

## Pyrimidine Acyclo-C-Nucleosides by Ring Transformations of 2-Formyl-L-arabinal

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**Abstract:** The protected 2-formyl-L-arabinal **2** reacted with thiourea and cyanamide in the presence of sodium hydride to afford via ring transformations the 5-[1*R*,2*S*-1,2-bis(benzyloxy)-3-hydroxypropyl]-1,2-dihydropyrimidines **3** and **4**, respectively. Similarly, treatment of **2** with 3-amino-2*H*-1,2,4-triazole yielded 6-[1*R*,2*S*-1,2-bis(benzyloxy)-3-hydroxypropyl][1,2,4]-triazolo[1,5-*a*]pyrimidine (**5**).

**Keywords:** Glycals, nucleoside analogues, pyrimidines, fused pyrimidines, push-pull alkenes, ring transformation

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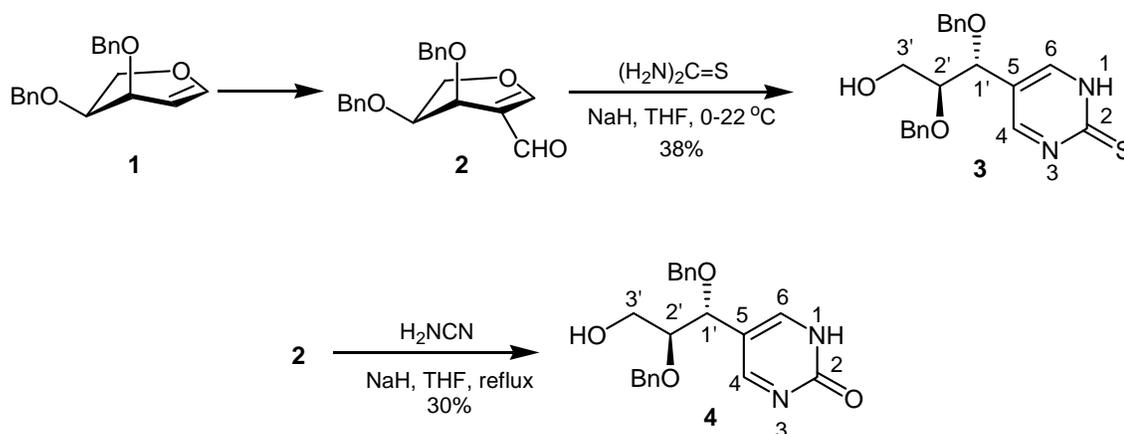
### Introduction

Because of their *push-pull* activated carbon-carbon double bond [1–3], 2-formylglycals [4–6] synthesized by a Vilsmeier-Haack reaction of *O*-benzyl protected hexose glycals are a versatile class of compounds which allowed a nucleophilic attack of dinucleophiles at C-1 under ring opening of the glycals followed by recyclization involving the formyl group to give acyclo-*C*-nucleoside analogues [6–12]. The protected 2-formyl-L-arabinal **2** was easily prepared from *O*-benzylated L-arabinal (**1**) by treatment with phosphoryl chloride and *N,N*-dimethylformamide [12]. In this paper we report the ring transformation reactions of 2-formyl-L-arabinal **2** with some nitrogen nucleophiles to obtain pyrimidine *C*-nucleoside analogue and related compounds.

## Results and Discussion

Thiourea was reacted with 2-formyl-L-arabinal **2** in the presence of sodium hydride in tetrahydrofuran to afford through the sequential combination of addition-elimination and ring closure reaction the required pyrimidine C-nucleoside analogue **3** in 38% yield as a pale yellow syrup (Scheme 1).

Scheme 1.



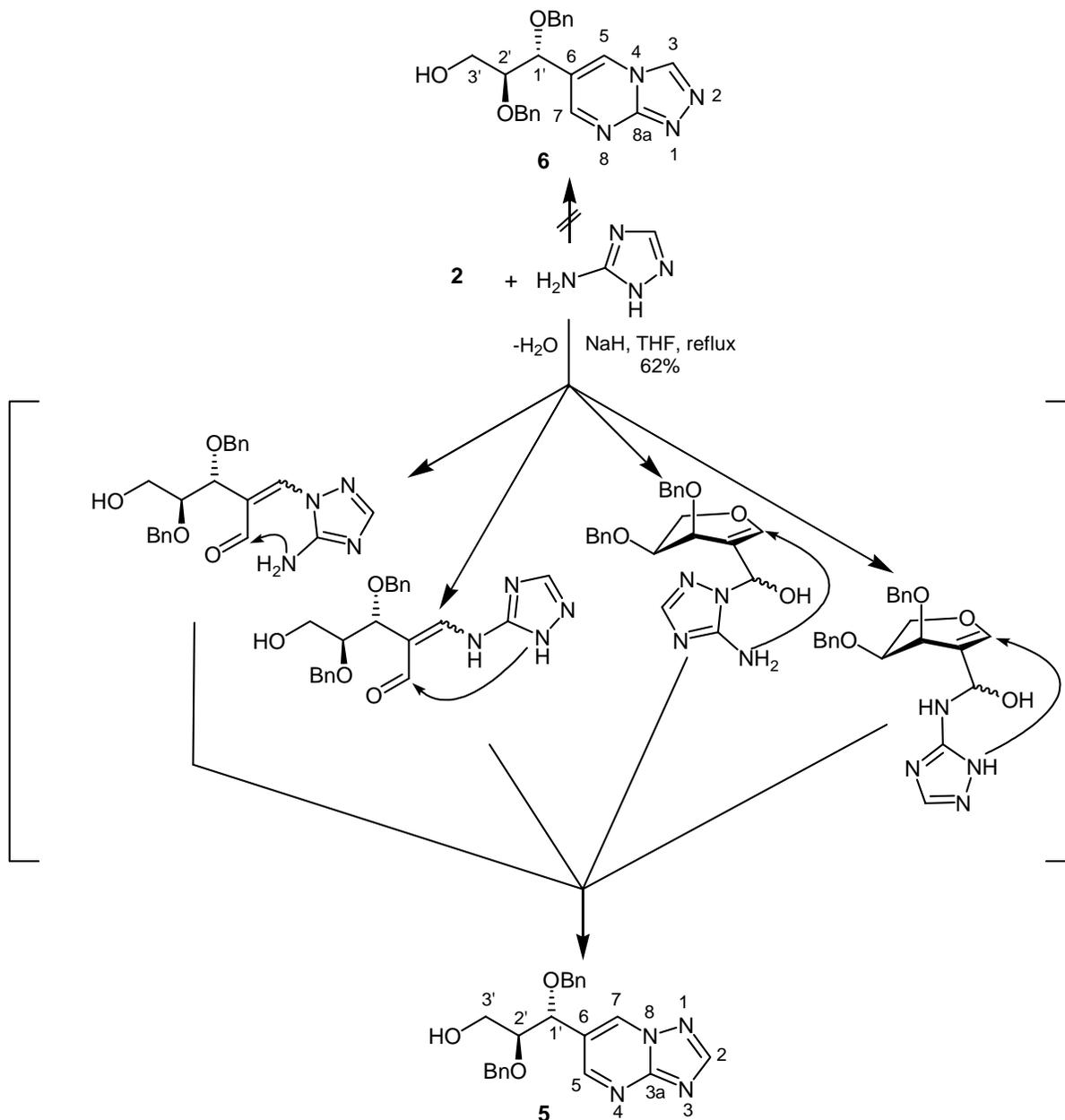
Compound **3** was characterized by  $^1\text{H}$ - and  $^{13}\text{C}$ -NMR as well as IR and mass spectroscopy. In the  $^1\text{H}$ -NMR spectrum, H-4 and H-6 exhibit only one signal at  $\delta = 8.50$ , which is due to the tautomerization involving the proton on the N atoms. In the  $^{13}\text{C}$ -NMR spectrum C-4 and C-6 appeared also as one signal at  $\delta = 157.0$ . For C-2 a chemical shift of  $\delta = 170.7$  was found. The IR spectrum showed a typical absorption for the associated OH group at  $3418\text{ cm}^{-1}$ . Furthermore, the mass spectrum confirmed the presence of the thiourea structural element by giving a  $[\text{M}+\text{H}]^+$  peak at 383.

In order to prepare the corresponding 5-[1*R*,2*S*-1,2-bis(benzyloxy)-3-hydroxy-propyl]-1,2-dihydropyrimidin-2-one (**4**), formylarabinal **2** was treated under reflux with an excess of cyanamide in tetrahydrofuran for 30 hours in the presence of sodium hydride. As first step in this reaction we assume a nucleophilic attack of the cyanamide amino group at formyl carbon followed by hydrolysis of the nitrile to give the carboxamide group. Ring transformation to **4** then occurred through carboxamide attack at C-1.

The carbonyl resonance was found at  $\delta = 162.8$  in the  $^{13}\text{C}$ -NMR spectrum, while H-4 and H-6 appeared as a common singlet at  $\delta = 8.25$  in  $^1\text{H}$ -NMR spectrum. In the same way C-4 and C-6 gave a signal at  $\delta = 158.1$  in the  $^{13}\text{C}$ -NMR spectrum. Furthermore, the IR spectrum showed typical absorption for the associated NH group of the pyrimidinone.

Next, 2-formyl-L-arabinal **2** was reacted with 3-amino-1,2,4-triazole in the presence of sodium hydride in tetrahydrofuran to give a fused pyrimidine. After 30 h under reflux 6-[1*R*,2*S*-1,2-bis(benzyloxy)-3-hydroxy-propyl][1,2,4]triazolo[1,5-*a*]pyrimidine (**5**) could be isolated in 62% yield (Scheme 2).

Scheme 2.



The <sup>1</sup>H-NMR spectrum displayed two doublets for H-5 and H-7, each having a coupling constant of 2.2 Hz due to coupling over four bonds (W type). In order to distinguish between the alternative structures **5** and **6** a NOESY spectrum was recorded. The absence of cross peaks between H-3 and H-5 protons, as expected for structure **6**, confirmed structure **5**. On the other hand, a correlation between H-7 and H-1' was found.

Four conceivable pathways can be formulated for this reaction. The first nucleophilic attack of 3-amino-1,2,4-triazole may occur with the ring NH and NH<sub>2</sub> groups, respectively, at C-1 of the 2-formyl-L-arabinal resulting in cleavage of the pyranose ring. After that recyclization is possible by reaction of the NH<sub>2</sub> or ring NH group with the formyl function to yield [1,2,4]triazolo[1,5-*a*]pyrimidine **5**. On the other hand, the first nucleophilic attack of ring NH and NH<sub>2</sub> groups, respectively,

could take place at the carbonyl group of 2-formyl-L-arabinal **2** followed by ring transformation including now the NH<sub>2</sub> and ring NH groups. Finally, all these reaction pathways result in the formation of the same [1,2,4]triazolo[1,5-*a*]pyrimidine **5**.

## Acknowledgments

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## Experimental

### General

Solvents were distilled and if necessary, dried using standard procedures. TLC was carried out on silica gel 60 GF<sub>254</sub> (Merck) with detection by UV light ( $\lambda = 254$  nm) and/or by charring with 10% sulfuric acid in methanol. Silica gel 60 (70-230 mesh) (Merck) was used for column chromatography. Melting points were determined by using a Boetius melting point apparatus and are corrected. Specific rotations were determined with a Gyromat HP (Dr. Kernchen). IR spectra were recorded with a Nicolet 205 FT-IR spectrometer. <sup>1</sup>H-NMR spectra (250.13 MHz and 300.13 MHz, respectively) and <sup>13</sup>C-NMR spectra (62.9 MHz and 75.5 MHz, respectively) were recorded on Bruker instruments AC 250 and ARX 300, with CDCl<sub>3</sub> as solvent. The calibration of spectra was carried out on the solvent signals ( $\delta(^1\text{H}) = 7.25$ ;  $\delta(^{13}\text{C}) = 77.0$ ). The <sup>1</sup>H- and <sup>13</sup>C-NMR signals were assigned by DEPT and two-dimensional <sup>1</sup>H,<sup>1</sup>H COSY and <sup>13</sup>C,<sup>1</sup>H correlation spectra (HETCOR). The mass spectra were recorded on an AMD 402/3 spectrometer (AMD Intectra GmbH). Elemental analyses were performed on a CHNS automatic elemental analyser Flash EA 1112 (ThermoQuest).

### 5-[1*R*,2*S*-1,2-Bis(benzyloxy)-3-hydroxy-propyl]-1,2-dihydro-pyrimidine-2-thione (**3**)

To a vigorously stirred solution of **2** (100 mg, 0.3 mmol) and thiourea (46 mg, 0.60 mmol) in anhyd. THF (6 mL) was added NaH (14 mg, 0.58 mmol) at 0 °C. After 30 min the reaction mixture was allowed to warm up to room temperature and stirring was continued until no starting material could be observed in the TLC (approx. 40 h). Methanol (3 mL) was added and the mixture was stirred for further 5 min. The resulting solution was diluted with water (25 mL) and extracted with chloroform (2×25 mL). Then the combined organic layers were dried (Na<sub>2</sub>SO<sub>4</sub>), filtered and evaporated under reduced pressure. The residue obtained was purified by column chromatography (toluene/EtOAc 6:4) to give a yellow syrup. Yield: 45 mg (38%); R<sub>f</sub>: 0.35 (toluene/EtOAc 6:4);  $[\alpha]_{\text{D}}^{24}$ : -41.1 ( $c = 1.0$ , CHCl<sub>3</sub>); IR (capillary, cm<sup>-1</sup>): 3418 (OH); <sup>1</sup>H-NMR (250.13 MHz, CDCl<sub>3</sub>):  $\delta = 2.07$  (br, 1H, OH), 3.56 (dt, 1H,  $J_{1',2'} = 8.0$  Hz,  $J_{2',3'} = 4.0$  Hz, H-2'), 3.83 (d, 2H, H-3'), 4.31 (d, 1H,  $J = 11.5$  Hz, CHHPh), 4.35 (d, 1H,  $J = 11.5$  Hz, CHHPh), 4.42 (d, 1H, H-1'), 4.50 (d, 1H,  $J = 11.5$  Hz, CHHPh), 4.51 (d, 1H,  $J = 11.5$  Hz, CHHPh), 5.00 (s, 1H, NH), 7.01–7.06 (m, 2H, Ph), 7.23–7.38 (m, 6H, Ph), 8.50 (s, 2H, H-4, H-6); <sup>13</sup>C-NMR (62.9 MHz, CDCl<sub>3</sub>):  $\delta = 61.2$  (C-3'), 71.6, 72.7 (CH<sub>2</sub>Ph), 76.7 (C-1'), 80.9 (C-2'), 125.3 (C-5), 128.0, 128.1, 128.1, 128.2, 128.5, 128.6 (Ph), 136.8,

136.9 (*i*-Ph), 157.0 (C-4, C-6), 170.7 (C-2); MS (CI):  $m/z$  (%): 383 (62)  $[M+H]^+$ , 91 (100); Anal. Calcd. for  $C_{21}H_{22}N_2O_3S$ : C, 60.95; H, 5.80; N, 7.32; S, 8.38. Found: C, 65.86; H, 6.12; N, 7.09; S, 7.95.

5-[1*R*,2*S*-1,2-Bis(benzyloxy)-3-hydroxy-propyl]-1,2-dihydro-pyrimidin-2-one (4)

To a vigorously stirred solution of **2** (100 mg, 0.3 mmol) and cyanamide (29 mg, 0.69 mmol) in anhyd. THF (5 mL) was added NaH (11 mg, 0.45 mmol) at 0 °C. After 30 min the reaction mixture was allowed to warm up to room temperature and heated under reflux for 5 h. Then more cyanamide (29 mg, 0.69 mmol) was added and the mixture was refluxed for 30 h. Methanol (3 mL) was added and the mixture stirred for further 5 min. The resulting solution was diluted with water (25 mL) and extracted with chloroform (2×25 mL). The combined organic layers were dried ( $Na_2SO_4$ ), filtered and evaporated under reduced pressure. The residue was purified by column chromatography (toluene/EtOAc 6:4) to give a colourless syrup. Yield: 34 mg (30%);  $R_f$ : 0.35 (toluene/EtOAc 6:4);  $[\alpha]_D^{24}$ :  $-46.1$  ( $c = 1.0$ ,  $CHCl_3$ ); IR (capillary,  $cm^{-1}$ ): 3448 (OH, NH);  $^1H$ -NMR (250.13 MHz,  $CDCl_3$ ):  $\delta = 2.75$  (br, 1H, OH), 3.58 (dt, 1H,  $J_{1',2'} = 7.8$  Hz,  $J_{2',3'} = 4.5$  Hz, H-2'), 3.80 (d, 2H, H-3'), 4.32 (d, 1H, H-1'), 4.32 (d, 1H,  $J = 11.6$  Hz, CHHPh), 4.36 (d, 1H,  $J = 11.6$  Hz, CHHPh), 4.49 (d, 1H,  $J = 11.6$  Hz, CHHPh), 4.50 (d, 1H,  $J = 11.6$  Hz, CHHPh), 5.43 (s, 1H, NH), 7.07–7.11 (m, 2H, Ph), 7.24–7.36 (m, 8H, Ph), 8.25 (s, 2H, H-4, H-6);  $^{13}C$ -NMR (62.9 MHz,  $CDCl_3$ ):  $\delta = 61.6$  (C-3'), 71.0, 72.9 ( $CH_2Ph$ ), 77.2 (C-1'), 81.4 (C-2'), 121.8 (C-5), 127.9, 127.9, 128.0, 128.0, 128.4, 128.6 (Ph), 137.2, 137.3 (*i*-Ph), 158.0 (C-4, C-6), 162.8 (C=O); MS (EI):  $m/z$  (%): 366 (62)  $[M]^+$ , 215 (85) ( $[M]-HOCH_2CHOBn$ )<sup>+</sup>, 91 (89); Anal. Calcd. for  $C_{21}H_{22}N_2O_4$ : C, 68.84; H, 6.05; N, 7.65. Found: C, 68.58; H, 6.59; N, 7.69.

6-[1*R*,2*S*-1,2-Bis(benzyloxy)-3-hydroxy-propyl][1,2,4]triazolo[1,5-*a*]pyrimidine (5)

To a vigorously stirred solution of **2** (100 mg, 0.3 mmol) and 3-amino-1,2,4-triazole (63 mg, 0.61 mmol) in anhyd. THF (5 mL) was added NaH (8 mg, 0.33 mmol) at 0 °C. After 30 min the reaction mixture was allowed to warm up to room temperature and then heated under reflux for 30 h. Methanol (3 mL) was added and the mixture was stirred for further 5 min. The resulting solution was diluted with water (25 mL) and extracted with chloroform (2×25 mL). Then the combined organic layers were dried ( $Na_2SO_4$ ), filtered and evaporated under reduced pressure. The residue obtained was purified by column chromatography (toluene/EtOAc 6:4) to give a colourless syrup. Yield: 75 mg (62%);  $R_f$ : 0.50 (toluene/EtOAc 6:4);  $[\alpha]_D^{22}$ :  $-30.1$  ( $c = 1.0$ ,  $CHCl_3$ ); IR (capillary,  $cm^{-1}$ ): 3373 (OH);  $^1H$ -NMR (250.13 MHz,  $CDCl_3$ ):  $\delta = 3.21$  (br, 1H, OH), 3.60 (dt, 1H,  $J_{1',2'} = 8.0$  Hz,  $J_{2',3'} = 4.0$  Hz, H-2'), 3.88 (m, center of AB part of ABX,  $J_{3'a,3'b} = 11.8$  Hz, 2H, H-3'a, H-3'b), 4.28 (d, 1H,  $J = 11.9$  Hz, CHHPh), 4.43 (d, 1H,  $J = 11.3$  Hz, CHHPh), 4.53 (d, 1H,  $J = 11.9$  Hz, CHHPh), 4.53 (d, 1H,  $J = 11.3$  Hz, CHHPh), 4.63 (d, 1H, H-1'), 6.90–6.93 (m, 2H, Ph), 7.00–7.09 (m, 3H, Ph), 7.22–7.33 (m, 5H, Ph), 8.49 (s, 1H, H-2), 8.65 (d, 1H, H-7), 8.74 (d, 1H,  $J_{5,7} = 2.2$  Hz, H-5);  $^{13}C$ -NMR (62.9 MHz,  $CDCl_3$ ):  $\delta = 60.6$  (C-3'), 72.1, 72.5 ( $CH_2Ph$ ), 75.9 (C-1'), 80.5 (C-2'), 123.3 (C-6), 127.9, 128.0, 128.2, 128.2, 128.3, 128.7 (Ph), 134.5 (C-7), 136.4, 136.5 (*i*-Ph), 154.8 (C-3a), 155.5 (C-5), 156.1 (C-2); MS (CI):  $m/z$  (%): 391 (90)  $[M+H]^+$ , 91 (100); Anal. Calcd. for  $C_{22}H_{22}N_4O_3$ : C, 67.69; H, 5.64; N, 14.35. Found: C, 67.35; H, 5.71; N, 14.45.

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*Sample Availability:* Not available.