Supplementary Materials: Investigating Glycol-Split-Heparin-Derived Inhibitors of Heparanase: Study of Synthetic Trisaccharides

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Figure S1. ¹H-NMR spectra used to monitor the periodate oxidation of **1**. The circled signal that disappears throughout the reaction corresponds to H-2' of GlcA (3.44 ppm).



Figure S2. Backbone computed dihedral angles: α/β , γ , δ/ϵ , ω/w for compound **2** (top) and the comparable α/β , ω/w for compound **1** (bottom) sampled during selected MD simulations steps at temperature 300 K (step 1, black dots), 400 K (step 6, red dots) and 300 K (step 11, blue dots). The dihedral angles pairs: α/β , γ , δ/ϵ , ω/w are reported in Ramachandran diagrams, while γ is reported as a function of simulation time. Two possible conformations for compound **2**, characterized by different glycosidic state ω/w (see Table S2) are found (A, and B). Yellow stars in Ramachandran plots indicate torsional states defined by averaging the corresponding dihedral angle pairs for a suitable amount of simulation time.



Figure S3. 2D HSQC spectra of compound **1** (green) and **2** (blue). (no quote, quote and double quote respectively refer to reducing-end unit, central unit and non reducing end unit).



Figure S4. Plot of the chemical shift of exchangeable protons in 2 vs. temperature (K).



Figure S5. Intra-residue experimental (empty symbols) and simulated (black line) 2D NOEs build up curves for the glucosamine units in **2**. The simulation was performed using the A conformer (see text).



Figure S6. Selected 2D NOEs. Upper part: Selected inter-glycosidic 2D NOEs build up curves for the two (see Figure S2, **A** and **B**) predicted conformations of **2** at mixing time between 0.2 to 1.5 s. Comparison of the middle and right panels shows the better fit obtained between experimental and modelled curves when the **A** conformer is used for the calculation. Lower part: A licorice representation of **2** in conformation **A** showing the previously discussed inter-residue NOEs. On the right a good fit is obtained between experimental and calculated NOEs for the obtained conformation of **1**.



Dipole moment vector							
Molecule	Мx	My	Mz	M (e Å)			
2 (A)	2.8	-2.1	-1.2	3.7			
2 (B)	-0.9	2.6	-2.9	4.0			
1	4.3	3.9	-4.7	7.4			

Figure S7. Distribution charge properties: dipole moment lenght and components, estimated for **1** and **2** in the previously determined conformations using the approach AM1-BCC (see text). On the the left, the non-reducing end residues of **1** and **2** have been superposed.

Table S1. Summary of the MD simulation thermal history for the model of compound 1 and 2.

MD Simulation Step	1	2	3	4	5	6	7	8	9	10	11
Temperature (K)	300	320	340	360	380	400	380	360	340	320	300
Time Length (ns)	5K										

Table S2. Backbone torsional angles for **1** and **2**. Values in **black** refer to the two initial conformations setted for each glycans (**1.1–1.2** and **2.1–2.2**) (black coloured lines). Values in red are the torsional angles obtained. Compound **2** significantly populates two states (A and B) at the torsional degree of freedom ω/w .

		Torsional Angles (°)					
		τ	α/β	γ	δ/ε	ω/w	
	Initial 1.1	-20	-40/-26	-	-	53/9	
1	Final 1.1	-17	-44/-24			49/26	
1	Initial 1.2	-60	-20/26	_	-	-30/-30	
	Final 1.2	-15	-45/-30	-	-	46/21	
2	Initial 2.1	-61	-52/-26	180	50/51	56/7	
	Final 2.1	-19	-42/-27	-59	44/34	49/33 (A) -37/-16 (B)	
	Initial 2.2	180	-30/-50	61	0/30	-40/10	
	Final 2.2	-20	-44/-20	-69	46/37	39/30 (A) -46/-18 (B)	

Table S3. Temperature coefficients $\Delta\delta$ (ppb·K⁻¹) estimated by linear regression. The estimated errors ($\Delta\Delta\delta$) is shown on the last decimal digit. The linear correlation coefficient (*R*) is reported

H-Bond	Δδ (ΔΔδ)	R
OH'-3	-9.8 (2)	-0.9996
OH'-2	-13.0 (3)	-0.9991
N H ''	-8.09 (3)	-0.99997
NH	-5.7 (5)	-0.9891