

Article

# Chemical Constituents from the Stems of Diospyros maritima

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Received: 1 November 2009; in revised form: 7 December 2009 / Accepted: 14 December 2009 / Published: 15 December 2009

**Abstract:** A new phenolic, bis(6-hydroxy-2,3,4-trimethoxylphen-1-yl)methane (1) and a new butanedioate, butylmethyl succinate (2), along with twenty-nine known compounds including one naphthoquinone derivative, two chromanones, eight benzenoids, one lignan, one tocopherol, and sixteen triterpenoids were isolated from the stems of *Diospyros maritima. epi*-Isoshinanolone (3) was isolated in pure form for the first time. In addition, 5,7-dihydroxy-2-methylchomanone (4) was isolated from a natural source for the first time. Their structures were established on the basis of spectroscopic data as well as direct comparison with authentic samples.

**Keywords:** *Diospyros maritima*; Ebenaceae; bis(6-hydroxy-2,3,4-trimethoxylphen-1-yl)methane; butylmethyl succinate

#### **1. Introduction**

Diospyros maritima Blume (Ebenaceae), a medium-sized shrub, is widely distributed throughout Southern Asia. The genus *Diospyros* is well-known to produce various naphthoquinone derivatives, some of which exhibit cytotoxic, ichthyotoxic, germination inhibitory, and antifungal activities [1–4]. The stems of *D. maritima* has been used in Taiwan as a folk medicine as a traditional treatment for rheumatic disease [5]. In previous studies, many naphthoquinones and triterpenes were isolated from its bark, root, fruits, leaves, and twigs by Tezuka [6] and Higa [1,2]. More recently, we also have reported some new naphthoquinones [7,8], triterpenes, steroids [3,9–13], phenolic and aliphatic components [14] from the stems of this plant and found that some naphthoquinones showed strong antitumor activity [3]. In continuation of our work on the discovery of secondary metabolites from this plant, we have also isolated a new phenolic, bis(6-hydroxy-2,3,4-trimethoxylphen-1-yl)methane (1), and a new butanedioate, butylmethyl succinate (2), along with twenty-nine known compounds including one naphthoquinone derivative, epi-isoshinanolone (3) [15], two chromanones, 5,7dihydroxy-2-methylchromanone (4) [16] and 5-hydroxy-2-methylchromanone (5) [17], eight benzenoids, 1-(4,6-dihydroxy-2-methylphenyl)ethanone (6) [18], ethyl 2,4-dihydroxy-6-methylbenzoate (7) [19], 4-hydroxybenzaldehyde (8) [20], vanillin (9) [21], 4-hydroxy-3,5-dimethoxybenzaldehyde (10) [20], acetovanillone (11) [22], trans-coniferylaldehyde (12) [23], and (E)-3-(4acetyloxy-3,5-dimethoxyphenyl)-2-propenal (13) [24], one lignan, 4-ketopinoresinol (14) [25], one tocopherol,  $\alpha$ -tocopherol (15) [20], and sixteen triterpenoids: squalene (16) [20], lupeol caffeate (17) [26], betulin-28-acetate (18) [27], (E)-betulin-3 $\beta$ -p-coumarate (19) [28], (Z)-betulin-3 $\beta$ -p-coumarate (20) [28], betulinaldehyde (21) [29], 3-oxo-20(29)-lupen-28-oic acid (22) [20], betulinic acid (23) [30], betulic acid acetate (24) [31], 3-O-betulinic acid p-coumarate (25) [32], 3-O-palmitoylerythrodiol (26) [33], 28-O-acetylerythrodiol (27) [34], 3β-acetoxyolean-12-en-28-oic acid (28) [35], 3β-acetoxyurs-12-en-28-oic acid (29) [36], 3β-hydroxyurs-12-en-28,13-olide (30) [37], and 3β-hydroxytaraxastan-28, 20β-olide (**31**) [38].

#### 2. Results and Discussion

The EtOH extracts of the stems of *D. maritima* was concentrated to give a black residue which was suspended in water and partitioned successively with *n*-hexane and *n*-BuOH. The combined *n*-BuOH soluble layer was subjected to repeated chromatography using silica gel and further purification by semipreparative HPLC to furnish two new compounds, bis(6-hydroxy-2,3,4-trimethoxylphen-1-yl)methane (1) and butylmethyl succinate (2), in addition to twenty-nine known compounds. The identification of the known compounds were performed by comparing their physical and spectral data (IR, UV, MS, and NMR) with literature values. This paper deals with the structural elucidation of compounds 1-4.

The HR-EI-MS of **1** exhibited a molecular ion peak at m/z 380.1476, which is corresponded to the molecular formula  $C_{19}H_{24}O_8$  and indicated eight degrees of unsaturation. The IR spectrum showed the presence of hydroxy (3,337 cm<sup>-1</sup>) and phenyl (1,619 and 1,500 cm<sup>-1</sup>) functionalities. The UV spectrum displayed the aromatic maximum absorption peak at 283 nm. In the <sup>1</sup>H-NMR and DEPT spectra of **1**, the signals for five quaternary carbons ( $\delta_C$  110.4, 135.0, 149.5, 151.6, 152.8) and one tertiary carbon ( $\delta_C$  97.7) attributed a pentasubstituted benzene ring. The five substituents of the benzene ring included

three methoxy groups ( $\delta_H$  3.75, 3.77, 4.07;  $\delta_C$  55.9, 61.1, 62.1), a hydroxyl ( $\delta_H$  8.10, disappeared on D<sub>2</sub>O exchange), and a methylene ( $\delta_H$  3.67;  $\delta_C$  18.3). From the above evidence, compound **1** was proposed to be a biphenyl derivative linked by a methylene group. The relative positions of those substituents on the benzene rings were determined by the long-range correlations between H-5 ( $\delta_H$  6.30) and C-1 ( $\delta_C$  110.4), C-3 ( $\delta_C$  135.0), and C-6 ( $\delta_C$  151.6); H-7 ( $\delta_H$  3.67) and C-1 ( $\delta_C$  110.4), C-2 ( $\delta_C$  149.5), and C-6 ( $\delta_C$  151.6) in the HMBC spectrum of **1**. Moreover, 2-OMe, 3-OMe, 4-OMe, and 6-OH also exhibited HMBC correlations with C-2, C-3, C-4, and C-6, respectively. The nOe correlations between H-7/6-OH ( $\delta_H$  8.10) and H-5/4-OMe ( $\delta_H$  3.77) further assured this proposed structure. Thus, compound **1** was elucidated as bis(6-hydroxy-2,3,4-trimethoxylphen-1-yl)methane.

The IR spectrum of **2** exhibited an absorption band (1735 cm<sup>-1</sup>) for an ester functionalty. The <sup>13</sup>C-NMR and DEPT spectra indicated the presence of one methyl ( $\delta_C$  13.7), one methoxy ( $\delta_C$  51.8), five methylenes [ $\delta_C$  19.0, 28.9 (×2), 30.6, 64.6], and two quaternary carbons ( $\delta_C$  172.3, 172.8). A butoxyl group was elucidated by the COSY correlations between H-1' ( $\delta_H$  4.07) and H-2' ( $\delta_H$  1.57); H-2' and H-3' ( $\delta_H$  1.34); H-3' and H-4' ( $\delta_H$  0.88). From the above observations, compound **2** was considered as a butanedioate derivative with both methyl and butyl groups. The chemically equivalent signals of two methylene groups ( $\delta_H$  2.60, s) were assigned to be located between C-1 ( $\delta_C$  172.3) and C-4 ( $\delta_C$  172.8) according to their HMBC relationship. The proposed structure of **2** was also supported by the molecular ion peak at m/z 188 and the fragmental ion peak at m/z 157 [M-OMe]<sup>+</sup>, 129 [M-COOMe]<sup>+</sup>, 115 [M-OCH<sub>2</sub> CH<sub>2</sub> CH<sub>2</sub> Me]<sup>+</sup>, 101 [M-CH<sub>2</sub>CH<sub>2</sub>COOMe]<sup>+</sup>, and 73 [M-COCH<sub>2</sub>CH<sub>2</sub>COOMe]<sup>+</sup> in the EI-MS spectrum. Compound **2** was accordingly identified as butylmethyl succinate.

The IR spectrum of **3** showed absorption bands at 3,360 (OH), 1,640 (conjugated ketone), and 1,580 (aromatic) cm<sup>-1</sup>. The <sup>1</sup>H-NMR and <sup>1</sup>H-<sup>1</sup>H COSY spectra displayed the signals for a set of aromatic ABX coupling system [ $\delta_{H} 6.89$  (1H, d, J = 7.4 Hz, H-7), 7.09 (1H, d, J = 7.4 Hz, H-5), 7.49 (1H, t, J = 7.4 Hz, H-6)] and a phenolic proton with strong intramolecular hydrogen bond [ $\delta_{H} 12.35$  (1H, s, 8-OH)]. In addition, the NMR signals for a complex coupling system attributing to a methine [ $\delta_{H} 2.30$  (1H, m, H-3)], which was coupled with methyl [ $\delta_{H} 1.17$  (3H, d, J = 6.6 Hz, 3-CH<sub>3</sub>)], oxymethine { $\delta_{H} 4.49$  (1H, dd, J = 6.6, 6.9 Hz, H-4) coupled with 4-OH [ $\delta_{H} 1.92$  (1H, d, J = 6.9 Hz)]}, and methylene [ $\delta_{H} 2.42$  (1H, dd, J = 10.1, 17.3 Hz, H<sub>a</sub>-2); 2.90 (1H, dd, J = 4.0, 17.3 Hz, H<sub>b</sub>-2)] was observed. The above features were almost identical to those of the known compound, *epi*-isoshinanolone [15]. Thus, compound **3** was determined as *epi*-isoshinanolone. It was isolated in pure form for the first time.

Analysis the IR spectrum of **4** suggested that it contains hydroxyl (3,400 cm<sup>-1</sup>), conjugated ketone (1,668 cm<sup>-1</sup>), and aromatic (1,620, 1,600 cm<sup>-1</sup>) functionalities. The <sup>1</sup>H-NMR spectrum of **4** revealed the signals for an oxymethine [ $\delta_{\rm H}$  4.66 (1H, m, H-2)] coupling to both methyl [ $\delta_{\rm H}$  1.48 (3H, d, J = 5.7 Hz, 2-CH<sub>3</sub>)] and methylene [ $\delta_{\rm H}$  2.84 (2H, m, H-3)], a phenolic proton with strong hydrogen-bonding [ $\delta_{\rm H}$  11.19 (1H, s, 5-OH)] and a free phenolic proton [ $\delta_{\rm H}$  5.90 (1H, s, 7-OH)], as well as two *meta*-positioned aromatic protons [ $\delta_{\rm H}$  6.18 (1H, s, H-8); 6.29 (1H, s, H-6)]. The above spectral information was similar to that of 5,7-dihydroxy-2-methylchromanone previously reported in the literature [17]. Thus, compound **4** was characterized as 5,7-dihydroxy-2-methylchromanone. Compound **4** was isolated from the natural product for the first time in the present investigation.

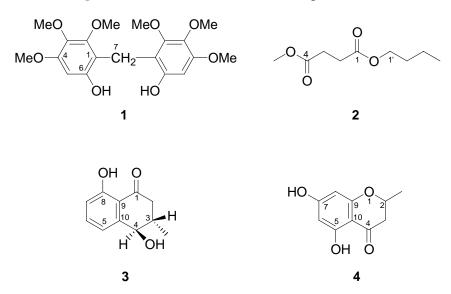


Figure 1. Chemical structures of compounds 1-4.

Table 1. <sup>1</sup>H- and <sup>13</sup>C-NMR spectral data for 1 (300 and 75 MHz in CDCl<sub>3</sub>).

position	$\delta_{\rm C}$	$\delta_{\mathrm{H}}$	position	$\delta_{\rm C}$	$\delta_{\mathrm{H}}$
1, 1'	110.4		7	18.3	3.67 s
2, 2'	149.5		2, 2' -OCH <sub>3</sub>	62.1	4.07 s
3, 3'	135.0		3, 3' -OCH <sub>3</sub>	61.1	3.75 s
4, 4'	152.8		4, 4' -OCH <sub>3</sub>	55.9	3.77 s
5, 5'	97.7	6.30 s	6, 6' <b>-</b> OH		8.10 s
6, 6'	151.6				

#### 3. Experimental

#### 3.1. General

Melting points were determined with a Yanagimoto (MP500D) micromelting point apparatus and are uncorrected. IR spectra were recorded on a Perkin-Elmer 781 spectrophotometer. <sup>1</sup>H- and <sup>13</sup>C-NMR spectra were obtained in CDCl<sub>3</sub> at a constant temperature controlled and adjusted to around 300 K on a Bruker AM-300 spectrometer, and the residual proton resonance (CHCl<sub>3</sub>) of CDCl<sub>3</sub> was used as internal shift reference. The 2D NMR spectra were recorded on a Bruker DMX-300 spectrometer by using standard pulse sequences. EI-MS, FAB-MS, HR-EI-MS, UV spectra, and specific rotations were recorded on a JEOL JMS-HX 300, a JEOL JMS-HX 110, a JEOL SX-102A, a Hitachi S-3210 spectrometer, and a JASCO DIP-1000 digital polarimeter, respectively. TLC was performed by using Si gel 60  $F_{254}$  plates (Merck). Column chromatography was performed on silica gel (Merck 9385, 70-230 mesh). HPLC was performed by using a Lichrosorb Si gel 60 (5µm) column (250 × 10 mm).

#### *3.2. Plant material*

The stems of *D. maritima* were collected in Lin-Ko, Taiwan, in 1993. The plant material was identified by Muh-Tsuen Gun, formerly a technician of the Department of Botany, National Taiwan University, and a voucher specimen has been deposited at the National Research Institute of Chinese Medicine, Taipei, Taiwan, R.O.C.

## 3.3. Extraction and isolation

The stems of D. maritima (16 kg) were extracted three times with EtOH (160 L) at 60 °C for 10 h each time. The EtOH extract was evaporated under reduced pressure, yielding a black residue, which was suspended in H<sub>2</sub>O (12 L), and then partitioned (×5) with 1 L of *n*-hexane. The aqueous layer was partitioned (×4) again with 1 L of n-BuOH. The combined n-BuOH extracts (180 g) was then subjected to column chromatography over silica gel  $(120 \times 6 \text{ cm})$  eluted with a setpwise gradient mixture of hexane and EtOAc as eluent. Seven fractions were collected as follows: 1 [3000 mL, hexane], 2 [4000 mL, hexane-EtOAc (9:1)], 3 [4000 mL, hexane-EtOAc (8:2)], 4 [4000 mL, hexane-EtOAc (7:3)], 5 [4000 mL, hexane-EtOAc (5:5)], 6 [3000 mL, hexane-EtOAc (3:7)], and 7 (6000 mL, EtOAc). Fraction 1 (8.5 g) was further purified through a silica gel column eluted with n-hexane/EtOAc (95/5) to yield 16 (6.2 mg). Fraction 3 (27.2 g) was further chromatographed on a silica gel column eluted with n-hexane/EtOAc (8/2) and semipreparative HPLC eluted with CH<sub>2</sub>Cl<sub>2</sub>/nhexane/EtOAc (8/3/1) to obtain 8 (5.3 mg), 9 (3.0 mg), 10 (3.6 mg), 11 (5.1 mg), and 15 (4.2 mg). Fraction 4 (38.5 g) was subjected to column chromatography over silica gel eluted with *n*-hexane/EtOAc (7/3) and semipreparative HPLC eluted with  $CH_2Cl_2/n$ -hexane/EtOAc (3/3/1) to yield 3 (3.8 mg), 5 (3.5 mg), 6 (6.1 mg), 7 (5.0 mg), 12 (12.4 mg), 17 (3.0 mg), 18 (6.3 mg), 19 (4.0 mg), 20 (10.1 mg), 21 (4.7 mg), 22 (5.1 mg), 27 (4.1 mg), and 30 (3.0 mg). Fraction 5 (32.1 g) was further chromatographed on silica gel eluted with n-hexane/EtOAc (5/5) and semipreparative HPLC eluted with CH<sub>2</sub>Cl<sub>2</sub>/n-hexane/EtOAc (2/3/1) to yield 1 (5.2 mg), 2 (10.5 mg), 4 (3.2 mg), 13 (3.1 mg), 14 (6.2 mg), 23 (3.1 mg), 24 (3.2 mg), 25 (3.0 mg), 26 (3.5 mg), 28 (4.7 mg), 29 (2.8 mg), and 31 (4.0 mg).

# 3.4. Spectroscopic data

*Bis*(6-hydroxy-2,3,4-trimethoxylphen-1-yl)methane (1): Gum; UV (logɛ) (MeOH)  $\lambda_{max}$ : 283 (3.4) nm; IR  $\nu_{max}$  cm<sup>-1</sup>: 3337, 1619, 1500, 1208, 1082, 897; <sup>1</sup>H- and <sup>13</sup>C-NMR (CDCl<sub>3</sub>): see Table 1; EI-MS (70 eV) *m/z* (rel. int. %): 380 [M]<sup>+</sup> (38), 211 (23), 197 (39), 184 (100), 169 (42), 57 (25); HR-EI-MS *m/z* 380.1476 (calcd for C<sub>19</sub>H<sub>24</sub>O<sub>8</sub>, 380.1471).

*Butylmethyl succinate* (**2**): Colorless oil; IR  $v_{max}$  cm<sup>-1</sup>: 1735, 1155, 1016; <sup>1</sup>H-NMR (CDCl<sub>3</sub>):  $\delta$  0.88 (3H, t, *J* = 7.6 Hz, H-4'), 1.34 (2H, sex, *J* = 7.6 Hz, H-3'), 1.57 (2H, quin, *J* = 7.6 Hz, H-2'), 2.60 (4H, s, H-2, 3), 3.67 (3H, s, -OCH<sub>3</sub>), 4.07 (2H, t, *J* = 7.6 Hz, H-1'); <sup>13</sup>C-NMR (CDCl<sub>3</sub>):  $\delta$  13.7 (C-4'), 19.0 (C-3'), 28.9 (C-2, 3), 30.6 (C-2'), 51.8 (-OCH<sub>3</sub>), 64.6 (C-1'), 172.3 (C-1), 172.8 (C-4); EI-MS (70 eV) *m/z* (rel. int. %): 188 [M]<sup>+</sup>(1), 157 (8), 129 (38), 115 (78), 101 (100), 87 (38), 73 (9), 55 (58); HR-EI-MS *m/z* 188.0940 (calcd for C<sub>9</sub>H<sub>16</sub>O<sub>4</sub>).

*epi-Isoshinanolone* (**3**): Amorphous solid; IR (dry film)  $v_{max}$  cm<sup>-1</sup>: 3360, 1640, 1580; <sup>1</sup>H-NMR (CDCl<sub>3</sub>):  $\delta 1.17$  (3H, d, J = 6.6 Hz, 3-CH<sub>3</sub>), 1.92 (1H, d, J = 6.9 Hz, 4-OH), 2.30 (1H, m, H-3), 2.42 (1H, dd, J = 10.1, 17.3 Hz, H<sub>a</sub>-2), 2.90 (1H, dd, J = 4.0, 17.3 Hz, H<sub>b</sub>-2), 4.49 (1H, dd, J = 6.6, 6.9 Hz, H-4), 6.89 (1H, d, J = 7.4 Hz, H-7), 7.09 (1H, d, J = 7.4 Hz, H-5), 7.49 (1H, t, J = 7.4 Hz, H-6), 12.35 (1H, s, 8-OH); EI-MS (70 eV) m/z [M]<sup>+</sup> 192 (90), 177 (20), 150 (45), 121 (100).

5,7-*Dihydroxy*-2-*methylchromanone* (**4**): Amorphous solid; IR (dry film)  $v_{max}$  cm<sup>-1</sup>: 3400, 1668, 1620, 1600; <sup>1</sup>H-NMR (CDCl<sub>3</sub>):  $\delta$  1.48 (3H, d, *J* = 5.7 Hz, 2-CH<sub>3</sub>), 2.84 (2H, m, H-3), 4.66 (1H, m, H-2), 5.90 (1H, s, 7-OH), 6.18 (1H, s, H-8), 6.29 (1H, s, H-6), 11.19 (1H, s, 5-OH); EI-MS (70 eV) *m/z* 194 [M]<sup>+</sup> (40), 185 (100), 149 (69).

# 4. Conclusions

Thirty-one compounds were isolated from the stems of *D. maritima*. Among them, bis(6-hydroxy-2,3,4-trimethoxylphen-1-yl)methane (1) and butylmethyl succinate (2) are new compounds and *epi*-isoshinanolone (3) was isolated in pure form for the first time. In addition, 5,7-dihydroxy-2-methylchromanone (4) was isolated from the natural source for the first time. This investigation of secondary metabolites may contribute to better understanding on the chemical characteristics of *D. maritima*.

## Acknowledgements

This research was supported by the National Science Council, Republic of China and China Medical University (CMU98-CT-01). We thank Shou-Ling Huang and Shu-Yun Sun for the NMR data acquisition and HR-EI-MS measurement in the Instrumentation Center of the College of Science, National Taiwan University. We are grateful to the National Center for High-performance Computing for computer time and facilities.

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Sample Availability: Samples of the compounds are available from the authors.

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