) flow rate 16 mL/min (t_R 11.302 min).

3.4. Cytotoxicity Test

Cytotoxicity of two compounds against HCT-116, MKN-45 and MCF-7 cancer cell lines (American Type Culture Collection, ATCC, Manassas, VA, USA), normal liver cell L-O2 (Zhongqiaoxinzhou Biotech, Shanghai, China) and gastric epithelial cell GES-1 (Nanjingkebai Biotech, Nanjing, China) were evaluated with the MTT (3-(4,5-dimethylthiazol-2-yl)-2,5-diphenylt-etrazolium bromide) method as described in the literature [20,21]. All cells were cultured in Dulbecco's modified Eagle's medium (DMEM) supplemented with 10% fetal bovine serum, three cancer cells were seeded in 96-well plate at a concentration of 8×10^4 cells/mL, two normal cells L-O2 and GES-1 were seeded in 96-well plate at a concentration of 1×10^4 cells/mL and 1×10^5 cells/mL respectively [21], and incubated for 24 h in humidifyed atmosphere with 5% CO₂ at 37 °C. One hundred microliters of cells were treated with compound **1** and **2** at different doses of 1.25, 2.5, 5, 10, 20 and 40 µg/mL in dimethyl sulfoxide (DMSO) in triplicate for 48 h at 37 °C with 5% CO₂. Then, each of them were added with 20 µL of MTT (5.0 mg/mL) and incubated for further 4 h, the growth medium was removed from all the wells. Finally, 150 µL DMSO were added to every sample. Cisplatin (Qilu pharmaceutical, Jinan, China) and 5-fluorouracil (5-Fu) (Sichuan Kangyi, pharmaceutical, Chengdu, China) served as positive control. Absorbance was determined by a microplate spectrophotometer at 490 nm.

4. Conclusions

The article reported two compounds, **1** and **2**, which are triterpenes, as well as twelve other known compounds (**3–14**). Compound **1** is a new tirucallane-type triterpene named tirucalla-8,24-diene-3 β ,11 β -diol-7-one. Compound **2** was isolated from *Euphorbia kansui* for the first timeand named eupha-8,24-diene-3 β ,11 β -diol-7-one. They also display effective anticancer activities against HCT-116, MKN-45 and MCF-7 cells. As we all know, *Euphorbia kansui* has pharmacological activities including tumor inhibition and antiviral effects [22,23]; this study further confirmed kansui may be a potential candidate for anticancer including colon cancer, gastric cancer and breast cancer and inferred that compounds **1** and **2** may be the main material basis of anticancer for *Euphorbia kansui*.

Some features about compounds **1**, **2**, **8**, **9**, **13** and **14** may be drawn based on their chemical structures. Above all, the CH₃-21 of compounds **9** and **14** were all in β -orientation and their ¹H-NMR chemical shift of CH₃-21 at ($\delta_{\rm H}$ 0.94, d, *J* = 6.6 Hz), whereas the CH₃-21 of compounds **8** and **13** were all in α -orientation and their ¹H-NMR chemical shift of CH₃-21 at ($\delta_{\rm H}$ 0.88, d, *J* = 6.4 Hz) [4,18,19]. The C-20 and C-22 of compound **9** had a chemical shift greater than compound **8**. The C-20 and C-22 of compound **9** had a chemical shift greater than compound **8**. The C-20 and C-22 of compound **13** (Table 5), which were identical with the ¹H-NMR and ¹³C-NMR of compounds **1** and **2**, respectively. Then, the polarity order of compound **9** was larger than compound **8**. Compound **14** is also greater than **13** in the same way, which were also the same as the polarity order of compounds **1** and **2**. Thus, the different positions of CH₃-21 of compounds **1** and **2**, **8** and **9**, and **13** and **14** may have a good correlation with their polarity order.

Position	Compound 8	Compound 9	Compound 13	Compound 14
1	34.61	34.61	35.26	35.26
2	27.40	27.41	27.95	27.92
3	78.07	78.06	79.00	79.03
4	38.82	38.83	38.94	38.94
5	48.19	48.24	50.97	50.97
6	35.77	35.78	18.95	18.95
7	198.37	198.35	27.68	27.67
8	138.94	138.93	134.03	134.08
9	165.46	165.48	133.55	133.51
10	39.27	39.27	37.24	37.27
11	23.73	23.67	21.53	21.45
12	29.95	29.87	30.90	30.80
13	44.62	44.61	44.12	44.11
14	47.68	47.61	50.03	50.12
15	31.39	31.45	29.77	28.83
16	28.67	28.65	28.14	28.05
17	48.24	48.76	49.64	49.96
18	15.73	15.54	15.63	15.52
19	18.60	18.61	20.15	20.14
20	35.65	36.16	35.88	36.33
21	18.90	18.75	18.92	18.69
22	35.52	36.35	35.43	36.40
23	24.74	24.91	24.77	24.94
24	125.07	125.12	125.22	125.26
25	131.04	131.01	130.08	130.90
26	25.76	25.73	17.69	17.62
27	17.71	17.65	25.74	25.71
28	27.28	27.29	27.92	27.92
29	15.07	15.07	15.53	15.43
30	24.42	24.31	24.47	24.36

Table 5. ¹³C-NMR data of compounds 8, 9, 13 and 14.

Record in CDCl₃, 100 MHz for ${}^{13}C$, δ in ppm, J = Hz.

Supplementary Materials: Supplementary materials are available online.

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