9-Vinylanthracene Based Fluorogens: Synthesis, Structure-Property Relationships and Applications

Supplementary Information

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Figure S1. (**A**) normalized absorption spectra of dye **1** in different solvents; (**B**) emission spectra of dye **1** in different solvents. Concentration = $10 \mu M$; λ_{ex} : 385 - 395 nm.



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Figure S4. Photographs of dyes **1-3** taken under UV radiation (366 nm) from a hand-held UV lamp. Left: dyes **1-3** in solid state; right: dyes **1-3** dissolved in acetone compared to their solid state.



Figure S5. Dynamic Light Scattering analysis of (A) dye **1** and (B) dye **2** particles via the Zetasizer Ver 7.01 program.



Figure S6. (**A**) emission spectra of dye 2 in water with increasing concentration of TFA. Concentration = 20 μ M; λ_{ex} : 392 nm; (**B**) emission intensity of dye **2** at 488 nm vs. proton concentration.



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Tuble 01 . Fluorescence decay parameters of dye 1 .						
Species ^[a]	$A_1^{[b]}$	$A_2^{[b]}$	$A_{3}^{[b]}$	τ1 [ns] ^[b]	τ2 [ns] ^[b]	τ3 [ns] ^[b]
Dye 1/DMSO	100			3.690		
Dye 1/H ₂ O	61.5	28.4	10.1	4.849	2.230	1.068

Table S1. Fluorescence decay parameters of dye 1

 $\label{eq:constraint} \begin{tabular}{l} [1] = 10 \mu M. \begin{tabular}{l} [b] Determined from Equation 1, where A and τ are the fractional amount and fluorescence lifetime of different species, respectively. \end{tabular}$

$$I = A_1 e^{-\frac{t}{\tau_1}} + A_2 e^{-\frac{t}{\tau_2}} + A_3 e^{-\frac{t}{\tau_3}} \quad \text{eq.1}$$