



Article Simultaneous Synthesis of Single- and Multiple-Contaminant Water Networks Using LINGO and Excel Software

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Abstract: Controlling the distribution of water and wastewater between industrial processes is vital to rationalize water usage and preserve the environment. In this paper, a mathematical technique is proposed to optimize water-wastewater networks, and a nonlinear program is introduced to minimize the consumption of freshwater and, consequently, the flowrate of wastewater discharge. A general mathematical model, able to handle industrial plants containing up to eight sources and eight sinks, is developed using LINGO optimization software to facilitate dealing with complex case studies. The introduced model can handle single-contaminant networks as well as multiplecontaminant ones. The optimal water network is synthesized through two steps; the first step involves the introduction of the case study data into the developed mathematical model. The second step considers using the optimal solution produced after running the developed LINGO model as feed data for a pre-designed Excel sheet able to deal with these results and simultaneously draw the optimal water-wastewater network. The proposed mathematical model is applied to two case studies. The first case study includes actual data from four fertilizer plants located in Egypt; the water resources and requirements are simultaneously integrated to obtain a sensible cutting in both freshwater consumption (lowered by 52.2%) and wastewater discharge (zero wastewater discharge). The second case study regards a Brazilian petrochemical plant; the obtained results show noticeable reductions in freshwater consumption by 12.3%, while the reduction percentage of wastewater discharge is 4.5%.

Keywords: wastewater networks; multiple contaminants; mathematical model; optimization

1. Introduction

The careful use of water in mass-exchange networks between processes or plants will reduce both freshwater consumption and wastewater discharge. Various mathematical programming-based methodologies were presented in the last decades for minimizing both freshwater consumption and wastewater discharge. A relaxed linear model aiming to design an optimal wastewater distribution network was firstly proposed in 1998 by Galan and Grossmann; they used different treatment technologies to decrease the contaminant levels in the network and achieve the maximum reuse of wastewater streams in the plant [1]. Gabriel and El-Halwagi introduced a linear program aiming to decrease the concentrations of contaminants and maximize the reuse and recycle of wastewater through the optimal design of a water interception network [2], while a mixed integer linear program, directed to design water–wastewater networks, was presented by Faria and Bagajewicz for the best handling of water allocation problems [3]. A mixed integer nonlinear programming (MINLP) model was proposed by Galan and Grossmann to design



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Copyright: © 2024 by the authors. Licensee MDPI, Basel, Switzerland. This article is an open access article distributed under the terms and conditions of the Creative Commons Attribution (CC BY) license (https:// creativecommons.org/licenses/by/ 4.0/). a wastewater network that included five contaminants with thirteen types of technologies to decrease the concentration of the contaminants [4]; their proposed model was applied on a mining plant in Turkey, and an optimal wastewater network with minimum cost was designed. Many other researchers have optimized water networks by MINLP, but for different applications, such as Buabeng and Majozi [5] who aimed to maximize the reuse of wastewater by regeneration-reuse and regeneration-recycling using a hollow-fiber reverse-osmosis membrane to minimize the cost of freshwater consumption. A MINLP model was also proposed by Padron et al. for the design of a wastewater network that included treatment units to decrease the concentration of contaminants and achieve the minimum consumption of freshwater [6]; they presented a case study of Mexico City to show the validity and effectiveness of their presented model. Many other mathematical methodologies have been introduced over the years. Nejad et al. [7] presented a mathematical model and applied it in a case study of a Tehran oil refinery to minimize water and wastewater flowrates; their study included three parameters of water quality, i.e., chemical oxygen demand, suspended solids and hardness. Hansen et al. [8] introduced a mathematical program directed to minimize freshwater consumption between several operations, including cooling systems and washing processes; they applied their proposed model in a petrochemical plant. The priority of source-sink matches was determined by using a ranking matrix technique [9]; three case studies of single and multiple contaminants were introduced to show the applicability of that technique. A closed-loop supply chain was proposed by Mohammad et al. [10] to control the shortage of water in an Azerbaijan city in Iran by using a Social Engineering Optimizer (SEO); their model showed a lowering in freshwater consumption. Arola et al. [11] presented a mathematical approach to maximize the reuse of wastewater by decreasing the concentration of contaminants such as phosphorous, chemical oxygen demand (COD) and total oxygen carbon (TOC) in wastewater; the study included a membrane bioreactor performing a regeneration process to decrease the concentration of contaminants. A mathematical technique was suggested by Chin et al. [12] to optimize the total cost of freshwater consumption and design an optimal water integration network in multiple-contaminant systems. Tuba et al. [13] presented a review of mathematical programming-based methodologies used for water minimization in industrial processes. Ho et al. [14] described a fuzzy multi-objective optimization to minimize freshwater consumption in oil industries in the presence of a single contaminant. A mathematical program was suggested by Liang et al. [15] to allocate wastewater and minimize the consumption of freshwater in a water allocation system in the Zhanghe River basin. Gajendra et al. [16] used the GAMS software (version 23.5.2) to allocate water and wastewater in water networks; the proposed model was formulated as linear Equations.

Graphical-based techniques were also considered in several studies to improve the distribution of water and wastewater in water networks; Irene et al. [17] used a graphical technique to minimize waste flowrates and fresh resources in the design of an inter-plant resource conservation network. Sahu et al. [18] introduced a graphical method to add a treatment unit to minimize the concentrations of contaminants and maximize the reuse of wastewater, while minimizing the costs of treatment. Another graphical technique was proposed by Farrag et al. [19] for the analysis and design of wastewater networks using the composition driving forces; they studied the effect of the mass load of contaminants on the maximum mass recovery. Yamin et al. [20] proposed a graphical technique based on six scenarios in the design of wastewater networks; the technique was applied to agricultural irrigation systems. A study of a water distribution system was presented by Praint et al. [21] to minimize the consumed freshwater in Bota the River area; this study used a graphical technique. The pinch technique was presented also to minimize wastewater discharge from industrial plants. Many other pinch-based techniques were introduced in the last decades to solve the problem of wastewater network optimization and minimize freshwater use and wastewater discharge in industrial plant operations [22–30]. However, pinch and graphical methods are only suitable for simple case studies; when a plant includes a high number of streams, it is difficult to optimize a water-wastewater network via graphical methods. Another group of research works used combined pinch and mathematical programming techniques to overcome the limitations of graphical methods in optimizing water–wastewater networks [31–35].

To date, no simple ready model capable of managing simple as well as complex cases, having a large number of streams and contaminants, and simultaneously drawing the optimum water–wastewater network is available. In this paper, a generalized mathematical model is proposed to minimize both freshwater consumption and the flowrate of wastewater discharge; nonlinear equations based on overall mass balance and component mass balance between several sources (up to eight sources) and several sinks (up to eight sinks) are formulated to be solved by LINGO software (version 14); Excel software (version 2010) is used to draw the water–wastewater network with the automated technique. Two case studies involving a single and multiple contaminants are presented to show the effectiveness of the suggested model in minimizing the flowrates of total freshwater consumption and wastewater discharge in the network. In the future, the proposed model can be applied to agriculture networks to maximize the reuse of irrigation water and minimize freshwater consumption in the mixing units that decreases the concentration of contaminants. Furthermore, the proposed model can be applied to drinkable water networks.

2. Methods

We considered a set of source streams, including up to eight sources, with flowrate F_{SK} , where K is the source number in the network design. It was assumed that each source contained multiple contaminants (up to three contaminants) with concentrations X_{SKA} , X_{SKB} , and X_{SKC} . The contaminants' concentration limits in the discharged wastewater with flowrate $g_{k-waste}$ were $X_{SAwaste}$, $X_{SBwaste}$, and $X_{SCwaste}$.

We considered a set of sink streams, including up to eight sinks, with flowrates G_Q , where Q is the sink number in the network design; each sink was assumed to contain up to three contaminants with concentrations Z_{QAin} , Z_{QBin} , and Z_{QCin} . Freshwater with flowrate F_{WQ} was distributed to each sink according to the mass load limit, with concentrations X_A , X_B , and X_C . Figure 1 presents the flowchart of the proposed model. The positive or zero values of the outlet flowrates for the sources and the inlet flowrates of the sinks are presented in the proposed model as assumptions; also, we assumed positive contaminant inlet concentrations for the sinks and outlet concentrations for the sources in the proposed model.

The procedure for obtaining the optimum design of a water–wastewater network is shown in Figure 1; the overall mass balance was applied for each source stream with flowrate F_{SK} , which was equal to the sum of the flowrates from that source to the sinks (g_{K-Q}) and of that of the portion of that stream which could be discharged as waste ($g_{K-Waste}$).

The overall mass balance was applied to each sink; the inlet flowrate for each sink is G_Q , which depends on the flowrate of freshwater fed to each sink (F_{WQ}) and the flowrate from each source to that sink, g_{K-Q} . The component mass balance of three contaminants (A, B, and C) was applied to each sink, and the concentration limits of the contaminants A, B, and C in each sink were Z_{QAin} , Z_{QBin} , and Z_{QCin} respectively. The freshwater flowrate (F_{WQ}) was determined assuming that the freshwater was mixed with the feed water in each sink to optimize the inlet mass load.

The overall mass balance was applied to determine the total waste discharge flowrate (G_{Waste}), which depended on the discharge waste flowrate from each source ($G_{K-Waste}$); the component mass balance was applied to the discharged wastewater, with the contaminants A, B, and C having the concentrations $X_{SAwaste}$, $X_{SBwaste}$, and $X_{SCwaste}$, respectively. The contaminant concentration limits in discharged wastewater are established by environmental laws.



Figure 1. Procedure for achieving the optimum design of a water–wastewater network.

3. Case Studies

The proposed mathematical model was applied to two case studies to show its effectiveness in minimizing freshwater consumption and reducing the total wastewater discharge flowrate. The following subsections describe in detail the considered case studies.

3.1. Case Study 1

The first case study in this work included four fertilizer plants (phosphoric acid plant, concentrated-phosphoric acid plant, single-superphosphate plant, and triple-superphosphate plant), which contained eight contaminant sources of hydrofluorosilicic acid (H_2SiF_6), sulfuric acid (H_2SO_4), and phosphorus pentoxide (P_2O_5). In the presented case study, some source streams were completely discharged as waste. Furthermore, several sinks were completely supplied for their water needs by freshwater; this consequently resulted in a higher consumption of freshwater and in an extra discharge of wastewater. Figure 2 shows the source and sink streams that were not integrated into the network in the current actual case study.



Figure 2. Source and sink streams were not integrated into the existing water network.

As shown in Figure 2, before applying the proposed mathematical model, source 5 (separator of phosphoric acid) discharged all its wastewater ($20 \text{ m}^3/\text{h}$) to waste with its high concentration of H₂SiF₆. Four sinks (dilution mixer of sulfuric acid in the phosphoric acid plant, dilution mixer of sulfuric acid in the single-superphosphate plant, washing filter cake in the phosphoric acid plant, and washing filter in the phosphoric acid plant) were supplied only with freshwater with a flowrate of 270 m³/h.

The flowrates and concentrations of the eight sources are shown in Table 1; the flowrate of source 1 (condenser of the phosphoric acid plant) was 280 m³/h, and the source had normal concentrations of H_2SiF_6 , H_2SO_4 , and P_2O_5 (14, 0.3, and 4 wt% respectively); the flowrates of sources 2, 3, and 4 (reaction vacuum pump 1 of the phosphoric acid plant, filter vacuum pump 1 of the phosphoric acid plant, and filter vacuum pump 2 of the phosphoric acid plant) were 15, 18, and 18 m³/h, respectively, and these sources had the same concentrations of H_2SiF_6 , H_2SO_4 , and P_2O_5 (8, 0.1, and 3 wt% respectively).

Source 5 (separator of the phosphoric acid plant) had a high concentration of H_2SiF_6 (30 wt%); so, its wastewater, with flowrate of 20 m³/h, was discharged to waste. Source 6 (cooling water of the phosphoric acid plant) had a wastewater flowrate of 120 m³/h and concentrations of H_2SiF_6 , H_2SO_4 , and P_2O_5 of 10, 0.6, and 5 wt% respectively.

Source 7 (cooling water of the single-superphosphate plant) had a flowrate of 160 m³/h and concentrations of H₂SiF₆, H₂SO₄, and P₂O₅ of 12, 0.4, and 5 wt%, respectively, while source 8 (condenser of the concentrated unit in the phosphoric acid plant) had a high flowrate (250 m³/h), a high concentration of H₂SiF₆ (20 wt%), and normal concentrations of H₂SO₄ and P₂O₅ (0.2, and 4 wt%, respectively).

Sources	Flowrate (m ³ /h)	H ₂ SiF ₆ (wt%)	H ₂ SO ₄ (wt%)	P ₂ O ₅ (wt%)
Source 1 (condenser of phosphoric acid plant)	280	14	0.3	4
Source 2 (reaction vacuum pump 1 of phosphoric acid plant)	15	8	0.1	3
Source 3 (filter vacuum pump 1 of phosphoric acid plant)	18	8	0.1	3
Source 4 (filter vacuum pump 2 of phosphoric acid plant)	18	8	0.1	3
Source 5 (separator of phosphoric acid plant)	20	30	0.5	4
Source 6 (cooling water of phosphoric acid plant)	120	10	0.6	5
Source 7 (cooling water of single-superphosphate plant)	160	12	0.4	5
Source 8 (condenser of the concentrated unit in phosphoric acid plant)	250	20	0.2	4

Table 1. Data regarding the sources in the investigated fertilizer plants.

The flowrate and contaminant concentration limits (H_2SiF_6 , H_2SO_4 , and P_2O_5) for the seven sinks are shown in Table 2; the flowrate of sink 1 (dilution mixer of sulfuric acid in the phosphoric acid plant) was 10 m³/h, with concentration limits for H_2SiF_6 , H_2SO_4 , and P_2O_5 of 12, 0.5, and 4.5 wt%, respectively.

Table 2. Data regarding the sinks in the considered fertilizer plants.

Sinks	Flowrate (m ³ /h)	H ₂ SiF ₆ (wt%)	H ₂ SO ₄ (wt%)	P ₂ O ₅ (wt%)
Sink 1 (dilution mixer of sulfuric acid in phosphoric acid plant)	10	12	0.5	4.5
Sink 2 (dilution mixer of sulfuric acid in single-superphosphate plant)	140	12	0.8	3.69
Sink 3 (washing filter cake in phosphoric acid plant)	50	10	3	5
Sink 4 (gas scrubber in phosphoric acid plant)	280	14	3	5
Sink 5 (gas scrubber in single-superphosphate plant)	280	16	3	6
Sink 6 (gas scrubber in triple-superphosphate plant)	180	11	3	5
Sink 7 (washing filter in phosphoric acid plant)	70	14	5	4

The data for the three gas scrubbers in the three different plants are shown in Table 2; sink 4 (gas scrubber in the phosphoric acid plant) and sink 5 (gas scrubber in the single-superphosphate plant) had the same flowrate of 280 m³/h, but the concentration limits of H₂SiF₆, H₂SO₄, and P₂O₅ in sink 4 and sink 5 were 14, 3, 5 wt% and 16, 3, 6 wt%, respectively. Sink 6 (gas scrubber in the triple-superphosphate plant) had a flowrate of 180 m³/h with concentration limits of H₂SiF₆, H₂SO₄, and P₂O₅ of 11, 3, and 5 wt%, respectively.

The washing filter in the phosphoric acid plant (sink 7) had a flowrate of 70 m³/h, with concentration limits of H_2SiF_6 , H_2SO_4 , and P_2O_5 of 14, 5, and 4 wt%, respectively.

The objective function of the model constructed for handling these fertilizer plants was to minimize freshwater consumption as well as wastewater discharge.

3.2. Case Study 2

The Brazilian petrochemical plant presented by Hansen et. al. [8] was the chosen second case study in the current research work. This case study included eight sources (clarified water, filtered water, cooling water 1, cooling water 2, cooling water 3, cooling

water 4, bearing water 1, and bearing water 2) and six sinks (cooling water 1, cooling water 2, cooling water 3, cooling water 4, bearing water 1, and bearing water 2). We considered the COD parameter as the only indicator of contamination. The flowrate and COD limits in each plant stream are shown in Table 3.

Sources	Flowrate (m ³ /h)	COD (mg/L)	Sinks	Flowrate (m ³ /h)	COD (mg/L)	
Clarified water	1131	0	Cooling water 1	717	3.7	
Filtered water	21	0	Cooling water 2	277	4.7	
Cooling water 1	717	30	Cooling water 3	92	5.8	
Cooling water 2	277	25	Cooling water 4	45	6.1	
Cooling water 3	92	25	Boaring water 1	6	14 7	
Cooling water 4	45	25	- Dearing water 1		1 1.7	
Bearing water 1	6	20	Boaring water 2	15	35	
Bearing water 2	15	20	- Dearing water 2	10	0.0	

Table 3. Data regarding the sources and sinks in the considered Brazilian petrochemical plant.

4. Results and Discussions

After applying the data regarding the sources and sinks of the two investigated case studies to the suggested mathematical model (using LINGO software), the obtained results showed the optimum flowrates from sources to sinks, the freshwater consumption in each sink, and the wastewater flowrate from each source. After that, the obtained results were analyzed simultaneously by the predesigned Excel software, automatically achieving the drawing of the optimal wastewater network. This approach aimed to minimize freshwater consumption and reduce wastewater discharge in the two case studies under investigation.

4.1. Results and Discussion for Case Study 1

The data regarding the fertilizer plants (case study 1) were introduced into the designed mathematical model and solved by LINGO optimization software. The obtained results showed the flowrates from sources to sinks (G_{k-Q}), the flowrates from sources to waste, and the freshwater flowrates to the sinks; these results are listed in Table 4.

Table 4. The optimum flowrates from sources to sinks and to waste of the studied fertilizer plants.

Stream	Flowrate (m ³ /h)	Stream	Flowrate (m ³ /h)	Stream	Flowrate (m ³ /h)	
Fw	129	G ₄₋₆	9.0697	G ₈₋₂	42.29	
G _{Waste}	Zero	G ₅₋₄	2.7824	G ₈₋₃	19.84	
G ₁₋₂	25.966	G ₅₋₅	17.217	G ₈₋₄	59.69	
G ₁₋₄	93.598	G ₆₋₄	68.7	G ₈₋₅	78.04	
G ₁₋₅	65.056	G ₆₋₅	37.34	G ₈₋₆	44.15	
G ₁₋₆	25.378	G ₆₋₆	13.97	F_{W1}	4.0201	
G ₁₋₇	70	G ₇₋₂	23.15	F _{W2}	24.8	
G ₂₋₂	5.9302	G ₇₋₃	8.425	F _{W3}	21.738	
G ₂₋₆	9.0697	G ₇₋₄	41.05	F_{W4}	14.175	
G ₃₋₂	8.9302	G ₇₋₅	55.66	F_{W5}	26.684	
G ₃₋₆	9.0697	G ₇₋₆	31.71	Fwa	37 581	
G ₄₋₂	8.9302	G ₈₋₁	5.98	- 000	07.001	



Figure 3. Design of water-wastewater network for the fertilizer plants studied.

Source 1 supplied sink 5, sink 6, and sink 7, with flowrates of 65.056, 25.378, and 70 m³/h, respectively. Source 2 fed sink 2 and sink 6, with flowrates of 5.93 and 9.07 m³/h, respectively; also source 3 supplied sink 2 and sink 6, with flowrates of 8.93 and 9.07 m³/h, respectively. Source 4 fed sink 2, with a flowrate of 8.93 m³/h, and sink 6, with a flowrate of 9.07 m³/h. Source 5 supplied sink 4 and sink 5, with flowrates of 2.7824 and 17.217 m³/h,

respectively, while source 6 supplied sink 4, sink 5, and sink 6, with flow rates of 68.7, 37.34, and 13.97 m^3/h , respectively.

Source 7 fed sink 2, sink 3, sink 4, sink 5, and sink 6, with flowrates of 23.15, 8.425, 41.05, 55.66, and 31.71 m³/h, respectively, while source 8 supplied sink 1, sink 2, sink 3, sink 4, sink 5, and sink 6, with flowrates of 5.98, 42.29, 19.84, 59.69, 78.04, and 44.15 m³/h, respectively. Zero wastewater discharge was obtained in this case study.

By comparing the obtained results with the original case study data, freshwater consumption was reduced from 270 m³/h to 129 m³/h, with a reduction percentage of 52.2%, and the total flowrate of wastewater discharge decreased from 20 m³/h to zero discharge.

4.2. Results and Discussion for Case Study 2

By feeding the data regarding the second case study (Brazilian petrochemical plant) to the proposed model, the optimum flowrates from sources to sinks and to waste could be determined. The obtained results, listed in Table 5, were used as feed data for the Excel software to obtain the automated drawing of the optimum water–wastewater network, shown in Figure 4.

Table 5. Flowrates from sources to sinks and to waste for the studied Brazilian petrochemical plant.

	Sources Flowrates (m ³ /h) Distributed between Sinks and Waste								
Sinks	S1 (Clarified Water)	S2 (Filtered Water)	S3 (Cooling Water 1)	S4 (Cooling Water 2)	S5 (Cooling Water 3)	S6 (Cooling Water 4)	S7 (Bearing Water 1)	S8 (Bearing Water 2)	
K1	623.57	0	78.43	0	0	0	0	15	
K2	233.6	0	43.4	0	0	0	0	0	
K3	57.213	15	13.79	0	0	0	6	0	
K4	35.85	0	9.15	0	0	0	0	0	
K5	3.06	0	2.94	0	0	0	0	0	
K6	13.25	0	1.75	0	0	0	0	0	
Waste	0	0	567.54	277	92	45	0	0	
Source Total Flowrate	966.54	15	717	277	92	45	6	15	

The clarified water source and filtered water source were considered as freshwater sources because the concentration of contaminants in these sources was zero. Regarding the results reported in Table 5, the clarified water flowrate of 966.54 m³/h was distributed between sinks K1, K2, K3, K4, K5, and K6, with flowrates of 623.57, 233.6, 57.213, 35.85, 3.06, and 13.25 m³/h, respectively. However, the filtered water source fed only the third sink (k3), with a flowrate of 15 m³/h. The cooling water 1 source stream supplied six sinks (K1, K2, K3, K4, K5, and K6) and was in part discharged as waste, with flowrates of 78.43, 43.4, 13.79, 9.15, 2.94, 1.75, and 567.54 m³/h, respectively. On the other hand, the cooling water 2, cooling water 3, and cooling water 4 sources discharged their water to waste, with flowrates of 277, 92, and 45 m³/h, respectively. The bearing water 1 source fed only K3, with a flowrate 6 m³/h, while the bearing water 2 source fed only sink 1, with a flowrate of 15 m³/h. The total wastewater discharge, with a flowrate of 982 m³/h, was collected from the S3, S4, S5, S6, S7, and S8 sources, with flowrates of 567.54, 277, 92, and 45 m³/h, respectively.



Figure 4. Design of a water-wastewater network for a Brazilian petrochemical plant (case study 2).

By comparing the results of the proposed mathematical model with those of the original case study, it was observed that the clarified water flowrate was reduced from $1102 \text{ m}^3/\text{h}$ to $966.54 \text{ m}^3/\text{h}$, with a reduction percentage of 12.3%. It was also noticed that the filtered water flowrate was reduced from $18 \text{ m}^3/\text{h}$ to $15 \text{ m}^3/\text{h}$, with a reduction percentage of 16.5%. Additionally, the wastewater discharge decreased by 4.5%, from $1029 \text{ m}^3/\text{h}$ to $982 \text{ m}^3/\text{h}$.

Regarding the abovementioned results of the two investigated case studies, it is clear that the introduced approach is an effective and economical technique for designing an optimal water–wastewater network. These results showed that the achieved water– wastewater networks for the considered case studies are economically effective and more profitable compared to the current networks, minimizing freshwater consumption and reducing wastewater discharge. Moreover, the proposed technique can be applied to other case studies to optimize their existing water networks.

5. Conclusions

The water requirement of industrial processes in chemical plants is high; in the industry, water is used for washing, cooling, diluting, and processing in chemical plants. In this work, a mathematical model (LINGO optimization software, version 14.0) was proposed to design optimal mass-exchange networks; the presented model was formulated as a nonlinear program to minimize both freshwater consumption and wastewater discharge. Two sequential steps were established to design optimal water-wastewater networks; the first step consisted in applying the collected data into LINGO software; in the second step, the obtained results were simultaneously analyzed by the proposed Excel software, which was responsible for drawing an automated optimal water-wastewater network. Two case studies (fertilizer plants and a Brazilian petrochemical plant) were presented to show the validity of the proposed model. The obtained results for the abovementioned two case studies showed a reduction in freshwater consumption by 52.2 and 12.3%, respectively. Furthermore, the wastewater discharge was reduced to zero in the fertilizer plants studied, while the reduction percentage of wastewater discharge for the Brazilian petrochemical plant was 4.5%. It was also noticed that the obtained water-wastewater networks are more efficient and profitable than the present networks. This can be attributed to the reduction in freshwater consumption and wastewater discharge. Furthermore, the proposed approach is easy to use and can be applied to different industries, including fertilizer plants and petrochemical plants.

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